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Shigenobu Okuda,\*<sup>1</sup> Yūya Nakayama,\*<sup>2</sup> and Kyosuke Tsuda\*<sup>1</sup>: Studies  
on Microbial Products. I. Helvolic acid and Related Compounds. I.  
7-Desacetoxyhelvolic Acid and Helvolinic Acid.

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Cephalosporin P<sub>1</sub> (I),<sup>1)</sup> fusidic acid (II)<sup>2)</sup> and helvolic acid (III)<sup>3)</sup> have hitherto been reported as the interesting antibiotics with unique protolanostane type framework. This paper deals with the isolation of additional homologues, 7-desacetoxyhelvolic acid (IV) and helvolinic acid (V), from a culture broth of *Cephalosporium caerulens* besides helvolic acid (III).

### A) 7-Desacetoxyhelvolic Acid

Ethyl acetate extract of the broth filtrate was evaporated and the residue was triturated with methanol to remove a greater part of helvolic acid (III). The helvolic

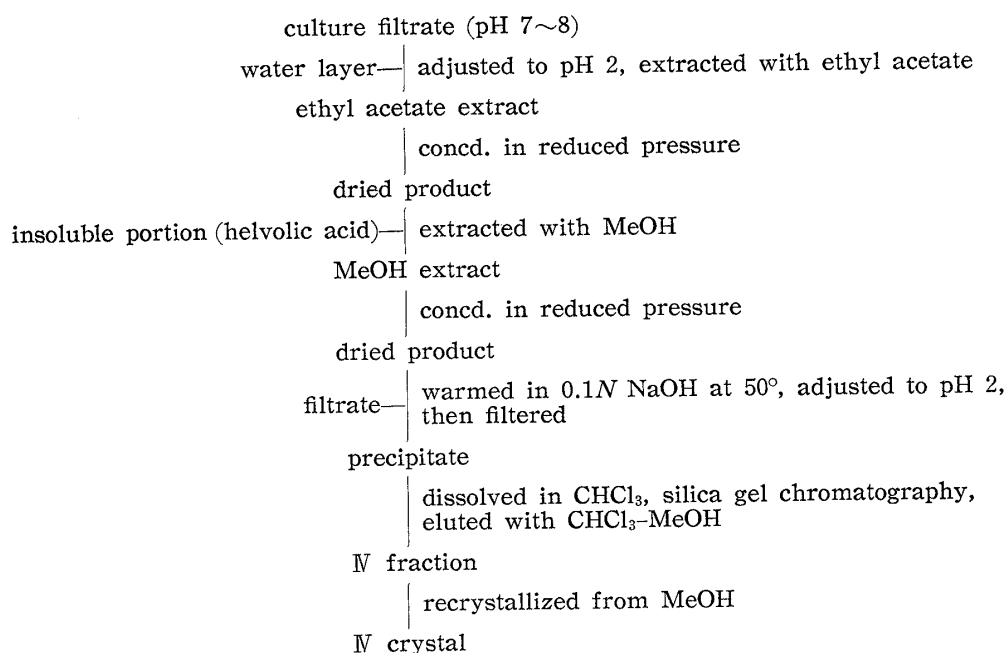


Fig. 1. Separation of 7-Desacetoxyhelvolic Acid (IV) from Culture Broth

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acid deficient methanol solution was evaporated and the residue was treated in 0.1N sodium hydroxide solution at 50° for 15 min., whereupon the remaining helvolic acid (III) was hydrolyzed into helvolinic acid (V). After acidification, the resulting precipitate was collected by filtration, dissolved in chloroform and purified through silica gel chromatography to afford white fine needles (IV), m.p. 214.5~215°,  $[\alpha]_D^{25}$  -51.7°, a yield of 30 mg./1000 ml.

This material is easily soluble in ethyl acetate, methanol, ethanol, acetone, chloroform, while it is insoluble in petroleum ether. This compound is also soluble in 0.1N sodium hydroxide and the resulting solution gave a recovery of original compound in a quantitative yield after heating at 50° for 15 min. followed by acidification. On the other hand, helvolic acid (III) is converted into helvolinic acid (V) under these conditions.

A treatment of this acid with diazomethane afforded methyl ester (VI) m.p. 178~179°,  $[\alpha]_D^{25}$  -54°. The analytical data of the acid (IV) and methyl ester (VI), were in good agreement with the molecular formulas,  $C_{31}H_{42}O_6$  and  $C_{33}H_{44}O_6$ , respectively. Infrared (IR) spectrum of IV by potassium bromide disk method exhibits absorptions ( $\nu$ ,  $cm^{-1}$ ) at 1240 (-OAc), 1250 (>C=O), 1670 ( $\alpha\beta$ -unsaturated ketone), 1700 (>C=O,  $\alpha\beta$ -unsaturated acid) and 1730 (-OAc). In nuclear magnetic resonance (NMR) spectrum of VI, the signals corresponding to  $C_7$ -H and  $C_7$ -acetoxy methyl were not observed, compared with that of helvolic acid methyl (VII) (cf. Table I). UV spectrum of IV showed the absorption maximum corresponding to  $\alpha\beta$ -unsaturated ketone at 232.5  $m\mu$  (cf. Table II).

TABLE I. Comparison of Nuclear Magnetic Resonance Spectrum

	$C_1, C_2-H$	$C_7-H$	$C_{16}-H$	$C_{24}-H$	$\begin{array}{c} -C-OCH_3 \\    \\ O \end{array}$	$C_7-OAc$	$C_{16}-OAc$	$\begin{array}{c} C_{26}, \\ C_{27}-CH_3 \end{array}$	$C_4-CH_3$	$\begin{array}{c}   \\ -C-CH_3 \\   \end{array}$
Helvolic acid methyl (VII)	2.70	4.79	4.17	4.89	6.37	7.90	8.05	8.30	8.72	8.55
	4.17									8.82
	doublet		singlet						doublet	9.08
	J=11.4 c.p.s.									
7-Desacetoxy-helvolic acid methyl (VI)	2.70	—	4.22	4.94	6.38	—	8.06	8.31	8.84	8.68
	4.16									8.89
	doublet		doublet						doublet	9.10
	J=11.0 c.p.s.			J=8.0 c.p.s.						

Assignment of Protons:  $\tau$  value

TABLE II. Comparison of Ultraviolet Spectrum

	Absorption $\lambda_{max}^{EtOH}$ $m\mu$	Absorption intensity $\log \epsilon$
Helvolic acid (III)	231.0	4.24
7-Desacetoxyhelvolic acid (IV)	232.5	4.20

Since this compound is produced by a helvolic acid producing strain, the facts above mentioned strongly suggested the absence of the  $C_7$ -acetoxy grouping in IV. Then the interrelation between III and IV was attempted, according to the scheme in Chart 1.

Catalytic hydrogenation of IV with palladium-charcoal in methanol solution afforded tetrahydro derivatives (IX), m.p. 219°,  $[\alpha]_D^{25.5}$  -13.2°. This compound is completely identical with 7-desacetoxytetrahydrohelvolic acid (IX), prepared from III *via* zinc acetic acid treatment followed by catalytic hydrogenation.

Consequently the structure of the above newly isolated acid, m.p. 214.5~215°, is represented by IV.

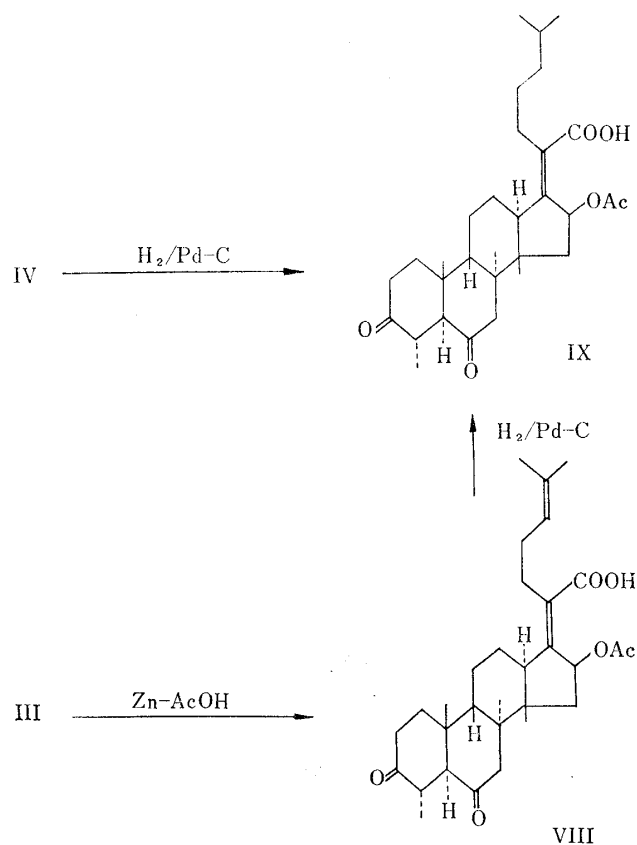
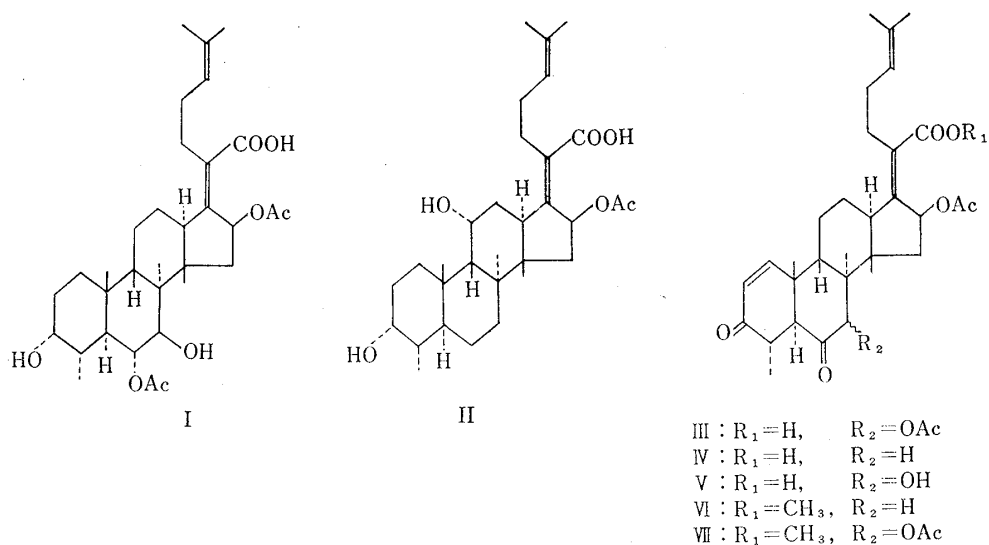


Chart 1.

### B) Helvolinic Acid

Cram and his coworkers<sup>3b)</sup> reported that mild hydrolysis of III with 0.1*N* sodium hydroxide solution gave rise to the removal of one of the two acetyls yielding helvolinic acid (V), which regenerated III by acetylation.

After cautious separation of metabolites from *Cephalosporium caerulens* especially preventing hydrolysis of acetoxy group was carried out, the methanol soluble fraction (a mixture of III, IV and V) was purified through silica gel chromatography (cf. Fig. 2). White fine needles, m.p. 202.5~203°,  $[\alpha]_D^{25} -88.9^\circ$ , thus obtained, was identical with an authentic specimen of helvolinic acid (V).

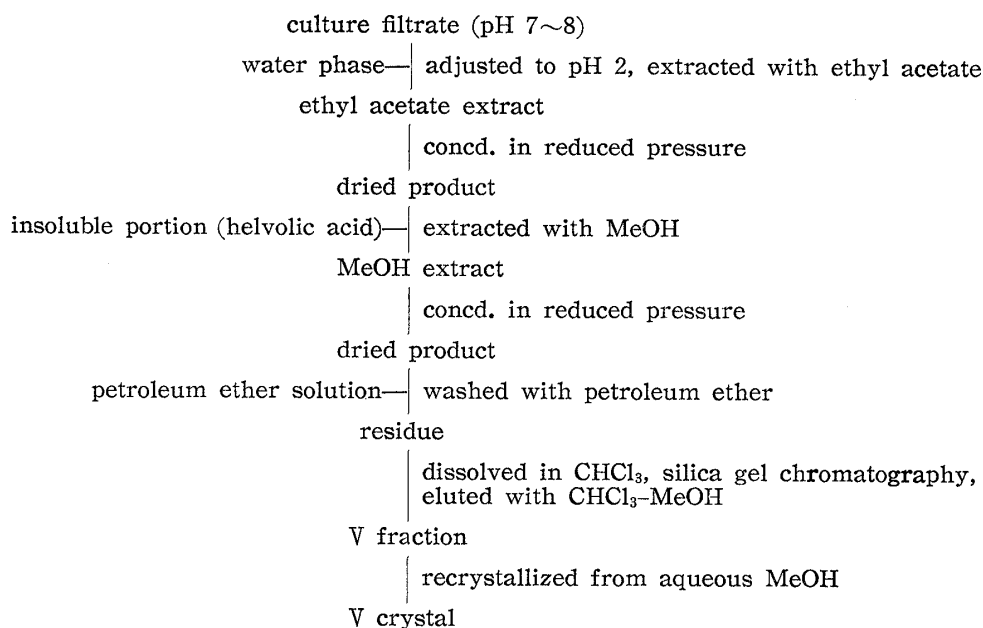


Fig. 2. Separation of Helvolic Acid (V) from Culture Broth

The fact that IV and V were isolated from a culture broth of *Cephalosporium caeruleus* indicates some possibilities that these compounds might serve as precursors of helvolic acid (III).

### C) Antibiotic Activity of IV

7-Desacetoxyhelvolic acid exhibits a similar antibiotic spectrum as in the case of helvolic acid (III) or helvolic acid (V) but the activity against *Staphylococcus aureus* of this compound is around one half compared with III (cf. Table III).

TABLE III. Biological Activities of IV and III

Sample No. Test strain	IV	III
<i>Staphylococcus aureus</i> 209P	12.5 25.0	6.25 12.5
<i>Staphylococcus</i> 76 (Penicillin resistant)	6.25 12.5	1.56 6.25
<i>Bacillus anthracis</i>	12.5 25.0	6.25 25.0
<i>Salmonella enteritidis</i> 1891	>100 >100	>100 >100
<i>Salmonella typhi</i> (Yomo)	>100 >100	>100 >100
<i>Salcina lutea</i> 9341	3.12 25.0	0.8 12.5
<i>Escherichia coli</i>	>100 >100	>100 >100
<i>Proteus vulgaris</i> X-28	>100 >100	>100 >100

MIC value:  $\gamma$ /ml.Method: agar streak in Buillon agar  
Lower: 48 hr., Upper: 24 hr.

## Experimental

### Production and Isolation of 7-Desacetoxyhelvolic Acid (IV)

The fermentation medium contained 5% glucose, 2% cornsteep liquor, 0.2% NaCl, 0.3% KCl, 0.1%  $K_2HPO_4$ , 0.05%  $MgSO_4$ , 0.01%  $ZnSO_4$ , 0.001%  $MnSO_4$ , 0.0001%  $CuSO_4$ , 0.4%  $CaCO_3$ . The strain (*Cephalosporium caerulens*) on an agar slant was inoculated into the medium.

After the fermentation at 27° for 7 days in shaking flasks, the mycelium were removed from the culture broth by filtration. The filtrate 5.51 L. thus obtained was adjusted to pH 2 and extracted twice with 1.2 L. each of ethyl acetate. The ethyl acetate extracts were combined and concentrated to dryness under reduced pressure. The syrupy residue was added with 50 ml. of MeOH and then the resulting precipitates were filtered off. After the filtrate was evaporated under reduced pressure, the residue was dissolved in 250 ml. of 0.1N NaOH and allowed to stand at 50° for 10 min. The reaction mixture was acidified to pH 2 and the resulting solids were collected by filtration.

This dried product (3.6 g.) was dissolved in 36 ml. of  $CHCl_3$  and chromatographed in a column packed with 100 g. of silica gel. The column was washed with  $CHCl_3$  and the elution was carried out with  $CHCl_3$  containing 1% MeOH. IV and V were eluted in this order. The fractions containing IV were dried *in vacuo* and the residue obtained was crystallized from MeOH to give crystals of IV, 80 mg. White needles, m.p. 214.5~215°,  $[\alpha]_D^{27} -51.7^\circ$  (c=0.5,  $CHCl_3$ ). Anal. Calcd. for  $C_{31}H_{42}O_6$ : C, 72.91; H, 8.29. Found: C, 72.66; H, 8.27.

### IV-Methyl Ester (VI)

IV (50 mg.) was esterified with diazomethane in  $CHCl_3$ . The resulting solution was evaporated to dryness and the residue was recrystallized from aqueous acetone to give crystals of VI 30 mg. White needles, m.p. 178~179°,  $[\alpha]_D^{33} -54.0^\circ$  (c=0.5,  $CHCl_3$ ). Anal. Calcd. for  $C_{32}H_{44}O_6$ : C, 73.25; H, 8.45. Found: C, 73.08; H, 8.43.

### 7-Desacetoxytetrahydrohelvolic Acid (IX)

A) Onto a refluxing solution of 1 g. of III in 150 ml. of AcOH-H<sub>2</sub>O (1:1), 100 g. of Zn-dust were added with vigorous stirring. Heating and stirring were continued for 15 min. UV absorption due to  $\alpha,\beta$ -unsaturated ketone completely disappeared within 10 min. under these conditions. After cooling, Zn-dust was removed by filtration and the filtrate was concentrated until  $Zn(OAc)_2$  was precipitated. The precipitated  $Zn(OAc)_2$  was filtered and washed with acetone. The filtrate and washing were combined and concentrated again until another  $Zn(OAc)_2$  was precipitated. After several similar treatments, the final concentrated solution was added with H<sub>2</sub>O and the precipitated solid was extracted with  $CH_2Cl_2$ . The  $CH_2Cl_2$  extract was washed with H<sub>2</sub>O,  $NaHCO_3 \cdot H_2O$ , and H<sub>2</sub>O, and then evaporated *in vacuo* after drying over  $Na_2SO_4$ . Recrystallization of the residue from acetone-H<sub>2</sub>O afforded 600 mg. of prisms (VIII), m.p. 202~203°. Anal. Calcd. for  $C_{31}H_{44}O_6$ : C, 72.62; H, 8.65. Found: C, 72.60; H, 8.67.  $[\alpha]_D^{18} +3.6^\circ$  (c=0.56, EtOH).

The solution of 600 mg. of VIII, obtained above, in 30 ml. of MeOH was shaken with 30 mg. of 10% Pd-C under atmosphere of hydrogen until 1.2 equivalent molecules of hydrogen were absorbed. A usual work up gave rise to crystalline material which was purified *via* recrystallization from MeOH to afford needles, m.p. 219°. Anal. Calcd. for  $C_{31}H_{46}O_6$ : C, 72.34; H, 9.01. Found: C, 72.38; H, 8.94.

B) 50 mg. of IV was dissolved in 10 ml. MeOH and hydrogenated at an atmospheric pressure in the presence of 5 mg. of 5% Pd-C. After 2 hr., 2 moles of hydrogen were absorbed, and then the catalyst was filtered off. Evaporation of the filtrate gave crude material, which was recrystallized from MeOH affording 41 mg. of crystalline K. White needles m.p. 219°,  $[\alpha]_D^{25.5} -13.2^\circ$  (c=0.76,  $CHCl_3$ ). Anal. Calcd. for  $C_{31}H_{46}O_6$ : C, 72.34; H, 9.01. Found: C, 72.51; H, 9.28. The IR spectrum (Nujol) was completely identical with the standard sample prepared as described above. The mixed melting point showed no depression.

### Production and Isolation of Helvolic Acid (V)

The fermentation was carried out similarly to the case of the production of IV. The broth filtrate 1.5 L. was adjusted to pH 2, which was extracted twice with 300 ml. each of ethyl acetate. The ethyl acetate extract thus obtained was evaporated under reduced pressure and the residue was washed by petroleum ether. The resulting solid was collected by filtration and dried.

The dried product (1.3 g.) was dissolved in 14 ml. of  $CHCl_3$  and chromatographed on a column of silica gel (40 g.). After the column was washed with 160 ml. of  $CHCl_3$ , it was eluted with  $CHCl_3$  containing 2% of MeOH. Evaporation of the fraction containing V gave crude powder (200 mg.), which was recrystallized from H<sub>2</sub>O-MeOH affording 21 mg. of crystalline V. White needles, m.p. 202.5~203°,  $[\alpha]_D^{24} -88.9^\circ$  (c=1,  $CHCl_3$ ). Anal. Calcd. for  $C_{31}H_{42}O_7$ : C, 70.09; H, 8.04. Found: C, 70.05; H, 8.01.

This sample of V showed the same IR spectrum as the standard sample prepared from helvolic acid by the method of Cram, *et al.*<sup>3b)</sup> and the mixed melting point showed no depression.

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### Summary

Two new antibiotics, 7-desacetoxyhelvolic acid and helvolinic acid were isolated from the culture broth of *Cephalosporium caerulens*. Characterization and structural studies of these substances are described.

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