

1-(Cholest-4-en-3 β -yl)-2,2,2-trichloroethanimidate
tert-butyl methyl ether hemisolvateSimon Fielder,^a Daryl D.
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Key indicators

Single-crystal X-ray study

 $T = 173\text{ K}$ Mean $\sigma(\text{C}-\text{C}) = 0.010\text{ \AA}$

Disorder in solvent or counterion

 R factor = 0.078 wR factor = 0.221

Data-to-parameter ratio = 8.2

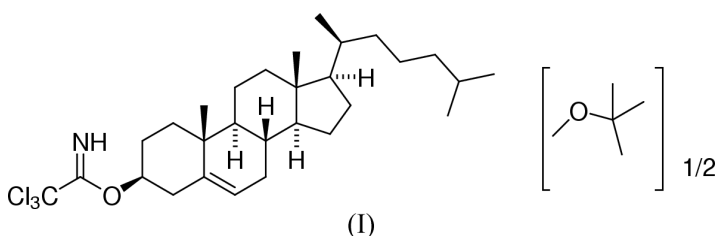
For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The title compound, $\text{C}_{29}\text{H}_{45}\text{Cl}_3\text{NO} \cdot 0.5\text{C}_5\text{H}_{12}\text{O}$, is a rare
example of a crystalline trichloroacetimidate and has a
structure similar to that reported for analogous species.

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Comment

As part of our studies on the synthesis and biological prop-
erties of certain terpene peroxides (Fielder *et al.*, 1996, 1998),
we investigated the preparation of hydroperoxides from tri-
chloroacetimidates by a modification of the reported proce-
dure for dialkyl peroxides (Bourgeois *et al.*, 1993). The X-ray
structure determination of the title compound, (I), was
performed as a means of structural identification, since (i) the
compound could not be transformed into peroxide analogues,
and (ii) the compound was crystalline, evidently a very
unusual property for trichloroacetimidates (Clegg *et al.*, 1995).The structure was solved in C2 with $Z = 2$ and so (I) is chiral:
the correct enantiomer was determined unambiguously. While
there are very few structurally characterized trichloro-
acetimidates (Nakano *et al.*, 1988; Clegg *et al.*, 1995; Ernst *et al.*,
1998), they all show generally similar features to the title
compound. The C_3NO skeleton has essentially a planar
configuration with a r.m.s. deviation from planarity of 0.024 \AA .

Experimental

The title compound was prepared in 67% yield, using a modification
of the standard procedure (Overman, 1976). IR max (KBr disc): 3342
(N—H), 1654 (C=NH) cm^{-1} ; ^1H NMR (270 MHz, CDCl_3): δ 8.24
(1H, s, =NH), 5.42 (1H, m, =CH—), 4.77 [1H, m, —CH(OR)—], 2.47
(2H, m, —CH₂—C=), 2.05–0.92 (26H, m), 1.07 (3H, s, —CH₃), 0.93 [3H,
d, $J = 6.4\text{ Hz}$, —CH(CH₃)—], 0.88 [3H, *d*, $J = 6.6\text{ Hz}$, —CH(CH₃)CH₃],
0.87 [3H, *d*, $J = 6.6\text{ Hz}$, —CH(CH₃)CH₃], 0.69 (3H, s, —CH₃) p.p.m.

Crystal data

 $\text{C}_{29}\text{H}_{45}\text{Cl}_3\text{NO} \cdot 0.5\text{C}_5\text{H}_{12}\text{O}$ $M_r = 574.08$

Monoclinic, C2

 $a = 29.037(13)\text{ \AA}$ $b = 9.072(6)\text{ \AA}$ $c = 12.628(6)\text{ \AA}$ $\beta = 109.14(7)^\circ$ $V = 3143(3)\text{ \AA}^3$ $Z = 4$ $D_x = 1.213\text{ Mg m}^{-3}$ Mo $K\alpha$ radiationCell parameters from 25
reflections $\theta = 9\text{--}16^\circ$ $\mu = 0.32\text{ mm}^{-1}$ $T = 173(2)\text{ K}$

Plate, colorless

 $0.50 \times 0.50 \times 0.30\text{ mm}$

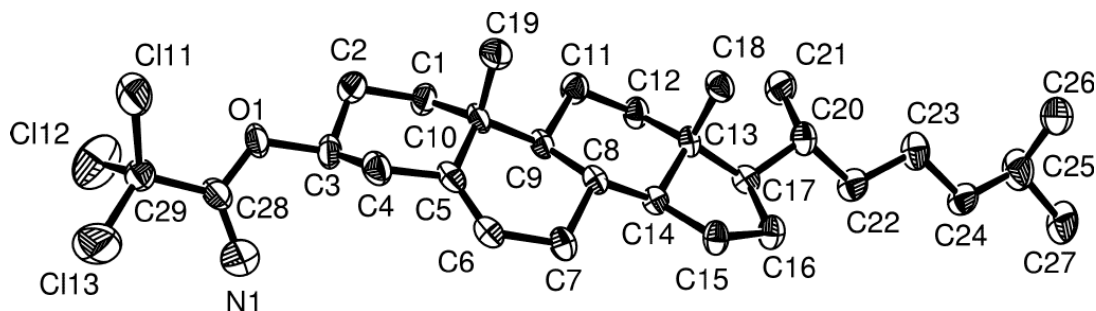


Figure 1

The molecular structure of the title compound showing 50% probability displacement ellipsoids (Farrugia, 1997). The *tert*-butyl methyl ether solvate and all H atoms have been omitted for clarity.

Data collection

Enraf–Nonius CAD-4 diffractometer
 ω scans
 Absorption correction: empirical (North *et al.*, 1968)
 $T_{\min} = 0.857$, $T_{\max} = 0.911$
 3008 measured reflections
 2926 independent reflections
 2640 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.022$
 $\theta_{\max} = 25.0^\circ$
 $h = -33 \rightarrow 0$
 $k = 0 \rightarrow 10$
 $l = -12 \rightarrow 14$
 3 standard reflections
 every 100 reflections
 intensity decay: $<2\%$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.078$
 $wR(F^2) = 0.221$
 $S = 1.03$
 2926 reflections
 358 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1249P)^2 + 13.5363P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.048$
 $\Delta\rho_{\max} = 0.97 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.72 \text{ e } \text{\AA}^{-3}$
 Absolute structure: Flack (1983)
 Flack parameter = 0.00 (18)

Data collection: *CAD-4/PC* (Enraf–Nonius, 1993); cell refinement: *CAD-4/PC*; data reduction: *XCAD4B* (Harms, 1994); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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A disordered half molecule of the crystallization solvent, *tert*-butyl methyl ether, is present in the lattice and therefore the C—C distances and the C—O distances were fixed at 1.55 (1) and 1.45 (1) Å, respectively, using the command *DFIX* (Sheldrick, 1997). All non-H atoms are refined anisotropically. All H atoms were placed in calculated positions, refined using a riding model, and given an isotropic displacement parameter equal to 1.2 times the equivalent isotropic displacement parameter of the atom to which they are attached.