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Tatsuhiko Nakano and Mikio Uchiyama: Studies on the Alkaloids of Menispermaceous Plants. CXXXVIII.¹⁾ Alkaloids of Cocculus laurifolius DC. (Suppl. 9).²⁾ Isolation of Magnoflorine.*

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At an earlier date, Tomita and Kusuda³⁾ recorded the isolation of quaternary alkaloids, laurifoline (I) and a base (picrate, m.p. 88°), from *Cocculus laurifolius* DC. At that time, they obtained, after separating laurifoline as the crystalline chloride, the remaining uncrystallizable alkaloid as the mercuric salt, but left it unexamined. This investigation has been resumed using the material which was left by them.

Decomposition of this mercuric chloride of the quaternary base with hydrogen sulfide and subsequent conversion of the resulting chloride into the styphnate yielded yellow needles, m.p. 230~231°(decomp.). The iodide also crystallized in colorless pillars, m.p. 249°(decomp.). Their molecular formulae and melting points were found to be in accord with those given for the corresponding derivatives of magnoflorine⁴⁾(II), and this similarity was also confirmed by infrared spectra.

Since it was known so far that *Cocculus laurifolius* DC. contains coclaurine (\mathbb{II}) and coclanoline (\mathbb{IV}), besides laurifoline (\mathbb{I}), this proof as to the occurrence of magnoflorine (\mathbb{II}) in the same plant is of significance in considering the biogenesis of these series of alkaloids.

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Experimental⁵⁾

Isolation of Magnoflorine from Cocculus laurifolius DC.—2.0 g. of the quaternary base mercuric chloride was dissolved in MeOH and decomposed with H_2S . The filtrate from HgS precipitate was evaporated to dryness in vacuo, the residue was dissolved in a small portion of water, and treated with aq. sodium styphnate. The resulting precipitate, after crystallization from acetone, yielded 1.8 g. of yellow needles, m.p. $230\sim231^\circ$ (decomp.), which was shown by infrared spectrum to be identical with magnoflorine styphnate. Anal. Calcd. for $C_{20}H_{24}O_4N \cdot C_6H_2O_8N_3$: C, 53.24; H, 4.47.

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- 1) Part CXXXVII, M. Tomita, T. Kugo: J. Pharm. Soc. Japan, 76, 857(1956).
- 2) (Suppl. 8). M. Tomita, I. Kikkawa: This Bulletin, 4, 230(1956).
- 3) M. Tomita, F. Kusuda: *Ibid.*, 1, 1, 5, 55(1953).
- 4) T. Nakano: *Ibid.*, 2, 326, 329(1954).
- 5) All melting points are uncorrected. We are indebted to Dr. K. Hozumi, Mr. K. Imaeda, and Miss F. Tanase of the Central Analysis Room of this Institute for the microanalytical data.

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