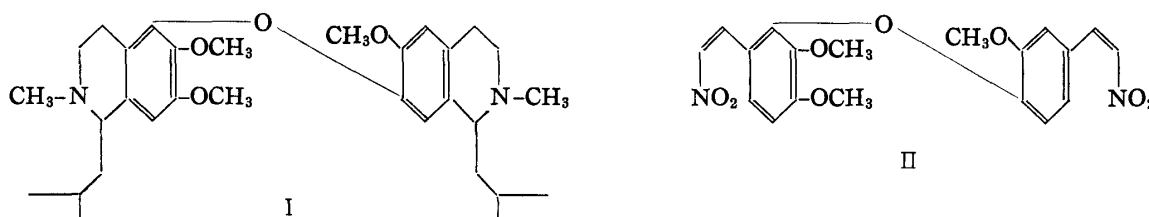


**229. Kiyoshi Bessho<sup>\*1,\*2</sup>: Studies on Pilocereine and Related Compounds. IV.<sup>\*3</sup> Synthesis of O-Methylisopilocereine. (1).**

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In order to synthesize I, which was supposed to be O-methylisopilocereine, an important intermediate (II) was prepared.<sup>\*3</sup> However, the poor yield of II showed that this synthetic route was unpractical. The present paper deals with the synthesis of I



according to an alternative route, *i.e.* an Ullmann condensation between two kinds of isoquinoline bases.

1-Isobutyl-2-methyl-5-bromo-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (VIII) was synthesized as shown in Chart 1, and was characterized as an oxalate, m.p. 164~165° (decomp.), and a picrate, m.p. 172~174°. Condensation of this compound with lophocereine (IX) was carried out in pyridine solution in the presence of potassium carbonate,

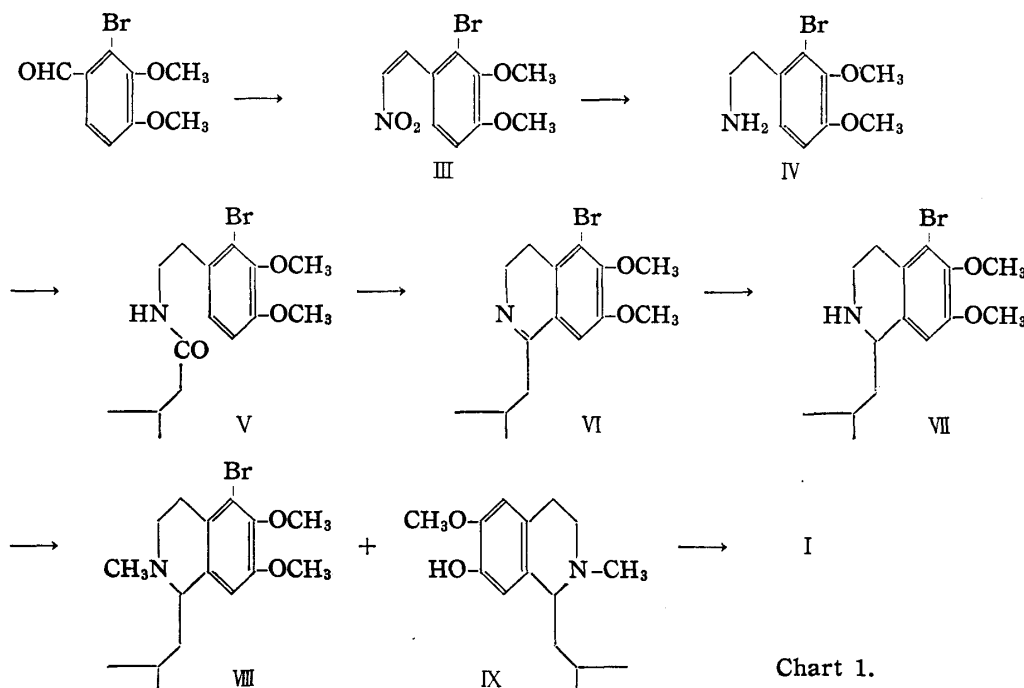


Chart 1.

potassium iodide, and copper powder. After removal of residual starting materials and the debrominated compound (X) by vacuum distillation, an oily base was obtained. This product showed a single spot on PPC, MPC, and PEP, but two spots were seen on

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TLC.\*<sup>4</sup> Repeated alumina chromatography resulted in the separation of two bases (I-A and I-B) corresponding to the two spots on TLC. These bases had identical ultraviolet spectra and their infrared spectra were also substantially the same (Fig. 1). These materials were considered to correspond to the two racemic diastereomers of I.

Although both bases of I coincided well with O-methylisopilocerine on PPC, MPC, and PEP, and also in regard to their ultraviolet spectra, several discrepancies were observed in their infrared spectra in chloroform solutions. This indicated that I was not identical with O-methylisopilocerine.

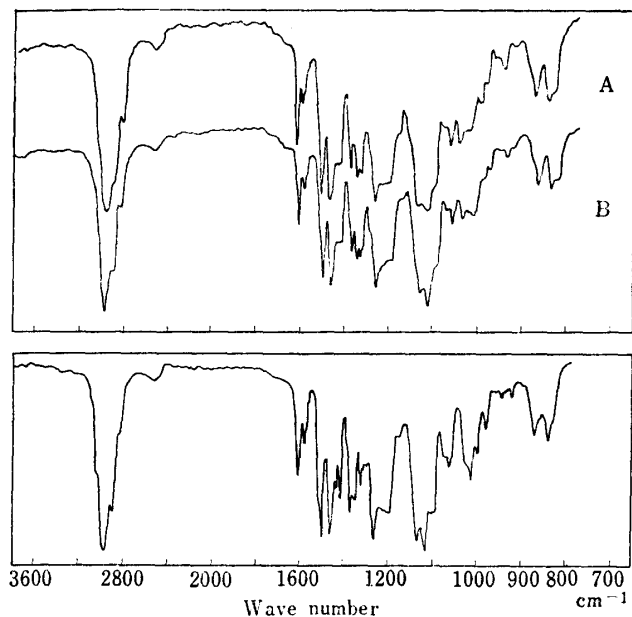
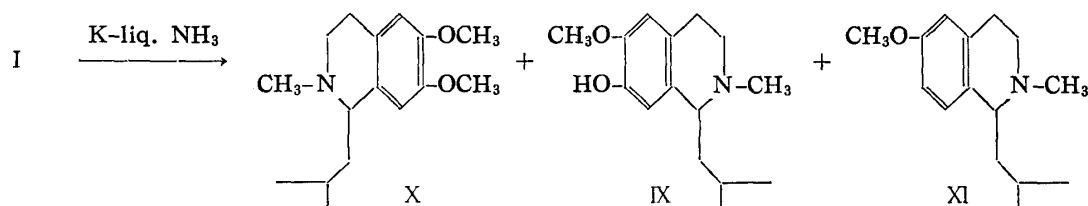


Fig. 1. Infrared Absorption Spectra (in  $\text{CHCl}_3$ )

Above : Synthesized base  
A: I-A; B: I-B  
Below : O-Methylisopilocerine

To verify the structure of the base (I) synthesized, the potassium-liq. ammonia cleavage reaction was carried out, where the base—a 1:1 mixture of I-A and I-B—gave O-methylpilocerine (X) and pilocerine (IX) nearly quantitatively with a trace of the 6-methoxy compound (XI). Considering the synthetic route, this result confirms that the structure of the base synthesized is I.

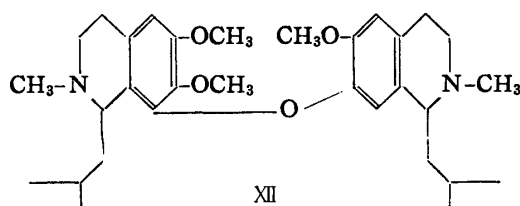


From the result of potassium-liq. ammonia cleavage<sup>1)</sup> of O-methylisopilocerine, it was deduced that the structure of the base should be I or XII. However, since O-methylisopilocerine differed from the synthesized I, it could not have structure (I) and so it must have formula (XII), which is that assigned to O-methylpilocerine and O-methylpiloceredine by Djerassi, *et al.*<sup>2)</sup>

\*<sup>4</sup> PPC : paper chromatography. MPC : multi-buffered paper chromatography. PEP : paper electrophoresis. TLC : thin layer chromatography. Cf. the footnote in Part II.<sup>1)</sup>

1) Part II. This Bulletin, 11, 1477 (1963).

2) C. Djerassi, S. K. Figdor, J. M. Bobbitt, F. X. Markley : J. Am. Chem. Soc., 79, 2203 (1957); C. Djerassi, T. Nakano, J. M. Bobbitt : Tetrahedron, 2, 58 (1958).



### Experimental\*<sup>5</sup>

**$\beta$ -Nitro-2-bromo-3,4-dimethoxystyrene (III)**—A solution of 30 g. of 2-bromoveratraldehyde, 30 ml. of  $\text{CH}_3\text{NO}_2$ , and 10 g. of  $\text{AcONH}_4$  in 80 ml. of  $\text{AcOH}$  was refluxed for 2 hr. The reaction mixture was then poured into 1 L. of ice water to form a crystalline precipitate. Recrystallization from  $\text{Me}_2\text{CO}$ - $\text{MeOH}$  gave 27.0 g. of yellow plates, m.p.  $96\sim 98^\circ$ , (77%). *Anal.* Calcd. for  $\text{C}_{10}\text{H}_{10}\text{O}_4\text{NBr}$ : C, 41.68; H, 3.52. Found: C, 41.48; H, 3.53.

**2-Bromo-3,4-dimethoxyphenethylamine (IV)**—Electrolytic reduction of 10 g. of III (Anode solution: 20%  $\text{H}_2\text{SO}_4$ . Cathode: Lead plate, 100  $\text{cm}^2$ . Cathode solution: A mixture of 260 ml. of  $\text{MeOH}$ , 100 ml. of  $\text{AcOH}$ , 320 ml. of tetrahydrofuran, and 40 ml. of conc.  $\text{HCl}$ . Current: 7 A. 1.5 hr.  $17\sim 20^\circ$ ) gave an oily base, which was converted into the oxalate and recrystallized from  $\text{MeOH}$  to yield 6.05 g. of IV oxalate as colorless plates, m.p.  $205\sim 206^\circ$  (decomp.), (50%). *Anal.* Calcd. for  $\text{C}_{10}\text{H}_{14}\text{O}_2\text{NBr}\cdot(\text{COOH})_2$ : C, 41.16; H, 4.61. Found: C, 41.33; H, 4.97.

The neutral oxalate was also found on treatment of the amine with a half mole of oxalic acid and recrystallized from  $\text{MeOH}$  as colorless needles, m.p.  $234\sim 235^\circ$  (decomp.). *Anal.* Calcd. for  $\text{C}_{10}\text{H}_{14}\text{O}_2\text{NBr}\cdot\frac{1}{2}(\text{COOH})_2$ : C, 43.29; H, 4.96. Found: C, 43.13; H, 5.04.

**N-(2-Bromo-3,4-dimethoxyphenethyl)isovaleramide (V)**—To a stirred mixture of 10%  $\text{KOH}$  and an ethereal solution of IV (regenerated from 5.5 g. of the oxalate) was added dropwise an ethereal solution of 2.2 g. of isovaleryl chloride with stirring. After 1 hr., the  $\text{Et}_2\text{O}$  layer was separated from the aqueous layer, washed successively with dil.  $\text{HCl}$  and dil.  $\text{KOH}$ , dried over anhyd.  $\text{MgSO}_4$ , and evaporated. The product was recrystallized from  $\text{Me}_2\text{CO}$ - $\text{EtOH}$  yielding 5.2 g. of colorless needles, m.p.  $104\sim 105^\circ$ , (96%). *Anal.* Calcd. for  $\text{C}_{15}\text{H}_{22}\text{O}_3\text{NBr}$ : C, 52.34; H, 6.45. Found: C, 52.08; H, 6.39.

**1-Isobutyl-5-bromo-6,7-dimethoxy-3,4-dihydroisoquinoline (VI)**—A solution of 6.8 g. of V and 30 ml. of  $\text{POCl}_3$  in 30 ml. of dry toluene was refluxed for 3 hr. The solvent and the excess of  $\text{POCl}_3$  were removed by evaporation under reduced pressure. After addition of 10%  $\text{KOH}$ , the product was taken up in  $\text{Et}_2\text{O}$ , and the  $\text{Et}_2\text{O}$  solution was dried over anhyd.  $\text{K}_2\text{CO}_3$  and evaporated. The residue crystallized on treatment with  $\text{Et}_2\text{O}$ -hexane to give 5.2 g. of colorless plates, m.p.  $76\sim 77^\circ$ , (81%). *Anal.* Calcd. for  $\text{C}_{15}\text{H}_{20}\text{O}_2\text{NBr}$ : C, 55.22; H, 6.18. Found: C, 55.52; H, 6.14.

Hydrochloride: Recrystallized from  $\text{MeOH}$ . Colorless plates, m.p.  $174\sim 176^\circ$  (decomp. darkened from  $165^\circ$ ). *Anal.* Calcd. for  $\text{C}_{15}\text{H}_{20}\text{O}_2\text{NBr}\cdot\text{HCl}$ : C, 49.67; H, 5.85; N, 3.86. Found: C, 49.37; H, 5.86; N, 3.70.

**1-Isobutyl-5-bromo-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (VII)**—A solution of 5.2 g. of VI in 100 ml.  $\text{MeOH}$  was treated with 2.5 g. of  $\text{NaBH}_4$ . The solution was stirred at room temperature for 1 hr. The solvent was removed under reduced pressure and after addition of  $\text{H}_2\text{O}$  the product was extracted with  $\text{Et}_2\text{O}$ . The  $\text{Et}_2\text{O}$  extract was dried over anhyd.  $\text{K}_2\text{CO}_3$  and evaporated leaving 5.0 g. of a colorless, oily product. It was converted into the oxalate and recrystallized from  $\text{MeOH}$ - $\text{EtOH}$  to yield 6.0 g. of colorless needles, m.p.  $205\sim 206^\circ$  (decomp.), (90%). *Anal.* Calcd. for  $\text{C}_{15}\text{H}_{22}\text{O}_2\text{NBr}\cdot(\text{COOH})_2$ : C, 48.81; H, 5.79. Found: C, 48.77; H, 5.69.

**1-Isobutyl-2-methyl-5-bromo-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (VIII)**—A mixture of 4.7 g. of VII (liberated from 6.0 g. of the oxalate), 40 ml. of 30%  $\text{HCHO}$ , and 40 ml. of 98%  $\text{HCOOH}$  was heated for 4 hr. in a boiling water bath. Usual acid-alkali treatment of the reaction mixture afforded 6.0 g. of the N-methyl compound, which was converted into the oxalate and recrystallized from  $\text{EtOH}$  to give 5.2 g. of colorless, rhombic plates, m.p.  $164\sim 165^\circ$  (decomp.), (84%). *Anal.* Calcd. for  $\text{C}_{16}\text{H}_{24}\text{O}_2\text{NBr}\cdot(\text{COOH})_2$ : C, 50.01; H, 6.07. Found: C, 49.96; H, 6.29.

The picrate was prepared in the usual manner and recrystallized from  $\text{EtOH}$  as yellow, rhombic plates, m.p.  $172\sim 174^\circ$ . *Anal.* Calcd. for  $\text{C}_{16}\text{H}_{24}\text{O}_2\text{NBr}\cdot\text{C}_6\text{H}_3\text{O}_7\text{N}_3$ : C, 46.24; H, 4.77. Found: C, 46.49; H, 4.86.

**1,1'-Diisobutyl-2,2'-dimethyl-6,6',7-trimethoxy-1,1',2,2',3,3',4,4'-octahydro-5,7'-oxydiisoquinoline (I)**—A mixture of 5.12 g. of VIII (regenerated from 6.48 g. of the oxalate), 4.25 g. of lophocericine (IX) (regenerated from 5.00 g. of the oxalate), 20 ml. of dry pyridine, 4.0 g. of finely pulverized anhyd.  $\text{K}_2\text{CO}_3$ , 0.5 g. of  $\text{Cu}$  powder, and 0.2 g. of  $\text{KI}$  was heated in a slow stream of  $\text{N}_2$  in an oil bath ( $153\sim$

\*<sup>5</sup> All melting points are uncorrected.

155°). After 8 hr., 0.2 g. of Cu and 1.0 g. of  $K_2CO_3$  were added; after 30 hr., 0.5 g. of Cu was added. The reaction mixture was examined by PPC from time to time. At the beginning of the reaction, the mixture showed two spots at  $R_f$  0.80 (VIII) and 0.70 (IX). As the reaction proceeded, the spot at  $R_f$  0.80 became weaker and a new spot appeared at  $R_f$  0.63 (Debromo-compound (X) shows a spot at the same position as IX,  $R_f$  0.70). After 35 hr., the spot at  $R_f$  0.80 almost disappeared. The reaction mixture was filtered, and the residue was washed with  $CHCl_3$ . The combined filtrates were evaporated under reduced pressure, and the residue was dissolved in  $Et_2O$ . The solution was fractionated in the usual manner to afford 5.71 g. of the non-phenolic base and 1.83 g. of the phenolic base. The latter, when treated with oxalic acid in EtOH, gave 1.81 g. of IX oxalate, m.p. 213~214° (decomp.).

Vacuum distillation of the non-phenolic, basic fraction gave 1.90 g. of a low-boiling material (b.p. 160~165°), which was converted into the oxalate yielding 2.17 g. of the debromo-compound (X) oxalate, m.p. 136~137°. The distillation residue was chromatographed on  $Al_2O_3$  (1 × 10 cm.) with a hexane- $Et_2O$  (1:1) mixture. The eluted base was dissolved in  $CHCl_3$ , washed with pH 5.7 buffer solution and then extracted with pH 3.9 buffer solution. Basification of the pH 3.9 buffer extracts and extraction with  $Et_2O$  gave an oily residue, which was again chromatographed on  $Al_2O_3$  (1 × 30 cm.). After a trace of X had been eluted with hexane, successive elution with hexane, hexane-benzene, benzene, and benzene- $Et_2O$  afforded forty-three fractions (Fract. 9~51) showing a same spot on PPC ( $R_f$  0.63). Total weight of the base was 1.59 g. (24%). These fractions also showed a single spot on MPC (pH 3.6) and on PEP, but two spots were observed on TLC. Therefore the fractions were further submitted to chromatographic separation.

Fractions 9~34 were combined (0.35 g.) and chromatographed on  $Al_2O_3$  (0.8 × 20 cm.). The first two fractions eluted with a hexane- $Et_2O$  mixture (19:1) showed a single spot (the higher one) on TLC and gave 33 mg. of a colorless, oily base (I-A).

Fractions 46~51 (0.37 g.) were also chromatographed on  $Al_2O_3$  (0.8 × 20 cm.) and the last four fractions eluted with a hexane- $Et_2O$  mixture (9:1) gave 12 mg. of a colorless, oily base which showed a single spot (the lower one) on TLC (I-B).

These two bases were considered to be the diastereomers of I, because their IR spectra (in  $CHCl_3$ ) were essentially identical. UV:  $\lambda_{max}^{EtOH}$  283  $m\mu$  ( $\log \epsilon$  3.72) (I-A). Crystallizations of them and of their derivatives were attempted without success.

**Cleavage of I by Potassium in Liquid Ammonia**—Fractions 35~45 of the above described chromatography of I afforded 864 mg. of the oily base which was shown by TLC to be a nearly 1:1 mixture of diastereomers I-A and I-B.

A 200 mg. portion of the oily base was allowed to react with K (100 mg.) in 200 ml. of liq.  $NH_3$  at -58~-53° for 3 hr. Treatment of the reaction mixture in the usual manner gave 94 mg. of non-phenolic, basic and 95 mg. of phenolic, basic fractions.

The non-phenolic, basic fraction was treated with picric acid and recrystallized from EtOH to yield 150 mg. of O-methyllophocerine (X) picrate as yellow plates, m.p. 184~185°, identity of which was shown by mixed melting point and IR comparison (in Nujol) with an authentic sample. *Anal.* Calcd. for  $C_{18}H_{25}O_2N \cdot C_6H_3O_7N_3$ : C, 53.65; H, 5.73. Found: C, 53.91; H, 5.96.

The free base (15 mg.), recovered from the mother liquor, was chromatographed on  $Al_2O_3$  (0.4 × 10 cm.) and eluted with hexane- $Et_2O$  (19:1). Fract. 1~2 were converted into the picrate and washed with EtOH. The picrate (1 mg.) melted at 153~154° and showed no depression of the melting point on admixture with a specimen of 1-isobutyl-2-methyl-6-methoxy-1,2,3,4-tetrahydroisoquinoline (XI) picrate (m.p. 154~155°). Fract. 3~6 afforded additional 15 mg. of X picrate, m.p. 184~185°.

The IR spectrum (in  $CHCl_3$ ) of the bisected, phenolic base was shown to be identical with that of lophocerine (IX). When treated with picric acid and recrystallized from EtOH, 172 mg. of IX picrate was obtained as yellow, rhombic plates, m.p. 194~195°. Identity was confirmed by mixed melting point and IR comparison (in Nujol). *Anal.* Calcd. for  $C_{15}H_{23}O_2N \cdot C_6H_3O_7N_3$ : C, 52.71; H, 5.48. Found: C, 52.43; H, 5.77.

Further examination of the mother liquor gave no other base.

### Summary

Base (I), which had been presumed to be O-methylisopilocereine, was synthesized and two racemic diastereomers were isolated. However, the fact that these bases are not O-methylisopilocereine suggested that the latter should be represented by formula (XII), which had been assigned to O-methylpilocereine and O-methylpiloceredine.

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