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100. Akira Takamizawa, Kentaro Hirai, Yoshio Hamashima, and Hisako Ito: Studies on Pyrimidine Derivatives and Related Compounds. XLV.*¹ On Some Synthetic Methods of N-[1-(2-Oxo-1,3-oxathian-4-ylidene)ethyl]-N-[(2-methyl-4-amino-5-pyrimidinyl)-methyl]formamide.*²

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The reaction of phosgene with thiamine chloride ($\mathbb{I}a$) or nitrate ($\mathbb{I}b$) gave O-chlorocarbonylthiamine (\mathbb{X}) in good yield. Treatment of \mathbb{X} with alkali afforded N-[1-(2-oxo-1,3-oxathian-4-ylidene)ethyl]-N-[(2-methyl-4-amino-5-pyrimidinyl)methyl]formamide ($\mathbb{X}\mathbb{I}$). The reaction of thiamine sodium salt (\mathbb{I}) with aryloxycarbonyl chloride (phenoxy-, p-methyl-phenoxy, p-methoxyphenoxy) gave corresponding S-aryloxycarbonylthiamine ($\mathbb{X}a\sim c$), and further treatment of $\mathbb{X}a\sim c$ with alkali afforded $\mathbb{X}\mathbb{I}$ in good yield. On the other hand, \mathbb{X} reacted with sodium salts of phenol, p-cresol, and p-methoxyphenol to give corresponding O-aryloxycarbonylthiamine ($\mathbb{X}a\sim c$). Treatment of $\mathbb{X}a\sim c$ with alkali also gave $\mathbb{X}\mathbb{I}$ in low yield. Therefore, it was assumed that \mathbb{X} and \mathbb{X} gave $\mathbb{X}\mathbb{I}$ in independent way by alkali treatment

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It has been reported¹⁾ that the reaction of thiamine sodium salt (III) with phosgene gave S-chlorocarbonylthiamine (\mathbb{N}) and O,S-bis(chlorocarbonyl)thiamine (\mathbb{N}), and treatment of \mathbb{N} and \mathbb{N} with ethanol afforded S-ethoxycarbonylthiamine (\mathbb{N}) and O,S-bis-(ethoxycarbonyl)thiamine (\mathbb{N}), respectively.

Recently, Murakmi, et al.²⁾ reported the synthesis of N-[1-(2-oxo-1,3-oxathian-4-ylidene)ethyl]-N-[(2-methyl-4-amino-5-pyrimidinylmethyl)]formamide (XIII) from the reaction of phosgene with thiamine in alkaline solution. It would be reasonable to consider the route via $\mathbb N$ as an intermediate in this reaction. Then isolated $\mathbb N$ was treated with alkali to give XIII as expected. The structure of $\mathbb N$ was also confirmed by the fact that $\mathbb N$ reacted with morpholine to give S-morpholinocarbonylthiamine (VIII). Previously, it was found that the reaction of thiamine hydrochloride (I) with phosgene in chloroform gave O-chlorocarbonylthiamine hydrochloride (X). It was also expected to form XIII from $\mathbb K$, therefore, $\mathbb K$ was treated with alkali to give XIII as expected.

Since this method was considered to be useful for obtaining XIII industrially from I, an attempt to improve the yield of K was made. I was sparingly soluble in organic solvents and had a tendency to remain unreacted, but the reaction of thiamine chloride (IIa) with phosgene in acetic anhydride gave pure K as the crystals of m.p. 225° (decomp.). The elemental analyses of these crystals were in agreement with the value of $C_{13}H_{17}O_2N_4SCl_3$, infrared (IR) spectrum showed the bands at 1765 ($\nu_{c=0}$) and 1149 cm⁻¹ ($\nu_{0-c=0}$), and the reaction with ethanol gave O-ethoxycarbonylthiamine (X). Treatment of this product with alkali gave pure XIII in good yield. Similarly, thiamine nitrate (IIb) gave K in good yield.

^{*1} Part XLIV, A. Takamizawa, K. Hirai, T. Ishiba, Y. Matsumoto: This Bulletin, 15, 731 (1967).

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¹⁾ A. Takamizawa, K. Hirai, Y. Hamashima: This Bulletin, 11, 882 (1963).

²⁾ M. Murakami, K. Takahashi, Y. Hirata, H. Iwamoto: Vitamin, 34, 71 (1966).

It was previously reported³⁾ that the reaction of \mathbb{II} with alkyl chloroformate gave S-alkoxycarbonylthiamine and its rearrangement to O-alkoxycarbonylthiamine by alkaline catalysis. In this rearrangement, the intermediate XVI' was assumed to be present in the course of the reaction. It should be possible to form XIII by the preferential removal of the alkoxyl group of XVI' (route B), and the scission of the C-S bond of XVI' should give O-alkoxycarbonylthiamine (route A). When R of OR group was alkyl group, the reaction proceeded through route A. However, when R is aryl group, its electron-attracting effect and resonance effect would make it possible to take route B. The reaction of III with phenoxycarbonyl chloride gave S-phenoxycarbonylthiamine (XIIa) as the crystals of m.p. 112° (decomp.). The structure of this product was

³⁾ A. Takamizawa, K. Hirai: This Bulletin, 10, 1102 (1962).

$$H_{3}C \longrightarrow NH_{2} \longrightarrow N$$

confirmed by its elemental analysis and IR spectrum showing the band at 1730 cm⁻¹ due to S-C=O group. Treatment of Ma with sodium ethoxide followed by the reaction with methylphenylcarbamoyl chloride gave the crystals of m.p. 143°. The elemental analysis of this product was in agreement with the value of $C_{23}H_{29}O_5N_5S$. The IR comparison with O-ethoxycarbonyl-S-methylphenylcarbamoylthiamine (XV) obtained from Ma and methylphenylcarbamoyl chloride by sodium ethoxide catalyzed reaction showed the identical spectra. It was assumed that Ma was first formed from Ma and then converted into O-ethoxycarbonylthiamine sodium salt (XIV). Then Ma was treated with 5% sodium bicarbonate in chloroform to give Ma in high yield. The reaction with p-methylphenoxycarbonyl chloride and p-methoxyphenoxycarbonyl chloride gave S-p-methylphenoxycarbonylthiamine (Ma) and S-p-methoxyphenoxycarbonylthiamine

(Mc), respectively. Similar treatment of Mb, c with sodium bicarbonate also gave XIII in high yield.

Chart 3.

In the course of this reaction, the route via O-aryloxycarbonylthiamine (X) from S-aryloxycarbonylthiamine (Ma \sim c) would not be eliminable. Then O-phenoxycarbonylthiamine (Ma) obtained from X by the action of sodium phenolate was treated with alkali to give XIII. Similarly, O-p-methylphenoxycarbonylthiamine (Mb) and O-p-methoxyphenoxycarbonylthiamine (Mc), which were obtained from the reactions of X with sodium salts of p-cresol and p-methoxyphenol, were treated with alkali to afford XIII. However, the yields of XIII from Ma \sim c were worse than those from Ma \sim c. Therefore, it was concluded that Ma \sim c and Ma \sim c gave XIII independently.

Experimental*4

S-Chlorocarbonylthiamine (IV)—Into a suspension of 3.22 g. of thiamine sodium salt (II) in 30 ml. of MeCN, 1.0 g. of phosgene was passed through with stirring at $-40\sim-45^{\circ}$. After 2 hr., the reaction mixture was concentrated *in vacuo* to give crystalline solid $\mathbb N$. It was failed to detect XII (Rf 0.35) on TLC (SiO₂, Me₂CO). IR $\nu_{\max}^{\text{Nujol}}$ cm⁻¹: 1708 (S-C=O).

S-Morpholinocarbonylthiamine (VIII)—To a solution of N obtained above in 30 ml. of MeCN, 3 g. of morpholine was added with cooling, then stirred for 2 hr. at 20°. The reaction mixture was concentrated in vacuo, and the residue was extracted with CHCl₃. The CHCl₃ extract was washed with H₂O, dried over Na₂SO₄, and evaporated. The residue was chromatographed on SiO₂ with Me₂CO to give 0.8 g. of colorless prisms, m.p. $169 \sim 170^{\circ}$ (decomp.), which proved to be identical with WI by IR comparison with an authentic sample.¹⁾

^{*4} All melting points are uncorrected.

- O-Chlorocarbonylthiamine Chloride Hydrochloride (IX)—a) Into a suspension of 1.5 g. of IIa in 20 ml. of Ac₂O, excess of phosgene was passed through with stirring at $-5\sim-10^{\circ}$. After 2 hr. in cooling, stirring was continued for 15 hr. at room temperature. The reaction mixture was concentrated *in vacuo* to give 1.7 g. (85%) of crystalline (K), m.p. 225°(decomp.). *Anal.* Calcd. for C₁₃H₁₇O₂N₄SCl₃: C, 39.03; H, 4.28; N, 14.02. Found: C, 39.35; H, 4.56; N, 14.31. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 1765 (C=O), 1147 (O-C=O).
- b) Into a suspension of 1.7 g. of thiamine nitrate (IIb) in 20 ml. of Ac_2O , excess of phosgene was passed through at -10° , and worked up as above to give 1.65 g. (88%) of K. IR spectrum of this product was completely in accordance with that of K obtained above a).
- O-Ethoxycarbonylthiamine Chloride Hydrochloride (X)—A suspension of 4 g. of X in 20 ml. of EtOH was stirred for 3 hr. at $50\sim60^{\circ}$. The reaction mixture was concentrated *in vacuo*, and the residue was recrystallized from EtOH to give 3.2 g. (74.8%) of X, m.p. $206\sim207^{\circ}$ (decomp.), which showed identical spectrum with that of an authentic sample.⁴⁾
- O-Phenoxycarbonylthiamine Chloride Hydrochloride (XIa)—To a solution of 1 g. of K in 10 ml. of H_2O , 100 ml. of M_2O was added. To this solution, the solution of 0.87 g. of sodium phenolate in 3 ml. of H_2O was added dropwise at -40° . The reaction mixture showed pH 7 after stirring for 4 hr., and was adjusted to pH 2 with addition of conc. HCl at -20° . The solvent was removed *in vacuo*, the concentrate was adjusted to pH 4.5 with 10% NaOH, and extracted with CHCl₃. The aqueous layer was adjusted to pH 1.5 with dil. HCl and concentrated *in vacuo*. The residue was recrystallized from EtOH to give 0.5 g. (51.8%) of colorless scales, m.p. $210\sim211^\circ$ (decomp.). Anal. Calcd. for $C_{19}H_{22}O_3N_4SCl_2$: C, 49.88; H, 4.85; N, 12.25; Cl, 15.50. Found: C, 49.52; H, 5.04; N, 12.36; Cl, 15.80. IR ν_{max}^{Najol} cm⁻¹: 1767 (C=O), 1245 (O-C=O). The CHCl₃ extract was dried over Na_2SO_4 , evaporated, and the residue was added ether to give the crystals. Recrystallization from AcOEt gave 0.2 g. (3.7%) of XII, m.p. $182\sim183^\circ$ (decomp.).
- O-p-Methylphenoxycarbonylthiamine Chloride Hydrochloride (XIb)—To a solution of 1 g. of \mathbb{K} in 10 ml. of H_2O , 100 ml. of M_2CO was added and a solution of 0.98 g. of sodium p-cresolate in 4 ml. of H_2O was added dropwise at $-40\sim-50^\circ$. After stirring for 2.5 hr., the reaction mixture was worked up as above to give 0.51 g. (43%) of colorless prisms, m.p. $206\sim207.5^\circ$ (decomp.). Anal. Calcd. for $C_{20}H_{24}O_3N_4SCl_2$: N, 11.84; S, 6.80. Found: N, 11.40; S, 6.88. IR $\nu_{\max}^{N_{4}}$ is $\nu_{\max}^{N_{4}}$ in $\nu_{\max}^{N_{4}}$ in $\nu_{\max}^{N_{4}}$ in $\nu_{\max}^{N_{4}}$ is $\nu_{\max}^{N_{4}}$ in $\nu_{\infty}^{N_{4}}$ in $\nu_$
- O-p-Methoxyphenoxycarbonylthiamine Chloride Hydrochloride (XIc)—To a solution of 3 g. of K in 30 ml. of H₂O, 300 ml. of Me₂CO was added and a solution of 3.36 g. of sodium p-methoxyphenolate in 10.5 ml. of H₂O was added dropwise at $-40\sim-50^{\circ}$. Treatment of this reaction mixture as above gave 1.768 g. (48.4%) of colorless prisms, m.p. 195 \sim 196.5°(decomp.). Anal. Calcd. for C₂₀H₂₄O₄N₄SCl₂: N, 11.50; Cl, 14.55. Found: N, 12.08; Cl, 14.71. IR $\nu_{\text{max}}^{\text{Nulol}}$ cm⁻¹: 1768 (C=O), 1251 (O-C=O).
- S-Phenoxycarbonylthiamine (XIIa)—To a solution of $32\,\mathrm{g}$. of II in $250\,\mathrm{ml}$. of EtOH, $17.5\,\mathrm{g}$. of ClCOOC₆H₅ was added dropwise at -10° and stirred for $30\,\mathrm{min}$. After stirring was continued for $1.5\,\mathrm{hr}$. at 20° , the reaction mixture was concentrated *in vacuo* to give crystalline residue. The crystals were collected and washed with a small amount of cold water to give $20.8\,\mathrm{g}$. (quantitative) of colorless prisms, m.p. 80. IR $\nu_{\mathrm{max}}^{\mathrm{Nujol}}$: $1730\,\mathrm{(S-C=O)}$. This product was unstable and used in the following step without further purification.
- S-p-Methoxyphenoxycarbonylthiamine (XIIc)—To a solution of 3.04 g. of II in 25 ml. of EtOH, 1.87 g. of p-methoxyphenylchloroformate was added and stirred for 30 min. After stirring for 1 hr. at 20°, the reaction mixture was concentrated *in vacuo* and the separated crystals were collected to give 3.8 g. (87%) of crystalline powder, m.p. $96 \sim 97^{\circ}$ (decomp.). IR $\nu_{\text{max}}^{\text{Nulol}}$ cm⁻¹: 1731 (S-C=O). This product was unstable and used in the following step without further purification.
- O-Ethoxycarbonyl-S-methylphenylcarbamoylthiamine (XV)—a) To a solution of 0.127 g. of metalic Na in 20 ml. of EtOH, 2.01 g. of XIIa was added and stirred for 10 min. at 20°. Methylphenylcarbamoyl chloride (0.94 g.) was added and stirred for 1 hr. at $40\sim45^\circ$. The reaction mixture was concentrated in vacuo and the residue was extracted with CHCl₃. The CHCl₃ extract was washed with H₂O, dried over Na₂SO₄, and evaporated. The crystalline residue was recrystallized from Me₂CO-ether to give 2 g. (82.2%) of colorless rhombs, m.p. 143°. Anal. Calcd. for C₂₃H₂₉O₅N₅S: C, 56.66; H, 6.00; N, 14.37; S, 6.57. Found: C, 56.78; H, 6.15; N, 14.07; S, 6.52. IR $\nu_{\rm max}^{\rm Nujol}$ cm⁻¹: 3325, 3180, 1746, 1662, 1654, 1251.
- b) To a solution of 0.253 g. of metalic Na in 30 ml. of EtOH, S-ethoxycarbonylthiamine (MI, 3.544 g.) was added to become yellow solution. To this solution, 1.867 g. of methylphenylcarbamoyl chloride was added to separate NaCl. After 2 hr., the reaction mixture was concentrated *in vacuo* and the residue was extracted with CHCl₃. The CHCl₃ extract was washed with dil. NaHCO₃ solution, dried over Na₂SO₄, and evaporated. The residue was recrystallized from Me₂CO to give 3.903 g. (80%) of colorless rhombs, which were identified in IR spectrum with XV obtained above a).
- N-[1-(2-oxo-1,3-oxathian-4-ylidene)ethyl]-N-[(2-methyl-4-amino-5-pyrimidinyl)methyl] formamide (XIII)—a) To a suspension of N obtained from 4.5 g. of II and 1.0 g. of phosgene in CHCl₃, saturated solution of NaHCO₃ was added and shaked. The CHCl₃ layer was dried over Na₂SO₄, and evaporated. The residual crystals were recrystallized from AcOEt to give 0.6 g. of colorless prisms, m.p. 182~183°(decomp.).

⁴⁾ A. Takamizawa, K. Hirai, Y. Hamashima: This Bulletin, 10, 1107 (1962).

Anal. Calcd. for $C_{13}H_{16}O_3N_4S$: C, 50.64; H, 5.23; N, 18.18. Found: C, 50.70; H, 5.18; N, 17.79. IR $\nu_{\max}^{\text{Nujol}}$ cm⁻¹: 1686 (C=O), 1167, 1027.

- b) K (4 g.) was dissolved in a solution of 10 ml. of H_2O and 30 ml. of Me_2CO was added at $-30\sim-40^\circ$, and a solution of 1.2 g. of NaOH in 2 ml. of H_2O was added. After stirring for 1.5 hr. at -40° , the reaction mixture was concentrated *in vacuo* (at $<30^\circ$). The residue was extracted with CHCl₃, and the CHCl₃ extract was washed with H_2O , dried over Na_2SO_4 , and evaporated. The residual crystals were recrystallized from AcOEt to give 1 g. of colorless prisms, m.p. $182\sim183^\circ$ (decomp.), which were identified in IR spectrum with XIII obtained above a).
- c) To a solution of $0.4\,\mathrm{g}$. of XIIa in $25\,\mathrm{ml}$. of CHCl₃, $10\,\mathrm{ml}$. of 5% NaHCO₃ (or 5% Na₂CO₃) was added and shaked for $30\,\mathrm{min}$. The CHCl₃ layer was dried over Na₂SO₄, and evaporated. The crystalline residue was recrystallized from AcOEt to give $0.26\,\mathrm{g}$. (84%) of colorless prisms, m.p. $182\sim183^\circ$ (decomp.), which were identified in IR spectrum with XIII obtained above a).
- d) MIb (0.4 g.) was treated as above c) to give 0.26 g. (88%) of MII. Identity was confirmed by IR comparison with an authentic sample.
- e) \mathbb{M} c (0.4 g.) was treated as above c) to give 0.25 g. (87.5%) of \mathbb{M} . Identity was confirmed by IR comparison with an authentic sample.
- f) $\text{Ma} (0.8 \, \text{g.})$ was dissolved in a solution of $2 \, \text{ml.}$ of H_2O and $6 \, \text{ml.}$ of Me_2CO , and a solution of $0.2 \, \text{g.}$ of NaOH in $0.5 \, \text{ml.}$ of H_2O was added at $-30 \sim -35^\circ$. After stirring for $3 \, \text{hr.}$ at $0 \sim 7^\circ$, Me_2CO was removed in vacuo (at 25°) and the residue was extracted with CHCl₃. The CHCl₃ extract was dried over Na₂SO₄ and evaporated. The residual crystals were recrystallized from AcOEt to give $0.11 \, \text{g.}$ (20%) of XII. Identity was confirmed by IR comparison with an authentic sample.
- g) Xb (0.975 g.) was treated as above f) to give 0.135 g. (21.2%) of XIII. Identity was confirmed by IR comparison with an authentic sample.
 - h) Mc (1 g.) was treated as above f) to give 0.114 g. (18%) of XIII, m.p. $182 \sim 183^{\circ}$ (decomp.).

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