Summary

- (1) The antibacterial action of organic arsenical compounds was inhibited by thiol compounds but not by sulfanilamide-antagonists.
- (2) The activity of succinic dehydrogenase of various bacteria was inhibited by organic arsenical compounds. This inhibitory effect was considered parallel to the bacteriostatic activities of organic arsenical compounds.

(Received November 5, 1953)

5. Takeo Ueda and Tadakazu Tsuji: Arsenical Chemotherapeutic Drugs. XV. Studies on Arsenical Compounds as Anthelmintics.

(Pharmaceutical Institute, Keio-Gijuku University*)

Many informations, regarding the trypanocidal, spirocidal, and antibacterial activities of the arsenical drugs, have been gained, but up to date no observations have been reported regarding their anthelmintic activities.

Studies to determine the most suitable type of chemical structure among the arsenical compounds and the most appropriate method of the anthelmintic tests for them were undertaken before proceeding with their detailed studies. Several arsenical compounds were examined as to their anthelmintic properties by the usual method. This paper describes the anthelmintic activities of several arsenical compounds against Ascaris lumbricoides.

Compounds Used to Test Anthelmintic Action Three series of arsenical compounds, viz. the diarylarsinic acids, the arylarsenous acids and the diarylarsinous acids, as shown in the following tables, were tested for their anthelmintic properties. All of the compounds are known. Among them, the diarylarsinic acids and the diarylarsinous acids were prepared by Ueda-Takahashi method¹⁾.

Anthelmintic Action According to the Lamson-Nakamura method²⁾, these arsenical compounds were examined as to their anthelmintic activities by observing the kinetic state of *Ascaris lumbricoides* exposed in dilution of 1:1,000 Bunge's solution of the chemicals at 38°.

Results are given in Tables I, II, and III. Numbers under the heading "Paralysis" denote the time after which the ascarid showed no movement without outside stimulus, and under "Death", the time after which the ascarid showed no movement under any stimulus.

As shown in the tables, none of the diarylarsinic acids showed any action, though some of the arylarsenous acids and the diarylarsinous acids exerted considerable effect: Phenylarsenous acid (No. 16), 4-tolylarsenous acid (No. 17), 4-hydroxyphenylarsenous acid (No. 18), and 4-hydroxydiphenylarsinous acid (No. 22) showed marked ascaricidal effect and 3-nitro-4-hydroxyphenylarsenous acid (No. 19), 3-amino-4-hydroxyphenylarsenous acid (No. 20), diphenylarsinous acid (No. 21), 4-nitrodiphenylarsinous acid (No. 23), 4-chlorodiphenylarsinous acid (No. 24), 2-carboxydiphenylarsinous acid

^{*)} Shinano-machi, Shinjuku-ku, Tokyo (上田武雄, 辻 忠和).

¹⁾ K. Takahashi: J. Pharm. Soc. Japan, 72, 533 (1952).

²⁾ E. Nakamura: Sei-I-Kai Med. J., 65, 3, 181 (1951).

29

4					T	ABLE I.	•					
			•				R	/			. *	
		Antl	helmintic	Action	of Diaryl	arsinic Ac		\	- As(D_2H	<u>.</u>	
							R'			-		
						•	*	\	_/			
No.		1 25	Struct	ural for	mula			. <u> </u>	Sta	te	<u> </u>	
		R		*	R′		Para	lysis	(hr.)		Death	(hr.)
1		\mathbf{H}			H			65			72	
2		H			4'-NO ₂			72			84	
3		H H			4'-NH ₂			42 65	•		68 7 2	
4 5		H			4'-CH ₃ 2'-CH ₃			39			65	
6		H			4'-COOH			72			84	
7		H			2'-COOH		* . *	42			68	
8		H			4'-OH, 3			39			78	
9		H	* *		4'-OH, 3			72			84 84	
10 11		$_{ m 4-NH_2}$			4'-CH ₃ , 3 4'-NH ₂	7-NO2		69 72			84	
12		4-CH ₃			4'-CH ₃			72			84	
13		2-CH ₃			2′-CH ₃			24			65	
14		4-COO	H		4'-COOH			72			84	
15		$4-NO_2$			4'-OH, 3'	-NO ₂	**	72			84	
									1			
					T_{A}	BLE II.		. *				
			e perteñajo. O de la estado de			And the						• •
			Anthelmi	intic Ac	tion of Ar	ylarsenous	3 R		>-AsO			
			Acids (A	rylarse	ne Oxides) ,					. *	
	No.		S	Structur	al formula				State	100		
	No.		S			ı	ralvsis	(hr.)		 Death	(hr.)	
			S	1	al formula R	ı	ralysis	(hr.)		Death	(hr.)	
	No.		S	H	R	ı	ralysis 1 1.5	(hr.)		Death	(hr.)	•
	16 17 18		\$	H 4-CH 4-OH	R [3	ı	1 1.5 9	(hr.)			2 2 15	
	16 17 18 19		\$	H 4-CH 4-OH 4-OH	R [₃ [[, 3-NO ₂	ı	1 1.5 9 16	(hr.)			2 2 15 39	•
	16 17 18		\$	H 4-CH 4-OH 4-OH	R [3	ı	1 1.5 9	(hr.)			2 2 15	
	16 17 18 19		\$	H 4-CH 4-OH 4-OH	R [₃ [[, 3-NO ₂	ı	1 1.5 9 16	(hr.)			2 2 15 39	
	16 17 18 19		\$	H 4-CH 4-OH 4-OH	R [[, 3-NO ₂ [, 3-NH ₂	ı	1 1.5 9 16	(hr.)			2 2 15 39	
	16 17 18 19		\$	H 4-CH 4-OH 4-OH	R [[, 3-NO ₂ [, 3-NH ₂	Pa	1 1.5 9 16 16				2 2 15 39	
	16 17 18 19			H 4-CH 4-OH 4-OH	R [3 [4] 3-NO ₂ [4] 3-NH ₂	Pa	1 1.5 9 16 16				2 2 15 39	
	16 17 18 19	Antl		H 4-CH 4-OH 4-OH	R [3 [4] 3-NO ₂ [4] 3-NH ₂	Pa	1 1.5 9 16 16 16				2 2 15 39	
	16 17 18 19	Antl		H 4-CH 4-OH 4-OH	R [3 [4] 3-NO ₂ [4] 3-NH ₂	Pa	1 1.5 9 16 16				2 2 15 39	
No.	16 17 18 19	Antl	helmintic	H 4-CH 4-OH 4-OH 4-OH	R [3 [7 [7 [7 [7 [7 [7 [7 [7 [7 [7 [7 [7 [7	Pa	1 1.5 9 16 16 16			sOH.	2 2 15 39	
No.	16 17 18 19		helmintic	H 4-CH 4-OH 4-OH	R [3 [4 [5 [7 [7 [7 [7 [7 [7 [7 [7 [7	Pa	1 1.5 9 16 16 R.c.ids		A	sOH.	2 2 15 39 65	
	16 17 18 19	R	helmintic	H 4-CH 4-OH 4-OH 4-OH	R $\begin{bmatrix} 3 \\ 1 \\ 1 \\ 3 \\ 3 \\ NO_2 \\ 1 \\ 3 \\ NH_2 \end{bmatrix}$ Tate of Diaryla mula	Pa	1 1.5 9 16 16 R.c.ids	lysis	A	sOH.	2 2 15 39 65	
21	16 17 18 19	R H	helmintic	H 4-CH 4-OH 4-OH 4-OH	R I, 3-NO ₂ I, 3-NH ₂ Tan of Diaryla mula R' H	Pa	1 1.5 9 16 16 R.c.ids	lysis	A	sOH.	2 2 15 39 65)
21 22	16 17 18 19	R H H	helmintic	H 4-CH 4-OH 4-OH 4-OH	TAI of Diaryla mula R' H 4'-OH	Pa	1 1.5 9 16 16 R.c.ids		A	sOH.	2 2 15 39 65 Death 39 12) -
21 22 23 24	16 17 18 19	R H H H	helmintic	H 4-CH 4-OH 4-OH 4-OH	TAI of Diaryla mula R' H 4'-OH 4'-NO ₂ 4'-Cl	Pa	1 1.5 9 16 16 R.c.ids	lysis 4 7 16 16	A	sOH.	2 2 15 39 65 Death 39 12 39 36)
21 22 23 24 25	16 17 18 19	R H H H H	helmintic	H 4-CH 4-OH 4-OH 4-OH	TA: of Diaryla mula R' H 4'-OH 4'-NO ₂ 4'-Cl 2'-COOH	Pa	1 1.5 9 16 16 R.c.ids	lysis 4 7 16 16 16 16	A	sOH.	2 2 15 39 65 65 Death 39 12 39 36	
21 22 23 24 25 26	16 17 18 19	R H H H H H	helmintic	H 4-CH 4-OH 4-OH 4-OH	TA: of Diaryla mula R' H 4'-OH 4'-NO ₂ 4'-Cl 2'-COOH 3'-NH ₂ •H	Pa	1 1.5 9 16 16 R.c.ids	lysis 4 7 16 16 16 72	A	sOH.	2 2 15 39 65 65 Death 39 36 42 84)
21 22 23 24 25 26 27	16 17 18 19	R H H H H H 4-NO ₂	helmintic Struct	H 4-CH 4-OH 4-OH 4-OH	Ta: of Diaryla mula R' H 4'-OH 4'-NO ₂ 4'-Cl 2'-COOH 3'-NH ₂ •H 4'-NO ₂	Pa	1 1.5 9 16 16 R.c.ids	lysis 4 7 16 16 16 72 16	A	sOH.	2 2 15 39 65 65 Death 39 12 39 36) ; ; ;
21 22 23 24 25 26	16 17 18 19	R H H H H H	helmintic Struct	H 4-CH 4-OH 4-OH 4-OH	TA: of Diaryla mula R' H 4'-OH 4'-NO ₂ 4'-Cl 2'-COOH 3'-NH ₂ •H	Pa	1 1.5 9 16 16 R.c.ids	lysis 4 7 16 16 16 72	A	sOH.	2 2 15 39 65 65 Death 39 36 42 84 39) ; ; ;

TABLE IV.

Toxicity	of	Arsenical	Compounds	in	Mice

No.	The second second	M.T.D $(i.v.)$ $(mg./kg.)$
16		1.57
17		1.72
18	e de la companya del companya de la companya del companya de la co	2.0
19		7.5
20		40.0
21		10.0
22		2.5
25		10.0
26		20.0
29		20.0
30		40.0

(No. 25), 4,4'-dinitrodiphenylarsinous acid (No. 27), 4-carboxy-4'-nitrodiphenylarsinous acid (No. 28) and diphenyl-2-carboxyphenylthioarsinite (No. 29) showed marked paralysing effect. However, none of them showed any excitation and curling-motion.

Discussion Although the arsenicals are of interest as invaders against SH-ferments in various kinds of living bodies, their anthelmintic properties have not yet been discussed.

This is the first paper to show that arsenicals possess activity against the ascarid, as described in the experimental part.

Relationship between chemical structure and anthelmintic property of the arsenicals is deduced by the fact that the quinquevalent arsenicals did not show any anthelmintic activity, while the trivalent arsenicals exerted considerable effect against the ascarid. This coincides with findings that the trivalent arsenicals possess, in general, far stronger microbial effects than the quinquevalent arsenicals.

The effects of the trivalent arsenicals were observed to depend on their substituent groups. It was observed that the arylarsenous series were, in general, more effective than the diarylarsinous series. From Tables II and III, compounds No. 16, 17, 18, and 22 showed remarkable ascaricidal activity, while compounds No. 19, 20, 21, 23, 24, 25, 27, and 28, showed only moderate activity. Effectivity was observed to be qualitatively parallel to their toxicity. Their effectivity was lowered by introduction of thiol-residue into the arsinous acids for the purpose of reducing toxicity. A good balance of effectivity and toxicity, among the diarylarsinous acid series, seems to be the basis of a promising drug.

At the present stage, it is impossible to conclude what type of the compound is of promise. Additional studies are necessary to determine compounds possessing higher effectivity and lower toxicity.

In view of the mode of anthelmintic action of the diarylarsinous acids against Ascaris lumbricoides, it is of interest that, though these arsenicals are not capable of

killing the ascarid as rapidly as alkylresorcinols, they paralyse the ascarid after a shorter interval. This property of the diarylarsinous acids is a distinguishing characteristic of their mode of action, in contrast with those of santonin and alkylresorcinols. From this finding, it may be requisite to obtain more appropriate methods for evaluating anthelmintic properties of arsenicals. Further work on this problem will be reported in this Bulletin in the future.

Our sincere thanks are due to Prof. Dr. T. Nakao and Dr. E. Nakamura of Jikei-Kai Medical College for their kind help in testing the anthelmintic action.

Summary

- (1) Three series of arsenical compounds, viz. the diarylarsinic acids, arylarsenous acids, and diarylarsinous acids were examined as to their anthelmintic activities against *Ascaris lumbricoides* by the Lamson-Nakamura method.
 - (2) Diarylarsinic acids did not show any anthelmintic effect.
- (3) Some of the arylarsenous acids and diarylarsinous acid exerted considerable effects against the ascarid but did not produce any curling-motion effect.

(Received November 5, 1953)

6. Takeo Ueda and Tadakazu Tsuji: Studies on Anthelmintics. II. Studies on Anthelmintic Compounds Related to Santonin.

(Pharmaceutical Institute, Keio-Gijuku University*)

These studies were undertaken with the aim of establishing the relationship between the chemical constitution and the activity of santonin as well as its analogs. As described in the previous paper, 1) al-hydroxytetralins have been found to possess weak anthelmintic activity against Ascaris lumbricoides.

On the basis of these results, it has been assumed that the activity of santonin might be due, at least partially, to the hydroxyl group in its 1-position. That is, the lactone-ring of santonin might contribute to the increase of anthelmintic activity as well as the hydroxyl group in the 1-position.

Though numerous studies^{2~5)} on the mode of action of santonin have been conducted, conclusions obtained from them have not always coincided with one another, and are contrary to our assumption.

This paper describes the relationship between the chemical structure and anthelmintic activity of several compounds possessing a lactone-ring.

Compounds Used to Test Anthelmintic Action 1-Hydroxytetralin-2-propionic acid lactone derivatives (I), 4-methyl coumarin derivatives (II), 2-methylchromone derivatives (III), 2-methylchromane derivatives (IV), and phenylbutyrolactone derivatives (V), as shown in the following tables, were employed to test their curling-motion

^{*} Shinano-machi, Shinjuku-ku, Tokyo (上田武雄, 辻 忠和).

¹⁾ T. Ueda, T. Tsuji: This Bulletin, 1, 32(1953).

²⁾ T. Nakamura: Sei-I-Kai Med. J., 52, 218 (1933).

³⁾ A. Shirane: Tokyo Med. J., 53, 9, 122 (1949).

¹⁾ K. Mineshita: Folia Pharmacol. Japonica, 44, 3, 26 (1948).

⁵⁾ K. Matsuda: *Ibid.*, 44, 3, 26 (1948).