

2'-Amino-5'-bromoacetophenone

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Key indicators

Single-crystal X-ray study

T = 291 K

Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$

R factor = 0.024

wR factor = 0.061

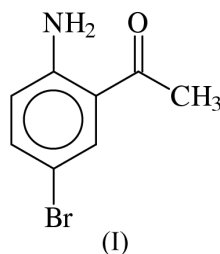
Data-to-parameter ratio = 10.9

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The synthesis and structure of a brominated derivative of 2'-aminoacetophenone, $\text{C}_8\text{H}_8\text{BrNO}$, is described. The conformation observed in the solid state was found to be the same as that previously determined by NMR studies.

Comment

As part of a project to prepare novel analogues of the cytotoxic marine alkaloid ascididemin (Lindsay *et al.*, 2000), we required derivatives of 2'-aminoacetophenone. Mild bromination of 2'-aminoacetophenone with pyridinium tribromide (Fieser & Fieser, 1967) afforded the desired product, 2'-amino-5'-bromoacetophenone (I). The conformation of the molecule (Fig. 1) was shown to have the carbonyl group directed towards the amino group so as to maximize H-bonding, and the acetyl methyl group directed towards H-6. A similar solution conformation for this compound has been previously deduced by NMR (Batts *et al.*, 1977).



There is an intramolecular hydrogen bond (Table 1) between an amine proton and the carbonyl O atom, $\text{N} \cdots \text{O}$ 2.679 (3) \AA . There is an additional weak interaction between the other amine proton and the ketone oxygen of an adjacent molecule, $\text{N} \cdots \text{O}$ 3.177 (3) \AA .

Experimental

Reaction of 2'-aminoacetophenone with pyridinium tribromide (Fieser & Fieser, 1967) in CH_2Cl_2 yielded 2'-amino-5'-bromoacetophenone (80%) with only trace amounts of the dibrominated product 2'-amino-3',5'-dibromoacetophenone being present (Baker *et al.*, 2001). Chromatography on silica gel (CH_2Cl_2 -hexane) afforded pure title compound that was recrystallized from EtOH as pale yellow tablets. M.p. 355–356 K [literature (Simpson *et al.*, 1945) 357–358 K, (Gibson & Levin, 1931) and 359–361 K]. ^1H NMR (CDCl_3 , 200 MHz) δ 7.77 (1H, *d*, $J = 2.2$ Hz, H-6), 7.29 (1H, *dd*, $J = 8.8, 2.2$ Hz, H-4), 6.53 (1H, *d*, $J = 8.8$ Hz, H-3), 6.16 (2H, *bs*, NH_2), 2.53 (3H, *s*, Ac). ^{13}C NMR (CDCl_3 , 50 MHz) δ 199.4 (C-7), 148.9 (C-2), 136.7 (C-4), 133.8 (C-6), 119.0 (C-1), 118.7 (C-3), 106.2 (C-5), 27.5 (C-8). EIMS 215 / 213 (100%), 200 / 198, 172 / 170. HREIMS m/z 214.9773 (calculated for $\text{C}_8\text{H}_8^{81}\text{BrNO}$, 214.9769), 212.9794 (calculated for $\text{C}_8\text{H}_8^{79}\text{BrNO}$, 212.9789). Analysis calculated for $\text{C}_8\text{H}_8\text{BrNO}$: C, 44.89; H, 3.77; N,

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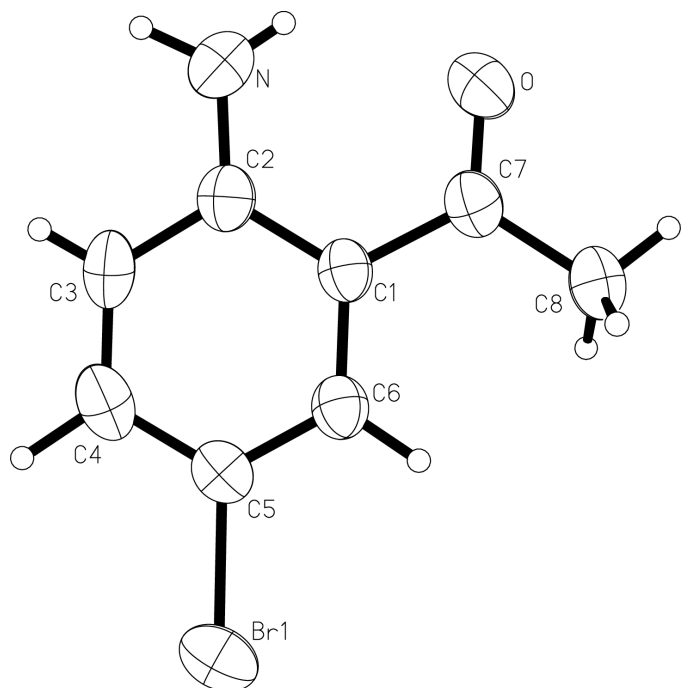


Figure 1
Structure of (I) showing 50% probability displacement ellipsoids and H atoms as arbitrary spheres (Siemens, 1994).

6.54; found C, 44.82; H, 3.51; N, 6.61%.

Crystal data

C_8H_8BrNO
 $M_r = 214.06$
Orthorhombic, $Pna2_1$
 $a = 21.9206(6) \text{ \AA}$
 $b = 7.3583(2) \text{ \AA}$
 $c = 5.0596(2) \text{ \AA}$
 $V = 816.11(4) \text{ \AA}^3$
 $Z = 4$
 $D_x = 1.742 \text{ Mg m}^{-3}$

Data collection

Siemens SMART diffractometer
Area detector ω scans
Absorption correction: multi-scan
(Blessing, 1995)
 $T_{\min} = 0.229$, $T_{\max} = 0.468$
4318 measured reflections
1347 independent reflections

Mo $K\alpha$ radiation
Cell parameters from 3623 reflections
 $\theta = 2-25^\circ$
 $\mu = 4.97 \text{ mm}^{-1}$
 $T = 291(2) \text{ K}$
Prism, yellow
 $0.42 \times 0.22 \times 0.18 \text{ mm}$

1278 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$
 $\theta_{\max} = 25.6^\circ$
 $h = -26 \rightarrow 23$
 $k = -8 \rightarrow 8$
 $l = -6 \rightarrow 4$
Intensity decay: $<2\%$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.024$
 $wR(F^2) = 0.061$
 $S = 1.08$
1347 reflections
124 parameters
H atoms treated by a mixture of constrained and independent refinement

$w = 1/[\sigma^2(F_o^2) + (0.0374P)^2 + 0.0296P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.007$
 $\Delta\rho_{\max} = 0.18 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.41 \text{ e \AA}^{-3}$
Absolute structure: Flack (1983)
Flack parameter = 0.059 (13)

Table 1

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N-H0b\cdots O$	0.86	2.05	2.676 (3)	129
$N-H0b\cdots O^i$	0.86	2.50	3.177 (3)	137

Symmetry code: (i) $1-x, -y, z-\frac{1}{2}$.

H atoms were placed geometrically and coordinates, apart from those of the methyl group, allowed to refine with isotropic displacement parameter riding on U_{eq} of the carrier atom. The methyl group was refined using a riding model.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1994); software used to prepare material for publication: *SHELXL97*.

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