

*Crystallographic report***Bis(indole-3-acetato)(1,10-phenanthroline)
cadmium(II)****Zhen-Feng Chen^{1*}, Ling Huang¹, Hong Liang^{1**}, Rui-Xiang Hu¹, Shao-Ming Shi¹
and Hoong-Kun Fun²**¹College of Chemistry and Chemical Engineering, Guangxi Normal University, Guilin 541004, P. R. China.²X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 Penang, Malaysia

Received 30 July 2003; Revised 8 August 2003; Accepted 9 August 2003

The structure of Cd(phen)(indole-3-acetato)₂ has twofold symmetry and features a six-coordinated distorted octahedral geometry around cadmium(II), defined by an N₂O₄ donor set, with Cd–O distances ranging from 2.214(3) to 2.526(3) Å. Copyright © 2003 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; cadmium; indole-3-acetic acid**COMMENT**

Indole-3-acetic acid (IAH), a phytohormone of the auxin series and a substance of essential and multifunctional biological significance, has been established as an essential component of associative plant–microbe interactions¹ and has an influence on the availability of cadmium in soils. Although a few crystallographic studies on metal complexes of indole-3-acetate have been performed, such as on [Pd(IA)(py)]₂·4CHCl₃² and [Pt(bpm)(L-Ala)](IA)·7H₂O,³ thus far, IA complexes with the toxic heavy-metal cadmium(II) have not yet been characterized by X-ray crystallography. The investigation of the crystal structure of cadmium(II) complexes with IAH and derivatives is of fundamental importance in aiding our understanding of its biological function, such as acting as phytochelatin to decrease the toxicity of cadmium.⁴ The structure of Cd(phen)(IA)₂ (Fig. 1) has crystallographic twofold symmetry and looks like a

butterfly, similar to Cd(S₂CNEt₂)₂(2,9-Me₂-1,10-phen),⁵ by virtue of the presence of chelating IAH ligands, and the cadmium center exists in a distorted octahedral environment defined by an N₂O₄ donor set with Cd–O distances ranging from 2.214(3) to 2.526(3) Å.

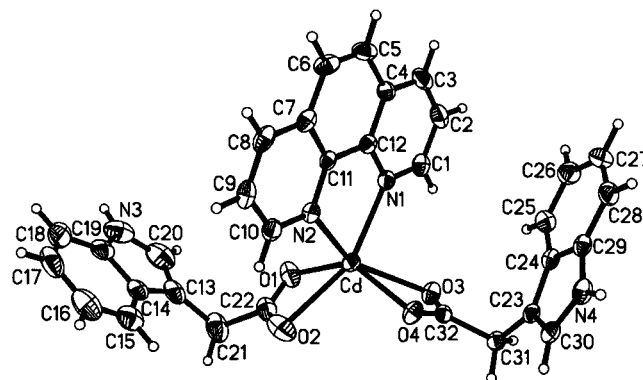


Figure 1. Molecular structure of Cd(phen)(IA)₂. Key geometric parameters: Cd–O1 2.234(4), Cd–O2 2.439(4), Cd–O3 2.526(3), Cd–O4 2.214(3), Cd–N1 2.308(4), Cd–N2 2.309(4) Å; O1–Cd–O2 55.81(17), O1–Cd–O3 105.93(13), O1–Cd–O4 136.12(15), O1–Cd–N1 113.37(15), O1–Cd–N2 105.23(14), O2–Cd–O3 115.35(15), O2–Cd–O4 94.41(17), O2–Cd–N1 156.68(16), O2–Cd–N2 89.41(15), O3–Cd–O4 54.58(11), O3–Cd–N1 86.87(13), O3–Cd–N2 147.67(12), O4–Cd–N1 104.90(13), O4–Cd–N2 105.81(12), N1–Cd–N2 73.01(14)°.

*Correspondence to: Zhen-Feng Chen, Department of Chemistry and Chemical Engineering, Guangxi Normal University, Guilin 541004, People's Republic of China.

E-mail: chenzfgxnu@yahoo.com

**Correspondence to: Hong Liang, Department of Chemistry and Chemical Engineering, Guangxi Normal University, Guilin 541004, People's Republic of China.

E-mail: chenzfgxnu@yahoo.com

Contract/grant sponsor: YSF of Guangxi Province of China.

Contract/grant sponsor: NSF of Guangxi Province of China.

Contract/grant sponsor: Project of One Hundred of Youth Academic Subject Leaders of Guangxi Universities, China.

Contract/grant sponsor: Malaysian Government/University Sains Malaysia; Contract/grant number: 305/PFIZIK/610961.

EXPERIMENTAL

To a methanolic solution (5 ml) containing IAH (0.2 mmol), which was adjusted to pH = 7 using 1 mol l^{-1} NaOH, was added an aqueous $\text{Cd}(\text{OAc})_2$ (0.2 mmol) solution (10 ml) with stirring at 60°C . After 3 h, a methanolic solution (10 ml) of 1,10-phenanthroline (0.1 mmol) was added with 6 h stirring. The reaction mixture was filtered and the filtrate was allowed to stand in air; yellow, block crystals were obtained within 3 weeks. Yield 60% (based on IAH). Anal. Found: C, 60.04; H, 3.85; N, 8.69%. Calc. for $\text{C}_{32}\text{H}_{24}\text{CdN}_4\text{O}_4$: C, 59.96; H, 3.77; N, 8.74. X-ray diffraction data were collected at 293(2) K on a Siemens SMART CCD area detector diffractometer using graphite-monochromated $\text{MoK}\alpha$ radiation on a block $0.20 \times 0.34 \times 0.40 \text{ mm}^3$. Crystallographic data: $\text{C}_{32}\text{H}_{24}\text{CdN}_4\text{O}_4$, $M = 640.95$, monoclinic, space group $P2_1/c$, $a = 12.4378(8)$, $b = 12.3442(8)$, $c = 18.2971(11) \text{ \AA}$, $\beta = 106.223(1)^\circ$, $V = 2697.4(3) \text{ \AA}^3$, $Z = 4$, $D_c = 1.578 \text{ Mg m}^{-3}$, 16 625 reflections collected, 6668 unique and 4752 with $I > 2\sigma(I)$. R (obs. data on F^2) 0.070, wR (all data) 0.132, $\rho_{\text{max}} = 1.03 \text{ e}^- \cdot \text{ \AA}^{-3}$. Programs used: SHELXTL97, ORTEP. CCDC deposition number: CCDC 216 213.

Acknowledgements

The Youth Science Foundation of Guangxi of the People's Republic of China, the State Key Laboratories of Coordination Chemistry

Visitor Program of the People's Republic of China, and the Project of One Hundred of Youth Academic Subject Leaders of Guangxi Universities of China are thanked for support. We also thank the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 305/PFIZIK/610961.

REFERENCES

1. Kamnev AA, Shchelochkov AG, Tarantilis PA, Polssiou MG, Perfiliev YD. *Monatsh. Chem.* 2001; **132**: 675.
2. Takani M, Masudu H, Yamauchi O. *Inorg. Chim. Acta* 1995; **235**: 367.
3. Yajima T, Maccarrone G, Takani M, Contino A, Arena G, Takamido R, Hanaki M, Funahashi Y, Odani A, Yamauchi O. *Chem. Eur. J.* 2003; **9**: 3341.
4. Johanning J, Strasdeit H. *Angew. Chem. Int. Ed. Engl.* 1998; **37**: 2464.
5. Lai CS, Tiekink ERT. *Appl. Organometal. Chem.* 2003; **17**: 139.