violet spectra. Furthermore, it quantitatively gave codeine by reduction with sodium borohydride.

Experimental

14-Bromocodeinone (II)—Thebaine (62.2 g.) was treated with N-bromosuccinimide in 2:1 (by volume) Me₂CO-H₂O mixture under the condition of Conroy's report,³⁾ and then gave 14-bromo-codeinone, yellow crystals. The yield, after drying at 60°, was 54.0 g. (71.8%).

Codeinone (III) To a solution of 14-bromocodeinone (15 g., 0.04 mol.) in 40 ml. of AcOH, Zn-dust $(3.12 \text{ g.}, 0.04 \times 1.2 \text{ mol.})$, activated by dil. HCl, was added in small portions with stirring during 15 min., maintaining at $15 \sim 17^{\circ}$ by external cooling. After addition, the mixture was stirred for 3.5 hr. at room temperature $(20 \sim 22^{\circ})$ and then poured into 250 ml. of ice-water. This mixture was filtered to remove a trace of unreacted Zn-dust and the filtrate was made alkaline with conc. NH₄OH, extracted with three 150 ml. portions of CHCl₃. The organic phase was washed with H₂O, dried over Na₂SO₄ and evaporated under reduced pressure. The residue was dissolved in 50 ml. of benzene and to this solution, 400 ml. of petr. ether was added. After the separation of solid, the filtrate was concentrated in vacuo. The residue was crystallized, m.p. 163~166°. After recrystallization from AcOEt, it showed m.p. $181 \sim 182^{\circ}$, undepression on admixture with authentic sample^{4,5}) and IR spectra of the two samples were identical. The total yield was 3.55 g. (30.0%). Anal. Calcd. for C₁₈H₁₉NO₃ : C, 72.73; H, 6.40; N, 4.71. Found : C, 73.14; H, 6.39; N, 5.02.

Codeine (IV)——To the solution of codeinone (4.0 g.) in 20 ml. of MeOH, NaBH₄ (10 g.) in 240 ml. of MeOH was added, and allowed to stand at room temperature $(20\sim25^{\circ})$ for 1.5 hr., then concentrated under reduced pressure to about 1/3 volume, and diluted with 200 ml. of 10% NaOH. The solution was heated on the steam-bath until boiling, diluted with 300 ml. of H₂O and extracted with four 100 ml. portions of CHCl₃. The extract was washed with H₂O, dried over Na₂SO₄, and the solvent was removed by distillation *in vacuo*. The residue was crystallized, m.p. 149~152°. After recrystallization from benzene, it showed m.p. 154~156°, undepressed on admixture with natural codeine and IR and UV spectra of both samples were identical. The yield was 3.7 g. (91.8%).

The author expresses his gratitude to Prof. K. Tsuda of Tokyo University, to Mr. M. Matsui, the Director of this Laboratory, and to Dr. I. Iwai, the Chief of the Section, for their kind guidance in this work. He is also indebted to Mr. H. Higuchi and N. Higosaki for measurement of infrared and ultraviolet spectra, and to Mr. T. Onoe, Misses T. Furukawa and H. Ohtsuka for elemental analyses.

Summary

Treatment of 14-bromocodeinone, prepared from the baine, with zinc-dust in glacial acetic acid gave codeinone, m.p. $181 \sim 182^{\circ}$. Codeinone was quantitatively reduced to codeine with sodium borohydride.

(Received May 22, 1961)

UDC 582.682.581.19:615.783.1

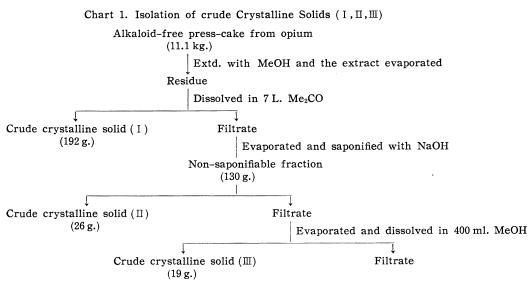
Katsuaki Matsui, Saburo Akagi, and Fukutaro Ugai: Chemical Constituents of the Alkaloid-free Fraction from Opium.

(Shinagawa Plant, Sankyo Co., Ltd.*1)

The alkaloid-free press-cake, which obtained from opium in the usual manner, was extracted with methanol and the extract was evaporated under a reduced pressure.

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The residue gave three crude crystalline solids (I, II, III) as shown in Chart 1.



Acetylation of the crude crystalline solid (I) followed by chromatography gave an acetate (V), m.p. $121\sim122^{\circ}$, $[\alpha]_{D}^{25} -20^{\circ}(c=2, \text{ CHCl}_{3})$, and an acetate (VI), m.p. $164\sim165^{\circ}$, $[\alpha]_{D}^{25} -25^{\circ}(c=2, \text{ CHCl}_{3})$.

Hydrolysis of (V) or (VI) with methanolic potassium hydroxide gave white prisms (VII), m.p. $288 \sim 290^{\circ}$. The Liebermann-Burchard test and the reduction of Fehling's solution indicated that (VII) was a steroidal glucoside.

TABLE I. Comparison of Constants obtained in the Present Work with those of Salway¹) Matlack² Kondo³ Swift⁴ Joshi⁶ on Corresponding Compounds

	Present work		Salway ¹⁾		Matlack ²⁾	
	m.p. (°C)	$[\boldsymbol{\alpha}]_{\mathrm{D}}^{25}$	m.p. (°C)	[<i>a</i>] _D	m.p. (°C)	[<i>a</i>] _D
β-Sitosteryl glucoside	$288 \sim 290$		$295{\sim}300$		280	
β-Sitosteryl glucoside tetraacetate	$164{\sim}165$	-25	$166 {\sim} 167$	-22.9	$164{\sim}165$	
β-Sitosteryl glucoside tetrabenzoate	196~198	+16	198	+18.3	198	_
β -Sitosterol	$136 \sim 137$	35	138	-32.2	$136 \sim 137$	
β -Sitosteryl acetate	$127 \sim 128$	-41			$124 \sim 125$	_
β -Sitosteryl benzoate	$146 \sim 147$	-13.5				
	Kondo ³⁾		Swift ⁴)		Joshi ⁶)	
	m.p. (°C)	$[\alpha]_{D}^{25}$	m.p. (°C)	$(\boldsymbol{\alpha})_{\mathrm{D}}$	m.p. (°C)	[a] _D
β-Sitosteryl glucoside	$252 \sim 254$	_	298	-40	$272 \sim 275$	
β-Sitosteryl glucoside tetraacetate	166~166.5	—	171	-33.7	167~168	-25.1
β-Sitosteryl glucoside tetrabenzoate	—	—	201	+15.9	—	
β-Sitosterol	138		$137 {\sim} 138$	-38.2		
β -Sitosteryl acetate			$125 {\sim} 126$	-40.3		_
β -Sitosteryl benzoate			$147 {\sim} 148$	-13.6	_	

1) A.H. Salway: J. Chem. Soc., 1913, 1022.

2) M.B. Matlack: J. Am. Pharm. Assoc., 18, 24 (1929).

3) K. Kondo: Nippon Kagaku Zasshi, 57, 1128 (1936).

4) L. J. Swift: J. Am. Chem. Soc., 74, 1099 (1952).

From the infrared spectrum,^{5,6}) melting point and optical rotation of (\mathbb{VI}), and those of its derivatives, (\mathbb{VI}) was identified as the β -sitosteryl-D-glucoside (Table I).

The identification of the β -sitosterol obtained by hydrolysis of (VII) was established on the basis of close agreement of its infrared spectrum,⁵⁾ melting point and optical rotation, and those of its derivatives with the values obtained by other investigators.

From the filtrate obtained by hydrolysis of (V), palmitic acid was isolated. On the basis of the carbon and hydrogen analysis, (V) was assigned to the molecular formula $C_{57}H_{96}O_{10}$. From these findings, (V) was assumed to be the monopalmityl-triacetyl-D-glucoside of β -sitosterol.

Chromatography of the crude crystalline solid (II) gave a white prisms (IX), m.p. $81 \sim 82^{\circ}$. and needles (X), m.p. $124 \sim 126^{\circ}$, $[\alpha]_{D}^{25} + 46^{\circ}(c=2, CHCl_{3})$. From the infrared spectrum, melting point and optical rotation of (IX) and (X), and those of their derivatives, (IX) was identified as (+)-10-nonacosanol, and (X) was identified as *cyclo*-laudenol, which isolated by Bentley *et al.*⁷⁾ from optium.

Chromatography of the crude crystalline solid (III) gave needles (XI), m.p. $112 \sim 115^{\circ}$, $(\alpha)_D^{25} + 54^{\circ}(c=2, CHCl_s)$, which gave a deep red solution with a strong green fluorescence in the Liebermann-Burchard test.

From the infrared spectrum,⁸⁾ melting point and optical rotation of (XI), and those of its derivatives, (XI) was identified as *cyclo*-artenol, which isolated by Bentley *et al.*⁹⁾ from the seed of *Strychnos nux-vomica* L.

TABLE II. Comparison of Constants obtained in the Present Work with

those of Bentley ⁹⁾ on Corresponding Compounds								
	Present	work	Betley ⁹⁾					
	m.p. (°C)	$[\boldsymbol{\alpha}]_{\mathbf{D}}^{25}$	m.p.(°C)	$[\boldsymbol{\alpha}]^{25}_{\mathbf{D}}$				
cyclo-Artenol	$112 \sim 115$	+52.5	115	+54				
cyclo-Artenyl acetate	$122 \sim 123$	+59	$122 \sim 124$	+59.5				
cyclo-Artenyl benzoate	$130 \sim 131$	+72	130	+76				
cyclo-Artanyl acetate	$130 \sim 131$	+58	$130 \sim 132$	+59				

Experimental

Extraction of the Crude Crystalline Solid—11.1 kg. of powdered alkaloid-free press-cake from opium was extracted twice with boiling MeOH with stirring, and the combined extracts were evaporated under a reduced pressure. The residue was dissolved in 7 L. of Me₂CO, and the solution was kept for 24 hr. in a refrigerator. The crude crystalline solid (I, 192 g.) that precipitated on cooling the solution was collected by filtration, and the filtrate was evaporated under a reduced pressure. The residue was dissolved in 2.24 L. of MeOH, and refluxed with a saturated NaOH (560 g.) solution for 3 hr. The non-saponifiable fraction (130 g.) was isolated by means of iso-propyl ether in the usual manner.

A solution of the non-saponifiable fraction in Me_2CO (600 ml.) was kept for 24 hr. in a refrigerator, and the separated crude crystalline solid (\square , 26 g.) was collected by filtration. The filtrate was evaporated under a reduced pressure, and the residue in MeOH (400 ml.) was kept for 24 hr. at room temperature. The crude crystallized solid (\square , 19 g.) was collected.

Acetylation and Chromatography of Crude Crystalline Solid (I)—A mixture of 192 g. of (I) in Ac_2O (200 g.) and dry pyridine (400 ml.) was kept for 24 hr. at room temperature, the solvents were removed by heating under a reduced pressure, and the residue was dissolved in 1.7 L. of MeOH. A precipitate that formed on cooling the MeOH solution for 24 hr. was collected, and thrice recrystallized from EtOH. The yield was 40.3 g. (IV), m.p. $132 \sim 142^{\circ}$.

A solution of (IV, 40.3 g.) in benzene was percolated through a column (4×40 cm.) of neutral

- 7) H.R. Bentley, J.A. Henry, D.S. Irvin, D. Mukerji, F.S. Spring: J. Chem. Soc., 1955, 596.
- 8) A.R.H. Cole: J. Chem. Soc., 1954, 3810.
- 9) H.R. Bentley, J.A. Henry, D.S. Irvin, D. Mukerji, F.S. Spring: Ibid., 1953, 3673.

⁵⁾ R. M. Ma, P. S. Schaffer: Arch. Biochem. Biophys., 47, 419 (1953).

⁶⁾ D.V. Joshi, S.F. Boyce: J. Org. Chem., 22, 95 (1957).

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alumina (400 g.), and the chromatogram eluted with a) benzene (3.6 L.), b) benzene- Et_2O (1 : 1, 3.6 L.), c) benzene-MeOH (1 : 1, 2 L.). Evaporation of fraction a) gave a solid 14.3 g. (IVa), m.p. 110~113°, and fraction b) gave a solid 12.3 g. (IV b.), m.p. 160~163°.

Monoparmityl-triacetyl-D-glucoside of β -Sitosterol (V)—A solution of (IVa, 14.3 g.) in benzene was percolated through a column (3×30 cm.) of neutral alumina (175 g.), and the chromatogram eluted with benzene (1.2 L.). Evaporation of benzene gave a solid, which was twice recrystallized from EtOH to white crystalline mass, m.p. 121~122°, (α)²⁵_D -20° (c=2, CHCl₃), yield, 4.5 g. Anal. Calcd. for C₅₇H₉₆O₁₀: C, 72.72; H, 10.28. Found C, 72.67; H, 10.28.

The filtrate, which obtained by the hydrolysis of (V) (3.0 g.) with MeOH-KOH (1.5 g.) was evaporated under a reduced pressure, and the residue was diluted with water, and acidified with HCl. A white crystallized solid was collected, and washed with water, and recrystallized from MeOH to white crystalline scales, m.p. $57\sim60^{\circ}$, yield 0.2 g, which showed no depression in m.p. on admixture with palmitic acid.

 β -Sitosteryl-D-glucoside tetraacetate (VI)—12.3 g. of (N b) was thrice recrystallized from EtOH to plates, m.p. 164 \sim 165°, (α)²⁵_D -25° (c=2, CHCl₃), yield, 9.8 g. *Anal.* Calcd. for C₄₃H₆₈O₁₀: C, 69.30; H, 9.20. Found : C, 69.21; H, 8.96.

 β -Sitosteryl-D-glucoside (VII) — Hydrolysis of (V, 3.0 g.) or (VI, 3.0 g.) with MeOH-KaOH (1.5 g.) gave a white prisms, which were insoluble in most of the organic solvents, but soluble in pyridine, m.p. 288 \sim 290° (decomp.), yield, 2.2 g. Anal. Calcd. for C₃₅H₆₀O₆: C, 72.85; H, 10.49. Found: C, 72.52; H, 10.46.

(VII) gave positive Liebermann-Burchard reaction.

 β -Sitosteryl-D-glucoside tetrabenzoate One g. of (VII) in pyridine (10 ml.) was heated with BzCl (7 ml.) for 30 min. The benzoate recrystallized from EtOH to needles, m.p. 196 \sim 198°, $[\sigma]_{25}^{25}$ +16° (c=2, CHCl₃), yield, 1.4 g. Anal. Calcd. for C₆₂H₇₆O₁₀ : C, 76.21; H, 7.71. Found : C, 76.47; H, 7.96.

Hydrolysis of the β -Sitosteryl-D-glucoside (VII) — Three g. of (VII) was hydrolyzed by refluxing for 2 hr. with a mixture of 60 ml. of AmOH and 20 ml. of 16% HCl. Most of the alcohol was removed by heating under a reduced pressure, and the aqueous acid solution was diluted with water, and extracted with several portions of Et₂O. The Et₂O solution was washed with water and the washings were added to the main solution. The aqueous solution was neutralized with dil. NaOH and concentrated to 10 ml. under a reduced pressure. Fehling's solution was reduced with this concentrated solution.

 β -Sitosterol (VII)—The Et₂O extract of the hydrolysis of (VII) was dried over Na₂SO₄ and evaporated. The residue was recrystallized from EtOH to plates, m.p. 136 \sim 137°, $[\alpha]_D^{25}$ -35° (c=2, CHCl₃), yield, 0.9 g. Anal. Calcd. for C₂₉H₅₀O: C, 83.99; H, 12.15. Found : C, 83.65; H, 11.98.

 β -Sitosteryl acetate was prepared in the usual way, m.p. 127 \sim 128°, $[\alpha]_D^{25}$ -41° (c=2, CHCl₃). Anal. Calcd. for C₃₁H₅₂O₂: C, 81.52; H, 11.48. Found: C, 81.45; H, 11.21.

 β -Sitosteryl benzoate was prepared in the usual way, m.p. 146 \sim 147°, (α)²⁵_D -13.5° (c=2, CHCl₃). Anal. Calcd. for C₃₈H₅₄O₂: C, 83.34; H, 10.49. Found: C, 83.12; H, 10.30.

Chromatography of Crude Crystalline Solid (III)—A solution of the crude crystalline solid (III, 10.0 g.) in benzene was percolated through a column $(3 \times 26 \text{ cm.})$ of alumina (150 g.), and the chromatogram eluted with a) benzene (1.2 L.), b) benzene-Et₂O (19:1, 700 ml.), c) benzene-MeOH (19:1, 400 ml.), d) MeOH (200 ml.).

cyclo-Artenol (XI)—Evaporation of the solution b) gave a solid, which was recrystallized from MeOH to needles, m.p. 99 \sim 102°. and raised to m.p. 112 \sim 115° after drying for 12 hr. at 65° *in vacuo*. $[\alpha]_{25}^{\infty}$ +52.5° (c=2, CHCl₃). Yield, 3.4 g. Anal. Calcd. for C₃₀H₅₀O: C, 84.44; H 11.81. Found: C, 84.16; H, 11.58.

cyclo-Artenyl acetate was prepared in the usual way, m.p. $122\sim123^{\circ}$, $[\alpha]_D^{25}$ +59.5° (c=2, CHCl₃). Anal. Calcd. for $C_{32}H_{52}O_2$: C, 81.99; H, 11.18. Found : C, 82.17; H, 11.20.

cyclo-Artenyl benzoate was prepared in the usual way, m.p. $130 \sim 131^{\circ}$, $[\alpha]_D^{25} + 58^{\circ}$ (c=2, CHCl₃). Anal. Calcd. for $C_{37}H_{54}O_2$: C, 83.72; H, 10.25. Found : C, 84.05; H, 10.16.

cyclo-Artanyl acetate — Hydrogenation of cyclo-artenyl acetate in presence of Pt catalyst in AcOH gave cyclo-artanyl acetate, as needles from MeOH-CHCl₃, m.p. $130\sim131^{\circ}$, $[\alpha]_D^{25}$ +58° (c=2, CHCl₃). Anal. Calcd. for $C_{32}H_{54}O_2$: C, 81.64; H, 11.56. Found: C, 81.36; H, 11.31.

The authors are grateful to Mr. T. Onoe, and Miss H. Otsuka for elemental analysis, and to Mr. N. Hayashi and Mr. M. Kishi for spectral measurements.

Summary

From the alkaloid-free fraction of opium, β -sitosteryl-D-glucoside and its monopalmitate, (+)-10-nonacosanol, *cyclo*-laudenol and *cyclo*-artenol were isolated.

(Received May 26, 1961)