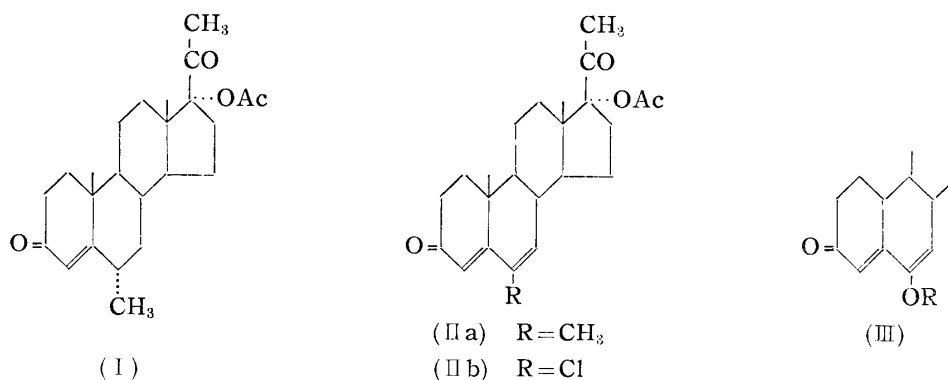


V

## Hiromu Mori : Studies on Steroidal Compounds. IV.<sup>1)</sup> Synthesis of 6-Alkoxy-3-oxo-4,6-diene Steroidal Hormones.

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Since it has been reported in 1951 that 19-norprogesterone has high progestational activity,<sup>2)</sup> brilliant progress has been made in the field of progestational 19-norsteroids.<sup>3)</sup> On the other hand, high progestational activity was reported for 6 $\alpha$ -methyl-17 $\alpha$ -acetoxy-pregn-4-ene-3,20-dione (I).<sup>4)</sup> Being prompted by this discovery, many substituted 17 $\alpha$ -acetoxy-pregn-4-ene-3,20-dione derivatives having high activity were synthesized<sup>5-7)</sup> and a new field was developed in the studies of progestational substances. It was shown by the Syntex group that 6-methyl-17 $\alpha$ -acetoxy-pregna-4,6-diene-3,20-dione<sup>6)</sup> (IIa) and 6-chloro-17 $\alpha$ -acetoxy-pregna-4,6-diene-3,20-dione<sup>7)</sup> (IIb) are the highest progestational substances known to date.



It is generally recognized that the substitution of a hydrogen atom at 6-position by some group and simultaneous introduction of the double bond at 6-7 position increase progestational activity of the parent compounds. From this point of view, it seemed of interest to prepare 6-alkoxy-3-oxo-4,6-diene steroidal hormones (III) and the present report concerns the synthesis of such steroidal hormones.

In order to investigate the most suitable method to prepare the above-mentioned structure (III), testosterone acetate derivatives were examined. 6 $\beta$ -Hydroxy-17 $\beta$ -acetoxyandrost-4-en-3-one (IV) was obtained from testosterone acetate by the method of Romo<sup>8)</sup> and converted into 17 $\beta$ -acetoxyandrost-4-ene-3,6-dione (V) by oxidation with chromium trioxide

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2) C. Djerassi, L. Miramontes, G. Rosenkranz : J. Am. Chem. Soc., **73**, 3540 (1951); **75**, 4440 (1953).

3) C. Djerassi, L. Miramontes, G. Rosenkranz, F. Sondheimer : *Ibid.*, **76**, 4092 (1954); H. J. Ringold, G. Rosenkranz, F. Sondheimer : *Ibid.*, **78**, 2477 (1956); D. A. McGinty, C. Djerassi : Ann. N. Y. Acad. Sci., **71**, 500 (1958).

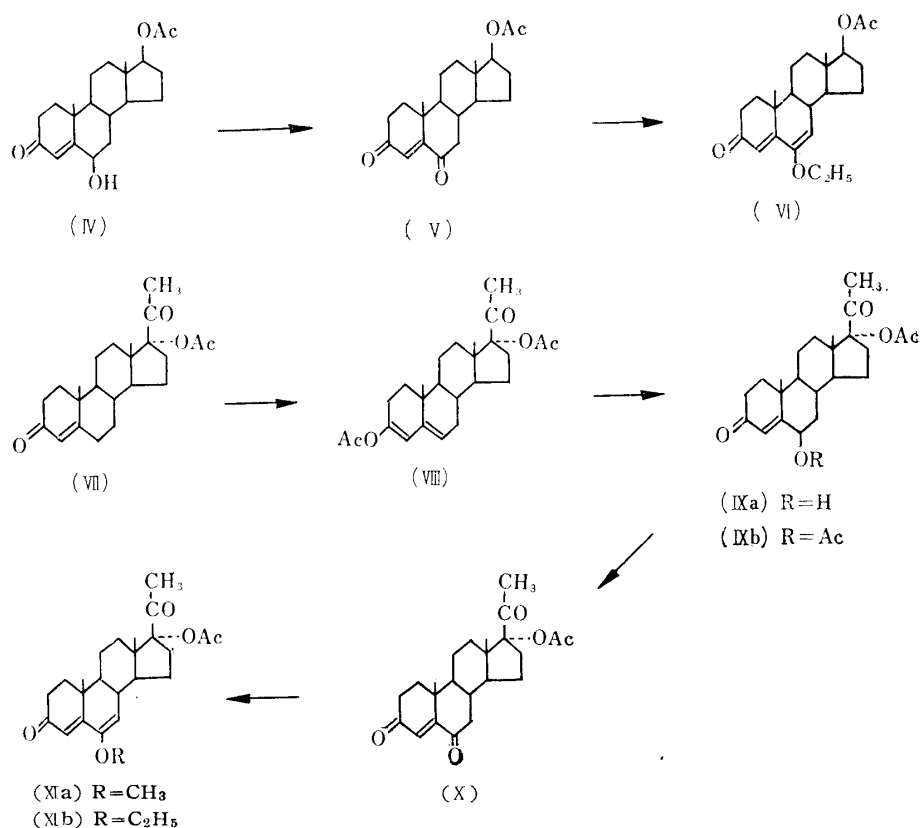
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in acetic acid. In a classical paper of 1904, Windaus<sup>9)</sup> showed that cholest-4-ene-3,6-dione was convertible by ethanol and hydrogen chloride into an enol ethyl ether, which was prepared also by Nakanishi<sup>10)</sup> in 1953 and identified as 6-ethoxycholesta-4,6-dien-3-one by Fieser.<sup>11)</sup> In the case of (V), however, milder condition is desirable to prevent hydrolysis of 17 $\beta$ -acetoxy group. The desired enol ethyl ether (VI) of m.p. 157.5~159° was obtained by stirring the diketone (V) in ethanol at room temperature in the presence of a large amount of *p*-toluenesulfonic acid monohydrate. The structural assignment of the enol ethyl ether as 6-ethoxy-17 $\beta$ -acetoxyandrosta-4,6-dien-3-one (VI) was based on ultraviolet and infrared absorption spectra and elementary analysis. (VI) showed strong absorption at 304 m $\mu$  in its ultraviolet spectrum and it is known that 6-ethoxycholesta-4,6-dien-3-one shows absorption at 302 m $\mu$ .<sup>12)</sup> The infrared absorption spectrum of (VI) did not show the

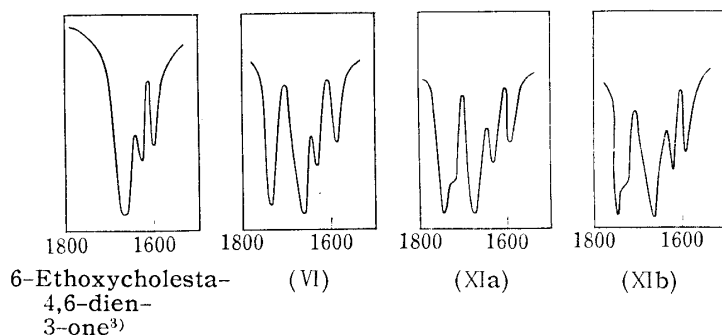


Fig. 1. Infrared Absorption Spectra of 6-Alkoxy-4,6-dien-3-ones (in CHCl<sub>3</sub>)

- 9) A. Windaus : Ber., **39**, 2249 (1906).  
 10) K. Nakanishi : Unpublished work.  
 11) L. F. Fieser : J. Am. Chem. Soc., **75**, 4386 (1953).  
 12) L. F. Fieser, M. Fieser : "Steroids," 45 (1959). Reinhold Publishing Corp., New York.

hydroxyl band but acetoxy band was observed at  $1725\text{ cm}^{-1}$ , and the spectral curve in the region of  $1600\sim 1700\text{ cm}^{-1}$  was identical with that of 6-ethoxycholesta-4,6-dien-3-one<sup>13)</sup> (Fig. 1).

Attempt was made to prepare 6-alkoxy-17 $\alpha$ -acetoxypregna-4,6-diene-3,20-dione (XI) via 17 $\alpha$ -acetoxypregn-4-ene-3,6,20-trione (X). The synthesis of (X) was made in accordance with the modified method of Romo.<sup>8)</sup> The 3-enol acetate (VIII) prepared from 17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (VII) by reaction with isopropenyl acetate was oxidized with monoperphthalic acid to 6 $\beta$ -hydroxy-17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (IXa). Oxidation of (IXa) with chromium trioxide in acetic acid led to 17 $\alpha$ -acetoxypregn-4-ene-3,6,20-trione (X). Table I shows molecular rotation differences by introduction of  $\beta$ -hydroxyl, keto, and 6-ethoxy-6-ene group in 6-position of 3-oxo-4-ene steroids.

TABLE I. Molecular Rotation Differences of 6-Oxygenated Steroids

	(A)	(B)	(C)	(D)			
		OH	O	OC <sub>2</sub> H <sub>5</sub>			
	M <sub>DA</sub>	M <sub>DB</sub>	M <sub>DC</sub>	M <sub>DD</sub>	M <sub>DA</sub> -M <sub>DB</sub>	M <sub>DA</sub> -M <sub>DC</sub>	M <sub>DA</sub> -M <sub>DD</sub>
5 $\alpha$ -Cholestane	+342 <sup>a)</sup>	+108 <sup>b)</sup>	-140 <sup>b)</sup>		+234	+482	
17 $\beta$ -Acetoxy-5 $\alpha$ -androstane	+288 <sup>c)</sup>	+94 <sup>b)</sup>	-162 <sup>d)</sup>	-102	+194	+450	+390
Pregnan-20-one	+557 <sup>e)</sup>	+354 <sup>b)</sup>	+100 <sup>b)</sup>		+203	+457	
17 $\alpha$ -Acetoxypregnan-20-one	+209 <sup>f)</sup>	+27	-208	-147	+182	+417	+356

a) A. Butenandt, A. Wolff : Ber., **68**, 2091 (1935).

b) L. F. Fieser : J. Am. Chem. Soc., **75**, 4377 (1953).

c) Codex Français, 933 (1949).

d) A. Butenandt, B. Riegel : Ber., **69**, 1163 (1936).

e) L. F. Fieser, M. Fieser : "Steroids," 541 (1959). Reinhold Publ. Corp., New York.

f) R. B. Turner : J. Am. Chem. Soc., **75**, 3489 (1953).

Treatment of triketone (X) with ethanol and *p*-toluenesulfonic acid monohydrate gave 6-ethoxy-17 $\alpha$ -acetoxypregna-4,6-diene-3,20-dione (XIa). By the use of methanol instead of ethanol, 6-methoxy-17 $\alpha$ -acetoxypregna-4,6-diene-3,20-dione (XIb) was obtained.

(X) and (XIb) were inactive in Clauberg test when assayed in a dose of 1 mg. From Burn's hypothesis<sup>14)</sup> that inactivation of steroidal hormones would also occur through metabolic oxygenation at 6-position, it is reasonable that (X) is inactive in Clauberg test. In the case of (XIb), the inactivity will probably be due to its easy hydrolysis. (VI) was also inactive in androgenic and anabolic tests using a mouse in a total dose of 14 mg. (2mg. per day).

### Experimental\*2

**17 $\beta$ -Acetoxyandrost-4-ene-3,6-dione (V)**—17 $\beta$ -Acetoxy-6 $\beta$ -hydroxyandrost-4-en-3-one<sup>8)</sup> (IV) (500 mg.) dissolved in AcOH (15 cc.) was oxidized with CrO<sub>3</sub> (250 mg.) in H<sub>2</sub>O (1 cc.) for 15 min. at room temperature. It was diluted with H<sub>2</sub>O and the product was extracted with Et<sub>2</sub>O. The organic layer was washed with 5% Na<sub>2</sub>CO<sub>3</sub> and H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and the solvent was evaporated. Recrystallization from Et<sub>2</sub>O-hexane mixture gave pale yellow prisms, m.p. 204~207°; yield, 350 mg. (reported<sup>1)</sup> m.p. 198~201°,  $[\alpha]_D^{25} -47.2^\circ \pm 1.4^\circ$  (Me<sub>2</sub>CO).

**6-Ethoxy-17 $\beta$ -acetoxyandrosta-4,6-dien-3-one (VI)**—To a solution of 17 $\beta$ -acetoxyandrost-4-ene-3,6-dione (V) (290 mg.) in dehyd. EtOH (100 cc.), *p*-toluenesulfonic acid monohydrate (2.0 g.) was added and the solution was stirred at room temperature for 5 hr. It was diluted with H<sub>2</sub>O and the product was extracted with Et<sub>2</sub>O. The organic fraction was washed with 5% Na<sub>2</sub>CO<sub>3</sub> and H<sub>2</sub>O, dried

\*2 All m.p.s are uncorrected.

13) K. Nakanishi : Jikken Kagaku Koza, **1**, [I], 358 (1957). Maruzen Co. Ltd., Tokyo.

14) D. Burn, B. Ellis, V. Petrow, I. A. Stuart-Webb, D. M. Williamson : J. Chem. Soc., **1957**, 4092.

over  $\text{Na}_2\text{SO}_4$ , and the solvent was evaporated. Recrystallization from MeOH containing one drop of pyridine gave (VI), m.p. 156~161°; yield, 250 mg. Further recrystallization from the same solvent gave white needles, m.p. 160~162°,  $[\alpha]_D^{20} -27^\circ$  ( $c=0.946$ ,  $\text{CHCl}_3$ ), UV:  $\lambda_{\text{max}}^{\text{EtOH}}$  304  $\text{m}\mu$  ( $\log \epsilon$  4.18), IR  $\nu_{\text{max}}^{\text{CHCl}_3} \text{ cm}^{-1}$ : (no hydroxyl band); 1725 (17-AcO), 1655~1650, 1621, 1586 ( $\Delta^{4,6}$ -3-CO).  $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$ : (no hydroxyl band), 1729 (17-AcO), 1662, 1624, 1586 ( $\Delta^{4,6}$ -3-CO). *Anal.* Calcd. for  $\text{C}_{23}\text{H}_{32}\text{O}_4$ : C, 74.16; H, 8.66. Found: C, 74.02; H, 8.38.

**6 $\beta$ -Hydroxy-17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (IXa)**—A solution of 17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (12.0 g.) and *p*-toluenesulfonic acid monohydrate (2.0 g.) was refluxed gently for 3 hr. After cool,  $\text{Et}_2\text{O}$  was added and the solution was washed with ice-cold 5%  $\text{NaHCO}_3$  and  $\text{H}_2\text{O}$ . After drying over  $\text{Na}_2\text{SO}_4$  and evaporation of the solvent, the residue was crystallized from MeOH to 3,17 $\alpha$ -diacetoxypregn-3,5-dien-20-one (VIII), m.p. 192~202°; yield, 11.4 g.

The crude enol acetate (VIII) (8.0 g.) was dissolved in  $\text{Et}_2\text{O}$  (800 cc.), a solution of monopero-phthalic acid in  $\text{Et}_2\text{O}$  (59.0 cc., 0.077 g./cc., 1.3 equivalent to 8.0 g. of (VIII)) was added and the solution was allowed to stand at room temperature in the dark for 3 days. After washing with ice-cold 5%  $\text{NaHCO}_3$  and  $\text{H}_2\text{O}$ , and drying over  $\text{Na}_2\text{SO}_4$ , the solvent was evaporated. Recrystallization of the residue from  $\text{Me}_2\text{CO}$ -hexane mixture gave (IXa), m.p. 235~241°; yield, 4.78 g. Further recrystallization from the same solvent gave white needles, m.p. 239~243°,  $[\alpha]_D^{20} +7^\circ$  ( $c=0.867$ , dioxane), UV:  $\lambda_{\text{max}}^{\text{EtOH}}$  238  $\text{m}\mu$  ( $\log \epsilon$  4.20). *Anal.* Calcd. for  $\text{C}_{23}\text{H}_{32}\text{O}_5$ : C, 71.10; H, 8.30. Found: C, 71.13; H, 8.46.

**6 $\beta$ ,17 $\alpha$ -Diacetoxypregn-4-ene-3,20-dione (IXb)**—To a solution of 6 $\beta$ -hydroxy-17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (IXa) (500 mg.) in pyridine (10 cc.),  $\text{Ac}_2\text{O}$  (5 cc.) was added and the solution was allowed to stand overnight at room temperature. It was diluted with  $\text{H}_2\text{O}$  and the product was extracted with  $\text{Et}_2\text{O}$ . After washing the  $\text{Et}_2\text{O}$  extract with 5%  $\text{Na}_2\text{CO}_3$  and  $\text{H}_2\text{O}$ , and drying over  $\text{Na}_2\text{SO}_4$ , the solvent was evaporated. Recrystallization from MeOH gave (IXb), m.p. 239~244°; yield, 330 mg. Further recrystallization from MeOH gave white prisms, m.p. 243~245°,  $[\alpha]_D^{20} +22^\circ$  ( $c=1.119$ ,  $\text{CHCl}_3$ ), UV:  $\lambda_{\text{max}}^{\text{EtOH}}$  236  $\text{m}\mu$  ( $\log \epsilon$  4.13). *Anal.* Calcd. for  $\text{C}_{25}\text{H}_{34}\text{O}_6$ : C, 69.74; H, 7.96. Found: C, 69.90; H, 7.73.

**17 $\alpha$ -Acetoxypregn-4-ene-3,6,20-trione (X)**—6 $\beta$ -Hydroxy-17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (IXa) (500 mg.) dissolved in AcOH (20 cc.) was oxidized with  $\text{CrO}_2$  (300 mg.) in  $\text{H}_2\text{O}$  (1 cc.) for 15 min. at room temperature. It was diluted with  $\text{H}_2\text{O}$  and the product was extracted with  $\text{Et}_2\text{O}$ . After washing the extract with 5%  $\text{Na}_2\text{CO}_3$  and  $\text{H}_2\text{O}$ , and drying over  $\text{Na}_2\text{SO}_4$ , the solvent was evaporated. Recrystallization from  $\text{Me}_2\text{CO}$  gave (X), m.p. 214~216.5°; yield, 360 mg. Further recrystallization from  $\text{Me}_2\text{CO}$  gave pale yellow needles, m.p. 215~218°,  $[\alpha]_D^{18} -54^\circ$  ( $c=1.095$ ,  $\text{CHCl}_3$ ). UV:  $\lambda_{\text{max}}^{\text{EtOH}}$  250  $\text{m}\mu$  ( $\log \epsilon$  4.11). *Anal.* Calcd. for  $\text{C}_{23}\text{H}_{30}\text{O}_5$ : C, 71.48; H, 7.82. Found: C, 71.16; H, 7.70.

**6-Methoxy-17 $\alpha$ -acetoxypregn-4,6-diene-3,20-dione (XIa)**—A solution of 17 $\alpha$ -acetoxypregn-4-ene-3,6,20-trione (X) (300 mg.) and *p*-toluenesulfonic acid monohydrate (2.0 g.) in MeOH (100 cc.) was stirred at room temperature for 5 hr. It was diluted with  $\text{H}_2\text{O}$  and the product was extracted with  $\text{Et}_2\text{O}$ . After washing  $\text{Et}_2\text{O}$  layer with 5%  $\text{Na}_2\text{CO}_3$  and  $\text{H}_2\text{O}$ , and drying over  $\text{Na}_2\text{SO}_4$ , the solvent was evaporated. Recrystallization from MeOH containing one drop of pyridine gave (XIa), m.p. 210~213°; yield, 230 mg. Further recrystallization from the same solvent gave white plates, m.p. 215~218°,  $[\alpha]_D^{18} -39^\circ$  ( $c=1.315$ ,  $\text{CHCl}_3$ ), UV:  $\lambda_{\text{max}}^{\text{EtOH}}$  304  $\text{m}\mu$  ( $\log \epsilon$  4.16). IR  $\nu_{\text{max}}^{\text{CHCl}_3} \text{ cm}^{-1}$ : (no hydroxyl band), 1732, 1710~1715 (17 $\alpha$ -AcO-20-CO), 1658, 1623, 1586 ( $\Delta^{4,6}$ -3-CO),  $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$ : (no hydroxyl band), 1732, 1715 (17 $\alpha$ -AcO-20-CO), 1662, 1625, 1589 ( $\Delta^{4,6}$ -3-CO). *Anal.* Calcd. for  $\text{C}_{24}\text{H}_{32}\text{O}_5$ : C, 71.97; H, 8.05. Found: C, 72.05; H, 7.79.

**6-Ethoxy-17 $\alpha$ -acetoxypregn-4-ene-3,20-dione (XIb)**—A solution of 17 $\alpha$ -acetoxypregn-4-ene-3,6,20-trione (X) (300 mg.) and *p*-toluenesulfonic acid monohydrate (2.0 g.) in dehyd. EtOH (100 cc.) was stirred at room temperature for 4 hr. It was diluted with  $\text{H}_2\text{O}$  and the product was extracted with  $\text{Et}_2\text{O}$ . After washing  $\text{Et}_2\text{O}$  layer with 5%  $\text{Na}_2\text{CO}_3$  and  $\text{H}_2\text{O}$ , and drying over  $\text{Na}_2\text{SO}_4$ , the solvent was evaporated. Recrystallization from MeOH containing one drop of pyridine gave (XIb), m.p. 177~179.5°; yield, 250 mg. Further recrystallization from the same solvent gave white needles, m.p. 181.5~183°,  $[\alpha]_D^{20} -35^\circ$  ( $c=0.792$ ,  $\text{CHCl}_3$ ), UV:  $\lambda_{\text{max}}^{\text{EtOH}}$  304  $\text{m}\mu$  ( $\log \epsilon$  4.17). IR  $\nu_{\text{max}}^{\text{CHCl}_3} \text{ cm}^{-1}$ : (no hydroxyl band), 1730, 1710~1715 (17 $\alpha$ -AcO-20-CO), 1650, 1620, 1585 ( $\Delta^{4,6}$ -3-CO).  $\nu_{\text{max}}^{\text{KBr}} \text{ cm}^{-1}$ : (no hydroxyl band), 1732, 1713 (17 $\alpha$ -AcO-20-CO), 1665, 1621, 1588 ( $\Delta^{4,6}$ -3-CO). *Anal.* Calcd. for  $\text{C}_{25}\text{H}_{34}\text{O}_5$ : C, 72.43; H, 8.27. Found: C, 72.70; H, 8.02.

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### Summary

Synthesis of some steroidal hormones having 6-alkoxy-4,6-dien-3-one structure was carried out.  $17\beta$ -Acetoxyandrost-4-ene-3,6-dione (V), prepared by oxidation with chromium trioxide of  $6\beta$ -hydroxy- $17\beta$ -acetoxyandrost-4-en-3-one (VI), was treated with ethanol and *p*-toluenesulfonic acid monohydrate to give 6-ethoxy- $17\beta$ -acetoxyandrosta-4,6-dien-3-one (VI). The treatment of  $17\alpha$ -acetoxypregn-4-ene-3,6,20-trione (X) with ethanol (or methanol) and *p*-toluenesulfonic acid monohydrate gave 6-ethoxy-(or methoxy)- $17\alpha$ -acetoxypregna-4,6-diene-3,20-dione (XIa or XIb). (X) was synthesized from  $17\alpha$ -acetoxypregn-4-ene-3,20-dione (VII) by the modified method of Romo.

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**Tatsuo Ohta and Susumu Mihashi: Furoquinolines. XXI.\*<sup>1</sup>**  
Synthesis and Alkoxy Interchange of 4-Ethoxy-  
furo[2,3-*b*]quinoline and its 2,3-Dihydro Analogs.

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Berinzaghi, Deulofeu, Labriola, and Muruzabal<sup>1)</sup> showed that, by the influence of alcoholic alkali, an interchange occurs between the methoxyl group at the 4-position of furoquinoline alkaloid and alkoxy group of the particular alcohol used. By this means, the ethoxyl and propoxyl analogs of skimmianine (Ia) and ethoxyl analog of  $\gamma$ -fagarine (IIa) were obtained.

The present paper deals with the synthesis of the title compounds and their alkoxy interchange reaction. Respective refluxing of 4-chloro-3-(2-chloroethyl)carbostyryl (I) and its 7-methoxy derivative (II) with ethanolic potassium hydroxide solution gave 4-ethoxy-2,3-dihydrofuro[2,3-*b*]quinoline (III) and its 7-methoxy derivative (IV). Dehydrogenation of (III) with *N*-bromosuccinimide in the presence of benzoyl peroxide, followed by heating of the product with collidine, yielded 4-ethoxyfuro[2,3-*b*]quinoline (V), identical with the ethoxyl analog<sup>2)</sup> of dictamnine obtained previously by alkoxy interchange of dictamnine.

Reconversion of (V) to dictamnine (VI) with 10% methanolic potassium hydroxide solution was concluded easily but 2,3-dihydrodictamnine (VII), 2,3-dihydroevolitrine (VIII), and 2,3-dihydroskimmianine (IX) did not undergo transformation, except (VII) into (III), into the corresponding ethoxyl analogs with 10% ethanolic potassium hydroxide solution. Further, an attempt was made to obtain (VII) and (VIII) respectively from (III) and (IV) with 10% methanolic potassium hydroxide solution, but it failed in spite of prolonged heating or the reaction in sealed tube.

From these experimental results, it may be concluded that the carbon atom in the 4-position of the furo[2,3-*b*]quinoline ring has small electron density due to the overlapping of the resonance effect of the furan and pyridine ring. Therefore, the alkoxy group in the 4-position of the furoquinoline alkaloids is quite easily substituted with other alkoxide ions.

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