(Chem. Pharm. Bull.) 11 (12) 1551 ~ 1556)

UDC 547.873.07:615.778

240. Sumiko Watanabe and Takeo Ueda: Syntheses and Antiviral Activity of 3-Thioxo-6-aryl-3,4-dihydro-as-triazin-5(2H)-one and 6-Aryl-as-triazine-3,5(2H,4H)-dione.

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In the previous paper,¹⁾ it was reported that, among the compounds of 3-mercapto-6-alkyl-as-triazine-5-ol, the 6-propyl derivative showed the most remarkable effect on the PR-8 strain of influenza A virus in chorioallantoic membrane culture. On the other hand, Hamre, et al.²⁾ reported that 4-methoxy- and 4-amino-benzaldehyde thiosemicarbazone showed very weak activities on vaccinia virus in embryonated egg, while Bauer³⁾ reported that isatine thiosemicarbazone exerted very weak in vivo effect on the virus in mice.

These findings suggest that 3-mercapto-6-aryl-as-triazine-5-ol and its preceding compound, arylpyruvic acid thiosemicarbazone, and their related compound are of interest to screen as to their antiviral effect for the purpose of searching antiviral agents. Thus, compounds of phenylpyruvic acid thiosemicarbazone substituted with alkyl group, hydroxy group, alkoxy group, or halogen atom in benzene ring, 3-thioxo-6-aryl-3,4-dihydro-as-triazin-5(2H)-one and their related were synthesized and screened as to their antiviral activity on polio virus.

This paper is concerned with the syntheses and antiviral activity of substituted phenylpyruvic acid thiosemicarbazone, 3-thioxo-6-aryl-3,4-dihydro-as-triazin-5(2H)-one and 6-aryl-as-triazine-3,5(2H,4H)-dione.

Syntheses of 3-Thioxo-6-aryl-as-triazin-5(2H)-one and 6-Aryl-as-triazine-3.5(2H.4H)-dione

Several compounds of 3-mercapto-as-triazin-5-ol substituted with aryl or alkyl group at 6-position were already synthesized by the treatment of corresponding α -keto-carbonic acid thiosemicarbazone with alkali by Girard, (Cattelain, Baugault, And Nakata.) By the modification of these methods, 3-thioxo-6-aryl-as-triazin-5(2H)-one was newly synthesized as follows, according to the scheme shown in Chart 1.

Phenylpyruvic acid substituted with alkyl group, hydroxy group, alkoxy group, or halogen atom at 2,3 and/or 4-position in benzene ring was employed as the starting material, which was prepared through the reaction of substituted benzaldehyde with hippuric acid or acetylglycine and acetic anhydride, $^{7)}$ followed by hydrolysis of the resulting azlactone. Then, these substituted phenylpyruvic acids were converted to corresponding thiosemicarbazones in the usual manner. The thiosemicarbazone was cyclized to 3-thioxo-6-aryl-3,4-dihydro-as-triazin-5(2H)-one by treating with potassium carbonate, according to the method of Nakata. Moreover, the resulting thioxo-triazin-one derivative was converted to 3-methylthio-6-aryl-as-triazin-5(4H)-one.

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¹⁾ I. Nakata, T. Ueda: Yakugaku Zasshi, 80, 1068 (1960).

²⁾ D. Hamre, K. A. Brownlee, R. Donovik: J. Immunol., 67, 305 (1951).

³⁾ D.J. Bauer: Brit. J. Exptl. Pathol., 36, 105 (1955).

⁴⁾ M. Girard: Compt. rend., 206, 1303 (1938).

⁵⁾ E. Cattelain: Bull. Soc. Chem.. 11, 256 (1944).

⁶⁾ J. Baugault: Compt. rend., 159, 83 (1915).; *Ibid.*, 186, 151 (1928).

⁷⁾ R.M. Herbst: Org. Synth., Col. Vol. II, p. 1; *Ibid.*, p. 691. J.S. Buck: *Ibid.*, p. 70. H.R. Snyer: *Ibid.*, p. 444.

Next, 6-aryl-as-triazine-3,5(2H.4H)-dione was prepared as follows. Some of these derivatives had been already synthesized by the following two methods; one was the cyclization of pyruvic acid semicarbazone derivative by the treatment with alkali⁸⁾, the other was the oxidation of 3-thioxo-3, 4-dihydro-as-triazin-5(2H)-one derivative with alkaline potassium permanganate, as reported by Nakata.

According to either the former or the latter method, as-triazinedione derivative was prepared, as shown in Chart 2.

Thus, twelve compounds of phenylpyruvic acid thiosemicarbazone, nine compounds of 3-thioxo-6-aryl-3, 4-dihydro-as-triazin-5(2H)-one, five compounds of 3-methylmer-capto-6-aryl-as-triazin-5(4H)-one, and five compounds of 6-aryl-as-triazine-3,5(2H,4H)-dione were newly synthesized, as shown in Table I \sim IV.

TABLE I. R-C-COOH

		[∥] NHCNH₂ [™] X			
R	X	m.p. (°C)	A n n a a m a a a	N (%)	
			Appearance	Calcd.	Found.
$HO-C_6H_4CH_2$	S	$194{\sim}195$	Colorless needles	16.60	16.60
$C_6H_5CH_2$	"	174	Colorless prisms	17.72	17.97
$C1-C_6H_4CH_2$	"	$193{\sim}194$	Colorless plates	15.47	15.23
$3,4-(CH_3O)_2C_6H_3CH_2$	"	$175 \sim 176$	Colorless prisms	14.14	14.19
$CH_3O-C_6H_4CH_2$	"	$205 \sim 206$	Colorless needles	15.73	15.76
$C_3H_7O-C_6H_4CH_2$	"	$154 \sim 155$	Colorless powder	14.23	14.23
$C_2H_5OC_6H_4CH_2$	"	$174 \sim 175$	Colorless prisms	14.94	15.22
$CH_3C_6H_4CH_2$	"		11	1 6. 73	16.54
$Br-C_6H_4CH_2$	"	$200 \sim 201$	Colorless plates	13.29	13.34
$2,4-Cl_2C_6H_3CH_2$	"	$205{\sim}206$	"	13.73	13.65
$3,4-(C_2H_5O)_2C_6H_3CH_2$	"		Colorless powder	12.92	12.84
C_6H_5	"	170	Colorless needles	22.21	22.05
$C_6H_5CH_2^{a)}$	O	179	Colorless prisms	19.00	18.91
$HO-C_6H_4CH_2$	"	$184 {\sim} 185$	"	17.72	17.60
a) Known compounds	S				

⁸⁾ P.K. Chang: J. Org. Chem., 23, 1953 (1958).

	Table II.	$R \xrightarrow[N]{N-NH} S$			
		NH NH			
		O	N	(%)	
R	m.p. (°C)	Appearance	,	(70)	
	• •		Calcd.	Found	
$C_3H_7OC_6H_4CH_2$	$186 \sim 187$	Colorless powder	15. 16	15. 15	
$C1-C_6H_4CH_2$	$226{\sim}227$	Colorless needles	16.56	16.44	
$3,4-(CH_3O)_2C_6H_3CH_2$	$204 \sim 205$	"	15.05	15.09	
$HO-C_0H_4CH_2$	$221 \sim 222$	<i>"</i>	17.87	17.90	
$C_2H_5OC_6H_4CH_2$	$198 \sim 199$	Colorless prisms	15.96	15.69	
$CH_3C_6H_4CH_2$	$182 \sim 183$	Colorless needles	18.02	18.03	
2,4-Cl ₂ C ₆ H ₃ CH ₂	$203 \sim 204$	Colorless prisms	15.55	15.42	
$Br-C_6H_4CH_2$	$230 \sim 232$	Colorless needles	15.00	14.71	
$3,4-(C_2H_5O)_2C_6H_3CH_2$	$156{\sim}157$	"	15.49	15.49	
$C_6H_5CH_2^{a)}$	$194 \sim 195$	Colorless plates	19.17	19. 16	
$\mathrm{CH_3O}\mathrm{C_6H_4CH_2}^{a)}$	$170 \sim 171$	<i>"</i>	16.86	17.09	
$C_6H_5^{a)}$	170	Colorless needles	20.48	19.72	
a) Known compound					
		NT NT			
	Тава Ш	R -≪N—N NH SCH₃			
	I ABLE III.	NH SCH ₃			
		Ö			
			N	(%)	
R	m.p. (°C)	Appearance		·	
	(0)		Calcd.	Found	
$C_6H_5CH_2^{a)}$	$200 \sim 201$	Colorless plates	18.02	17.93	
$3,4-(C_2H_5O)_2C_6H_3CH_2$	$183 \sim 184$	"	13.08	13. 11	
C_6H_5	$236 \sim 237$	"	19. 17	19.40	
$Br-C_6H_4CH_2$	$264 \sim 265$	"	13.46	13.37	
$2,4$ – $Cl_2C_6H_3CH_2$	$232\sim\!233$	"	13. 91	13.70	
$CH_3-C_6H_4CH_2$	$218{\sim}219$	"	17.00	16.79	
a) Known compound					
		N-NH			
	TABLE IV.	R - NH = O			
		NH O			
		O	N (%)		
R	m.p.	Appearance		(70)	
	(°C)		Calcd.	Found	
Cl-C ₆ H ₄ CH ₂	$234 \sim 235$	Colorless prisms	17.68	17.70	
3,4-(CH ₃ O) ₂ C ₆ H ₃ CH ₂	$236{\sim}237$	<i>"</i>	15.96	15.98	
$C_3H_7O-C_6H_4CH_2$	$196{\sim}197$	"	16.08	16.02	
$C_2H_5O-C_6H_4CH_2$	$205 \sim 206$	"	15.05	15.06	
$3,4-(C_2H_5O)_2C_6H_3CH_2$	$210 \sim 211$	Colorless plates	14.43	14.70	
$2,4-C1_2-C_6H_3CH_2$	$198{\sim}199$	<i>"</i>	15.44	15.29	
Br-C ₆ H ₄ CH ₂	$257 \sim 259$	Colorless prisms	14.90	14.75	
$C_6H_5\check{CH_2}^{a)}$	$208 \sim 209$	Colorless needles	20.68	20.75	
$C_{\scriptscriptstyle{6}}H_{\scriptscriptstyle{5}}{}^{a)}$	$255{\sim}256$	"	22.21	22.05	
a) Known compound					
, -					

It is considerable that 3-thioxo-6-aryl-as-triazin-5-(2H)-one could have two tautomeric structures, thiol form (II) and thioxo form (I). Baugault⁹⁾ assumed that these compounds should mainly have the tautomeric form (II). This assumption, however, seemed unreasonable, since, in general, most of keto-enol tautomers exist mainly in keto form. Therefore, the authors investigated the tautomerism of 3-thioxotriazinone derivative, only by the inspection of its infrared and ultraviolet absorption spectra.

⁹⁾ J. Baugault: Compt. rend., 186, 1216 (1928).

The infrared absorption spectra of 6-methyl-, 6-benzyl-, and 6-(p-bromobenzyl)-3-thioxo-3,4-dihydro-as-triazin-5(2H)-one showed characteristic absorptions in region of $1670 \sim 1690 \, \mathrm{cm^{-1}}$. (IR $\nu_{\mathrm{max}}^{\mathrm{Nujol}} \, \mathrm{cm^{-1}}$: 1674, 1678, 1686) and in $1530 \sim 1550$ (IR $\nu_{\mathrm{Nujol}}^{\mathrm{max}} \, \mathrm{cm^{-1}}$: 1548, 1545, 1528), the former assigned to C=O and the latter, to CS-NH- in the thioxo form (I). But any characteristic absorption due to -SH was not observed therein.

 $T_{\mbox{\scriptsize ABLE}}$ V. Antiviral Activity on the Mahoney Strain of Polio Virus Type 1

	R	Toxicity (M)	Control	Treated
R-C-COOH	HO-C ₆ H ₄ CH ₂	10-4	$0/2^{a_0}(2,2)^{b_0}$	0/2 (6, 6)
$\stackrel{\scriptscriptstyle{II}}{N}NHCSNH_2$	C ₆ H ₅ CH ₂	10-4	0/2	0/2
111110011112	Cl-C ₆ H ₄ CH ₂	10-4	0/2	0/2
	$3,4-(CH_3O)_2C_6H_3CH_2$	10 ^{- 3}	0/2	2/2
	CH ₃ O-C ₆ H ₄ CH ₂	10^{-3}	0/2 (3, 3)	0/2(6,7)
	$C_3H_7O-C_6H_4CH_2$	10^{-4}	0/2	0/2
	$C_2H_5O-C_6H_4CH_2$	10-4	0/2	0/2
	CH ₃ C ₆ H ₄ CH ₂	10 - 4	0/2	0/2
	Br-C ₆ H ₄ CH ₂	10 · 4	0/2	0/2
	$2,4-\text{Cl}_2\text{C}_6\text{H}_3\text{CH}_2$	10-4	0/2	0/2
	$3,4-(C_2H_5O)_2C_6H_3CH_2$	10^{-4}	0/2	0/2
	C_6H_5	10 - 5	0/2	0/2
R-C-COOH	$C_6H_5CH_2$	103	0/2	0/2
$\stackrel{\parallel}{ m N}{ m NHCONH}_2$	HO-C ₆ H ₄ CH ₂	10 - 3	0/2	0/2
N-NH	/ C ₃ H ₇ OC ₆ H ₄ CH ₂	10-5	0/2 (2, 2)	1/2 (4,7)
$R - \langle NH \rangle = S$	C1-C ₆ H ₄ CH ₂	10-5	0/2	0/2
`i⊢NH	3,4-(CH ₃ O) ₂ C ₆ H ₃ CH ₂	10^{-4}	0/2 (2, 2)	0/2 (6, 6)
O	HO-C ₆ H ₄ CH ₂	10-4	0/2	0/2
	$C_2H_5O-C_6H_4CH_2$	10-4	0/2	0/2
	CH ₃ C ₆ H ₄ CH ₂	10-4	0/2 (2, 2)	0/2 (3, 3)
	2,4-Cl ₂ C ₆ H ₃ CH ₂	10-5	0/2	0/2
	Br-C ₆ H ₄ CH ₂	10-4	0/2 (2, 2)	0/2 (3, 3)
	$3,4-(C_2H_5)_2C_6H_3CH_2$	10 - 5	0/2	0/2
	$C_6H_5CH_2$	10-4	0/2(2,2)	0/2(3,3)
	CH ₃ O-C ₆ H ₄ CH ₂	10-4	0/2(2,2)	0/2(3,3)
	C_6H_5	10^{-5}	0/2(2,2)	1/2(3,7)
NN	$C_6H_5CH_2$	10-5	0/2	0/2
R-\(\bigcap_{NH}\)\rightarrow SCH ₃	$3,4-(C_2H_5)_2C_6H_3CH_2$	10-4	0/2	0/2
)—NH	C_6H_5	10-3	0/2	0/2
O	Br-C ₆ H ₄ CH ₂	10^{-5}	0/2~(2,2)	0/2(3,3)
	$2,4$ - $Cl_2C_6H_3CH_2$	10^{-5}	0/2 (2, 2)	0/2(3,3)
	CH ₃ C ₆ H ₄ CH ₂	10-4	0/2	0/2
R-N-NH ONH=O	ClC ₆ H ₄ CH ₂	10-4	0/2 (2, 2)	1/2 (5,7)
R-($3,4-(CH_3O)_2C_6H_3CH_2$	10^{-4}	0/2~(2,2)	0/2 (3, 3)
NH O	$C_3H_7O-C_6H_4CH_2$	10^{-4}	0/2 (2, 2)	0/2(3,3)
U	$C_2H_5O-C_6H_4CH_2$	10^{-4}	0/2	0/2
	$3,4-(C_2H_5O)_2C_6H_3CH_2$	10^{-4}	0/2 (2, 2)	0/2(3,3)
	$2,4$ – $Cl_2C_2H_3CH_2$	10-4	0/2 (2, 2)	0/2 (3, 3)
	Br-C ₆ H ₄ CH ₂	10^{-4}	0/2	0/2
	C_6H_5	10-4	0/2 (2, 2)	0/2 (3, 5)

a) The numerator represents the number of tubes in which no CPE was observed and denominator, total tubes used.

b) Days at which CPE was observed.

This experiment was carried out by using $100\times TCID_{50}$ of polio virus.

The ultraviolet absorption spectra of the above three compounds in 10% ethanolic solution showed absorption maxima at 275 mm (ε 11,800) and resembled to those of in acidic 10% ethanolic solution (all in keto form), which showed maxima at 270 mm (ε 14,300), but differed from those in 10% alkaline ethanolic solution (all in enol form), which showed each maxima at 225 mm (ε 16,500), 245 mm (ε 13,300) and 255 mm (ε 14,300).

From the results, it may be assumed that these mercatotriazinone derivatives exist mainly in keto form (I).

Screening Test with the Compounds Synthesized

The compounds synthesized above, were examined as to their inhibitory effect on the Mahoney strain of polio virus type 1 in tissue culture, according to the method described in the previous paper.¹⁰⁾ The results are shown in Table V.

As can be seen from the table, it may be said that 3,4-dimethoxyphenylpyruvic acid thiosemicarbazone was effective on the virus, while 6-phenyl- and 6-(p-propoxy-benzyl)-3-thioxo-3,4-dihydroxy-as-triazin-5(2H)-one, and 6-(p-chloro-benzyl)-as-triazine-3,5(2H,4H)-dione were slightly effective, in which only one tube among the used showed the inhibitory effect on the development of the cytopathogenic effect. Three compounds of them, p-hydroxyphenyl- and p-methoxyphenylpyruvic acid thiosemicarbazone, and 6-(3,4-dimethoxybenzyl)-3-thioxo-3,4-dihydro-as-triazin-5(2H)-one also showed slightly the delay of the time necessary for the appearance of the cytopathogenic effect caused by polio virus.

For the second screening test, by using the end point estimating method, the effect of 3,4-dimethoxyphenylpyruvic acid thiosemicarbazone on the Mahoney strain was further examined in Hep. No. 2 cells. The experimental results shown in Table VI, indicates that $TCID_{50}$ of polio virus is reduced to 1/1000 by addition of 10^{-3} mole of the compound.

Table VI. Inhibitory Effect of 3,4-Dimethoxyphenylpyruvic Acid Thiosemicarbazone on the Multiplication of the Mahoney Strain of Polio Virus Type 1

 $TCID_{50}^{a)}$ $TCID_{50}^{a)}$ $TCID_{50}^{a)}$ Control 6.5 Treated 3.5

At the present stage, it is difficult to discuss the relationship between antiviral activity and chemical structure of the compounds of these two series, from the standpoint of medicinal chemistry. However, from the fact that 3,4-dimethoxyphenylpyruvic acid thiosemicarbazone was effective on polio virus, while its cyclic derivative, 6-(3,4-dimethoxybenzyl)-3-thioxo-3,4-dihydro-as-triazin-5(2H)-one was only slightly effective on the virus, it seems that the structure of ketocarbonic acid thiosemicarbazone might contribute to the generation of the antiviral activity, more than that of its closed form. The work of this problem will be reported in the near future.

Experimental

3-Thioxo-6-aryl-3,4-dihydro-as-triazin-5(2H)-one—A mixture of 0.02 mole of α -ketocarbonic acid thiosemicarbozone and a solution of 2.7 g. (0.02 mole) of K_2CO_3 dissolved in 15 ml. of H_2O was boiled for $2\sim3$ hr. After cooling, the reaction mixture was acidified with AcOH and the precipitate was recrystallized from H_2O -EtOH.

3-Methylthio-6-aryl-as-triazin-5(4H)-one—A solution of EtONa, 0.01 mole of thioxotriazinone derivative and 0.013 mole (1.8 g.) of CH_3I was refluxed for $2\sim3$ hr. After removal of the solvent by distillation, the residue was recrystallized from H_2O -EtOH.

¹⁰⁾ M. Muraoka, A. Takada, T. Ueda: Keio J. Med., 11, 95 (1962).

6-Aryl-as-triazine-3,5(2H,4H)-dione—To a solution of 0.02 mole of 3-thioxotriazinone derivative dissolved in 15 ml. of N NaOH, $5\% K_2 M n_2 O_7$ solution was added dropwise until no more color of $K_2 M n_2 O_7$ disappeared, and the mixture was warmed for 15 min. on a water bath. The cooled reaction mixture was filtered, the filtrate was acidified with AcOH and the precipitate was recrystallized from EtOH.

The authors express their deepest thanks to U.S. Army Research and Development Group (Far East) for the kind support to this study.

Summary

Twelve compounds of phenylpyruvic acid thiosemicarbazone derivatives substituted with alkyl group, hydroxy group, alkoxy group or halogen atom at 2, 3- and/or 4-position in benzene ring, nine compounds of 3-thioxo-6-aryl-3,4-dihydro-as-triazin-5(2H)-one, five compounds of 3-methylthio-6-aryl-as-triazin-5(4H)-one, and five compounds of 6-aryl-as-triazine-3,5(2H,4H)-dione were newly synthesized.

Among these compounds, 3,4-dimethoxyphenylpyruvic acid thiosemicarbazone showed remarkable activity on the multiplication of the Mahoney strain of polio virus type 1, while 6-phenyl-, and 6-(p-propoxybenzyl)-3,4-dihydro-as-triazin-5(2H)-one, and 6-(p-chlorobenzyl)-as-triazine-3,5(2H,4H)-dione were slightly effective on the virus by using $100 \times \text{TCID}_{50}$ of the virus.

(Received February 9, 1963)

(Chem. Pharm. Bull.) 11 (12) 1556 ~ 1563 UDC 547.538.2.07

241. Issei Iwai and Tetsuo Hiraoka: Studies on Acetylenic Compounds. XXXIV.*1 Rearrangement of Propargylammonium Halide Derivatives.

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In the preceding paper, 1) it was reported that propargylammonium salts derivatives react with the active methylene group to afford higher homolog of acetylenic compound involving the loss of trialkylamine.

At that time, it was shown that C-N bond of propargylammonium iodide ($RC \equiv CCH_2- \stackrel{\oplus}{N} \Leftarrow$) is rather stable against cleavage than that of benzylammonium iodide ($PhCH_2- \stackrel{\oplus}{N} \Leftarrow$). If the cleavage reaction of propargylammonium and benzylammonium derivatives proceeds through propargyl cation ($RC \equiv CCH_2$) and benzyl cation ($PhCH_2$), phenyl propargyl cation should be more stable than benzyl cation in accordance with the first order approximation because carbonium cation of $PhC \equiv CCH_2$ is stabilized by the distribution

^{*1} Part XXXIII This Bulletin, 11, 1049 (1963).

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¹⁾ I. Iwai, T. Hiraoka: This Bulletin, 10, 81 (1962).