

# Bis( $\mu$ -diphenylphosphido)bis{*trans*-tricarbonyl-bis[1-(4-ethoxybutoxy)pentylidene]rhenium}

Ulrich Flörke\* and Dina Petters

Fachbereich Chemie und Chemietechnik,  
Universität-GH Paderborn, Warburgerstraße  
100, D-33098 Paderborn, Germany

Correspondence e-mail:  
uf@chemie.uni-paderborn.de

## Key indicators

Single-crystal X-ray study

$T = 293\text{ K}$

Mean  $\sigma(\text{C}-\text{C}) = 0.009\text{ \AA}$

$R$  factor = 0.028

$wR$  factor = 0.066

Data-to-parameter ratio = 18.8

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The geometric centre of the title molecule,  $\text{Re}_2(\mu\text{-PPh}_2)_2(\text{CO})_6[\text{ax-C}(\text{''Bu})(\text{OBuOEt})]_2$  or  $[\text{Re}_2(\text{C}_{11}\text{H}_{22}\text{O}_2)_2(\text{C}_{12}\text{H}_{10}\text{P})_2(\text{CO})_6]$ , lies on a crystallographic inversion centre, realising a *trans*-position of the two *Z*-configured carbene ligands.

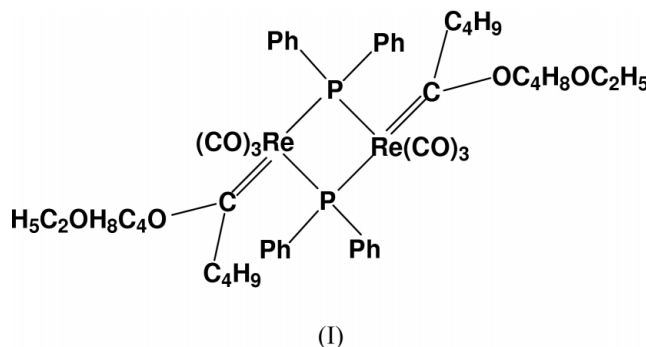
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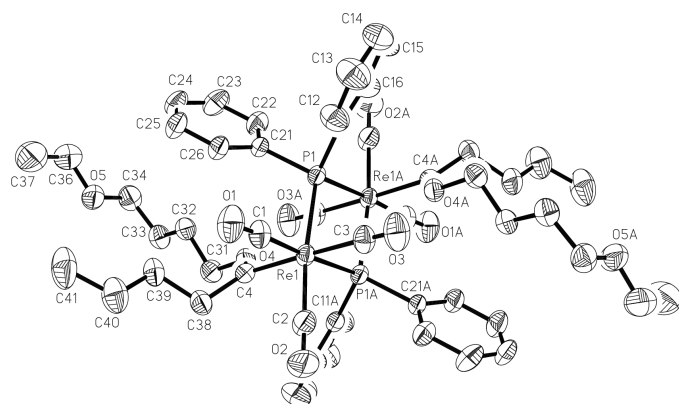
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## Comment

Recently, we reported the preparation and characterization of Fischer-type carbene complexes  $\text{Re}_2(\mu\text{-PPh}_2)_2(\text{CO})_{8-n}(\text{ax-C}(\text{R})\text{OR}')_n$  ( $n = 1, 2$ ;  $\text{R}, \text{R}' = \text{organic residue}$ ) (Haupt *et al.*, 1998). The title compound, (I), is another member of this family.



The centre of the molecule lies on a crystallographic inversion centre. As a consequence, the  $\text{Re}_2\text{P}_2$  ring is planar and the ax-carbene ligands at both Re atoms are in an *anti* position. Additionally, each rhenium is attached to three CO groups and the two common bridging  $\mu\text{-P}$  atoms. The resulting coordination geometry is a slightly distorted octahedron. The CO and carbene ligands show eclipsed arrangement along the  $\text{Re}\cdots\text{Re}$  vector, with torsion angles  $\text{C}-\text{Re}\cdots\text{Re}-\text{C}$  in the range  $0.3\text{--}0.6^\circ$ . The orientation of the carbene ligand along the  $\text{Re}=\text{C}$  bond shows a *Z* configuration, as is already known from the other mentioned compounds of this type, and the  $\text{Re}\cdots\text{Re}-\text{C4}-\text{O4}$  torsion angle of  $3.4(2)^\circ$  indicates the orientation of the  $\text{C}-\text{O}$  bond parallel to the  $\text{Re}\cdots\text{Re}$  vector. The position of both the  $\text{''Bu}$ , as well as the  $\text{OC}_4\text{H}_8\text{OC}_2\text{H}_5$ -chain of the carbene group, avoids intramolecular steric crowding with the phenyl and carbonyl ligands. The shortest non-bonding intramolecular distance is  $1.984(1)\text{ \AA}$  for  $\text{H31A}\cdots\text{H38B}$  and the shortest non-bonding intermolecular interaction is  $2.430(1)\text{ \AA}$  for  $\text{H36A}\cdots\text{H40A}(x-1, y, z)$ , both from the two branches of the carbene ligands. The  $\text{Re}=\text{C}$  carbene bond length is  $2.105(4)\text{ \AA}$  and the  $\text{Re}-\text{P}$  bonds are



**Figure 1**  
The molecular structure with H atoms omitted. Displacement ellipsoids are drawn at the 40% probability level.

2.5328 (15) and 2.5355 (14) Å (to P1 and P1', respectively). These values compare well with those known from the related compounds  $\text{Re}_2(\mu\text{-PPh}_2)_2(\text{CO})_6[\text{ax-C(Ph)OEt}]_2$  or  $\text{Re}_2(\mu\text{-PPh}_2)_2(\text{CO})_6[\text{ax-C(nBu)OMe}][\text{ax-C(Ph)OMe}]$  with  $\text{Re}=\text{C}$  in the range 2.08–2.131 Å and  $\text{Re}-\text{P}$  in the range 2.520–2.540 Å (Haupt *et al.*, 1998).

## Experimental

Reaction of  $\text{trans-Li}_2[\text{Re}_2(\mu\text{-PPh}_2)_2(\text{CO})_6\{\text{C(Bu)O}\}_2]$  with  $\text{CF}_3\text{SO}_3\text{Et}$  (molar ratio 1:2) in  $\text{CHCl}_3$ /pentane solution at room temperature. The synthesis of the precursor Li complex salt has been described elsewhere (Haupt *et al.*, 1998).

### Crystal data

$[\text{Re}_2(\text{C}_{11}\text{H}_{22}\text{O}_2)_2(\text{C}_{12}\text{H}_{10}\text{P})_2(\text{CO})_6]$   
 $M_r = 1283.37$   
 Triclinic,  $P\bar{1}$   
 $a = 10.294$  (6) Å  
 $b = 12.102$  (5) Å  
 $c = 12.230$  (6) Å  
 $\alpha = 69.86$  (2)°  
 $\beta = 86.58$  (2)°  
 $\gamma = 66.39$  (2)°  
 $V = 1305.3$  (11) Å<sup>3</sup>  
 $Z = 1$   
 $D_x = 1.633$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 25 reflections  
 $\theta = 8\text{--}15^\circ$   
 $\mu = 4.75$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
 Prism, yellow  
 $0.49 \times 0.41 \times 0.20$  mm

### Data collection

Bruker P4 diffractometer  
 $\omega$  scans  
 Absorption correction:  $\psi$  scan  
 (North *et al.*, 1968)  
 $T_{\min} = 0.150$ ,  $T_{\max} = 0.409$   
 5891 measured reflections  
 5615 independent reflections  
 4907 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.026$   
 $\theta_{\max} = 27.6^\circ$   
 $h = -13 \rightarrow 13$   
 $k = -14 \rightarrow 15$   
 $l = 0 \rightarrow 15$   
 2 standard reflections  
 every 400 reflections  
 intensity decay: <1%

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.028$   
 $wR(F^2) = 0.066$   
 $S = 1.04$   
 5615 reflections  
 299 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0371P)^2 + 0.5044P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 1.00$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.87$  e Å<sup>-3</sup>  
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.0081 (4)

**Table 1**

Selected geometric parameters (Å, °).

Re1—C4	2.105 (4)	C4—O4	1.304 (5)
Re1—P1 <sup>i</sup>	2.5328 (15)	C4—C38	1.514 (6)
Re1—P1	2.5355 (14)		
P1 <sup>i</sup> —Re1—P1	75.40 (4)	O4—C4—Re1	119.5 (3)
Re1 <sup>i</sup> —P1—Re1	104.60 (4)		

Symmetry code: (i)  $2 - x, -y, 1 - z$ .

Due to a generator failure during data collection about 400 intensities out of about 6000 were not recorded. The largest peak of residual electron density is close to a heavy atom, 0.85 Å from Re1. H atoms were refined at calculated positions riding on the parent C atoms with isotropic displacement parameters  $U_{\text{eq}}(\text{H}) = 1.2U_{\text{iso}}(\text{C})$ .

Data collection: *XSCANS* (Bruker, 1996); cell refinement: *XSCANS*; data reduction: *SHELXTL* (Bruker, 1998); program(s) used to solve structure: *SHELXTL*; program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

## References

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