U.D.C. 577.1:547.963.32:547.784.1.02

21. Teruhisa Noguchi: Studies on the Biosynthesis of Nucleic Acids. III<sup>1)</sup>.

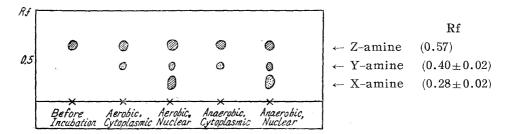
A New Arylamine found during Biosynthesis of DNA.

(Biochemical Institute, Medical Faculty, University of Tokyo\*)

During the course of studies on the incorporation of 4-aminoimidazole(4-14C)-5-carboxamide(AICA-14C)into polynucleotide purines with subcellular fractions of pigeon liver,<sup>2)</sup> the author noted the net incorporation of AICA-14C into deoxyribonucleic acid (DNA) without adding deoxyribose in the medium which contained only ribose-5'-phosphate and fructose-1,6-diphosphate as sole sources of sugar. On the assumption that AICA-14C should be bound with deoxyribose moiety, which would be derived from ribose-5'-phosphate or fructose-1,6-diphosphate before incorporation into polynucleotides, attempt was made to find these intermediates (e.g. "active" AICA deoxyriboside) by paper chromatography of the acid-soluble fraction of the incubation medium.

Actually, two or three spots were found in the acid-soluble fraction of each incubation. They were designated as X (Rf  $0.28\pm0.02$ ), Y (Rf  $0.40\pm0.02$ ), and Z (Rf 0.57) as demonstrated in Fig. 1.

Fig. 1. Paper Chromatogram of Acid-soluble Fraction after Incubation with Subcellular Fractions



As all these spots were positive to color reaction by the Tsuda reagent<sup>3)</sup> or the Pauly reagent<sup>4)</sup> on the paper chromatogram, it is postulated that these substances, X, Y, and Z, are arylamine derivatives.

In Table I and Fig. 1, it is demonstrated that the cytoplasmic fraction did not yield the X-amine spot but afforded only Y- and Z-amine spots (Pauly reaction: pale purple and blue, respectively). Under the same conditions, the incubation with nuclear fraction offered definite X-amine spot (Pauly reaction: yellow). It is of interest to note that, whereas formation of X-amine was stimulated in the aerobic glycolysis system, it was suppressed by anaerobic glycolysis incubation. However, formation of Y-amine increased in anaerobic glycolysis.

<sup>\*</sup> Hongo, Tokyo (野口照久).

<sup>1)</sup> Part II. T. Noguchi: This Bulletin, 4, 97(1956).

<sup>2)</sup> T. Noguchi, Y. Miura: J. Biol. Chem.(1956), in press.

<sup>3)</sup> K. Tsuda, et al.: J. Pharm. Soc. Japan, 62, 362(1942); 67, 239(1947).

<sup>4)</sup> B. N. Ames, H. K. Mitchell: J. Am. Chem. Soc., 74, 252(1952).

TABLE I.	Detection of	Each	Spot	after	Incubation	with	Pigeon
Liver Nuclear Fractions							

Medium	Spots	Rf	Tsuda test	Pauly test	Bial test	Tryptophan test	Seliwanoff test
Before incubation	AICA	0.57	+++	Blue(+++)			<u> </u>
After incubation in aerobic glycolysis	(X	$0.28 \!\pm\! 0.02$	++	Yellow(+)	Brown(++	) —	++
	{ Y	$0.40 \pm 0.02$	土	$Purple(\pm)$	$Green(\pm)$		
	L	0.57	+++	Blue(+++)	-	MALES .	
After incubation in anaerobic glycolysis	ſΧ	$0.28 \pm 0.02$	土	$Yellow(\pm)$	Brown(±)		+
		$0.40 \pm 0.02$	+	Purple(+)	Green(+)		
	$\iota_Z$	0.57	+++	Blue(+++)			

Sugar moiety of these spots was examined. X-Amine exhibited dark brick color with the Bial's orcinol-HCl test,<sup>5)</sup> the Seliwanoff's resorcinol test<sup>6)</sup> was positive, and diphenylamine reaction<sup>7)</sup> and tryptophan test<sup>8)</sup> were negative. Y-Amine exhibited green color with the Bial's test for a pentose moiety, and other reactions were negative. Z-Amine was negative to all sugar reactions. The detection of phosphoric acid moiety on the paper chromatogram by the Hans-Isherwood's procedure<sup>9)</sup> failed.

Paper with these spots was cut into narrow strips, each of which was eluted with 0.01N HCl, 0.02M phosphate buffer (pH 7.4), or 0.03N NaOH. Then each solution was placed in a 1-cm. cell and the ultraviolet absorption of these spots was measured by the Beckman spectrophotometer. The absorbance ratios in 0.01N HCl were calculated. The results are listed in Table II. Maximum absorption of X-amine appeared at  $261 \, \mathrm{m}\mu$  in 0.01N HCl, distinctly differing from those of Y- and Z-amines and of AICA.

Table II. Absorption Spectra and Absorbance Ratio of Each Spot  $(\lambda_{max} m\mu)$ 

Medium		X	$\mathbf{Y}$	Z	AICA
0.01N HCl		261	266	266	266
0.02M Phosphate buffer (p	H 7.4)	264	267	267	267
0.03N NaOH		270	278	278	278
Absorbance Ratio (0.01 N HCl)					
Amine	250/260		270/260		280/260
X-Amine	0.82		0.83		0.45
Y-Amine	0.74		1.00		0.51
Z-Amine	0.75		1.00		0.61
AICA	0.75		1.04		0.68

These spots were hydrolyzed with 0.1N or 1N HCl at  $100^{\circ}$  for  $5 \, \text{mins}$ . The solution was again chromatographed on paper and the spots were detected by the Pauly reagent. As demonstrated in Table III, X-amine was resistant to boiling in water and hydrolysis with weak acids but was hydrolyzed with 1N HCl, while Y-amine was decomposed to liberate pentose from the parent compound by acid hydrolysis and other moiety moved to the same Rf as that of AICA and Z-amine.

It is, therefore, concluded that X-amine is a kind of arylamine associated with a ketose, which is found to be neither ribose nor deoxyribose. From the above results, Y-amine was identified as AICA riboside in other experiments, <sup>10)</sup> and Z-amine agreed with AICA itself. Further research on X-amine, the structure of which is

<sup>5)</sup> M. Bial: Deut. Med. Wochschr. 29, 253, 477(1903); W. Mejbaum: Z. physiol. Chem., 258, 117 (1939).

<sup>6)</sup> T. Seliwanoff: Ber., 20, 181(1887).

<sup>7)</sup> F. B. Seibert: J. Biol. Chem., 133, 593(1940).

<sup>8)</sup> P. Thomas: Z. physiol. Chem., 199, 10(1931).

<sup>9)</sup> C.S. Hanes, F.A. Isherwood: Nature, 164, 1107(1949).

<sup>10)</sup> T. Noguchi, Y. Miura: Unpublished data.

not clear in this stage, should be of interest, if the amine is a new intermediate in the biosynthesis of DNA.

TABLE III. Paper Chromatogram of Each Spot after Hydrolysis

Amines	HCl added	Rf	Pauly test
AICA	None (H <sub>2</sub> O)	0.57	Blue
AICA	0.1N	0.57	<i>"</i>
AICA	1.0N	0.57	//
X-amine	None (H <sub>2</sub> O)	0.28	Yellw
X-amine	0.1N	0.28	<i>"</i>
X-amine	1.0N	0.7	Pink
Y-amine	None $(H_2O)$	0.40	Pale purple
Y-amine	0.1N	(0.40) + 0.57	(Pale purple) + Blue
Y-amine	1.0N	0.57	Blue
Z-amine	None $(H_2O)$	0.57	"
Z-amine	0.1N	0.57	. "
Z-amine	1.0N	0.57	"
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Hydrolysis: 100°, 5 mins. Solvent: BuOH:EtOH:H<sub>2</sub>O

The author wishes to express his appreciation to Prof. Y. Ito, Prof. N. Shimazono, and Prof. Y. Miura of the University of Tokyo. He is also grateful to Mr. T. Owada, Mr. K. Oga, Mr. E. H. Hibino, and Dr. H. Okeda of Nippon Soda Co. Ltd. for encouragement, and to Mr. S. Takeyama for technical assistance.

## Experimental

**Preparation of Subcellular Fractions**—For fractionation of subcellular components, the Hogeboom and Schneider's method<sup>11)</sup> was used. Starting with  $14{\sim}15\,\mathrm{g}$ . of pigeon liver, 20% liver homogenate was prepared in  $0.25\,M$  sucrose (Potter-Elvehjem type homogenizer was used). The nuclear fraction was obtained by three centrifugations at  $700\times g$  for  $10\,\mathrm{mins}$ ; the supernatant obtained from the first centrifugation was used for the cytoplasmic fractions.

Incubation of Subcellular Fractions—The incubation of nuclear fraction or cytoplasmic fractions was carried out at 37° for 100 mins. under aerobic or anaerobic conditions. The composition of the medium for anaerobic glycolysis were K-phosphate buffer (pH 7.4) 0.0024M, KHCO3 0.025M, nicotinamide 0.04M, K-ATP 0.00033M, DPN 0.00022M, fructose-1,6-diphosphate 0.002M, glucose 0.01M, MgCl2 0.007M, K-pyruvate 0.005M, KF 0.01M and Solution F. The medium for aerobic glycolysis contained K-phosphate buffer (pH 7.4) 0.0033M, KHCO3 0.0033M, fructose-1,6-diphosphate 0.01M, DPN 0.00022M, MgCl2 0.0033M, KF 0.01M and Solution F. The composition of Solution F: AICA-14C 0.06 mM, ribose-5'-phosphate 0.06 mM, Na-formate 0.06 mM, Ca leucovorin 50  $\gamma$  (total volume: 15 cc.).

Paper Chromatography and Identification of Acid-soluble Fractions—The acid-soluble fractions were obtained by the addition of conc.  $HClO_4$  to the medium (final concn. 4%). Then, the paper chromatography of this fraction was developed with BuOH:EtOH: $H_2O$  (50:15:35)(Markham-Smith's solvent<sup>12</sup>)), using acid-treated filter paper (Toyo Roshi No. 51-A). The ultraviolet light was used for the detection of spots.

## Summary

During the course of studies on the incorporation of 4-aminoimidazole[4-14C]-5-carboxamide into polynucleotide purines with subcellular fractions of pigeon liver, a new diazotizable amine was obtained by the paper chromatography of the acid-soluble fraction of the incubation medium. The amine is a kind of arylamine associated with a ketose, which is neither ribose nor deoxyribose.

(Received January 10, 1956)

<sup>11)</sup> G. H. Hogedoom, W. C. Schneider: J. Biol. Chem., 197, 611(1952).

<sup>12)</sup> R. Markham, J. D. Smith: Biochem. J. (London), 45, 294(1949).