

by the catalytic desulfurization, and a catalytic reduction of carbon-carbon unsaturated compounds, carbonyl and nitro compounds derivatives etc., in addition to biotin and its related compounds. Moreover, it will be very interesting to investigate the mechanisms of these reductive tritiation whether the tritium atoms introduced into *d*-dethiobiotin are derived from the solvent (water) or the catalyst (Raney nickel).

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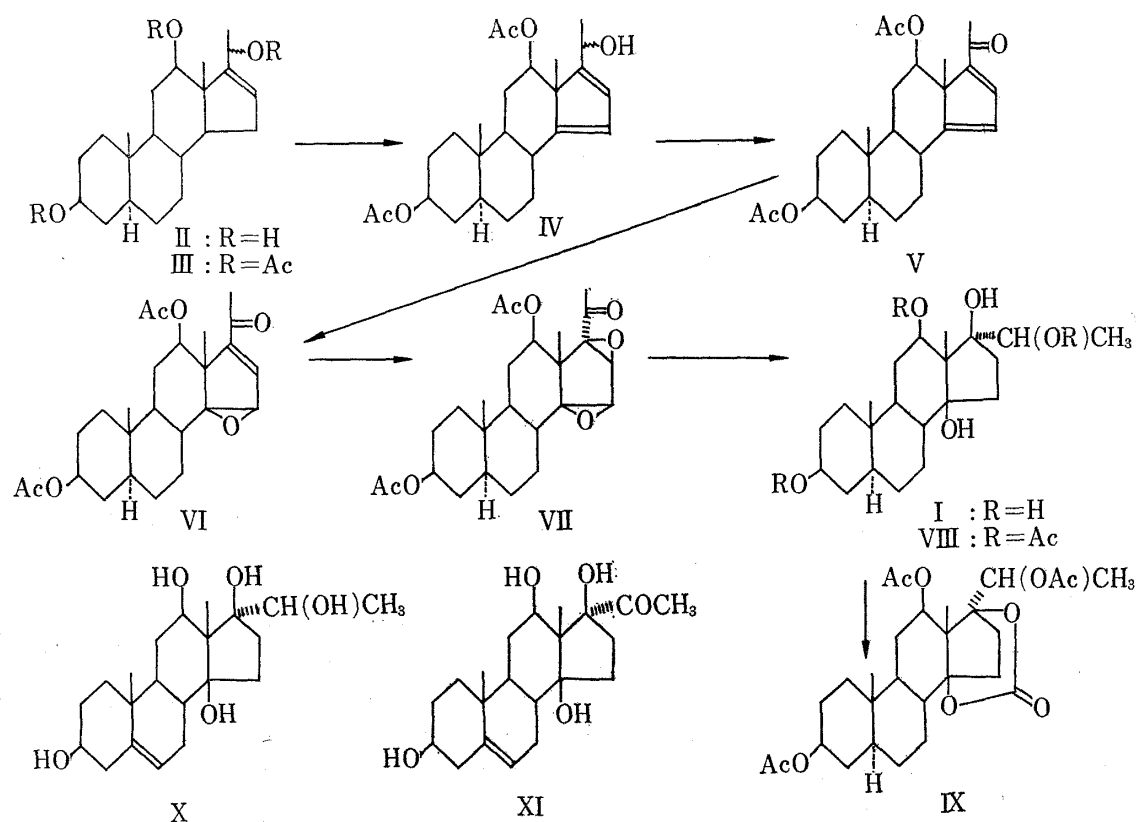
The Structure of Tomentogenin

Previously, Mitsuhashi, *et al.*,¹⁾ proposed the structure $3\beta,12\beta,14\beta,17\beta,20\xi$ -pentahydroxypregnane (I) for tomentogenin isolated from *Marsdenia tomentosa* DENCE (Asclepiadaceae, Japanese name: Kijoran), but there remains some ambiguity for the configuration of C-17 hydroxyl group.

We now wish to describe the establishment of the configuration of the C-17 hydroxyl group by partial synthesis of tomentogenin (I) from hecogenin acetate. Hecogenin acetate was converted to $3\beta,12\beta,20\xi$ -trihydroxy- 5α -pregn-16-ene (II)²⁾ according to the procedures by Cameron, *et al.*³⁾ and by Adams, *et al.*,⁴⁾ and the product, after acetylation with acetic anhydride-pyridine, was treated with dibromodimethylhydantoin⁵⁾ to give $3\beta,12\beta,20\xi$ -trihydroxy- 5α -pregna-14,16-diene 3,12-diacetate (IV), mp 176-178°, $[\alpha]_D^{25} +111.3^\circ$ ($c=0.88$, CHCl_3), UV $\lambda_{\text{max}}^{\text{EtOH}}$ m μ (log ϵ): 262 (3.98), IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3580 (OH), 1740, 1710, 1265, 1240 (OAc). Oxidation of IV with MnO_2 gave the dienone⁶⁾ (V), mp 155-157°, UV $\lambda_{\text{max}}^{\text{EtOH}}$ m μ (log ϵ): 305 (4.30). Epoxidation of V with monoperphthalic acid gave the $14\beta,15\beta$ -epoxide (VI),⁶⁾ mp 150-153°, UV $\lambda_{\text{max}}^{\text{EtOH}}$ m μ (log ϵ): 243 (3.65), and further epoxidation of VI with alkaline- H_2O_2 gave the bisepoxide (VII), mp 146.5-147.5°, ORD ($c=0.17$, MeOH) $[\alpha](\text{m}\mu)$: positive Cotton effect, +173.1° (285) (trough) +358.4° (314) (peak), +45.8° (589), IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1745, 1730, 1240 (OAc), 1710 (methylketone), NMR (CDCl_3) τ : 9.18 (19- CH_3), 8.57 (18- CH_3), 8.05, 8.02 (OAc), 8.00 (21- CH_3), 6.30 (1H, broad singlet, C-15-proton), 6.05 (1H, broad singlet, C-16-proton).

Djerassi, *et al.*,⁷⁾ reported that the Cotton effect of 16,17-epoxy-20-oxopregnane showed the opposite sign to the Cotton effect observed on the corresponding 20-ketopregnane. Since the compound (VII) showed the positive Cotton effect, while $3\beta,12\beta$ -diacetoxy- $14\beta,15\beta$ -epoxy- $5\alpha,17\alpha$ -pregnan-20-one⁸⁾ showed the negative Cotton effect, and since the compound

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(VII) could be converted to the cyclic carbonate (IX), the configuration of the epoxide at the position C-16 and 17 was considered to be 16 β ,17 β -type.

Successive reduction of VII with NaBH₄ and LiAlH₄, gave tomentogenin (I), mp 256.5—259.5°, $[\alpha]_D^{25} +40.0^\circ$ ($c=1.0$, MeOH), which was identical with an authentic sample¹⁾ by a mixed melting point, comparisons of thin-layer and paper chromatography, and IR spectra.

Acetylation of I with Ac₂O-pyridine gave the corresponding triacetate (VIII), mp 290—293°, ORD ($c=0.11$, MeOH) $[\alpha]$ ($m\mu$): a positive plane curve, +354.7° (210), +87.3° (230), 0° (300). Treatment of VIII with 30% phosgene-chloroform gave quantitatively a cyclo carbonate (IX), mp 261—263°, ORD ($c=0.11$, MeOH) $[\alpha]$ ($m\mu$): a negative Cotton effect,⁹⁾ 0° (210), -353.9° (228) (though), -88.4° (300), IR $\nu_{\max}^{\text{Nujol}}$ cm⁻¹: 1755, 1750, 1735, 1725, 1245, NMR (CDCl₃) τ : 9.18 (19-CH₃), 8.68 (18-CH₃), 8.64 (21-CH₃, d, $J=6$ cps), 7.97, 7.89 (OAc). These characteristics correspond to the formula IX.

These results established the structure of tomentogenin to be 3 β ,12 β ,14 β ,17 β ,20 ξ -pentahydroxy-5 α -pregnane (I).

Since the correlation of tomentogenin and utendin (X) and pergularin (XI) was reported by Abisch, *et al.*,¹⁰⁾ and by Mitsuhashi, *et al.*,¹¹⁾ the structures of utendin and pergularin were established to be the formula X and XI, respectively.

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Isolation of Insect Moulting Substances from *Osmunda japonica* and *Osmunda asiatica*

During our screening tests on vegetable materials by means of bioassay, it has been noticed that the crude extracts of ferns especially show insect-moulting hormone activity in high frequency. Therefore, we have investigated the constituents of varieties of ferns whose extracts exhibit the biological activity, and isolated a number of active steroids.¹⁾ We have further observed that the methanol extracts of *Osmunda japonica* THUNBERG and *O. asiatica* OHWI²⁾ (Osmundaceae) grown in Japan are positive in the *Sarcophaga* assay. The polar fraction of each extract was submitted to column chromatography resulted in the isolation of the following substances: ponasterone A, mp 266—268°, ecdysone, mp 241.5—242°, and ecdysterone, mp 239—240°, from *O. japonica*, and ponasterone A, mp 260—261°, ecdysone, mp 243—244°, and ecdysterone, mp 242—243°, from *O. asiatica*.

It is of interest to note that the known plant sources of ecdysone, the first insect-moulting hormone, are all ferns, *Pteridium aquilinum* KUHN,³⁾ *Polypodium vulgare* LINNÉ,⁴⁾ and *Lemma-phyllum microphyllum* PRESL.^{1a)}

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