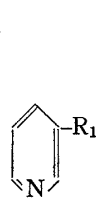
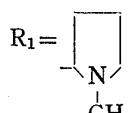

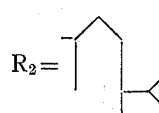
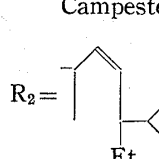
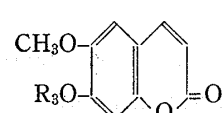
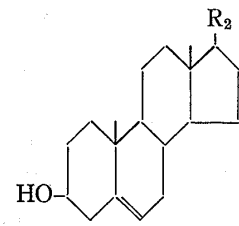
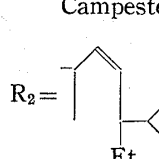
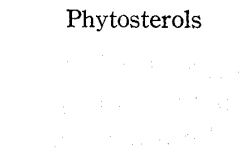
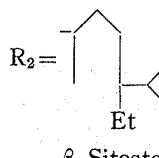
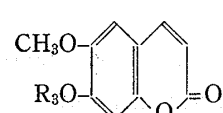


TABLE I. Constituents in Tobacco Callus

Compound		Callus	IAA	2,4-D	Compound		Callus	IAA	2,4-D
Alkaloids		R ₁ = 	+	-	Phytosterols		R ₂ = 	-	+
	Nicotine	Nicotine				Stigmasterol	R ₂ = 	-	+
Phenols		R ₃ = H	+	+	Phytosterols		R ₂ = 	-	+
	Scopoletin	Scopoletin					R ₂ = 	-	+
Phenols		R ₃ = Glucose	+	+					
	Scopolin	Scopolin							

tobacco callus have been shown to be markedly influenced by the supply of IAA and 2,4-D in the nutrient medium.

More detailed chemical and biosynthetic investigations are now in progress to determine the possible causal relationship between the action of IAA and that of 2,4-D in these phenomena.

We are grateful to Dr. E. Tamaki and Dr. E. Masuda of Japan Monopoly Corporation for authentic samples of tobacco alkaloids and tobacco plant, and to Dr. H. Itokawa of Tokyo College of Pharmacy for authentic samples of phytosterols.

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New Constituents of *Chamaecyparis formosensis* MATSUM.¹⁾

During the reinvestigation of the terpenoid constituents of the Benihi tree (*Chamaecyparis formosensis* MATSUM., Cupressaceae, grown in Taiwan), a novel nor-sesquiterpenoid, chamaecynone and related compounds were isolated.^{1,2)} Chamaecynone and freelinene reported by Massy-Westropp, *et al.*³⁾ are the first examples of acetylenic

1) Presented at the general local meeting of the Hokkaido district of the Chemical Society of Japan, at Sapporo, in July, 1965; abstract paper, pp. 11~12.

2) T. Nozoe, Y. S. Cheng, T. Toda: *Tetrahedron Letters*, **1966**, 3663.

3) R. A. Massy-Westropp, G. D. Reynolds, T. M. Spotswood: *Tetrahedron Letters*, **1966**, 1939.

CCl_4 , 9.25 (3H, s, $-\overset{|}{\underset{|}{\text{C}}}-\text{CH}_3$), 5.63 and 5.34 (1H each, m, $-\overset{|}{\underset{|}{\text{C}}}=\text{CH}_2$), 4.09 and 3.79 (1H each, broad s, $\text{CHO}-\overset{|}{\underset{|}{\text{C}}}=\text{CH}_2$), and 0.52 (1H, s, $\text{CHO}-\overset{|}{\underset{|}{\text{C}}}=\text{CH}_2$) τ . The above data suggest that II has the structure shown in Fig. 1; this was established by oxidation of sesquibenihiol¹⁰⁾ (or costol)¹⁵⁾ (III) with active manganese dioxide to afford an aldehyde whose IR and NMR spectra were identical with those of II. This is the first reported occurrence of II in nature.⁵⁾

The new alcohol (IV), $\text{C}_{15}\text{H}_{26}\text{O}$, m.p. 75° , $[\alpha]_D^{25} +5.02^\circ$ (MeOH), was identified with so-called cadinenol by comparison of its spectral data with those reported by Šorm and his colleagues,⁶⁾ and by Hirose and Nakatsuka.⁷⁾ This alcohol has also been obtained from *Torreya nucifera* SIEB. et ZUCC.¹⁶⁾ and *Schizandra nigra* MAXIM.¹⁷⁾ by Hirose and his coworkers, and from *Thujopsis dolabrata* SIEB. et ZUCC. by Endo.¹⁸⁾ Presumably, the distribution of this alcohol in nature is fairly common. Structure V is proposed for the alcohol on the basis of the following evidence: UV, end absorption only; IR (KBr pellet), 3370, 1670, 1390, 1385, 1365, 1075, 1023, 918, 888 and 851 cm^{-1} ; NMR in CCl_4 , 9.19 (d, $J=7.3$ c.p.s., isopropyl methyl), 9.13 (d, $J=7.3$ c.p.s., isopropyl methyl), 9.09 (d, $J=6.0$ c.p.s., $\text{H}-\overset{|}{\underset{|}{\text{C}}}-\text{CH}_3$), 8.34 (broad s, $-\text{CH}=\overset{|}{\underset{|}{\text{C}}}-\text{CH}_3$) and 4.62 (1H, m, $-\text{CH}=\overset{|}{\underset{|}{\text{C}}}-\text{CH}_3$) τ . Dehydrogenation of V gave cadalene,⁷⁾ therefore V is a cadinane

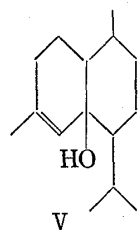


Fig. 2.

type sesquiterpene alcohol having one double bond with one vinyl proton. However, V does not give cadinene dihydrochloride. Since the alcohol is resistant to chromium trioxide oxidation in pyridine and the dihydro-derivative of V shows a new C-O stretching band at 1135 cm^{-1} with disappearance of the band at 1075 cm^{-1} in its IR spectrum, the alcohol group of V is tertiary and allylic. Furthermore, mild acid treatment of the alcohol afforded conjugate diene hydrocarbons, and main product ($\text{C}_{15}\text{H}_{24}$, by mass spectrum) possesses following spectroscopic properties: IR, 1635, 1625 cm^{-1} ; UV, 239 $\text{m}\mu$; namely, cadina-1,9-diene. Those facts lead to the conclusion that the structure of the alcohol is represented by formula IV or V (Fig. 2). Ozonolysis of the alcohol gave acetaldehyde,⁷⁾ suggesting that formula IV is more favorable than V, since the ozonide of IV would form an α -hydroxymethyl ketone which could afford acetaldehyde whereas the ozonide of V could not. The name "cadinenol" should be changed to "cadin-9-en-1-ol."

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15) V. Benesova, V. Sykora, V. Herout, F. Šorm: Chem. & Ind., 1958, 363.

16) T. Sakai, K. Nishimura, H. Chikamatsu, Y. Hirose: Bull. Chem. Soc. Japan, 36, 1261 (1963).

17) K. Morikawa, K. Nishimura, Y. Hirose: Nippon Kagaku Zasshi, 87, 591 (1966).

18) Endo isolated the alcohol as an oil, however its IR spectrum is identical with that of IV. K. Endo, D. Sc. thesis: Tohoku University, August, 1965.