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# 166. Tsutomu Momose, Yoshiko Yano (née Mukai), and Katsuko Ohashi: Organic Analysis. XLIV.\*1 A New Deproteinizing Agent for Determination of Blood Sugar.

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Several deproteinizing agents are now available for the determination of blood sugar. Those involve zinc sulfate and sodium hydroxide, 1) sodium tungstate and sulfuric acid, 2) and zinc sulfate and barium hydroxide. 3) These reagents consist of two components, respectively, which should exactly match with each other in a definite ratio, and should carefully be prepared, preserved and used. Among all, zinc sulfate and barium hydroxide (Somogyi's reagent) is preferable to give the "true sugar value" in blood, yet a more simple reagent is desirable for wide use in clinical tests. In the writers' laboratory, sodium tungstate and alum proved to fit for the purpose in the determination of blood sugar with 3,6-dinitrophthalic acid as the color developing agent.

## Experimental

#### Reagents

Sodium Tungstate Solution— $10.0 \,\mathrm{g}$ . of sodium tungstate, Na<sub>2</sub>WO<sub>4</sub>·2H<sub>2</sub>O, is dissolved in sufficient H<sub>2</sub>O to measure  $100 \,\mathrm{ml}$ .

Alum Solution—9.6 g. of alum,  $KA1(SO_4)_2 \cdot 12H_2O$ , is dissolved in sufficient  $H_2O$  to meaure 100 ml. The color developing agents<sup>4)</sup> are modified as follows.

3,6-Dinitrophthalic Acid Solution—1.50 g. of 3,6-dinitrophthalic acid monopyridinium salt (Ishizu Seiyaku Kabushiki Kaisha) is dissolved in sufficient  $H_2O$  to measure 500 ml., and stored in a light-resistant bottle, to which an automatic burette, 10 ml., is attached.

Alkaline Solution—125 g. of anhyd.  $K_2CO_3$  and 25.0 g. of sodium thiosulfate are successively dissolved in  $H_2O$  and made up to 500 ml., stored in a bottle, to which an automatic burette, 10 ml., is attached, and protected from  $CO_2$ .

Standard Procedure for Clinical Tests—0.100 ml. of blood is haemolysed with 2.90 ml. of  $H_2O$  in a test-tube, and 0.50 ml. of sodium tungstate solution is added. To the mixture, 0.50 ml. of alum solution is added, and the test-tube shaken vigorously. The mixture is then transferred to a centrifuge tube, stoppered with a piece of aluminium foil, and centrifuged. 2.00 ml. of the supernatant solution is pipetted into a test-tube, 1.00 ml. of 3,6-dinitrophthalic acid solution and 1.00 ml. of alkaline solution are successively added, developed, and diluted in the same way as in the previous paper.<sup>4)</sup> The absorption intensity is measured at 450 m $\mu$  with the reagent blank, and the blood sugar value is obtained from the calibration curve which is prepared by developing 2.00 ml. of standard glucose solutions,  $10 \sim 100 \, \gamma/\text{ml}$ . The deproteinizing operation should be avoided for glucose solutions.

In this study, blood was first diluted to an appropriate volume with  $H_2O$ , and a definite volume of the solution was pipetted which corresponded to 0.100 ml. of the blood to eliminate the error caused by using small pipettes. The pipetting of reagents and sample solutions was carried out by the rapid automatic pipettes.<sup>5)</sup>

## Results and Discussion

Aqueous sodium tungstate and alum precipitate very slightly soluble aluminium tungstate. (6) This compound removes proteins in blood in the presence of sulphate.

<sup>\*1</sup> Part XLII: Yakugaku Zasshi, 83, 724 (1963).

<sup>\*2</sup> Katakasu, Fukuoka (百瀬 勉, 矢野(旧姓向井)良子, 大橋勝子).

<sup>1)</sup> H.C. Hagedorn, B.N. Jensen: Biochem. Zschr., 135, 46 (1923).

<sup>2)</sup> O. Folin, H. Wu: J. Biol. Chem., 38, 81 (1919).

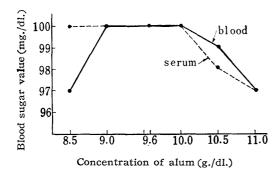
<sup>3)</sup> M. Somogyi: Ibid., 160, 69 (1945).

<sup>4)</sup> T. Momose, A. Inaba, Y. Mukai, M. Watanabe: Talanta, 4, 33 (1960).

<sup>5)</sup> T. Momose: Japan Analyst, 11, 688 (1962).

<sup>6)</sup> J. W. Mellor: Comprehensive Treatise On Inorganic And Theoretical Chemistry (Longmans, Green and Co.), 11, 789.

When the concentration of sodium tungstate is fixed as 10 g./dl., the equivalent concentration of alum is 9.6 g./dl. to match with each other in the same volume. In the deproteinization of blood or serum, however, the concentration of alum can be varied from 9 to 10 g./dl. for blood and from 8.5 to 10 g./dl. for serum to give the same blood sugar value as shown in Fig. 1. This large allowance proves the easiness of preparation and preservation of the reagents. Furthermore, it is important to note that there may be no possibility of failure in the deproteinization in clinical tests.



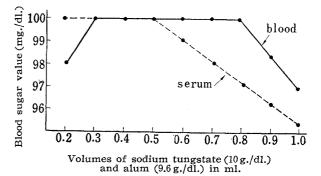


Fig. 1. Relationship between Blood Sugar Value and the Concentration of Alum which corresponds to Sodium Tungustate Solution (10 g./dl.) Blood sugar value is converted into 100.

Fig. 2. Relationship between Blood Sugar Value and the Volume of Deproteinizing Agent for 0.1 ml. of Blood or Serum. Blood sugar value is converted into 100.

The volumes of sodium tungstate and alum solutions which need for deproteinizing 0.1 ml. of blood or serum are shown in Fig. 2. A constant sugar value is given in the range of 0.3 to 0.8 ml. of the reagents for blood, and a little smaller volumes for serum. Therefore, each 0.5 ml. of the solutions was used in the standard method. The pH value of the deproteinized solution is about 4.7, and gives no effect on the color development.

A recovery test was carried out by adding known amount of glucose to blood solutions. Table I shows that the found values are in good agreement with the added amount. The mean value of recovery in per cent, 99.5, proves that the deproteinizing agent adsorbs no glucose in the reaction.

Table I. Recovery Test (Mean value of 4 determinations)

Initial blood sugar value (mg./dl.)	Glucose added (mg./dl.)	Total glucose found (mg./dl.)	Glucose recovered (%)
ſ	50	149	98
100 {	100	200	100
Į.	200	303	102
ſ	50	142	100
92 {	100	193	101
· ·	200	286	98
ſ	50	126	96
78 {	100	179	101
l	200	277	100
ſ	50	150	100
100 {	100	200	100
	200	295	98
ſ	50	163	100
113 {	100	211	98
l	200	313	100
		Mean	99 5%

Mean 99.5%

Now, blood sugar tests were carried out with the deproteinizing agent by adding some substances in blood solutions. Some other reducing substances other than glucose which are present in blood give practically no influence on the determination as shown in Table II. The amount of ascorbic acid in the table, which shows a little higher

Table II. Influence of Reducing Substances on Blood Sugar Value (Mean value of 4 determinations)

Initial blood	Blood sugar value found (mg./dl.)			
sugar value	Creatinine added	Creatine added	Glutathione added	Ascorbic acid added
(mg./dl.)	(10  mg./dl.)	(10  mg./dl.)	(10  mg./dl.)	(10  mg./dl.)
93	93	93	. 94	95

sugar value, is a large excess than the usual value in blood. The influence of some preservatives which may be added in blood is shown in Table III, and sodium fluoride only gives a lower value. This fact may be caused by the formation of soluble complex ion  $(AlF_{\theta})^{3-}$  in the deproteinized solution.

Table III. Influence of Preservatives on Blood Sugar Value (Mean value of 4 determinations)

Initial blood		Blood sugar value found (mg./dl.)			
	sugar value	Sodium citrate added	Sodium oxalate added	EDTA added	Sodium fluoride added
	(mg./dl.)	(5  mg./ml.)	(2  mg./ml.)	(1.5  mg./ml.)	(10  mg./ml.)
	153	153	152	153	146

To test the precision of this method, each 16 aliquots of a sample blood solution were deproteinized and developed for three times. All data are summarized in Fig. 3. The standard deviation of the test is 1.6 mg. for a mean sugar value of 188 mg./dl.

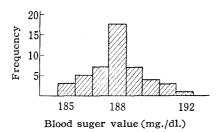


Fig. 3. Graph of Frequency for 48 Analyses by Standrad Method

The results of parallel tests with other deproteinizing agents are shown in Table IV. The present deproteinizing agent gives "true sugar values" which agree with Somogyi's reagent in an experimental error. The sugar values determined by the other methods than 3,6-dinitrophthalic acid are also shown in the table for reference.

# Micro-Method

The deproteinizing agent can successfully be used in a micro-method of determination of blood sugar.

### Reagents

Sodium tungstate solution—5 g./dl.

Alum solution—4.8 g./dl.

Color developing agent——The same volumes of 3,6-dinitrophthalic acid solution and alkaline solution under Standard method are mixed before use.

#### **Procedure**

0.020 ml. of blood is haemolysed with 0.58 ml. of  $H_2O$  in a centrifuge tube, and 0.20 ml. of sodium tungstate solution is added. To the mixture, 0.20 ml. of alum solution is added, and the tube shaken

	(Mean blood s	sugar value (mg., ui.,	or b determinations,	
Method of determination	3,6-Dinitrophthalic acid		Somogyi-Nelson	Hagedorn-Jensen
Deproteiniz- ing agent Blood	Sodium tungstate and alum	Barium hydroxide and zinc sulphate	Barium hydroxide and zinc sulphate	Sodium hydroxide and zinc sulphate
1	130	129	131	135
$\frac{1}{2}$	124	123	123	131
3	153	153	155	164
	203	205	206	212
4 5	203 76	203 77	79	91
			148	159
6	149	152	67	83
7	71	72		126
8	110	110	108	
9	75	77	77	94
10	173	169	173	192
11	93	90	<del></del>	$92^{a_{1}}$
12	89	87	<del></del>	$89^{a}$
Serum				
1	68	67	68	—
2	88	90	89	<del></del> ,
3	69	68	69	<del></del> ,
4	77	79	81	
5	84	84	84	· —.
6	110	113	116	<del></del> .
7	78	77	75	<del></del> ,
8	94	93	97	

Table IV. Parallel Test with Other Methods and Other Deproteinzing Agents (Mean blood sugar value (mg./dl.) of 3 determinations)

vigorously. The tube is stoppered with a piece of aluminium foil, and centrifuged. 0.50 ml. of the supernatant clear solution is pipetted into a small test tube of about  $11 \times 120$  mm., 0.50 ml. of the color-developing solution is added, and mixed well. The test tube is packed in a proper heating basket with the reagent blank, and heated in a boiling water bath for exactly 10 min. After cooling in running water, the mixture is diluted to 5.0 ml., and its absorption intensity is measured at 450 m $\mu$  with the reagent blank. The blood sugar value is read from a calibration curve which is prepared by developing 0.50 ml. of glucose solutions,  $10 \sim 100 \, \gamma/\text{ml}$ .

The precision of this method was examined by developing each 16 aliquots of a blood solution for three times. The data are summarized in Fig. 4. The standard deviation, 2.0 mg. for a mean blood sugar value of 190 mg./dl., shows that the method is accurate enough for usual clinical tests.

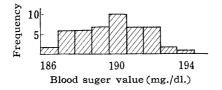


Fig. 4. Graph of Frequency for 48 Analyses by Micro-Method

# Powdered Deproteinizing Agent

The new deproteinizing agent can also be used in a powdered state as follows.

#### Reagent

To a hot solution of 50 g. of sodium tungstate in 200 ml. of  $H_2O$ , a hot solution of 48 g. of alum in 250 ml. of  $H_2O$  is added in one portion, and the mixture is heated on a water bath. 25 g. of  $K_2SO_{44}$  is added in the mixture, and the mixture evaporated to dryness. The white mass is then dried at  $100 > 105^\circ$  for 5 hr., powdered, and shifted with a 100-mesh sieve. It is stable in the air.

# Procedure

 $0.100 \,\mathrm{ml}$ . of blood is heamolysed with 3.90 ml. of  $\mathrm{H_2O}$  in a test-tube, about 0.1 g. of the deproteinizing agent is added, and the test-tube shaken vigorously. The mixture is then transferred in a cen-

a) Deproteinized with barium hydroxide and zinc sulphate

trifuge tube, centrifuged, and treated in the same way as in the standard method. For deproteinizing 0.100 ml. of serum, about 0.05 g. of the reagent is used.

The powdered deproteinizing agent gives the same blood sugar value when used in the range of 0.08 to 0.12 g. for 0.1 ml. of blood, and 0.04 to 0.06 g. for 0.1 ml. of serum. The blood sugar values obtained by this method agree with those of the standard method as shown in Table V. The powdered reagent is very simple to use, and may sometimes be suitable for clinical tests.

Table V. Parallel Test of Powdered Deproteinizing Agent with the Standard Method (Mean blood sugar value (mg./dl.) of 4 determinations)

Blood		Serum		
Standard method	Powdered deproteinizing agent (0.1 g.)	Standard method	Powdered deproteinizing agent (0.05 g.)	
95	94	204	203	
94	94	83	84	
57	58	99	98	
151	149	127	128	
70	71	136	135	

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## Summary

Aqueous sodium tungstate and alum were successfully used in the deproteinization of blood or serum to determine the sugar value with 3,6-dinitrophthalic acid. They were easily prepared and preserved, and fitted for clinical tests. The deproteinizing agent could also be used in a powdered state.

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