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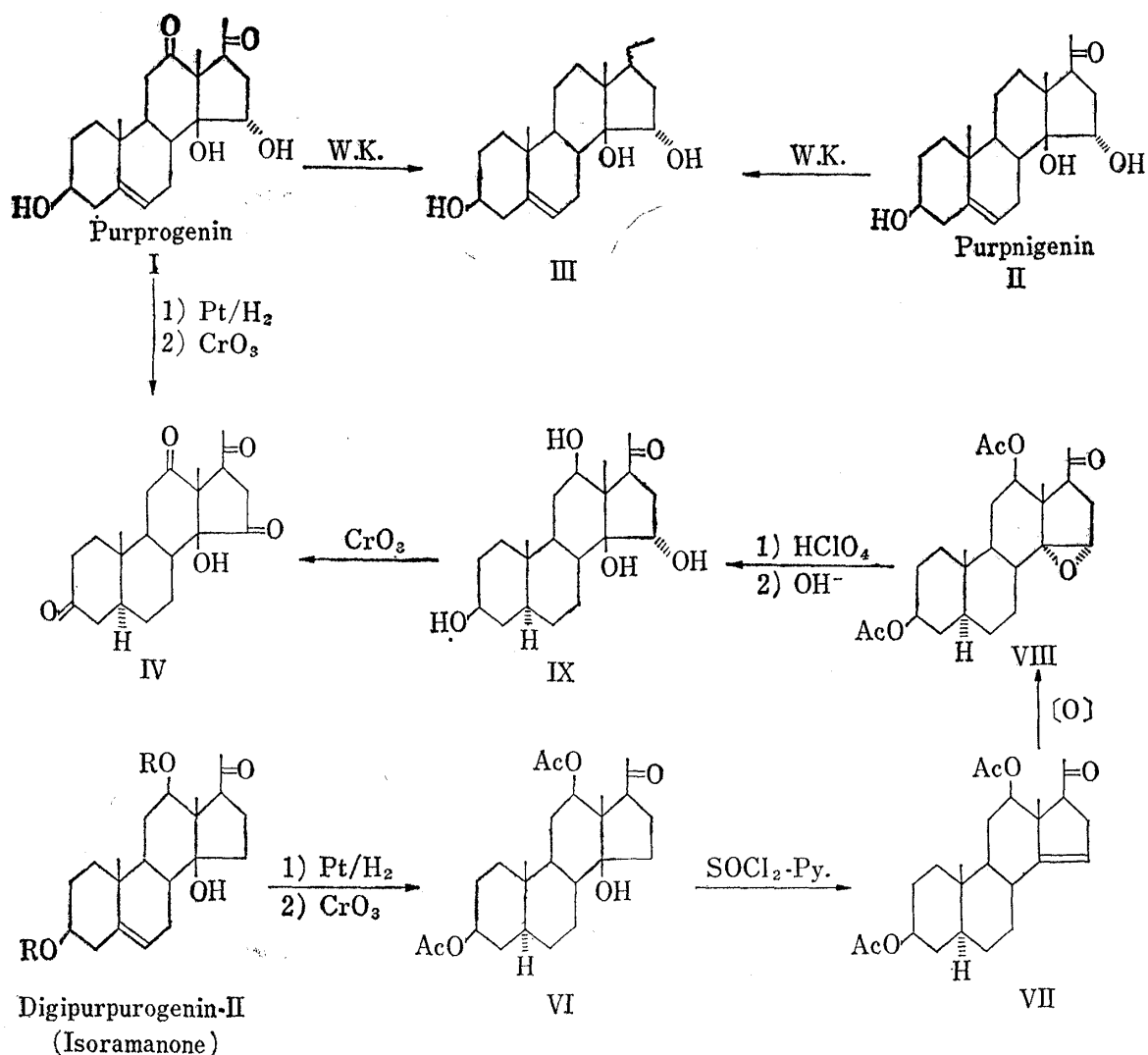
### Studies on Digitalis Glycosides. The Structure of Purprogenin

Previously, our group<sup>1-4)</sup> proposed the structure 3 $\beta$ ,14,15 $\alpha$ -trihydroxy-14 $\beta$ , 17 $\beta$ -pregn-5-ene-12,20-dione (I) for purprogenin, a C<sub>21</sub>-steroid of *Digitalis purpurea* L. leaves, but there remains the establishment of the location of a six-membered ring ketone and the configuration of C-17 side chain by stronger evidence. Recent works on the structure proof of purprogenin by the correlation of the genin and hecogenin<sup>5)</sup> as well as digacetigenin<sup>6,7)</sup> prompted us to report our own interconversion of purprogenin and digipurprogenin-II<sup>8)</sup> (isoramanone,<sup>8)</sup> V).

Purprogenin (I) was hydrogenated with platinum catalyst in acetic acid to saturate 5,6-double bond, giving, after oxidation with Kiliani's reagent, a hydroxy-tetraketone (IV), mp 262-265°, C<sub>21</sub>H<sub>28</sub>O<sub>5</sub>,  $[\alpha]_D^{25} +101.3 \pm 6.1^\circ (c=0.232, \text{CHCl}_3)$ ; NMR (CDCl<sub>3</sub>)  $\tau$ : 8.95 (19-CH<sub>3</sub>), 8.73 (18-CH<sub>3</sub>), 7.62 (21-CH<sub>3</sub>); IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3293 (14 $\beta$ -OH), 1747 (15-ketone), 1707, 1698 (20-ketone). Since the IR band at 1707 cm<sup>-1</sup> has twice intensity of absorption at 1698 cm<sup>-1</sup>, it is ascribable to 3-ketone and the six-membered ring ketone in question, which is probably located at C-12. A/B-ring juncture was shown by CD determination: the difference between the molecular ellipticity of IV ( $[\theta] +12417$  (291 m $\mu$ )) and that of its 3-hemiketal<sup>9,10)</sup> ( $[\theta] +9313$  (290 m $\mu$ )) is  $\Delta[\theta] +3104$  and this positive contribution due to 3-ketone indicates that A/B-ring juncture is *trans*. This hydroxy-tetraketone was the substance which we wish to derive from digipurprogenin-II (V) of known structure.

Digipurprogenin-II diacetate (Vb) was submitted to catalytic hydrogenation with platinum in acetic acid and the crude product was oxidized with Kiliani's reagent to give a dihydro derivative (VI), mp 165-167°, C<sub>25</sub>H<sub>38</sub>O<sub>6</sub>. The signal of 3 $\alpha$ -proton in NMR

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Va : R=H, Vb : R=Ac

spectrum appears as a broad multiplet, showing that the A/B-ring juncture of VI is *trans*. Dehydration of VI with thionyl chloride in pyridine afforded an amorphous 14-anhydro derivative (VII), UV  $\lambda_{\text{max}}^{\text{EtOH}}$   $m\mu$  ( $\epsilon$ ): 201 (6900), which was oxidized with *m*-chloroperbenzoic acid to give a 14,15-epoxide (VIII), mp 137–140°,  $\text{C}_{25}\text{H}_{36}\text{O}_6$ ,  $[\alpha]_{\text{D}}^{25} +34.1 \pm 3.3^\circ$  ( $c=0.226$ ,  $\text{CHCl}_3$ ); NMR ( $\text{CDCl}_3$ )  $\tau$ : 9.14 (18- $\text{CH}_3$ ), 9.08 (19- $\text{CH}_3$ ), 8.01, 7.98 (2Ac), 7.91 (21- $\text{CH}_3$ ), 6.60 (15 $\beta$ -H). Since oxidation of 14,15-unsaturated steroids having 17 $\beta$ -side chain with peracid has been known to afford 14 $\alpha$ ,15 $\alpha$ -epoxy derivatives,<sup>11-15</sup> the compound VIII must be 14 $\alpha$ ,15 $\alpha$ -epoxide. The above NMR data support this assignment. Cleavage of the epoxide ring in VIII with 0.7% perchloric acid in acetone at room temperature for four days<sup>16</sup> and subsequent deacetylation with 0.15% potassium hydroxide in methanol at the same temperature for two hours<sup>16</sup> afforded a crude tetrahydroxy-ketone (IX) as a main product,  $[\alpha]_{\text{D}}^{25} +94.2 \pm 4.5^\circ$  ( $c=0.257$ , MeOH). The 20-ketone in IX gave positive Cotton effect in

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16) Under these conditions, epimerization at C-17 was controlled to a negligible extent in parallel experiments using purpurogenin (II) and digipurpurogenin-II diacetate (Vb).

CD,  $[\theta] +676$  (285  $m\mu$ ), showing the 17-side chain to be  $\beta$ -oriented. As the cleavage of 14 $\alpha$ ,15 $\alpha$ -epoxide with acid generally gives rise to 14 $\beta$ ,15 $\alpha$ -glycol,<sup>11-15</sup> the structure IX was assigned to the tetrahydroxy-ketone.

Oxidation of IX with Kiliani's reagent gave a hydroxytetraketone, mp 262–266°, C<sub>21</sub>H<sub>28</sub>O<sub>5</sub>,  $[\alpha]_D^{25} +104.3 \pm 6.1^\circ$  ( $c=0.235$ , CHCl<sub>3</sub>). The IR spectrum in chloroform shows absorptions at 3281 (14 $\beta$ -OH), 1747 (15-ketone), 1707 (3- and 12-ketone) and 1697 cm<sup>-1</sup> (20-ketone). Inspection of the IR spectrum in detail showed the presence of a strong hydrogen bonding between 14 $\beta$ -hydroxyl group (3281 cm<sup>-1</sup>) and 20-carbonyl group (1697 cm<sup>-1</sup>), giving further evidence that 17-side chain has the  $\beta$ -configuration. This product having 12-ketone and 17 $\beta$ -methyl ketone grouping proved to be identical with the hydroxy-tetraketone (IV) derived from purprogenin by mixed melting point and comparisons of TLC and IR spectra. These results established the structure of purprogenin to be 3 $\beta$ ,14,15 $\alpha$ -trihydroxy-14 $\beta$ ,17 $\beta$ -pregn-5-ene-12,20-dione (I) as previously proposed by our group.

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### A Novel Rearrangement of a Quinol Acetate<sup>1)</sup>

Quinol acetates<sup>2)</sup> prepared by the reaction of phenols with Pb(OAc)<sub>4</sub> are known to be rearranged to hydroquinones<sup>3)</sup> under conditions of the Thiele reaction (Ac<sub>2</sub>O–concd. H<sub>2</sub>SO<sub>4</sub>). We attempted to apply this method to tetrahydroisoquinolines in order to introduce hydroxyl group to benzene ring and chose corypalline (I) as a starting material.

Unexpectedly, we encountered a novel rearrangement, in which acetoxy group moved to 4-position instead of 5-position (normal rearrangement).

A solution of I (200 mg, ca. 1 mmole) and Pb(OAc)<sub>4</sub> (680 mg, ca. 1.5 mmole) was stirred for 1.5 hour at room temperature, treated with ice-water, basified with NaHCO<sub>3</sub>, and extracted with CHCl<sub>3</sub>. Chromatography of the CHCl<sub>3</sub> layer on neutral Al<sub>2</sub>O<sub>3</sub> (Woelm, elution with CHCl<sub>3</sub>) gave N-methyl-10-acetoxy-6-methoxy-7-oxo- $\Delta^{5,6,8,9}$ -hexahydroisoquinoline (II), mp 118–120° from *n*-hexane, (72 mg, 20%) [Anal. Calcd. for C<sub>13</sub>H<sub>17</sub>O<sub>4</sub>N: C, 62.14; H, 6.82; N, 5.57. Found: C, 62.34; H, 6.94; N, 5.63. NMR  $\tau$ : 7.92 (s., OAc), 7.61 (s., =NMe), 7.06, 6.68 (each d., C-1 *gem.* protons,  $J=12$  cps), 6.34 (s., OMe), 4.20 (s., a vinylic proton), 3.82 (d., a vinylic proton,  $J=1.5$  cps). IR  $\nu_{\max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 1745 (OAc), 1675, 1655, 1630 (dienone)].

A mixture of quinol acetate (II) (275 mg, ca. 1.1 mmole) in Ac<sub>2</sub>O (4 ml) and concd. H<sub>2</sub>SO<sub>4</sub> (0.3 ml) in Ac<sub>2</sub>O (1 ml) was stood for 2 hours at room temperature. After adding crushed ice to the reaction mixture and extracting excess Ac<sub>2</sub>O with ether, H<sub>2</sub>O layer was

- 1) All melting points were uncorrected using Yanagimoto micro melting points measuring apparatus. All NMR spectra were measured at 60 Mc by JNR-C60S spectrometer in CDCl<sub>3</sub> using Me<sub>4</sub>Si as internal standard. Gas-liquid chromatography (GLC) was taken with Shimadzu GC-1C gas chromatograph equipped with a hydrogen flame ionization detector.
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