(Chem. Pharm. Bull.) 13(11)1285~1289(1965)

UDC 547.924.07:615.76

## 163. Takuichi Miki, Kentaro Hiraga, and Tsunehiko Asako:

An Improved Synthesis of Estrogens. \*1,2

(Chemical Research Laboratories, Research & Development Division, Takeda Chemical Industries, Ltd.\*3)

Recently, Ananchenko and Torgov<sup>1)</sup> reported an excellent synthesis of p-homosteroid using a new reaction found by themselves. This prompted us to apply their method to the synthesis of estrogens. Around the time when our short communication2) was published, the similar synthetic studies were reported by several groups3) independently.

<sup>\*1</sup> This paper constitutes Part XXXIV of Takeda Laboratories' series entitled "Steroids"; Part XXXII: This Bulletin, 12, 1189 (1964).

<sup>\*\*</sup> Preliminary communication, T. Miki, K. Hiraga, T. Asako: Proc. Chem. Soc., 1963, 139.

\*\* Juso-nishino-cho, Higashiyodogawa-ku, Osaka (三木卓一, 平賀謙太郎, 朝子典彦).

1) a) S. N. Ananchenko, V. N. Leonov, A. V. Platonova, I. V. Torgov: Doklady Akad. Nauk S. S. S. R., **153**, 73 (1960). b) S. N. Ananchenko, V. Y. Limanov, V. N. Leonov, V. N. Rzheznikov, I. V. Torgov: Tetrahedron, 18, 1355 (1962).

<sup>2)</sup> T. Miki, K. Hiraga, T. Asako: Proc. Chem. Soc., 1963, 139.

<sup>3)</sup> a) D. J. Crispin, J. S. Whitehurst: Proc. Chem. Soc., 1963, 22. b) H. Smith, et al.: Experientia, 19, 394 (1963). c) T.B. Windholz, J.H. Fried, A.A. Patchett: J. Org. Chem., 28, 1092 (1963). d) S. N. Ananchenko, I. V. Torgov: Tetrahedron Letter, 1963, 1553.

1286 Vol. 13 (1965)

Here will be described on what characterizes our work: the stereospecific reduction and the optical resolution mainly.

Instead of 2-methyl-1,3-cyclohexanedione which was used for the synthesis of p-homosteroids,<sup>1)</sup> we treated 2-methyl-1,3-cyclopentanedione (II) with 1-vinyl-6-methoxy-1-tetralol (I).

This reaction proceeded smoothly in the presence of base such as Triton B or triethylenediamine, and the resulting product, dl-3-methoxy-8,14-secoestra-1,3,5(10),9-tetraene-14,17-dione ( $\mathbb{H}$ ) was then converted by heating at 120° with phosphorus pentoxide or by warming in methanolic hydrochloric acid to the known dl-3-methoxyestra-1,3,5(10),8,14-pentaen-17-one ( $\mathbb{N}$ ).

When the secodione (II) was heated with phosphorus pentoxide at a temperature higher than  $140^{\circ}$  or with a large excess of phosphorus pentoxide at  $120^{\circ}$ , dl-isoequilenin 3-methyl ether (III) was obtained together with a small amount of the unknown compound melting at  $132^{\circ}$ . The latter was assigned to 1-methyl-7-methoxy-2,3-dihydro-1H-cyclopenta[a]phenanthrene (IX), since it showed no carbonyl absorption in infrared spectrum and an ultraviolet absorption curve similar to that of cyclopentenophenanthrene.

On the other hand, when the secodione ( $\mathbb{II}$ ) was treated with silica gel or with dilute hydrochloric acid solution, there was obtained dl-3-methoxy-14-hydroxyestra-1,3,5(10),9-tetraen-17-one (X), which was found to be the intermediate compound in the reaction from  $\mathbb{II}$  to  $\mathbb{V}$ . The stereochemistry of the hydroxyketone (X) will be discussed in the separate paper.

In our synthesis, the pentaenone ( $\mathbb{N}$ ) was, first of all, reduced with sodium borohydride in methanol at a low temperature ( $0\sim-30^\circ$ ) to yield dl-3-methoxyestra-1,3,5(10), 8,14-pentaen-17 $\beta$ -ol ( $\mathbb{N}$ ). When the reaction was carried out at the higher temperature, an oily substance was produced.

Catalytic hydrogenation of the double bond at C-14 of the pentaenol (V) over Raney nickel or palladium charcoal proved to be more stereospecific than the same reaction on pentaenone ( $\mathbb{N}$ ), by yielding dl-3-methoxyestra-1,3,5(10),8-tetraen-17 $\beta$ -ol ( $\mathbb{N}$ )

5) G. A. Hughes, H. Smith: Chem. & Ind. (London), 1960, 1022.

<sup>4)</sup> a) J.W. Cook, C.L. Hewett: J. Chem. Soc., 1933, 1098. b) H. Hilemann: Ber., 69, 2610 (1936).

in good yield. The tetraenol  $(\mathbb{V})$  was further reduced with potassium in liquid ammonia to give dl-estradiol 3-methyl ether which was identical with the material from natural origin in infrared\*4 and ultraviolet spectrum.

The optical resolution was performed by means of the ester formation of  $17\beta$ -hydroxyl group with l-menthoxyacetyl chloride. The crystals of l-menthoxyacetate (X) of the pentaenol (V) which precipitated from benzene-methanol were mainly levorotatory isomer. Hydrolysis of this ester should be carried out at room temperature in methanolic potassium hydroxide solution, since the pentaenol was unstable in hot alkaline solution. The l-pentaenol (V),  $[\alpha]_{\rm p} - 126^{\circ}({\rm CHCl_3})$  thus obtained, was reduced, through the l-tetraenol (V),  $[\alpha]_{\rm p} - 3^{\circ}({\rm CHCl_3})$ , to d-estradiol 3-methyl ether (VI)  $[\alpha]_{\rm p} + 80^{\circ}({\rm CHCl_3})$ , which was identical with the material of natural origin in all respects.

Dehydrogenation reaction of tetraenol ( $\mathbb{V}$ ) with selenium dioxide in acetic acid or *tert*-butyl alcohol preceded smoothly to yield dl-dihydroequilenin 3-methyl ether ( $\mathbb{X}$ ), which was then converted to dl-equilenin 3-methyl ether ( $\mathbb{X}$ ) by chromic acid oxidation.

Another route investigated was to oxidize  $17\beta$ -hydroxyl group of  $\mathbb M$  with chromic acid. The product obtained was found to be dl-equilenin 3-methyl ether ( $\mathbb M$ ), indicating that chromic acid caused dehydrogenation at 6,7-position, unexpectedly, in addition to oxidation of  $17\beta$ -hydroxyl group. The dehydrogenation reaction by chromic acid was observed at 6,7-position of tetraenone ( $\mathbb M$ ) to afford dl-equilenin 3-methyl ether.

## Experimental\*5

dl-3-Methoxy-8,14-secoestra-1,3,5(10),9-tetraene-14,17-dione (III)——A solution of 0.5 g. of 1-vinyl-6-methoxy-1-tetralol (I) in 15 ml. of xylene was added to a stirred solution of 0.27 g. of 2-methyl-1,3-cyclopentanedione (II) and 0.05 g. of triethylenediamine in 30 ml. of xylene at the refluxing temperature. After stirring for 1 hr. at the refluxing temperature, the reaction mixture was concentrated to the half volume, cooled, worked up with ether. Some unchanged 2-methyl-1,3-cyclopentanedione (II) was separated by filtration and washed with ether. The combined ether solutions were washed with 5% KOH solution and H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The oily residue deposited 0.25 g. of a crystalline material which was recrystallized from EtOH to yield 0.2 g. of V, [m.p. 79~80°. UV  $\lambda_{\rm max}^{\rm EtOH}$  mμ (ε): 266 (13000). Anal. Calcd. for C<sub>19</sub>H<sub>22</sub>O<sub>3</sub>: C, 76.48; H, 7.43. Found: C, 76.58; H, 7.48.

dl-3-Methoxyestra-1,3,5(10),8,14-pentaen-17-one (IV)—a) A mixture of 0.3 g. of  $\blacksquare$  and 0.3 g. of P<sub>2</sub>O<sub>5</sub> was heated at 120° for 7 min. under reduced pressure. After cooling, the mixture was shaken with ether. The ether solution was washed with saturated NaHCO<sub>3</sub> solution and H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated to yield 0.25 g. of a crystalline material. Recrystallization from EtOH afforded an analytical sample, m.p. 108°. UV  $\lambda_{\max}^{\text{EtOH}}$  mµ ( $\epsilon$ ): 310 (22700). Anal. Calcd. for C<sub>19</sub>H<sub>20</sub>O<sub>2</sub>: C, 81.43; H, 7.14. Found: C, 81.55; H, 7.08.

b) To a hot solution of 1 g. of  $\mathbb{I}$  in 30 ml. of MeOH was added 1 ml. of 4N HCl and the solution was allowed to stand at room temperature overnight to deposit a crystalline substance. Filtration and recrystallization from EtOH yielded 0.8 g. of  $\mathbb{N}$ .

Cyclization of  $\mathbb I$  at 150°—A mixture of 2 g. of  $\mathbb I$  and 2 g. of  $P_2O_5$  was heated at 150° for 7 min. under reduced pressure. After cooling, the mixture was worked up with ether and the ether solution was washed with saturated NaHCO<sub>3</sub> solution and H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated to yield 1.3 g. of an oily material, which was chromatographed on florisil. Elution with benzene-petr. ether (1:4) gave 0.1 g. of colorless crystals, m.p. 132°. UV  $\lambda_{max}^{\text{BIOH}}$  mµ ( $\epsilon$ ): 262 (22500), 284 (shoulder), 292 (4500), 302 (2300). Anal. Calcd. for  $C_{19}H_{16}O$ : C, 86.98; H, 6.91. Found: C, 86.71; H, 6.98.

The portion eluted with benzene-petr. ether (1:1) gave 0.32 g. of dl-isoequilenin methyl ether, m.p. 130°. UV  $\lambda_{\max}^{\text{EiOH}}$  m $\mu$  ( $\epsilon$ ): 230 (23600), 265 (2800), 320 (480), 335 (630).

dl-3-Methoxy-14-hydroxyestra-1,3,5(10),9-tetraen-17-one ( $\dot{X}$ )—To a solution of 0.5 g. of II in 40 ml. of tetrahydrofuran was added 1 ml. of 2N HCl and the solution was allowed to stand at room temperature for 20 hr. The solution was poured into H<sub>2</sub>O, shaken with AcOEt, and the AcOEt layer was washed with NaHCO<sub>3</sub> solution and H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated to yield 0.45 g. of an oily residue,

<sup>\*4</sup> Measured in CHCl<sub>3</sub>.

<sup>\*5</sup> All melting points are uncorrected.

<sup>6)</sup> D. K. Banerjee, S. Chatterjee, C. N. Pillai, M. V. Bhatt: J. Am. Chem. Soc., 78, 3769 (1956).

which was chromatographed on alumina. Elution with ether gave 0.02 g. of X, m.p. 165°. UV  $\lambda_{\text{max}}^{\text{Etoful}}$  m<sub> $\mu$ </sub> ( $\epsilon$ ): 267 (18000). IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3450, 1740. *Anal.* Calcd. for  $C_{19}H_{22}O_3$ : C, 76.48; H, 7.43. Found: C, 76.53; H, 7.33.

- dl-3-Methoxyestra-1,3,5(10),8,14-pentaen-17β-ol (V)—To a solution of 1.6 g. of N in 100 ml. of MeOH was added 0.3 g. of NaBH<sub>4</sub> at  $-30^{\circ}$ . After stirring for 1 hr., the excess NaBH<sub>4</sub> was destroyed with 3 drops of AcOH and 500 ml. of H<sub>2</sub>O. The mixture was shaken with ether and the ether solution was washed with saturated NaHCO<sub>3</sub> solution and H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated to yield 1.5 g. of a crystalline material. Recrystallization from benzene-hexane afforded 1.3 g. of V, m.p. 113°. UV  $\lambda_{\rm max}^{\rm EOH}$  mμ (ε): 310 (22200). IR  $\nu_{\rm max}^{\rm Nujol}$  cm<sup>-1</sup>: 3300. Anal. Calcd. for C<sub>19</sub>H<sub>22</sub>O<sub>2</sub>: C, 80.85; H, 7.80. Found: C, 81.16; H, 7.88.
- dl-3-Methoxyestra-1,3,5(10),8-tetraen-17β-ol (VI)—A solution of 0.5 g. of VII in 50 ml. of dioxane was hydrogenated over 0.5 g. of Raney Ni at room temperature and atmospheric pressure. The mixture was filtered and solvent was evaporated to afford 0.4 g. of a crystalline material, which was recrystallized from EtOH-H<sub>2</sub>O to give colorless prisms, m.p. 126°. UV  $\lambda_{\text{max}}^{\text{BiOH}}$  mμ (ε): 278 (14100). Anal. Calcd. for  $C_{19}H_{24}O_{2}\cdot \frac{1}{3}H_{2}O$ : C, 78.65; H, 8.51. Found: C, 78.51; H, 8.66.
- dl-3-Methoxyestra-1,3,5(10)-trien-17 $\beta$ -ol (VII)—To a solution of 1 g. of VI in 50 ml. of dioxane, 90 ml. of ether and 150 ml. of liq. NH<sub>3</sub> was added 1.5 g. of K at  $-50^{\circ}$ . After stirring for 2 hr., the reaction mixture was neutralized with NH<sub>4</sub>Cl, and NH<sub>3</sub> was evaporated. The residue was shaken with 500 ml. of H<sub>2</sub>O and 300 ml. of ether, and the ether layer was washed with H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated to yield 0.8 g. of a crystalline material, which was recrystallized from EtOH-H<sub>2</sub>O to afford 0.5 g. of W, m.p. 130°. Anal. Calcd. for C<sub>19</sub>H<sub>20</sub>O<sub>2</sub>: C, 79.72; H, 9.09. Found: C, 80.22; H, 9.18.
- *l*-3-Methoxyestra-1,3,5(10),8,14-pentaen-17 $\beta$ -ol *l*-Menthoxyacetate (XI)—To a solution of 1.7 g. of V in 8.5 ml. of pyridine and 12 ml. of dioxane was added dropwise 4.8 g. of *l*-menthoxyacetyl chloride, then the reaction mixture was allowed to stand at room temperature overnight. The mixture was poured into ice water and extracted with ether. The ether solution, after washing with dil. HCl, saturated NaHCO<sub>3</sub> solution, and H<sub>2</sub>O successively, was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated to yield 4 g. of an oily residue, which deposited 1 g. of crystalline *l*-menthoxyacetate (X). Filtration and recrystallization from MeOH-ether afforded 0.9 g. of pure X, m.p. 61°.  $(\alpha)_D^{\text{20}} 142^{\circ} (c=1, \text{CHCl}_3)$ . UV  $\lambda_{\text{max}}^{\text{BiOH}} \text{ m}_{\mu}$  (ε): 310 (34000). Anal. Calcd. for C<sub>31</sub>H<sub>42</sub>O<sub>4</sub>: C, 77.82; H, 8.76. Found: C, 78.46; H, 8.43.
- *l*-3-Methoxyestra-1,3,5(10),8,14-pentaen-17 $\beta$ -ol (V)—Under cooling with ice, 20 ml. of 1N ethanolic KOH was added to a solution of 0.65 g. of X in 30 ml. of EtOH, then the solution was stirred for 40 min. at room temperature. To the mixture was added 100 ml. of H<sub>2</sub>O and the mixture was shaken with ether. The ether solution was washed with H<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated to yield 0.2 g. of *l*-VII, m.p. 55°.  $[\alpha]_{2}^{2}$  -126°(c=1, CHCl<sub>3</sub>).
- *l*-3-Methoxyestra-1,3,5(10),8-tetraen-17β-ol (VI)—Catalytic reduction at the 14 double bond of l-V was carried out in the same manner as in the case of racemic one, giving l-VI, m.p. 125°.  $(\alpha)_{D}^{24}$  -3° (c=0.78, CHCl<sub>3</sub>).
- d-3-Methoxyestra-1,3,5(10)-trien-17 $\beta$ -ol (VII)—Reduction of l- $\mathbb{N}$  with K in liq. NH<sub>3</sub> gave d-estradiol 3-methyl ether ( $\mathbb{N}$ ), m.p. 98°. ( $\alpha$ )<sup>24</sup> +80° (c=1, CHCl<sub>3</sub>).
- dl-3-Methoxyestra-1,3,5(10),6,8-pentaen-17 $\beta$ -ol (XI)—A solution of 0.1 g. of dl-3-methoxyestra-1,3,5(10),8-tetraen-17 $\beta$ -ol (VI), 0.05 g. of SeO<sub>2</sub> and 0.02 ml. of pyridine in 10 ml. of t-BuOH was heated at refluxing temperature for 18 hr. The Se deposited was filtered off and the filtrate was concentrated in vacuo to yield an oily material, which was chromatographed on alumina. The fraction eluted with benzene afforded 0.02 g. of a colorless material melting at 132°, the structure of which is not investigated yet and the fraction eluted with benzene-ether (7:3) afforded 0.03 g. of crude crystals of N, which were recrystallized from acetone-MeOH to give colorless plates, m.p. 178°. UV  $\lambda_{max}^{BiOH}$  mμ (ε): 233 (20140), 255 (6800), 265 (7100), 279 (3835), 291 (3770), 306 (3350).
- dl-Equilenin 3-Methyl Ether (XIII)——a) From XI. To a solution of 0.02 g. of N in 50 ml. of acetone was added 0.05 ml. of Jones' reagent.\*6 After 2 min., the excess CrO₃ was destroyed with 10 ml. of MeOH. The mixture was diluted with 50 ml. of H₂O, and the resulting precipitate was filtered and recrystallized from acetone–MeOH to yield 0.01 g. of II, m.p. 187°. UV  $\lambda_{max}^{BIOH}$  mμ (ε): 230 (33600), 268 (3400), 278 (3400), 289 (2400), 323 (1400), 338 (1700).
- b) From V. To a solution of 0.1 g. of I in 10 ml. of acetone was added 0.3 ml. of Jones' reagent. After 2 min., this reaction mixture was worked up as above to give 0.036 g. of V as yellow scales, m.p.  $184 \sim 187^{\circ}$ .
- c) From XI. To a solution of  $0.02\,\mathrm{g}$ . of I in 2 ml. of acetone was added  $0.08\,\mathrm{ml}$ . of Jones' reagent. The reaction mixture was worked up as above to give  $0.006\,\mathrm{g}$ . of N melting at  $183\sim184^{\circ}$ .

<sup>\*6</sup> A solution of 26.72 g, of  $CrO_3$  in 23 ml, of  $H_2SO_4$  was diluted with  $H_2O$  to 100 ml, and it was used as a standard reagent.

The authors express their deep gratitude to Takeda Chemical Industries, Ltd. for permission to publish this report, to Dr. Y. Abe for his kind guidance. Thanks are also due to Mr. H. Kamio and his associates for spectral determinations and Mr. M. Kan and his associates for elemental analysis.

## Summary

A short step synthesis of estradiol was presented. 1-Vinyl-6-methoxy-1-tetralol (I) was condensed with 2-methyl-1,3-cyclopentanedione (II) in the presence of Triton B to give dl-3-methoxy-8,14-secoestra-1,3,5(10),9-tetraene-14,17-dione (II) which was converted by phosphorus pentoxide into the known dl-3-methoxyestra-1,3,5(10),8,14-pentaen-17-one (IV). The ketone (IV) was treated with sodium borohydride in methanol, yielding dl-3-methoxyestra-1,3,5(10),8,14-pentaen-17 $\beta$ -ol (V). Catalytic hydrogenation of this compound over Raney nickel was found to be stereospecific, yielding dl-3-methoxyestra-1,3,5(10), 8-tetraen-17 $\beta$ -ol(V). This compound was subjected to K-NH<sub>3</sub> reduction to give dl-estradiol 3-methyl ether (VII).

Racemic V was resolved through its 17 l-menthoxyacetate and d-estradiol 3-methyl ether obtained by the above series of reactions was identical with the material of natural origin.

dl-Equilenin 3-methyl ether was prepared from dl-3-methoxyestra-1,3,5(10),8-tetraen-17 $\beta$ -ol (V) by oxidation with Jones' reagent at one step.

(Received April 19, 1965)

(Chem. Pharm. Bull.) 13(11)1289~1294(1965)

UDC 547.924.07:615.76

164. **Kentaro Hiraga**: Syntheses of Racemic and Optically Active 13β-Ethylgonanes.\*<sup>1</sup>

(Chemical Research Laboratories, Research & Development Division, Takeda Chemical Industries, Ltd.\*2)

Recently, Smith, et al.<sup>1)</sup> reported that  $dl-17\beta$ -hydroxy- $13\beta$ ,  $17\alpha$ -diethylgon-4-en-3-one has an excellent anabolic activity. Quite independently, the author also synthesized the compound, and applied for patents already.

The synthetic route is similar to that of Smith, et al., but different in many steps, and this paper deals with the synthesis and the resolution of the  $13\beta$ -ethylgonane series (Chart 1).

Condensation<sup>2)</sup> of 1-vinyl-6-methoxy-1-tetralol (I) and 2-ethyl-1,3-cyclopentanedione (II) in boiling xylene without catalyst or in the presence of basic catalyst gave dl-3-methoxy-13-ethyl-8,14-secogona-1,3,5(10),9-tetraene-14,17-dione (II) in 40% yield.

The seco-compound (II) in methanolic hydrogen chloride solution cyclized easily to yield dl-3-methoxy-13 $\beta$ -ethylgona-1,3,5(10),8,14-pentaen-17-one (IV). As an intermediate compound of this reaction was obtained dl-3-methoxy-13 $\beta$ -ethyl-14 $\xi$ -hydroxygona -1,3,5(10),9-tetraen-17-one, on which will be described separately.

<sup>\*1</sup> This paper constitutes Part XXXV of Takeda Laboratories' series entitled "Steroids"; Part XXXIV: This Bulletin, 13, 1285 (1965).

<sup>\*2</sup> Juso-nishino-cho, Higashiyodogawa-ku, Osaka (平賀謙太郎).

<sup>1)</sup> H. Smith, et al.: Experientia, 19, 394 (1963); J. Chem. Soc., 1964, 4472.

<sup>2)</sup> S.N. Ananchenko, V.Y<sub>E</sub>. Limanov, V.N. Leonov, V.N. Rzheznikov, I.V. Torgov: Tetrahedron, 18, 1355 (1962).