

Three-Component Domino Reactions for Regioselective Formation of Bis-indole Derivatives

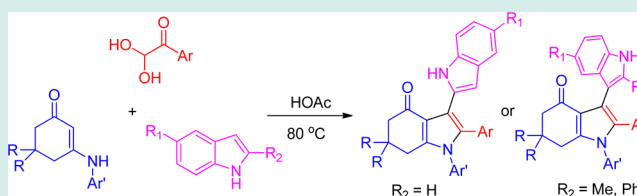
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Supporting Information

ABSTRACT: A microwave-assisted regioselective reaction dealing with arylglyoxal monohydrate, diverse *N*-aryl enaminones, and indoles to achieve 3,2'- and 3,3'-bis-indoles by varying a substituted indole substrate is reported. The 2-unsubstituted indoles resulted in the 3,2'-bis-indole skeleton, whereas indoles bearing a methyl or phenyl group at C2 led to the 3,3'-bis-indoles with simultaneous formation of three sigma-bonds. The procedures feature excellent regioselectivity, short reaction times, convenient one-pot manner, and operational simplicity.

KEYWORDS: multicomponent domino reaction, regioselective synthesis, indolation, enaminones, bis-indoles



INTRODUCTION

The search for an efficient construction of indole skeletons of chemical and biomedical importance has been an active theme in organic synthesis.^{1,2} Structurally diverse bis-indole skeletons commonly exist in nature and are represented by topsentins,³ isoborreverine,⁴ and indirubin (Figure 1)⁵ which exhibit a

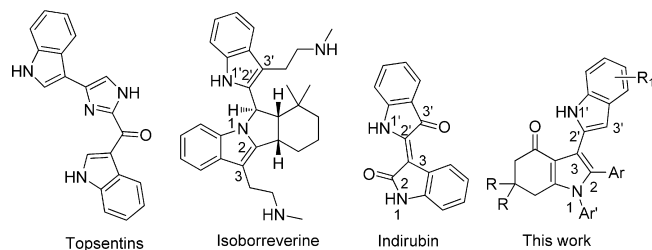


Figure 1. Structures of some bis-indole alkaloids.

broad range of biological activities. For example, topsentins and its dihydro analogues have shown antitumor, antiviral, and anti-inflammatory activities,³ isoborreverine has exhibited antimalarial activity,⁶ and indirubin and its analogues were found to target glycogen synthase kinase-3 (GSK-3),⁷ and aurora kinases,⁸ and acted as cyclin-dependent kinase (CDK) inhibitors⁶ and dioxin receptor.⁹ Thus, these complex architectures have inspired our interest in creating methodologies for their total synthesis.

Multicomponent domino reactions (MDRs) for use in total syntheses of natural products¹⁰ or natural-like structures were one of the key tools that allow the creation of several bonds in a single operation and offer remarkable advantages like convergence, operational simplicity, and facile automation.¹¹ These reactions can avoid time-consuming and costly processes for purification of various precursors and isolation of

intermediates, thereby minimized waste generation rendering the transformations green.¹² Recently, we have developed a series of new, multicomponent domino reactions (MDRs) that provide easy access to multiple functionalized ring structures of chemical and pharmaceutical importance.¹³ We would like to report another new synthetic strategy for the regioselective preparation of polyfunctionalized bis-indoles through a three component domino reaction. This reaction was achieved from the readily available starting materials such as arylglyoxal monohydrate (Figure 2), *N*-aryl enaminones (Figure 3), and

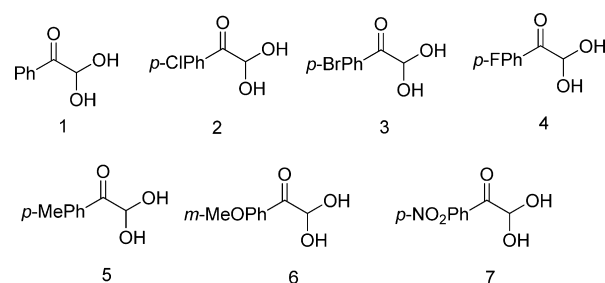


Figure 2. Diversity of arylglyoxal monohydrate 1{1–6}.

indoles (Figure 4) in HOAc under microwave heating (Scheme 1). The great aspect of the present domino reaction is shown by the fact that the synthesis of new polyfunctionalized bis-indoles, which connected through C3–C2' bond of the two indole units, and its polyfunctionalities at C3 and C2' positions were readily achieved via HOAc-promoted three-component

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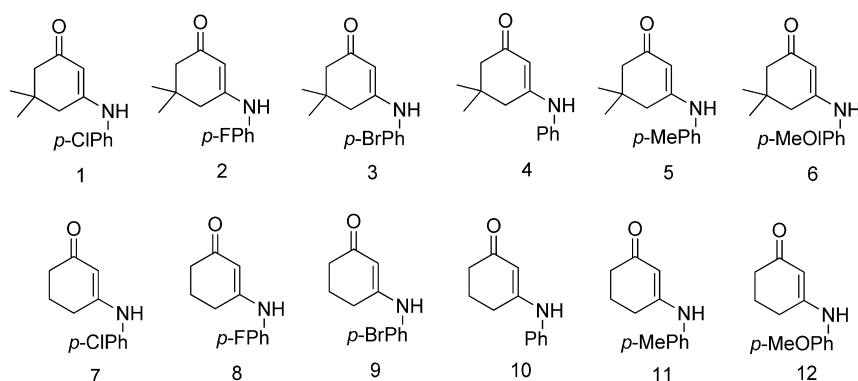


Figure 3. Diversity of enamines 2{1–12}.

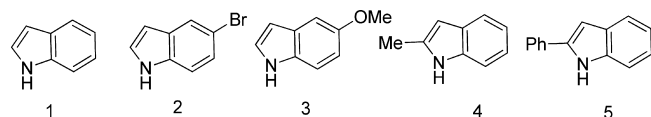
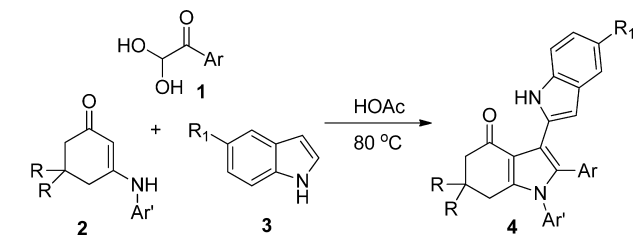


Figure 4. Diversity of substituted indoles 3{1–5}.

Scheme 1. Regioselective Synthesis of Bis-Indoles 4



reaction in a single step, and regioselectivity was controlled well in an intermolecular manner.

■ RESULT AND DISCUSSION

Indole, possessing three nucleophilic centers, is an important type of nucleophile. Indeed, C-nucleophilicity at the C3 position is stronger than that of the C2 position, which has been described widely in the literature.² The utilization of its C2 position as a nucleophilic center to the preparation of 2-substituted indoles has been neglected. Aryl glyoxal monohydrate, as readily available 1,2-biacceptors,¹⁴ can provide two active sites (C–O and C=O bonds), which can be attacked by the electron-rich β -C atom and NH group in β -enaminones¹⁵ for C–C and C–N bonds formation via [3 + 2] cyclization mechanism. Meanwhile, nucleophilic substitution between indole and aryl glyoxal monohydrate occurs, in which functionalities at C3 and C2' positions of two indole units can be realized through the control of an intermolecular hydrogen bond. Based on the above analysis, we started this study by subjecting a preformed *N*-*p*-ClPh substituted enamines 2{1} to the reaction with phenylglyoxal monohydrate 1{1} and indole 3{1} in HOAc at 80 °C under microwave irradiation. The reaction smoothly gave the desired product 4{1,1,1} in 84% chemical yield. Indole 3 has two C-electrophilic centers, which may lead to two different products 4{1,1,1} or 5{1,1,1} (Table 1). In this reaction, only product 4{1,1,1} was isolated; it is obvious that the reaction has high regioselectivity. An overview of the synthetic details is summarized in Table 1. This result prompted us to further optimize reaction conditions. The same reactions were

Table 1. Optimization for the Synthesis of 4{1,1,1} under MW

| entry | solvent | <i>T</i> (°C) | time (min) ^a | yield (%) ^c |
|-------|---------------------------------|---------------|-------------------------|------------------------|
| 1 | DMF | 80 | 20 | 20 |
| 2 | CH ₂ Cl ₂ | 80 | 20 | trace |
| 3 | EtCOOH | 80 | 20 | 42 |
| 4 | HCOOH | 80 | 20 | 58 |
| 5 | HOAc | 80 | 20 | 84 |
| 6 | HOAc | 100 | 20 | 63 |
| 7 | HOAc | 80 | 60 ^b | 67 |

^aMicrowave (MW) Irradiation. ^bClassical Heating (CH) conditions. ^cIsolated yields.

performed in dimethylformamide (DMF) and CH₂Cl₂, and both cases scarcely proceeded. We reasoned that the acidic solvent can be used as a Brønsted acid promoter and improve the yield of the desired product 4{1,1,1}. Next, this chemistry was carried out in EtCOOH and HCOOH. An incomplete reaction was observed using EtCOOH as an acidic solvent while HCOOH gave the 58% yield of 4{1,1,1}. Therefore, HOAc proved to be a best case (Table 1, entry 5). Subsequently, the reaction was performed in HOAc and repeated many times at different temperatures in a sealed vessel under microwave (MW) irradiation for 20 min. The lower yield of product 4{1,1,1} (63%) was obtained as the reaction temperature was increased to 100 °C. Subsequently, the identical reaction was investigated under classical heating (CH) conditions at 80 °C for 60 min, providing the desired product 4{1,1,1} in 67% chemical yield (Table 1, entry 7).

With these optimized conditions in hand, we examined the scope of this MDR process by using various easily available starting materials. As revealed in Table 2, the reaction resulted in corresponding desired products in good to excellent yields. The regioselective transformation is easy to perform simply by subjecting a mixture of aryl glyoxal monohydrate, various enamines 2, and indoles 3 in acetic acid to microwave heating. As shown in Table 2 (entries 1–6), we explored the enamine substrate scope, phenylglyoxal monohydrate 1{1} and indole 3{1} were used as model substrates, with the results

Table 2. Domino Synthesis of 3,2'-Bis-indoles 4 under MW

| entry | 4 | time/min | yield/% |
|-------|-----------|----------|---------|
| 1 | 4{1,1,1} | 20 | 84 |
| 2 | 4{1,2,1} | 30 | 71 |
| 3 | 4{1,3,1} | 20 | 81 |
| 4 | 4{1,4,1} | 30 | 82 |
| 5 | 4{1,5,1} | 24 | 77 |
| 6 | 4{1,6,1} | 22 | 85 |
| 7 | 4{1,1,2} | 30 | 83 |
| 8 | 4{1,2,2} | 25 | 86 |
| 9 | 4{1,3,2} | 28 | 81 |
| 10 | 4{1,4,2} | 30 | 76 |
| 11 | 4{7,4,2} | 28 | 88 |
| 12 | 4{1,5,2} | 26 | 89 |
| 13 | 4{1,6,2} | 28 | 83 |
| 14 | 4{1,5,3} | 32 | 70 |
| 15 | 4{1,6,3} | 28 | 65 |
| 16 | 4{1,7,1} | 36 | 76 |
| 17 | 4{1,8,1} | 38 | 75 |
| 18 | 4{1,9,1} | 30 | 80 |
| 19 | 4{1,10,1} | 36 | 63 |
| 20 | 4{1,11,1} | 30 | 78 |
| 21 | 4{1,12,1} | 30 | 82 |
| 22 | 4{1,7,2} | 36 | 79 |
| 23 | 4{1,8,2} | 34 | 73 |
| 24 | 4{1,9,2} | 40 | 79 |
| 25 | 4{1,10,2} | 36 | 68 |
| 26 | 4{1,12,2} | 32 | 72 |

indicating the tolerance of some functional groups in the substrates **2** including ether and C–Cl (or F, Br) bonds. 4-Nitrophenylglyoxal monohydrate amine **1**{7} was also suitable for this three-component domino reaction (entry 11). The 5,5-unsubstituted enaminones **2**{7–12} were subjected to this reaction providing corresponding bis-indoles **4**{1,7,1}–**4**{1,12,1} in 63–82% yields. Next, we further examined the generality of this system. Various indole substrates **3** bearing either electron-donating or electron-withdrawing functional groups such as bromo or methoxyl were used to give the corresponding polysubstituted bis-indoles through metal-free intermolecular indolation. It is worth noting that functional groups like bromide and chloride were well tolerated. These functional groups provide ample opportunity for further functional group manipulations, for example, by modern cross-coupling reactions. This result is significant since there is no literature precedent for the synthesis of such highly functionalized 3,2'-bis-indoles.

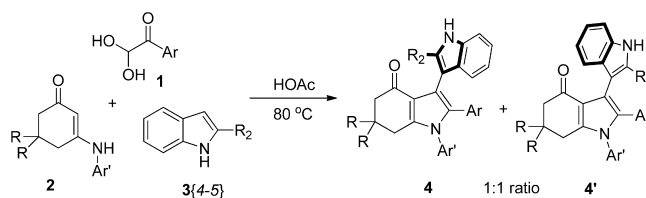
Since bis-3-indoles show important bioactivities as selective inhibitors of cyclin-dependent-kinases (CDK4)¹⁶ and anti-cancer agents,¹⁷ we turned our attention to investigate several differently 2-substituted indoles to synthesize a series of polyfunctionalized bis-3-indoles. Next, the reactions of arylglyoxal monohydrate **1**, *N*-aryl enaminones **2** with 2-substituted indoles **3**{4,5} in acetic acid were performed under the conditions described above for a short period (28–40 min) (Table 3). In all these cases, the reactions proceeded smoothly to give the corresponding bis-3-indoles in good yields of 60–88% (Scheme 2). The bulkiness of 2-phenyl indoles **3**{5} did not hamper the reaction process. Several different *N*-substituents were compared and substituents bearing electron-withdrawing (4-ClPh **2**{1} and 4-FPh **2**{2}) or electron-donating (4-MeOPh, **2**{6}) groups were found to be suitable

Table 3. Domino Synthesis of 3,3'-Bis-indoles 4 under MW

| entry | 4 and 4' | time/min | yield ^a /% |
|-------|-----------|----------|-----------------------|
| 1 | 4{1,1,4} | 30 | 73 |
| 2 | 4{1,2,4} | 32 | 60 |
| 3 | 4{1,4,4} | 30 | 79 |
| 4 | 4{1,5,4} | 30 | 80 |
| 5 | 4{1,6,4} | 32 | 78 |
| 6 | 4{1,2,5} | 38 | 70 |
| 7 | 4{2,1,5} | 35 | 78 |
| 8 | 4{2,2,5} | 38 | 75 |
| 9 | 4{3,1,5} | 34 | 79 |
| 10 | 4{3,5,5} | 30 | 83 |
| 11 | 4{4,2,5} | 28 | 78 |
| 12 | 4{4,5,5} | 30 | 88 |
| 13 | 4{5,5,5} | 38 | 71 |
| 14 | 4{1,7,5} | 40 | 73 |
| 15 | 4{1,9,5} | 38 | 75 |
| 16 | 4{1,12,5} | 36 | 70 |

^aIsolated yield of two isomers.

Scheme 2. Domino Synthesis of 3,3'-Bis-indoles 4



for this domino reaction. Impressively, the ¹H NMR analysis of the products **4**{1,1,4}–**4**{1,12,5} indicates the presence of a mixture of two isomers resulting from generation of axial chiral molecules due to large steric hindrance of 2-substituted indoles (Scheme 2). The ratio of the isomers was in 1:1 as demonstrated by ¹H NMR integration and HPLC analysis of the crude mixture (See Supporting Information). Indeed, the protocol provides a straightforward pathway to construct highly functionalized 3,2'- and 3,3'-bis-indoles. The resulting functionalities of these 3,2'- and 3,3'-bis-indoles offer a great flexibility for further structural modifications. These special 3,2'- and 3,3'-bis-indoles may be directly useful for drug design, discovery, and development.

In all cases, the functional complexity of the resulting products from this new reaction illustrates the remarkable regioselectivity of the sequence starting from very common and easily accessible starting materials. The structural elucidation and the attribution of regioselectivity were unequivocally determined by NMR spectroscopic analysis. To further ascertain structures of the newly synthesized bis-indole derivatives (**4**), single crystals of **4**{1,5,3} and **4**{4,2,5} were successfully obtained with their structures (Figures 5 and 6) unambiguously confirmed by X-ray diffraction analysis (see Supporting Information).

Similar to our previous multicomponent domino process,¹² the present reaction also showed the following attractive characteristics: (1) the environmentally friendly process in which water is the major byproduct; (2) the convenient workup which only needs simple filtration; (3) readily available starting materials of arylglyoxal monohydrate, preformed enaminones and indoles. Moreover, during these domino processes, the formation of indole skeleton and its indolation were readily achieved via regioselective three-component domino reaction

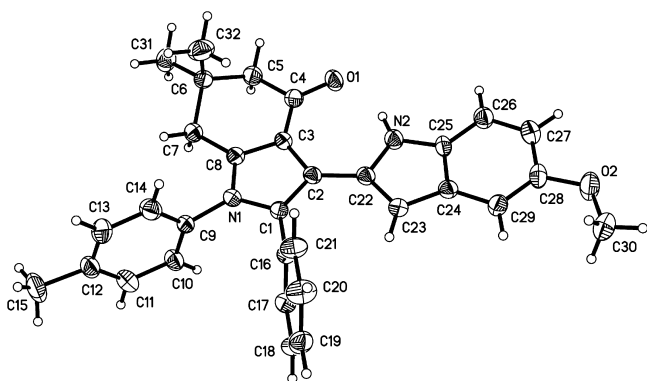


Figure 5. X-ray structure of product 4{1,5,3}.

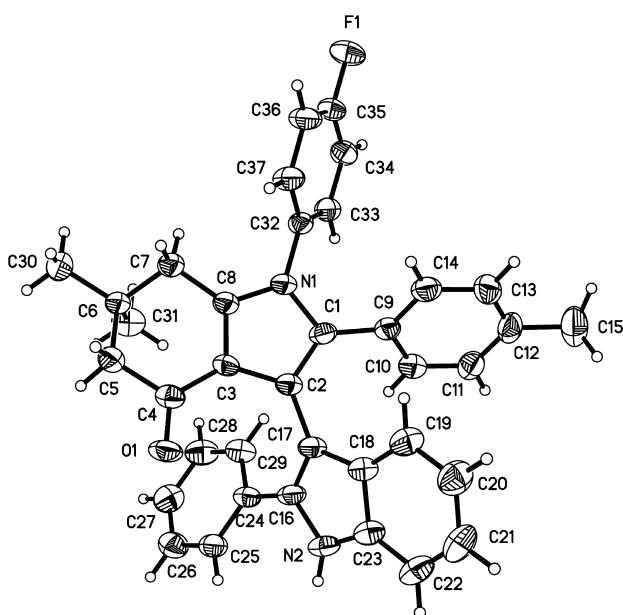


Figure 6. X-ray structure of product 4{4,2,5}.

in a one-pot operation. Up to three sigma-bonds were formed accompanied by the cleavage of C=O and C–O bonds of the aryl glyoxal via intermolecular indolation (Scheme 3).

On the basis of all the above results, a possible mechanism has been proposed for the formation of bis-indoles as shown in

Scheme 3. The reaction involves the ring closure cascade process that consist of initial protonation (1 to A), nucleophilic substitution (A to B), and subsequent second protonation (B to C), second nucleophilic substitution (C to D and E), intramolecular cyclization and final dehydration (E to 4). This regioselectivity is attributed to intramolecular hydrogen bond of intermediate B.

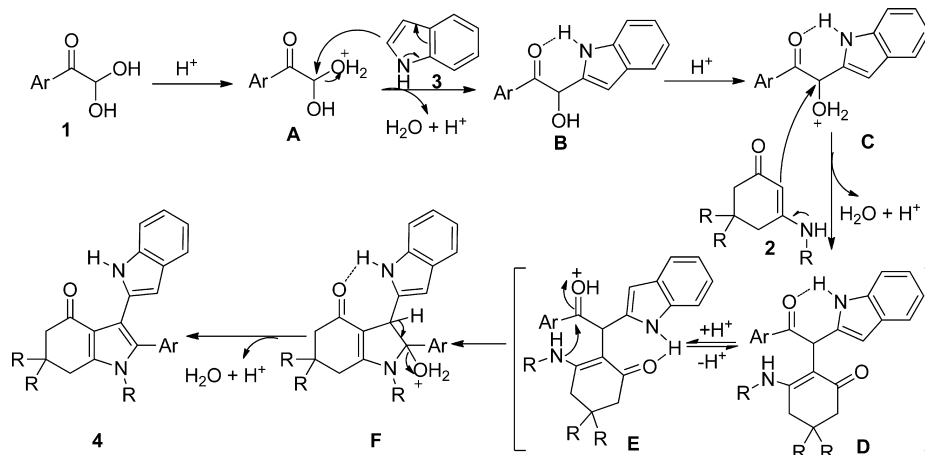
In conclusion, we have developed a three-component domino reaction (arylglyoxal monohydrates, enaminones, and indoles) as a novel method for regioselective synthesis of 3,2'- and 3,3'-bis-indoles by varying the substituted indole substrate. The reaction under acidic condition proceeds by selective [3 + 2] heterocyclization obtaining bis-indoles 4 in good yields, showing that the synthetic route allows us to build blocks of bis-indole derivatives with a wide diversity of substituents. This methodology is simple, practical, and is a regioselective alternative synthetic route to obtain good yields of bis-indole derivatives by microwave irradiation. Features of this strategy include the mild condition, convenient one-pot operation, and excellent regioselectivity. Further investigations are in progress in our laboratory to evaluate the process with a broader range of substrates, and to synthesize more complex products and test their biological activity.

EXPERIMENTAL PROCEDURES

Microwave irradiation was carried out with Initiator 2.5 Microwave Synthesizers from Biotage, Uppsala, Sweden. Melting points were determined in open capillaries and were uncorrected. IR spectra were taken on a FT-IR-Tensor 27 spectrometer in KBr pellets and reported in cm^{-1} . ^1H NMR spectra were measured on a Bruker DPX 400 MHz spectrometer in $\text{DMSO}-d_6$ with chemical shift (δ) given in ppm relative to TMS as internal standard. HRMS (ESI) was determined by using microTOF-QII HRMS/MS instrument (BRUKER). X-ray crystallographic analysis was performed with a Siemens SMART CCD and a Siemens P4 diffractometer.

Typical Procedure for the Preparation of Bis-indoles 4{1,1,1}. In a 10-mL reaction vial, phenylglyoxal monohydrate 1{1} (0.153 g, 1 mmol), 3-((4-chlorophenyl)amino)-5,5-dimethylcyclohex-2-enone 2{1} (0.25g, 1 mmol), indole 3{1} (0.12 g, 1 mmol), and HOAc (1.5 mL) were mixed and then capped (the automatic mode stirring helped the mixing and uniform heating of the reactants). The mixture was heated for 20 min at 80 °C under microwave irradiation. Upon

Scheme 3. Reasonable Mechanism for the Product 4



completion, monitored by TLC, the reaction mixture was cooled to room temperature. A suspension was formed which was then poured into water. The solid product was collected by Büchner filtration, subsequently washed with two different solvents of ethanol and ethylether in sequence to give the pure white solid (0.39 g, yield 84%). m.p.: 270–272 °C.

IR (KBr, ν , cm^{-1}): 3287, 1650, 1494, 1424, 1362, 1132, 1093, 1015, 834, 739, 700, 638, 519; ^1H NMR (400 MHz, DMSO- d_6) δ : 10.91 (s, 1H, NH), 7.48 (d, J = 8.8 Hz, 2H, ArH), 7.31–7.27 (m, 3H, ArH), 7.14 (s, 1H, ArH), 7.04 (t, J = 3.2 Hz, 3H, ArH), 6.98–6.95 (m, 4H, ArH), 6.75 (t, J = 7.2 Hz, 1H, ArH), 2.59 (s, 2H, CH_2), 2.31 (s, 2H, CH_2), 1.08 (s, 6H, CH_3); ^{13}C NMR (100 MHz, DMSO- d_6) δ : 192.2, 143.4, 136.1, 135.6, 132.7, 132.7, 131.4, 130.5, 129.9, 129.2, 127.6, 127.3, 126.9, 125.2, 120.3, 119.9, 118.1, 117.6, 113.9, 111.0, 1.7.6, 52.8, 36.5, 34.5, 28.1; HRMS (ESI): m/z calcd for: $\text{C}_{30}\text{H}_{25}\text{ClN}_2\text{NaO}$, 487.1548 $[\text{M}+\text{Na}]^+$, found: 487.1544.

■ ASSOCIATED CONTENT

■ Supporting Information

Crystallographic data in CIF format. Further details on the experimental procedures and results. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Author Contributions

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Notes

The authors declare no competing financial interest.

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