(Chem. Pharm. Bull.) 11 (8) 1037 ~ 1041)

UDC 547.92.07

176. Hiroko Hasegawa, Shigeo Nozoe, and Kyosuke Tsuda: Stereochemistry of Steroids containing Aromatic A-Ring. II.**1,1) Reaction of $9\alpha,11\alpha$ -Epoxyestrone.**2

(Institute of Applied Microbiology, University of Tokyo*3)

The previous paper¹⁾ of this series reported that the epoxide (III) obtained by the oxidation of 3-hydroxyestra-1,3,5(10),9(11)-tetraen-17-one acetate (I) with perbenzoic acid is identical with the epoxide obtained by alkali treatment of the bromohydrin (II) derived from I and that this substance is 9α , 11α -epoxide.

The present paper describes the steric structure of the estratriol, 1) obtained during the reductive cleavage of this epoxide (\mathbb{II}) with lithium aluminum hydride, as well as the configuration of hydrogen atom in 9-position of the 3-hydroxyestra-1,3,5(10)-triene-11,17-dione 1) obtained during the treatment of \mathbb{II} with acid or alkali, and the stability of the B/C ring junction in the steroids containing aromatic A-ring.

The reduction of the epoxide (III) with lithium aluminum hydride in tetrahydrofuran gave estratriol¹⁾ (IVa), m.p. 250°, which on acetylation formed triacetate (IVc), proving the compound (IVa) to be 3,11,17-triol. Methylation of IVa with dimethyl sulfate afforded methyl ether (IVb), m.p. $145\sim146^\circ$. Magerlein and $Hogg^2$) have synthesized estra-1,3,5(10)-triene-3,11 α , 17β -triol³⁾ from methyl 11α -hydroxy-3-oxopregna-1,4,17 (20)-cistrien-21-oate²⁾ by its pyrolysis, followed by ozonization and then reduction, whose methyl ether had m.p. $144\sim145^\circ$. Since they had the same melting points, thus in order to confirm their identity, an oxidation of methyl ether (IVb) was attempted with chromic anhydride in pyridine, which gave two products, 11,17-diketone (Vb) and 11-hydroxy-17-one (VIb). On the other hand, 11β -hydroxyestrone methyl ether prepared by Hogg's procedure²⁾ was oxidized with chromic anhydride in pyridine to a compound which was identical with Vb by the mixed melting point and comparison of their infrared spectra.

Since the inversion of the configuration at the α -position of carbonyl group has been now considered not to occur under oxidative condition with chromic anhydride in pyridine, the hydrogen atom at 9-position of IVa has α -orientation.

In general, the mechanism⁴⁾ of the reductive cleavage of epoxide ring with lithium aluminum hydride has been considered to be $S_{\rm N}2$ type reaction. However, the formation of a compound with 11α -hydroxy and 9α -hydrogen configuration by the cleavage of 9α , 11α -epoxide with lithium aluminum hydride is thought to be due to the attack of the hydride from the α -side at 9-position through a stabilized benzyl cation.

Partial acetylation of the triol (IVa) afforded a 3-monoacetate (IVd), whose oxidation with chromic anhydride in pyridine gave 11α -hydroxyestrone 3-acetate (VId).

Treatment of the epoxide (III) with dry hydrogen chloride gas in chloroform afforded the corresponding chlorohydrin (VIIa), whose acetylation formed a diacetate (VIIb), suggesting the structure of 9ξ -chloro- 11α -hydroxy compound. On the other hand, the

^{*1} This paper constitutes Part XLII of a series entitled "Steroid Studies" by K. Tsuda.

^{*2} This paper will constitute a part of the dissertation to be submitted by H. Hasegawa in partial fulfilment of the requirements for the Doctor's degree.

^{*3} Yayoi-cho, Bunkyo-ku, Tokyo (長谷川弘子, 野副重男, 津田恭介).

¹⁾ Part XLI. K. Tsuda, S. Nozoe, Y. Okada: This Bulletin, 11, 1022 (1963).

²⁾ E.G. Magerlein, J.A. Hogg: J. Am. Chem. Soc., 80, 2220 (1958).

³⁾ A. Bowers, J.S. Mills, C. Casas-Campillo, C. Djerassi: J. Org. Chem., 27, 361 (1962).

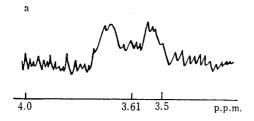
⁴⁾ R.E. Parker, N.S. Isaacs: Chem. Revs., 737 (1959).

1038 Vol. 11 (1963)

same treatment in methanol did not give a chlorohydrin but formed a phenol (IXa),¹⁾ which possessed no chlorine. Since infrared spectrum of this product showed the absorption of five- and six-membered ring ketones, it seemed that a fission of the epoxide ring^{5,6)} took place to form a 11-keto compound.

In order to confirm this point, 11β -hydroxyestrone²⁾ (Wa) was oxidized with Killiani's reagent to afford 3-hydroxyestra-1,3,5(10)-triene-11,17-dione (Va), which was clearly different from the product (IXa). Treatment of diketone (Va) with potassium hydroxide in methanol gave the compound (IXa), which indicated that with the alkaline treatment the hydrogen at 9-position of Va was converted to β -configuration. Thus the structure of IXa was assigned as 3-hydroxy-9 β -estra-1,3,5(10)-triene-11,17-dione. Treatment of epoxide (III) with alkali in methanol also gave IXa. The methylation of IXa with diazomethane afforded the amorphous methyl ether.

The nuclear magnetic resonance spectra of C-9 hydrogens of Va and IXa were shown in Fig. 1.*4 The coupling constants⁷⁾ between the C-9 hydrogens and the C-8 hydrogens of Va and IXa, 9 c.p.s. and 3 c.p.s., respectively, agree with the B/C-trans juncture of Va and with the B/C-cis juncture of IXa.



4.0 3.73 3.5 p.p.m.

Fig. 1. Nuclear Magnetic Resonance spectra of C-9 Hydrogens

a: 3-Hydroxyestra-1,3,5(10)triene-11,17-dione (Va)

b: 3-Hydroxy-9\beta-estra-1,3,5(10)-triene-11,17-dione (IXa)

In general, the steroid skeleton is the most stable when its ring juncture has the *trans-anti-trans* conformation. Elks, *et al.*,⁸⁾ however, have reported that the hydrolysis of 1-methyl-3,11-diacetoxy-1,3,5(10),9(11)-tetraene compound gave a mixture of the two C-11 carbonyl compounds, in which the isomer possessing B/C-*cis* juncture was main product.

The easy conversi on of 9α -H of 3-hydroxyestra-1,3,5(10)-triene-11,17-dione (Va) to 9β -H (IXa) clearly shows that even this compound which does not possess CH₃ group at C-1 position and has B/C-cis juncture is more stable than B/C-trans juncture.

^{*4} The NMR spectra were measured in pyridine with tetramethylsilane as internal standard, using a Varian A-60 instrument. All chemical shifts were reported in p.p.m. as δ-value.

⁵⁾ R.C. Cookson, J. Hudec: Proc. Chem. Soc., 24 (1957).

⁶⁾ H. Linde, K. Meyer: Helv. Chim. Acta., 42, 807 (1959).

⁷⁾ M. Karplus: J. Chem. Phys., 11 (1959).

⁸⁾ J. Elks, J.F. Oughton, L. Stephenson: Proc. Chem. Soc., 7 (1959).

Experimental*5

Cahrt 1.

Estra-1,3,5(10)-triene-3,11α,17β-triol (IVa)—To a solution of 900 mg. of 3-acetoxy-9α,11α-epoxy-estra-1,3,5(10)-triene-17-one (III) in 60 ml. of tetrahydrofuran was added 900 mg. of LiAlH₄. The mixture was refluxed for 24 hr. and after cooling, the excess reagent was decomposed by the addition of Me₂CO. The solvent was removed by distillation and 5% HCl was added to the residue. The resulting precipitate was filtered to give 513 mg. of IVa, m.p. 240~243° (decomp.). The analytical sample was prepared with recrystallization from Me₂CO, m.p. 249~250°, [α]_D -48° (c=1.09). UV: λ_{max} 279 mμ (ε 1930). IR ν_{max} cm⁻¹: 3405, 3190, 1620, 1591, 1500. Anal. Calcd. for C₁₈H₂₄O₃: C, 74.97; H, 8.39. Found: C, 74.88; H, 8.38.

3-Methoxyestra-1,3,5(10)-triene-11 α ,17 β -diol (IVb)—To a stirred (solution of 200 mg. of IVa in 15 ml. of EtOH was added 0.75 ml. of NaOH solution (10 g. of NaOH in 15 ml. of H₂O) and then 1 ml. of dimethyl sulfate dropwisely. After repeating the addition of NaOH solution and dimethyl sulfate three times, the reaction mixture was stirred for 15 min. The solvent was removed, washed with H₂O and dried to give 150 mg. of material. Recrystallization from Et₂O gave 90 mg. of IVb, m.p. $145\sim146^{\circ}$, [α]_D -44° (c=1.00). UV λ_{max} m μ (ϵ): 277 (1600), 283 (1470). IR ν_{max} cm⁻¹: 3545, 3280 \sim 3180, 1612, 1577, 1497. Anal. Calcd. for C₁₉H₂₆O₃: C, 75.46; H, 8.67. Found: C, 75.54, H, 8.60.

3-Methoxyestra-1,3,5(10)-triene-11,17-dione (Vb) and 3-Methoxy-11 α -hydroxyestra-1,3,5(10)-triene-17-one (VIb)—To a solution of 150 mg. of IVb in 2 ml. of pyridine was added 1.5 ml. of pyridine-CrO₃ complex (containing 150 mg. of CrO₃) and the mixture was allowed to stand at room temperature for 3 hr. It was poured into ice water and extracted with Et₂O. The ethereal solution was washed successively with 5% HCl, saturated Na₂CO₃ solution and H₂O. The extract was dried and evaporated to

^{*} All melting points were uncorrected. Optical rotations were measured in dioxane. Infrared spectra were obtained with Nujol and ultraviolet spectra were measured in methanol.

leave a solid residue. The residue was triturated with Et_2O and Et_2O soluble and insoluble portions were separated. Recrystallization of insoluble part from Me_2CO afforded 9 mg. of V b, m.p. 185° (soften), 200° (decomp.). This compound was identified by mixture melting point and infrared spectral comparison with V b, prepared by oxidation of VIIb.

The Et₂O soluble part was chromatographed over 3 g. of silica gel and eluted with Et₂O-benzene (1:99 \sim 1:2) to give 63 mg. of crude VIb. Recrystallization from Et₂O afforded 29 mg. of the analytical sample, m.p. 153 \sim 156°, [α]_D +24°(c=1.00). UV λ_{max} m μ (ϵ): 277 (1880), 283 (1680). IR ν_{max} cm⁻¹: 3540, 1730, 1608, 1583, 1500. *Anal.* Calcd. for C₁₉H₂₄O₃: C, 75.97; H, 8.05. Found: C, 76.03; H, 8.04.

3-Methoxyestra-1,3,5(10)-triene-11,17-dione (Vb) from VIIb—To a solution of 100 mg. of 3-methoxy-11\$\beta\$-hydroxyestra-1,3,5(10)-trien-17-one (VIIb)\$^2\$) in 2 ml. of pyridine was added 1 ml. of pyridine CrO₃ complex (containing 100 mg. of CrO₃) and the mixture was allowed to stand at room temperature for 3 hr. The reaction mixture was poured into ice water and extracted with Et₂O. The ethereal solution was washed successively with HCl, saturated NaHCO₃ solution and H₂O, and dried over anhyd. Na₂SO₄. The solvent was removed and the resulting residue was recrystallized from Me₂CO three times to give 8 mg. of V b, m.p. 184° (softed) 201° (decomp.). UV λ_{max} m μ : 277 (\$\epsilon\$ 1880), 284 (shoulder). IR ν_{max} cm⁻¹: 1743, 1715, 1612, 1582, 1502. Anal. Calcd. for C₁₉H₂₂O₃: C, 76.48; H, 7.43. Found: C, 76.25; H, 7.45.

Estra-1,3,5(10)-triene-3,11 α ,17 β -triol Triacetate (IVc)—A solution of 100 mg. of IVa in 0.5 ml. of pyridine and 0.5 ml. of Ac₂O was allowed to stand at room temperature for 15 hr. and then poured into ice water. The resulting precipitate was filtered and dried. Recrystallization from Et₂O gave 78 mg. of IVc, m.p. $132\sim135^{\circ}$, $(\alpha)_D -110^{\circ}(c=1.12)$. IR ν_{max} cm⁻¹: 1755, 1728. Anal. Calcd. for C₂₄H₃₀O₆: C, 69.54; H, 7.30. Found: C, 69.38; H, 7.26.

Estra-1,3,5(10)-triene-3,11α,17β-triol 3-Acetate (IVd)——A solution of 157 mg. of IVa in 2 ml. of pyridine and 76 mg. of Ac₂O was allowed to stand at room temperature for 15 hr. The mixture was poured into ice water and extracted with Et₂O. The ethereal solution was washed successively with 5% HCl, saturated NaHCO₃ solution and H₂O. The extract was dried over anhyd. Na₂SO₄ and the solvent was removed by distillation to give 147 mg. of solid. Recrystallization afforded IVd, m.p.154~155°. UV λ_{max} mμ (ε): 266 (536), 274 (445). IR ν_{max} cm⁻¹: 3300, 1763, 1583, 1493. Anal. Calcd. for C₂₀H₂₀O₄: C, 72.70; H, 7.93. Found: C, 72.82; H, 7.89.

3,11 α -Dihydroxyestra-1,3,5(10)-trien-17-one 3-Acetate (VId)—To a 0.5 ml. of pyridine-CrO $_3$ complex (containing 50 mg. of CrO $_3$) was added a solution of 50 mg. of IVd in 1.5 ml. of pyridine. The mixture was allowed to stand at room temperature for 3.5 hr., then poured into ice water and extracted with Et $_2$ O. The ethereal solution was washed successively with 5% HCl, saturated NaHCO $_3$ solution and H $_2$ O. The extract was dried over anhyd. Na $_2$ SO $_4$ and the solvent was removed. The residue was recrystallized from Et $_2$ O to give 14 mg. of VId, m.p. 156 \sim 158°. UV: λ_{max} 266 m μ (ϵ 565). IR ν_{max} cm $^{-1}$: 3520, 1768, 1740, 1603, 1583, 1488. Anal. Calcd. for C $_{20}$ H $_{24}$ O $_4$: C, 73.14; H, 7.37. Found: C, 73.29; H, 7.41.

9\$-Chloro-3,11a-dihydroxyestra-1,3,5(10)-trien-17-one 3-Acetate (VIIIa)—To a solution of 300 mg. of III in 6 ml. of CHCl₃ was introduced dry HCl gas at -7° for 2 hr. During this time, the solution changed its color to light yellow. The CHCl₃ solution was washed with cold saturated NaHCO₃ solution, and then H₂O. The solution was dried over anhyd. Na₂SO₄ and the solvent was removed in vacuo to give 330 mg. of powder. Recrystallization from Et₂O gave 83 mg. of WIa, m.p. 115° (soften), 133° (decomp.). IR ν_{max} cm⁻¹: 3550, 1753, 1736, 1613, 1585, 1493. Anal. Calcd. for C₂₀H₂₃O₄Cl: C, 66.20; H, 6.39. Found: C, 66.24; H, 6.44

9\$-Chloro-3,11 α -dihydroxy-1,3,5(10)-trien-17-one Diacetate (VIIIb)—A solution of 70 mg. of chlorohydrin (Wa) in 1 ml. of pyridine and 1 ml. of Ac₂O was allowed to stand at room temperature for 24 hr., then poured into ice water. The resulting precipitate was filtered to afford 34 mg. of crude material, m.p. 137~159°. Recrystallization from Et₂O gave 20 mg. of Wb, m.p. 149~159°. IR $\nu_{\rm max}$ cm⁻¹: 1766, 1747, 1613, 1587, 1493. *Anal.* Calcd. for C₂₂H₂₅O₅Cl: C, 65.26; H, 6.22; Cl, 8.76. Found: C, 65.36; H, 6.26; Cl, 8.94.

3-Hydroxy-9β-estra-1,3,5(10)-triene-11,17-dione (IXa)—To a solution of 1.2 g. of $\mathbb H$ in 25 ml. of MeOH was introduced dry HCl gas at -5° for 1.5 hr. A cooled saturated NaHCO₃ solution was added to the methanolic solution to give precipitate. Filtration of it gave 840 mg. of IXa, m.p. $201\sim205^\circ$. Recrystallization from Et₂O prepared the analytical sample, m.p. $204\sim207^\circ$ (decomp.), $[\alpha]_D$ +282° (c= 1.07). UV: λ_{max} 282 m $_{\mu}$ (ε 2070). IR ν_{max} cm⁻¹: 3360, 1730, 1700, 1611, 1502. Anal. Calcd. for C₁₈H₂₀O₃: C, 76.03; H, 7.09. Found: C, 75.93; H, 6.88.

3-Hydroxyestra-1,3,5(10)-triene-11,17-dione (Va)—To a solution of 500 mg. of 3,11 β -dihydroxyestra-1,3,5(10)-trien-17-one (VIa) prepared by the method of Hogg and Magerlein²⁾ in 20 ml. of Me₂CO was added dropwise 1.03 ml. of Kiliani solution (530 mg. of CrO₃ in 4.0 ml. of H₂O and 0.45 ml. of conc. H₂SO₄) at room temperature. The reaction mixture was poured into H₂O, and extracted with Et₂O. The ethereal solution was washed with H₂O and dried over anhyd. Na₂SO₄. Evaporation and recrystallization from Me₂CO afforded Va, m.p. 199~203°. [α]_D +374°(c=0.97). UV λ max m μ : 279 (ϵ 1950),

285 (shoulder). IR $\nu_{\rm max}$ cm⁻¹: 3350, 1740, 1715, 1610, 1510. Anal. Calcd. for $C_{18}H_{20}O_3$: C, 76.03; H, 7.09. Found: C, 76.08; H, 7.08.

3-Hydroxyestra-1,3,5(10)-triene-11,17-dione Acetate (Vd)—To a solution of 200 mg. of $11 \mbox{\ensuremath{\it E}}$ -hydroxyestrone 3-acetate (VId)²⁾ in 10 ml. of Me₂CO was added 0.37 ml. of Kiliani solution. After addition of small amount of aq. NaHCO₃, the mixture was poured into H₂O. Filtration and recrystallization from Me₂CO gave 50 mg. of Vd, m.p. $156\sim161^{\circ}$, (α)_D +326° (c=1.03). UV $\lambda_{\rm max}$ m μ (ϵ): 267 (609), 274 (580). IR $\nu_{\rm max}$ cm⁻¹: 1761, 1738, 1714, 1606, 1583, 1496. *Anal.* Calcd. for C₂₀H₂₂O₄: C, 73.60; H, 6.79. Found: C, 73.56; H, 6.81.

3-Methoxy-9 β -estra-1,3,5(10)-triene-11,17-dione (IXb)—To a solution of 300 mg. of 3-hydroxy-9 β -estra-1,3,5(10)-triene-11,17-dione (IXa) in 30 ml. of MeOH was added an excess of solution of diazomethane in Et₂O and the mixture was allowed to stand at room temperature for 14 hr. Removal of the solvent afforded 293 mg. of amorphous product. It was chromatographed over silica gel 12 g. and eluted with petr. ether-benzene (3:7) to give 100 mg. of amorphous product (IXb). UV λ_{max} m μ (ϵ): 279.7 (1830), 286.8 (1610). IR ν_{max}^{CHCL} cm⁻¹: 1744, 1710, 1610, 1577. Anal. Calcd. for $C_{19}H_{22}O_3$: C, 76.48; H, 7.43. Found: C, 76.12; H, 7.39.

Isomerization of Va to IXa—A solution of 105 mg. of Va in 15 ml. of 1% MeOH-KOH was refluxed for 70 min. under nitrogen atmosphere. The solution was neutralized with AcOH and MeOH was removed to give precipitate. The filtration of the precipitate afforded 83 mg. of material, m.p. $192\sim199^{\circ}$. Recrystallization from Et₂O gave 51 mg. of IXa. The result of the thin layer chromatography with elution solvent, benzene-Me₂CO (9:1), showed that the mother liquor contained only a little of Va.

Alkaline Treatment of IXa—A solution of 30 mg. of IXa in 5 ml. of 1% MeOH-KOH was refluxed 70 min. The solution was neutralized with AcOH and the solvent was removed. The resulting precipitate was filtered to give 13 mg. of starting material, m.p. $205\sim208^{\circ}$. The result of the thin layer chromatography with above solvent system showed that Va was not contained in it.

The authors express their gratitude to Dr. K. Takeda, Director of Research Laboratory, Shionogi & Co., Ltd., and his staff members for the measurement of the Nuclear magnetic resonance spectra. The authors are also grateful to Dr. Y. Kawazoe, National Cancer Center, for his assistance in analytical measurement of the nuclear magnetic resonance spectra. The authors are indebted to Misses H. Yamanouchi and K. Hayashi for the elemental analyses, and Miss K. Arimoto for the infrared spectra measurement

Summary

The reduction of 3-hydroxy- 9α , 11α -epoxyestra-1,3,5(10)-trien-17-one (III) with lithium aluminum hydride gave estra-1,3,5(10)-triene-3,11 α ,17 β -triol. The reaction of III with hydrogen chloride in methanol yielded 3-hydroxy- 9β -estra-1,3,5(10)-triene-11,17-dione (IXa). On the other hand, the reaction in chloroform afforded 9ξ -chloro-3,11 α -dihydroxyestra-1,3,5(10)-triene-17-one (IIIa). With alkali 3-hydroxyestra-1,3,5(10)-triene-11,17-dione (Va) is converted into 9β -isomer (IXa).

(Received March 20, 1963)