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Synthesis of Anomeric Pyrimidine Nucleoside 5'-Phosphates of L-Ribose, α-D-Cytidine 5'-Phosphate and 3-β-L-Ribofuranosyladenine 5'-Phosphate

In the course of an investigation of the synthesis of unnatural nucleosides and their 5'-phosphates, 1,2) it became desirable to synthesize anomeric pyrimidine nucleoside 5'-phosphates of L-ribose for examination of their biological activities.

The present report deals with the synthesis of new L- α - and L- β -nucleoside 5'-phosphates as a continuation of our study on the synthesis of nucleotides.

Table I. Physical Properties of Anomeric Nucleoside 5'-Phosphates of L-Ribose and p-ribose

Configuration Optical isomer Compound	α		β	
	D	L	D	L
5'-Uridylic acid (5'-UMP)				
$[\alpha]_{\mathbf{D}^{\mathbf{a}}}$	- 25.0	+ 26.9	- 12.9	+ 13.0
$C. S. b. (J_{1',2'})$	-148(4.0)	-148(4.0)	-135(4.5)	-138(5)
Rf ^{c)}	0.24	0. 24	0.24	0.24
Cotton effect (O.R.D.)	Θ	\oplus	\oplus	\ominus
UV $\lambda_{\max}(\varepsilon)$ pH 1	0	263 (9600)	•	262.5(9350)
pH 7		263.5(9700)		262.5(9350)
pH 13		262.0(7300)		261.5(7100)
5'-Cytidylic acid (5'-CMP)		/		20110(1200)
$[\alpha]_{\mathbf{p}^{a}}$	- 86.8	+ 87.4	+ 6.9	- 6.8
C. S. $^{b)}$ (J _{1',2'})	-148(4.0)	-148(4.0)	-137(4.0)	-137(4.0)
Rfc)	0.205	0. 205	0. 205	0. 205
Cotton effect (O.R.D.)	⊖	()	⊕	$\bigcirc 203$
UV $\lambda_{\max}(\varepsilon)$ pH 1	280 (14000)	280. 5 (14000)	279. 0(13300)	279. 0 (13300)
pH 7	271.5(9500)	271.5(10000)	271.0(9100)	271.0(9350)
pH 13	271.5(9500)	271.5(10000)	271.0(9100)	271.5(9300)
Ribofuranosylthymine 5'-phos	• /	211.5(3300)	271.0(3100)	211.3 (9300)
$[\alpha]_{\mathbf{D}^{\mathbf{a}}}$	- 41.0	+ 43, 2	- 12.3(HCl)	+ 13.0
$C. S. b (J_{1',2'})$	-146(4.5)	-147(4.5)	-136(5.0)	
Rf ^{c)}	0.31	0.31	0.31	-137(5.0)
Cotton effect (O. R. D.)	⊕ ⊖	⊕ ⊕		0.31
UV $\lambda_{\max}(\varepsilon)$ pH 1	Θ	269. 0(10300)	\oplus	967 5 (9650)
pH 7		269. 0 (10200)		267.5(9650)
pH 13				267.5(9550)
-	3 \	268.0(7500)		267.0(7100)
3-isoAdenylic acid (3-isoAMI	7)		0.0	. 7.0
$(\alpha)_{\mathbf{D}^{a}}$			- 6.6	+ 7.0
C. S. $^{b)}$ $(J_{1',2'})$			-250(1.0)	-249(1.0)
Rfc)			0.20	0.20
Cotton effect (O. R. D.)				
UV $\lambda_{\max}(\varepsilon)$ pH 1			275. 0 (16600)	275.0(18200)
pH 7			274.0(11800)	275. 5 (13800)
pH 13			275.0(11900)	275. 5 (13700)

a) Li salt (H2O).

b) Chemical shift of C-1' proton from dioxane as an internal standard in c.p.s.

c) Solvent system: iso-BuOH-H₂O-AcOH=5:3:2 (using Toyo Roshi No. 50 ascending at 20°).

d) Chemical shift of C-1' proton from aceton in c.p.s.

B. Shimizu, M. Asai, M. Miyaki, H. Okazaki: This Bulletin, 13, 616 (1965). Idem: Ibid., 15, 1863 (1967).

²⁾ T. Nishimura, B. Shimizu, M. Futai: Biochim. Biophys. Acta, 129, 654 (1966).

$$(CH_3)_3SiO N + Br \longrightarrow O CH_2OP (OC_6H_5)_2 \\ (CH_3)_3SiO N + Br \longrightarrow O CH_2OP (OC_6H_5)_2 \\ (CH_3)_3SiO N + Br \longrightarrow O CH_2OP (OC_6H_5)_2 \\ (CH_3)_3SiO N + Br \longrightarrow O CH_2OP (OC_6H_5)_2 \\ (OC_6H_5)_2 OB_2OB_2 \\ (OC_6$$

$$\begin{array}{c} NH_{2} \\ NH_{2$$

1- α -L-Ribofuranosyluracil 5'-monophosphate (α -L-UMP) and its β -anomer (β -L-UMP) were prepared by the "silyl method" starting from L-ribose³⁾ according to the procedure for α -D-UMP and its β -anomer.²⁾

When the trimethylsilyluracil (Ia)²⁾ was allowed to react at 100~110° with 2.3-di-O-benzoyl-5-diphenylphosphoryl-L-ribofuranosyl bromide (Ia),*1 then the protecting groups

^{*1} This bromide (Ia) was obtained from L-ribose³⁾ according to the procedure for the corresponding p-series (Ib).⁴⁾

³⁾ F. L. Humoller: Methods in Carbohydrate Chemistry, I, 83 (1962).

removed, α -L-UMP (Na) and β -L-UMP (Mb) were produced in 34.5 and 18.0% yields based on the bromide, respectively (see Fig. 1).

Methods for the synthesis of L- α -ribofuranosylthymine 5'-phosphate (α -L-5'-rTMP) and its β -anomer (β -L-5'-rTMP) were virtually the same as those of the foregoing experiment. Starting from trimethylsilylthymine ($\mathbb{I}b$)²) and the phosphorylated sugar bromide ($\mathbb{I}a$), α -L-rTMP ($\mathbb{I}b$) and β -L-rTMP ($\mathbb{I}b$) were obtained in 34.3 and 17.0% yields based on the bromide, respectively (see Fig. 1).

The nucleotides thus obtained were identical with those of the corresponding D-series except that the specific rotations were opposite in sign having the same numerical values, (see Table I).

The synthesis of α -D-cytidine 5'-phosphate was not reported yet. made successfully to prepare the nucleotide. Bistrimethylsilyl N-acetylcytosine (N) was similarly treated with 5-O-diphenylphosphoryl-2,3-di-O-benzoyl-p-ribofuranosyl bromide (Ib)4) by the same procedure described above, followed by treatment of the reaction mixture with aqueous ethanol to give protected anomeric nucleotides. Chromatography of the crude product mixture on a silicagel column afforded 1-(5-O-diphenylphosphoryl-2,3-di-O-benzoyl- α -D-ribofuranosyl)-N-acetylcytosine (VII) m.p., 202.5° and its β -anomer (V), m.p., 176.5°, in 32.1 and 15.6% yield respectively (see Fig. 2). The aqueous alkaline (LiOH) treatment of the products (IIc and IVc) gave 1-(5-phenylphosphoryl- α -p-ribofuranosylcytosine and its β -anomer which liberated the remaining phenyl group by incubation with phosphodiesterase obtained from trimeresurus flavoviridis (Hallowell) to yield the corresponding α -D-5'-CMP (K) and its β -anomer (β -D-5'-CMP) (W) in good yield respectively. Characterization of the former (X) as α -D-CMP(5') was readily achieved as follows: The mobilities of the former (K) on paper electrophoresis (acetate buffer, pH, 3.5) and thin-layer chromatography (DEAE-cellulose, 0.02M HCl) were almost identical to those of The ultraviolet absorption spectra curves for the product (X) at various pH values were closely similar to those for the β -anomer but the absorption maximum exhibited a slight bathochromic shift (approx., 1 m_{lb}) as compared with β -anomer. Furthermore, the product (X) consumed 1 mole of metaperiodate. In NMR spectra, the signals due to the C-1' proton of this product appeared at a lower magnetic field than that for the β -anomer^{5,6)} (see Table I). Recently, the usefulness of O.R.D. measurement was reported by Ulbricht⁷⁾ for the assignment of the configuration at the anomeric carbon; α -pyrimidine nucleotides show the negative Cotton effect and β -anomers the positive effect. (In the case of α -nucleotides prepared previously, almost same results were obtained with α -UMP and α -rTMP giving the negative Cotton curves. In the case of the product (α p-CMP), it also shows the negative Cotton effect and β -anomer the positive effect.*2)

The latter compound (W) was identical with the authentic sample of natural 5'-cytidylic acid (5'-CMP) in its physical properties.

In the same manner α -L-5'-CMP and β -L-5'-CMP were obtained starting from L-ribose. The nucleotides thus obtained were indicated to be α -L- and β -L-compound in physical properties as shown in Table I.

In addition, $3-\beta$ -L-ribofuranosyladenine 5'-phosphate (XIII) (L-3-isoAMP) was prepared by coupling of the bromosugar phosphate (Ia) with adenine (X), according to the procedure for that of p-isomer.^{8,9)} When adenine (X) was allowed to react at 40° with the bromide

^{*2} Details of the ORD studies on the anomeric nucleotides will be published elsewhere.

⁴⁾ T. Ukita, H. Hayatsu: J. Am. Chem. Soc., 84, 1879 (1962).

⁵⁾ T. Nishimura, B. Shimizu, I. Iwai: This Bulletin, 12, 1471 (1964).

⁶⁾ H.G. Gassen, H. Witzel: Biochim. Biophys. Acta, 95 244 (1965).

⁷⁾ T.L.V. Ulbricht, J.P. Jennings, P.M. Scopes, W. Klyne: Tetrahedron Letters, 1964, 695.

⁸⁾ M. Asai, M. Miyaki, B. Shimizu: Agr. Biol. Chem., 29, 170 (1965); Idem: Ibid., 31, 319 (1967).

⁹⁾ N. J. Leonard, R. A. Laursen: Biochemistry, 4, 354 (1965).

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$$(CH_3)_3 Sio(N) + (C_6H_5O)_2 OB_2OB_2$$

$$V Ib$$

$$V$$

(Ia) in acetonitrile (see Fig. 1), then the protecting groups removed, by usual method, 3β -L-ribofuranosyladenine 5'-phosphate (XII) and L-5'-adenylic acid (XIV) were obtained in 14.8% and 8.8% yield based on the bromide (Ia), respectively. As shown in Table I, the former (XIII) was indicated to be $3-\beta$ -L-AMP in physical properties.

The results thus obtained are listed in Table I. Studies on biochemical activities of these nucleotides are in progress in the laboratories.

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