3. In NaOH-MeOH solution, however, BH_3 formed at the first stage being captured by NaOH, reduction of enamine with BH_a, was prevented at the second stage of the reaction and the dihydro compounds remained unchanged.

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Gas Chromatography of the Morphine Alkaloids and the Related Compounds

The microanalytical separation of the comparatively high molecular weight compounds such as steroids and alkaloids by gas chromatography has been very rapidly developed.1~8)

In the present studies, the gas chromatographic analyses of forty three compounds, including the morphine alkaloids, sinomenine and its derivatives, were carried out in order to obtain the correlation between their structures and their retention times.

The compounds listed in the table gave single sharp peaks, consistent with the absence of decomposition, except for 8,14-dihydroxydihydrocodeinone (XXVII).*1 The relative retention times were calculated based on that of codeine (XIV) as a reference and the characteristics in the structure were illustrated comparing with dihydrodesoxycodeine D(II).

The retention times were increased as shown in the following cases: 1) opening of $C_4 \sim C_5$ ether bridge, (II)~(IV), (VI)~(XX), (XXI)~(XXXVII), (XXX)~(XLII); 2) hydroxylation at C₁₄, (I) \sim (IX), (II) \sim (VI), (IV) \sim (XX), (XI) \sim (XXVI), (XIV) \sim (XXVII), (XVII) \sim (XXXII), (XXI) \sim (XXX), $(XXXVII) \sim (XLII);$ 3) N-demethylation, $(XIV) \sim (XIX), (XXX) \sim (XL);$ 4) acetylation of 14-hydroxy, (VI) \sim (X), (XXIX) \sim (XXXI), (XXX) \sim (XXXVI); 5) oxidation of the axial type hydroxy at C₆position into the carbonyl,*² (XI) \sim (XXI), (XVI) \sim (XXIV), (XXVI) \sim (XXX). The retention times were decreased as observed in the following cases: 1) removal of the hydroxy or carbonyl group from C₆-position and the phenolic group from C₄-position, (III) or $(XIV) \sim (I)$, (XI) or (XXI) or (XVII) \sim (II), (XXVI) or (XXX) or (XXXI) \sim (VI), (XXVII) or (XXIX) \sim (IX), (XXXVII) - (VII) (XXXVI)~(X); 2) methylation or ethylation of C₃- or C₄-phenolic group, (XXXVI)~(XXIV), $(XLII) \sim (XLI), (XXII) \sim (XVII)$ or (XIV).

^{*1} This showed also a single sharp peak with the same retention time as that of 14-hydroxycodeinone (XXIX) and it is expected that this compound is dehydrated into (XXIX) in the flash heating step.

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Horning,⁹⁾ Clayton¹⁰⁾ and Tsuda¹¹⁾ have shown that the axial 3-hydroxysteroids were eluted faster than their epimers. The following compounds of morphine series which contain C_{6^-} or C_{8^-} axial type hydroxyl groups, isocodeine (III), allopseudocodeine (VII), dihydrocodeine (XI), dihydro- 6α -thebainol methyl ether (XVI) and 14-hydroxydihydrocodeine (XXVI), always exihibited lower retention times than the corresponding epimeric compounds, codeine (XIV), pseudocodein (XV), dihydroisocodeine (XVII), dihydro- 6β -thebainol methyl ether (XXII), dihydro- 6β -thebainol methyl ether (XXII).



^{*&}lt;sup>2</sup> The similar change of the equatorial type did not significantly effect the retention time, i.e. $(XXII) \sim (XXII) \sim (XXII) \sim (XXII)$.

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No. 8

XXXV	Dihydrothebainone ϕ	1.52	opening the ether ring, $\Delta_{5,6}$, $\Delta_{8,14}$, C ₄ -OH, C ₆ -OCH ₃	
XXXVI	14-Acetoxydihydrocodeinone	1.52	$C_6=O, C_{14}\beta-OAc$	
XXXVII	Dihydrothebainone	1.61	opening the ether ring, C_4 -OH, C_6 =O	
XXXVШ	N-Propalgyl-14-hydroxydihydro- norcodeinone	1.69	$C_6=O$, $C_{14\beta}$ -OH, N-CH ₂ C \equiv CH	
XXXIX	Sinomenine methyl ether	1.87	antipode, opening the ether ring, $\Delta_{7,8}$, C_4 -OCH ₃ C_6 =O, C_7 -OCH ₃	
XL	14-Hydroxydihydronorcodeinone	1.91	С ₆ =О, С ₁₄ <i>β</i> -ОН, N-Н	
XLI	Dihydrosinomenine	2.00	antipode, opening the ether ring, C_4 -OH, C_6 =O C_7 -OCH ₃	
XLII	14–Hydroxydihydrothebainone	2.45	opening the ether ring, C ₄ -OH, C ₆ =O, C ₁₄ -β-OH	
XLII	Sinomenine	2.83	antipode, opening the ether ring, C ₄ -OH, C ₆ =O C_7 -OCH ₃ , $\Delta_{7,8}$	
Apparatus: Barber Colman Model 10, Argon ionization detector, 6 ft×8 mm., 1% SE-30				
on gas-chrom P (100 \sim 140 mesh)				
Condition : Column temp. 185', Cell temp. 160', Flash heater temp. 280', Argon pressure				
2 kg./cm^2				
Sample : $0.5 \sim 1\%$ Me ₃ CO solution were used except in the case of morphine (XXI) in which				
0.5% MeOH solution was used				
S	Standard : Codeine = 4.71 min. = 1 R.R.T.			

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Teflon Coated Support for Gas Chromatography at a Lower Temperature

The gas chromatographic analysis has no doubt been a very effective tool for chemis ts, but there is a narrow limitation of tailings which occur inevitably when the analysis is done at a lower temperature, with strongly polar substances and/or on the packings containing minor amount of stationary liquid (e.g. $0.1 \sim 10\%$).

Many attempts which have been done to reduce such tailings, can be classified as follows; a) use of new support. e.g. salts,¹⁾ glass beads,²⁾ sea sand,³⁾ quartz. powder,⁴⁾ metal helices,⁵⁾ Tide⁶⁾ or Teflon powder,⁷⁾ b) improvement of firebrick or Celite, e.g.

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