

[Chem. Pharm. Bull.]
18(9)1724-1730(1970)]

UDC 547.68.02 : 548.73 : 615.332.011.5

Studies on Pillaromycin A. V.¹⁾ The X-Ray Analysis of Pillaronone Monobromoacetate

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(Received December 25, 1969)

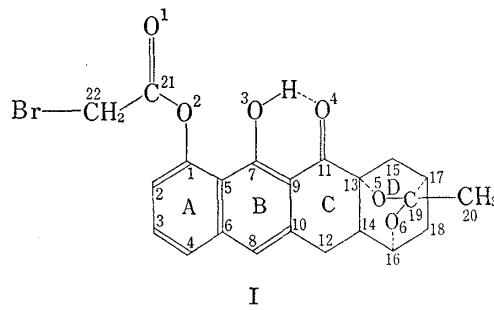
In order to confirm the chemical and stereochemical structure of pillaronone, a derivative of the aglycone obtained from pillaromycin A, an X-ray analysis of its monobromoacetate has been carried out. Pillaronone monobromoacetate crystallizes in the orthorhombic system, space group $P2_12_12_1-D_{2h}^4$, with four molecules of $C_{22}H_{19}O_6Br$ in a unit cell. Application of a three-dimensional Patterson superposition method facilitated the elucidation of the structure which is given in formula (I). The final R -value was 0.133. The absolute configuration was determined using the anomalous dispersion of the bromine atom. The molecular structure in the crystal was also discussed.

Pillaromycin A,³⁾ $C_{28}H_{28}O_{11}$, is an anti-tumor antibiotic compound produced by *Streptomyces flavovirens* No. 65786. The present X-ray work⁴⁾ has been carried out in order to establish the chemical and stereochemical structure of pillaronone, $C_{20}H_{18}O_5$, which is prepared by catalytic reduction of pillaromycinone. A preliminary study indicated that the monobromoacetate of pillaronone was the most promising for this purpose.

Experimental

Preparation of the Crystal—To a solution of 100 mg of pillaronone in 2 ml of benzene was added dropwise 0.2 ml of monobromoacetic acid anhydride. After being stirred for 6 hr at 70° to complete the reaction, the mixture was poured into ice water and extracted with benzene. The benzene solution was washed with water and dried over anhydrous sodium sulfate. The solution was concentrated under a reduced pressure, petroleum ether was added and crystals separated were recrystallized several times from a mixture of ethyl acetate, ethyl ether and petroleum ether to afford yellow prisms, mp 194°. *Anal.* Calcd. for $C_{22}H_{19}O_6Br$: C, 57.53; H, 4.17; Br, 17.40. Found: C, 57.56; H, 4.14; Br, 17.37. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1772 (s), 1630 (vs). UV $\lambda_{\text{max}}^{\text{EtOH}}$ m μ ($\epsilon \times 10^{-3}$): 270 (37), 290 (6.8), 300 (6.8), 395 (6.5).

Measurements of X-ray Diffraction—Oscillation and Weissenberg photographs were taken about the two principal crystal axes, using CuK α radiation. The specimens used had the following cross-sections at right angles to the axis of rotation: 0.2×0.3 mm for the c axis rotation, 0.2×0.2 mm for the a axis. Intensities of reflections were visually estimated from multiple film integrating Weissenberg photographs (8 layers about the c axis and 12 layers about the a axis, respectively) and obtained 2000 independent data (excluding zero intensity reflections). The linear absorption coefficient for CuK α radiation was estimated to be $\mu = 21.84 \text{ cm}^{-1}$. Thus the absorption and extinction errors were neglected. The absolute structure factors were deduced from these intensity data by the ordinary procedure, using a NEAC 2206 electronic computer.



- 1) Part IV: M. Asai, E. Mizuta, K. Mizuno, A. Miyake and S. Tatsuoka, *Chem. Pharm. Bull. (Tokyo)*, **18**, 1720 (1970).
- 2) Location: a) *Juso, Higashiyodogawa-ku, Osaka*; b) *Uegahara, Nishinomiya*.
- 3) M. Asai, *Chem. Pharm. Bull. (Tokyo)*, **18**, 1699 (1970).
- 4) Preliminary report, H. Matsuda, M. Nishikawa, K. Kamiya, M. Asai, Y. Tomie, I. Nitta, S. Yamamura, Y. Hirata; presented at 9th Symposium on the Chemistry of Natural Products, Osaka, Oct., 1965.

Result and Discussion

Unit Cell and Space Group

It has been found from X-ray examination that the crystal belongs to the orthorhombic system with the unit cell parameters, $a=12.46$, $b=19.54$, $c=7.85$ Å.

The space group was uniquely determined to be $P2_12_12_1-D_{2h}^4$. Judging from the cell size and the space group, it is clear that the unit cell contains four molecules as the calculated density becomes a reasonable value, 1.596 g cm $^{-3}$ ($D_0=1.591$ g cm $^{-3}$).

Determination of the Structure

As the first step of the structure analysis, a three-dimensional Patterson function $P(uvw)$ was calculated. The coordinates of the bromine atom were determined without ambiguity from

the three Harker sections, one of which at $v=1/2$ is shown in Fig. 1. With phase angles calculated from these coordinates, a three-dimensional Fourier synthesis was calculated. In co-operation with this analysis, a minimum function diagram was obtained with four superposition of bromine vectors. In spite of quite different principles, the results obtained by these two methods were fully consistent with each other and the positions of all light atoms except hydrogen were determined undoubtedly.

The refinement of the atomic positions was carried out using the least squares method, and in accordance, R -value fell to 0.133. Maps of the final electron density distribution are shown in Fig. 2a and 2b. The final x , y and z atomic parameters and temperature factors are listed in Table I. The observed and the calculated structure factors are listed in Table II.

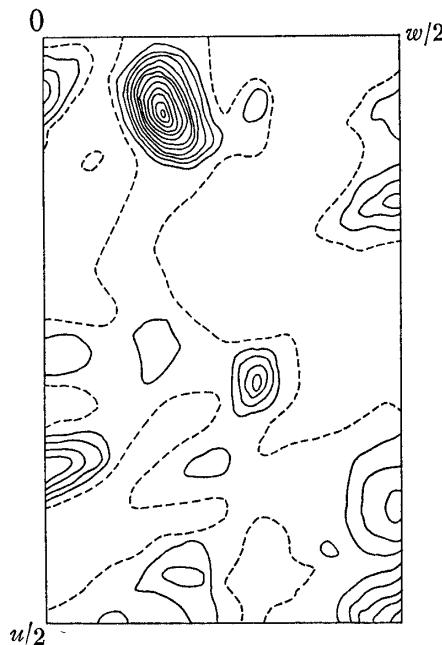


Fig. 1. Harker Section at $v=1/2$

Of the 20 pairs of reflections for which the intensity differences between $I(hkl)$ and $I(\bar{h}\bar{k}\bar{l})$ were expected to be discernible, 14 pairs showed significant differences in the 1st layer Weissenberg photographs. The results are shown in Table III. The comparison between observed and calculated intensities indicated that the assumed configuration was actually correct. The absolute configuration was then established as (I). In this paper, all figures are drawn in such a way that they represent the correct configuration.

Molecular and Crystal Structure

The bond lengths and angles, calculated from the atomic parameters of Table I, are shown in Fig. 3 and Table IV. The mean values of bond lengths are 1.40 Å for aromatic carbon bond, 1.53 Å for C-C single bond, 1.42 Å for C-O single bond and 1.24 Å for C=O double bond. These values seem to be quite reasonable.

The rings A and B form a naphthalene skeleton and C(11), C(12), C(13), O(2), O(3) and O(4) atoms are nearly on this plane. The OH group attached to the ring B seems to form a strong six-membered ring hydrogen bond with the carbonyl group at C(11), the bond between O(3) and C(7) being somewhat contracted.

The juncture between the ring C, which is puckered at C(14), and the ring D, which is of a chair form, is in a *trans*-fusion. The atoms, O(5), O(6) and C(19) are axially bonded to

TABLE I. The Final Atomic Coordinates and Temperature Factors

Atom	<i>a/x</i>	<i>b/y</i>	<i>c/z</i>	<i>b</i>
Br	0.0266	0.8425	0.1649	4.65
O(1)	0.4733	0.5466	0.1030	4.82
O(2)	0.3614	0.0445	0.2810	3.31
O(3)	0.3253	0.9376	0.4792	3.31
O(4)	0.2314	0.1672	0.1442	4.29
O(5)	0.2292	0.3042	0.1965	3.21
O(6)	0.1911	0.4144	0.1084	3.57
C(1)	0.3982	0.9960	0.1536	3.01
C(2)	0.4367	0.0200	0.0084	4.00
C(3)	0.0372	0.0239	0.3782	4.32
C(4)	0.0541	0.0945	0.3987	3.93
C(5)	0.3808	0.9248	0.1877	2.78
C(6)	0.4038	0.8792	0.0531	3.03
C(7)	0.3439	0.8963	0.3464	3.10
C(8)	0.3837	0.8078	0.0728	3.79
C(9)	0.3234	0.8246	0.3578	2.69
C(10)	0.3463	0.7806	0.2125	3.33
C(11)	0.2179	0.2038	0.0177	3.73
C(12)	0.3176	0.7044	0.2200	3.63
C(13)	0.2447	0.2795	0.0254	2.85
C(14)	0.3275	0.6785	0.4036	3.09
C(15)	0.1352	0.7066	0.4950	3.37
C(16)	0.2950	0.6002	0.4262	3.19
C(17)	0.3636	0.3691	0.0524	3.28
C(18)	0.1780	0.5888	0.4020	3.53
C(19)	0.2720	0.3742	0.1797	3.38
C(20)	0.2969	0.3998	0.3591	4.52
C(21)	0.4323	0.0654	0.3921	3.32
C(22)	0.3769	0.1141	0.5059	4.91

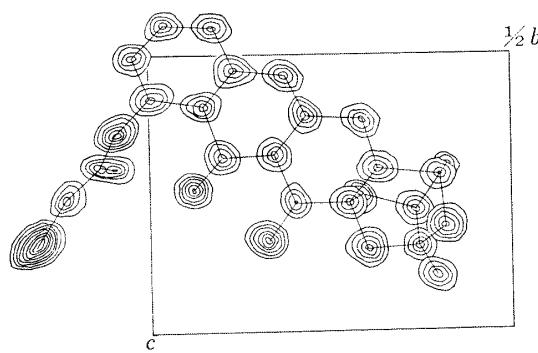


Fig. 2a. The Final Three-dimensional Electron Density Distribution shown by Means of Superimposed Contour Section Projected on (100)

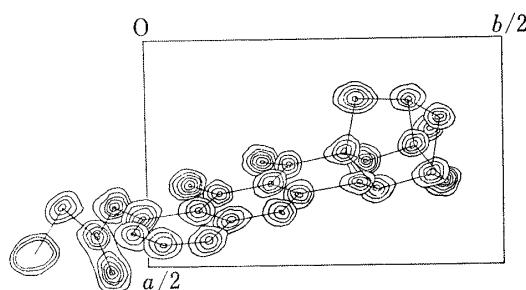


Fig. 2b. Projected on (001)

the ring D and form a ketal, leading to a cage structure of this part of the molecule. Thus considerable steric strains seem to exist at C(15) and C(18) so that the bond angles C(13)-C(15)-C(17) and C(16)-C(18)-C(17) are smaller than 100°.

TABLE II. Observed and Calculated Structure Factors

K	L	F _O	F _C	H K L	F _O	F _C	H K L	F _O	F _C	H K L	F _O	F _C	H K L	F _O	F _C	H K L	F _O	F _C	H K L	F _O	F _C	H K L	F _O	F _C					
4	0	26.9	13.4	03 6	13.1	11.0	120 2	7.4	5.5	112 8	3.1	1.9	214 4	4	23.9	20.6	391	40.0	334.5	312 6	11.1	10.1	141	340.3	35.4	54	056.9	56.4	
4	0	132.6	145.8	4 6	24.8	20.1	21 2	9.7	11.6	14 8	6.9	7.2	15 4	11.4	12.6	101	39.1	134.8	13 6	15.1	13.6	2	345.2	42.5	5	025.1	18.4		
0	6	52.3	57.0	5 6	19.5	14.4	22 2	6.6	5.8	15 8	3.7	4.8	16 4	8.4	9.7	111	23.6	619.8	18 6	20.6	18.6	3	319.9	19.2	6	012.6	6.3		
8	0	88.8	82.1	6 6	48.6	47.3	23 2	15.5	12.8	1 9	6.1	5.2	14 4	14.7	10.3	121	11.1	118.5	18 6	10.0	9.2	4	317.9	16.8	7	076.7	74.9		
10	0	34.2	9.0	8 6	6.5	5.7	24 2	10.9	10.5	3 9	10.9	14.7	19 4	16.8	13.6	131	26.0	24.4	0 7	27.1	31.7	5	351.6	58.8	8	014.7	20.0		
12	0	19.1	23.7	9 6	10.5	7.1	0 3	19.9	14.2	7 9	9.5	13.7	20 4	7.9	9.0	141	7.7	7.4	1.8	17	23.6	2.2	6	326.6	26.7	9	046.9	38.1	
14	0	7.5	8.5	10 6	24.9	19.4	1 3	36.6	34.0	8 9	2.8	5.7	24 4	5.9	5.6	151	23.9	92.2	1	27.1	12.2	2.4	7	319.2	14.2	10	012.9	18.1	
16	0	34.0	35.6	11 6	4.5	6.9	2 3	9.5	9.1	9.9	4.5	7.0	22 4	4.4	9.3	161	24.6	62.6	1	37	6.5	5.3	8	326.6	18.0	11	031.0	23.4	
18	0	20.2	20.2	12 6	17.9	16.2	3 3	85.7	83.8	1 0	4.6	7.7	0.5	1.7	3.4	171	5.3	6.1	1	47	22.5	2.8	9	329.5	25.2	12	013.0	10.3	
20	0	11.5	12.1	14 6	14.3	13.6	4 3	46.4	33.3	2 0	3.1	5.0	15 2	21.9	19.4	181	19.0	18.0	0	5.7	8.4	8.6	10	340.3	39.0	13	040.4	54.9	
22	0	12.4	13.0	16 6	11.5	11.6	5 3	25.5	19.7	3 0	3.6	7.0	2 5	33.2	22.9	191	4.7	7.8	6	6.7	4.1	3.3	11	325.2	23.8	14	019.6	17.2	
24	0	7.5	7.1	17 6	3.4	1.7	6 3	34.6	7.35.9	0 0	55.6	38.4	3 5	29.5	28.5	201	13.3	31.1	2	7.7	11.7	16.0	12	315.3	12.7	15	016.6	15.9	
1	1	54.7	41.7	14 6	12.0	12.6	7 3	73.7	70.5	1 0	57.2	59.0	5 5	41.2	41.5	211	6.1	7.1	8	8.7	4.6	4.3	13	320.0	21.8	16	014.1	10.3	
2	1	100.7	136.4	20 6	2.8	3.4	8 3	15.7	13.7	2 0	87.7	86.6	5 5	40.2	36.1	221	6.6	8.5	9	8.7	8.4	10.0	14	331.2	29.6	17	017.9	22.6	
3	1	39.2	42.3	2 7	39.8	53.4	9 3	70.5	56.9	2 0	18.9	16.0	6 5	6.1	8.4	231	6.6	6.6	10	17.8	21.6	15	318.3	15.1	18	014.1	14.6		
4	1	94.8	114.5	3 7	7.7	9.6	10 3	32.1	61.4	7 4	6.5	7.3	5 5	5.9	9.1	241	5.1	7.0	1	7	6.1	7.6	16	3	9.8	10.4	20	02.5	1.8
5	1	39.7	32.8	4 7	6.3	6.6	11 3	15.7	11.4	5 0	34.5	35.9	8 5	48.4	47.8	023	42.1	5	12	8.7	12.0	17	326.3	29.4	21	0 8.3	8.4		
6	1	78.7	78.7	5 7	19.7	24.8	12 3	32.4	0.19.2	6 0	102.8	81.0	5 0	27.4	19.7	127	7.7	6.6	10	18.0	2.1	3	32	1.0	23	0 3.5	6.9		
7	1	18.1	15.4	6 7	3.3	2.1	13 3	34.1	29.9	8 0	43.7	34.8	10 5	24.2	22.2	222	11.8	17.8	1	14.7	11.5	0.1	0.9	19.3	1.2	1	02.4	2.3	
8	1	7.3	12.3	7 7	7.8	8.2	14 3	11.5	7.7	8 0	88.0	75.7	11 5	16.9	16.4	323	6.6	13.1	1	15.7	5.0	4.7	21	31.1	10.3	0	166.1	15.2	
9	1	60.8	59.4	8 7	17.1	16.7	15 3	12.2	40.2	10 0	108.4	107.9	12 5	30.5	31.6	420	6.4	12.0	1	16.7	6.4	5.7	22	31.1	0.3	13.8	1	145.8	40.0
10	1	18.3	22.2	9 7	10.0	11.6	16 3	31.1	27.7	11 0	10.8	15.8	15 0	17.3	13.7	525	7.47.9	17.7	3.1	5.4	0	52.8	55.1	2	15.7	15.2			
11	1	22.7	28.2	10 7	6.3	5.8	17 3	7.7	8.7	12 0	39.8	38.7	14 5	28.1	26.8	628	5.7	0.8	18.4	8.12.6	1	4.2	29.8	27.4	3	118.9	28.2		
12	1	24.7	24.9	11 7	11.7	11.4	18 3	5.4	6.5	13 0	14.3	12.3	15 5	7.8	6.5	722	3.21.0	18.7	1.8	7.5	5.8	2	427.4	24.9	9	4 124.5	20.9		
14	1	52.6	35.9	12 7	13.5	11.5	19.3	18.3	18.0	14 0	13.2	10.4	16 5	4.3	2.3	822	5.42.4	3	8.8	9.8	10.0	3	43.3	36.3	5	149.7	45.6		
16	1	13.2	10.0	13 7	10.3	10.7	21 3	4.1	4.6	16 0	27.6	29.0	18 5	6.8	6.6	922	8.72.5	3	8.15	3.6	2.4	26.1	6	143.6	43.7	7	143.6	43.7	
17	1	16.0	15.6	14 7	6.5	6.4	23 3	7.5	9.5	9.3	17.0	14.9	20 5	13.3	13.3	102	22.0	0.21.3	4	8.4	6.1	5	44.6	4.9	3	7	141.8	35.0	
20	1	11.1	14.7	18 1	15.3	18.4	0 4	36.8	37.0	18 0	6.7	10.3	2 5	9.7	9.8	112	23.8	8.4	1.4	21.9	7.9	8	112.7	14.5					
21	1	10.5	12.1	18 1	2.8	6.6	9.8	1 4	44.8	35.2	19 0	6.6	7.6	6	16.5	20.9	122	22.3	31.5	8	8.8	4.10.5	7	43.4	27.1	9	160.1	61.2	
23	1	5.0	6.4	3 8	13.6	8.0	2 2	32.4	33.2	20 0	7.3	10.0	1 6	25.8	27.6	132	22.1	5.7	18.7	15.1	0.5	0.5	28.5	28.5	5	10.5	28.5		
24	1	6.2	10.1	4 8	13.5	14.4	3 4	36.6	30.3	22 0	16.4	16.6	2 6	22.7	24.2	142	4.7	6.4	2	14.4	2.4	2.4	2	35.8	11.1	2	143.2	42.3	
0	2	33.7	13.3	5 8	6.6	8.1	4 4	34.7	28.7	23 0	3.8	5.2	8	1.5	51.5	15	25.1	54.9	1	8.9	10.8	10.7	10	43.1	38.9				
1	2	21.0	7.0	6 8	6.3	6.6	5 4	74.6	74.9	24 0	2.0	1.8	4	6	24.8	24.3	162	21.3	9.14.0	10	8.1	11.1	1	30.3	12	1	140.1	38.9	
2	2	10.6	12.9	3 2	3.3	2.9	1 4	54.2	44.1	21 4	21.1	11.6	14 1	11.1	11.1	172	22.1	16.0	1	16.7	1.1	1.1	1	27.2	24.6	4	127.2	24.6	
3	2	10.6	12.9	13 0	5.4	5.2	1 4	54.2	44.1	21 4	21.1	11.6	14 1	11.1	11.1	172	22.1	16.0	1	16.7	1.1	1.1	1	27.2	24.6	4	127.2	24.6	
4	2	43.5	40.3	5 9	5.4	5.2	1 4	55.0	62.5	16 1	6.1	7.6	17.2	17.2	17.2	102	22.0	1.23.9	3	5.5	3.4	2.8	6	26.9	66.7	7	142.8	37.0	
5	2	4.0	5.0	48.1	39.2	0 5	0 55.5	62.5	16 1	6.1	7.6	17.2	17.2	17.2	102	22.0	1.23.9	3	5.5	3.4	2.8	6	26.9	66.7	7	142.8	37.0		
6	2	32.7	26.6	7 9	6.9	8.3	16 4	4.9	8.0	16.7	4.0	6.7	14.7	14.7	14.7	43	3.3	2.75.7	2	5.6	7.4	0 28.5	81.0						
7	2	13.5	17.4	17 0	27.5	22.8	14 5	8.9	8.8	42	36.2	28.4	17 7	8.1	8.1	43	5.7	3.5	2.7	24.9	1.1	1.1	1	22.7	21.5				
11	3	48.0	46.3	18 0	29.8	30.1	15 5	11.1	10.6	5 2	40.2	28.7	0 8	5.1	5.1	0 4	7.2	2.5	17.0	11.1	0.3	31.1	11.5						
13	3	21.7	16.6	19 0	15.3	12.5	16 5	19.0	17.6	6 2	18.3	19.7	18 1	18.5	18.7	143	5.7	1.4	57.6	1.2	12.5	2.2	8	8.1	11.5				
14	3	22.2	14.5	20 0	12.9	15.1	17 5	12.0	7.1	7 2	63.4	62.8	2 8	13.5	13.4	20	2.5	37.6	36.7	2	20	6.9	8.3	8.3	8.3	8.3			
15	3	49.9	51.3	21 0	12.1	10.2	19 5	9.1	8.4	9 2	30.7	31.3	13 0	12.4	11.4	34	19.1	3.81.5	0	2.1	22.7	21.5	2	22.7	21.5				
17	3	22.5	23.6	22 0	4.9	6.2	21 5	3.1	2.6	10 2	37.4	33.0	4 8	7.1	8.1	44	23.8	19.1	22.1	21.1	0	22.7							

TABLE II. (continued)

MKL	FO	FC	HNL	FO	FC	HNL	FO	FC	HNL	FO	FC	HNL	FO	FC	MKL	FO	FC	KL	FO	FC	KL	FO	FC		
865	24.5	22.8	652	30.4	427.0	678	4.5	6.271	527	627.0	4.819	2	2.0	2.393	114	312.5	104	20.0	20.0	101	12.2	7.4	2313	715.8	
751	16.6	14.1	62	22.2	216.4	812.6	12.2	510	71.4	20.10	11.1	4.1136	954.9	100	14.9	913.7	111	17.4	21.0	53	9.6	10.6	4316	419.0	
851	12.1	12.4	72	22.9	17.7	108	5.8	5.8	4.517	516.3	0	3.39	8.425	5127	453.0	110	18.7	822.0	121	10.3	8.7	4.1	3.5		
95	8.4	8.7	82	38.0	352.7	118	5.6	8.9	5.84	348.9	1	3.22	8.16.7	6115	516.3	120	10.3	510.1	131	6.1	5.6	5	5.40.8		
105	20.8	20.6	93	110.1	93.8	28	5.2	6.4	6.512	818.3	2	3.21	1.18.5	7116	515.1	170	11.3	510.5	141	9.0	8.8	6.3	9.11.2		
115	19.0	17.6	102	23.8	815.1	918.3	14.9	7.9	718.3	317.2	3	3.35	5.8	9.9	8122	225.4	180	7.2	3.6	151	7.2	7.7	73	7.2	
125	21.6	23.6	112	16.0	14.4	29.5	9.0	6.521	629.0	4	3.29	7.29.8	91	4.1	7.4	1120	287.5	171	1.6	3.3	8.3	6.9	4.8		
135	17.9	15.9	122	10.0	4	39.3	3.3	3.9	516.9	17.5	5	3.81	7.6	10113	7.7	7.6	2114	715.6	181	2.2	2.6	9.3	3.4	4.7	
145	12.5	11.5	132	18.8	17.1	49	4.6	6.4	6.410	515.1	11.9	6	3.34	8.38.0	11132	731.1	3126	932.8	02	35.941	7	10	313.0	212.7	
155	8.4	16.0	142	22.5	123.3	H59	4.2	6.311	516.8	417.7	7	3	3.9	7.2	12116	311.7	41	23.220	8.6	12	13.0	22.5	3.3		
165	12.5	12.1	152	17.8	15.9	69	6.6	0.012	512.5	510.0	8	3	3.9	9.1	131	82.6	6.7	5110.9	9.8	22	4.3	2.3	13.3	5.6	
175	8.	8.	162	22.2	119.7	27	0.6	15.9	913.5	512.0	101.0	9	9.3	6.2	2.1	1518.7	819.0	6114	14.317	3.4	32.16.0	16.5	24	9.6	13.8
195	4.8	5.6	172.3	15.8	3054	2.3	53.414	812.2	512.7	5	10	13.6	818.9	19.9	61	3.8	5.1	71	30.313	8	62	19.4	21.1	3.4	
205	3.3	1.1	182	21.1	215.0	40	5.5	12.116	5.5	2.6	7.0	11	320	3.18.0	17114	013.6	8121	1.6	8.4	414.9	11.4	8.8			
215	5.2	7.1	212	8.0	4.8	5037	3	29.017	5.8	6.8	8.8	12	317	0.17.9	181	3.8	6.5	9123	1.21.6	6	16.8	318.2	5.4		
26	12.3	11.8	03	49.0	052.0	6295	2.5	23.018	5.5	4.5	5.2	13	3	4.7	3.8	201	6.0	6.9	11115	6.10.7	72	18.6	21.2		
36	27	26.9	1.9	35.9	35.85	7037	9.3	9.6	0.6	5.4	3.2	14	3	5.4	7.2	0280	953.3	12	7.1	8.8	82	21.12.8	7.4	6.2	
46	14.5	16.7	23	17.6	61.8	8052	2.4	27.47	7	16.20	223.4	15	310.2	8.7	1237	335.9	131	18.2	21.1	92	18.017	8	8.4	8.42.1	
56	26.3	23.8	33	19.3	31.1	9355	2.3	35.7	6	612	515.1	16	311	9.12.9	2223	282.7	151	12.4	16.8	102	10.314	8.2	9.4	8.6	
66	11.1	7.7	4.7	35.6	939.0	11024	2.5	5.5	363	23.4	9	17	3	9.7	7.0	3217	92.1	16114	19.917	2.2	112	5.4	5.4	3512	164.1
76	25.5	28.4	53	36.2	23.2	12013	1	7.6	4616	510.5	18	3	6.3	7.9	4236	439.9	191	5.9	9.6	122	14.515	5.9	9.5	5.7	
86	20.5	19.6	63	33.8	84.8	13027	5.5	29.5	616	21.0	19	3	3.9	2.9	5218	81.6	1228	0.230	132	10.010	5	6.5	3.6	4.5	
96	20.1	19.9	73	24.8	25.5	140	5.9	7.3	616	919.5	20	3	5.0	7.6	6230	0.317	2210	0.12.4	142	2.8	4.5	7.5	0.7		
116	11.1	9.9	83	6.6	4.4	1501.5	6.5	15.5	7.6	7	0425	3.19.8	7219	915.2	32	5.5	6.2	152	7.1	9.0	10	5.5			
126	23.6	17.6	93	21.1	416.1	16013	4	18.2	8	6.1	5.1	5.8	1	4.8	8.4	829	9.31.4	4219	22.2	1	03	11.6	6.8	11.5	3.5
136	7.8	10.0	102	30.3	27.2	18014	0.9	9.4	618	11.8	3	20	4.0	2.1	210	3.21.2	9218	919.5	52	12.0	81.8	1.1	9.6	8.91.6	
146	13.3	13.7	113	327	223.8	19013	1	13.9	12.6	9.5	7.8	3	418	2.12.1	10226	4.07.6	62	5.3	3.0	23	16.1	21.9	2	012	883.8
156	9.8	7.1	123	50.0	364.9	21010	5.9	5.13	6	5.5	6.7	4	424	8.26.2	11223	819.5	7211	9.12.0	3.0	33	9.0	6.7	113	018.9	
166	8.8	8.9	13.9	28	61.7	0123	7.6	15.7	6.14	6.4	4.0	6.8	5422	1.21.4	12216	618.4	82	26.7	3.05	43	17.4	21.1	8	6.0	
176	8.7	8.6	14.5	6.3	6.7	1176	7.8	84.9	15.6	7.9	8.0	6	427	8.30.1	13218	919.5	92	20.4	11.6	14.5	7.7	7.0	2.02	2.2	
186	5.3	6.2	152	9.3	9.8	7.7	21	5.5	8.4	616	8.4	4.0	4.0	7.8	14.8	12111	6.13.7	12211	6.13.7	13	7.0	7.7	7.6		
17	10.3	12.0	173	2.8	6.6	4131	3	29.8	1.1	711	8.12.0	1	4	20	3.2	1	52	6.8	7.6	83	10.4	21.1	2	120.6	
27	18.3	18.2	183	9.9	9.3	5157	0.8	59.9	2	7	6.3	6.6	10	415	5.16.5	172	5.8	4.6	172	8.4	9.9	93	9.010	5	
37	6.8	8.0	193	5.6	5.2	616	0.8	28.4	4	412	818.9	1	4	23	8.1	6.6	182	8.8	9.4	8.5	0.011	5	4.1	13.5	
47	7.7	8.9	203	3.1	2.5	7146	7.4	45.4	5	7	8.6	10.8	12	41	6.13.0	192	9.4	8.5	0320	12.6	7.7	2.8	4.4		
57	17.0	20.7	213	7.1	8.8	8140	3	31	6.7	8.4	6.6	14	49	1.11.0	0315	1.9	1.1	128	28.5	6	123	21.3	3.2		
67	19.0	19.5	8417	21.9	5.5	9128	7.0	20	0	8712	512.1	1	4	49	9.19.5	1310	1	7.3	2323	2.12.5	133	8.11.0	7		
77	9.3	12.1	1434	31.1	10134	5.2	27.8	9	7.7	6.3	6.5	14	510	12.8	2327	220.8	33	8.9	3.8	143	12.21.9	9	9.1	5.5	
87	8.0	9.0	24	27.3	31.1	1123	2.8	26.6	10	7.6	5.8	8.3	16	8	1.5	5.1	83	8.18.6	153	1.8	5.4	5.0	3.0	101	3.8
97	10.8	12.6	31	29.7	26.6	12110	9.3	12.1	7.1	6.2	7.7	17	4	6.5	6.2	513	925.5	53	11.2	8.1	04	13.915	1	11	6.2
107	7.1	6.1	41	24.5	22.8	143	5.4	31.1	7.6	3.4	12.7	1	4	3.2	2.7	5321	11.54	63	26.6	42.7	3	4.5	1.4	2.6	
117	11.1	12.6	322	7.19	3.3	14139	1	35.3	313.7	3.6	5.3	5	516	5.18.5	6311	814.4	73	15.810	4	34	8.7	6.9	0212	8.81.2	
127	5.5	6.4	613	13.1	21.3	152	9.1	21.5	7.6	716.0	2.2	5	1515	6.12.6	7315	212.6	83	18.211	6.17.7	22	5.2	4.6			
137	9.8	11.1	7420	7.16	9.5	17123	9.9	22.8	2	8	2.7	5.5	2526	8.23.9	6333	13.84	93	9.3	9.8	132	23.1	3.3			
147	8.3	8.0	154	5.4	8.4	2230	2.3	3	2.5	6.2	6	12	513	2.10.2	183	7.2	9.6	4423	2.60.7	1	05	7.5	6.8		
158	9.1	18.4	14114	15.1	15.3	5230	30	2	2.0	29.3	30.6	1	416	2.20.8	8146	218.0	74	7.3	4.8	25	9.710	3.5	6.3	8.2	
168	6.5	9.2	819	3.6	25.28	4	8339	3.8	8.0	4	27.226	2.8	17	5.6	8.4	1415	0.616.3	80	10.4	9.912	33	1.4	1.1	8.3	
178	3.3	5.1	051	18.2	16.7	7235	3.4	32.0	8	4	3.2	3	618	6.18.6	2415	817.5	94	7.7	5.7	45	9.4	5.8	9.3	7.6	
199	9.1	9.1	1539	0.41	5	8237	2.8	5.8	5056	863.0	1	621	2.22.1	91521	1.22.2	10412	2.19.5	24	14.417	0.7	124	10.711	3.4		
209	8.8	8.3	83	35	22.8	1010	5.0	20	7.30	13.6	3	6	9.3	8.0	3426	22.95	108	5.0	5.7	12.4	3.4	8.10.8			
30	6.5	6.5	492	23.0	25.2	1236	9.3	10.9	3.9	2029	3.2	1	610	5.26.7	3421	7.19.8	124	8.1	8.8	6	6.510.4				
40	8.2	10.0	201	31.7	16.6	1223	2.2	23	5.13	3.0	2.6	4	612	7.5	3421	2.10.2	125	5.25.6	7.9	5010.4	8.9				
50	6.5	6.9	422	23.0	22.3	1230	7.3	2.9	8.1	210	3.13.6	1	613	8.19.6	3416	2.10.2	06	4.1	6.4	01	5.9	9.4			
60	5.7	6.3	413	21.5	21.2	1521	6.0	6.1	6.17.4	405	1.1	5	5.2	6.6	5151	5.18.8	7511	6.15.8	16	9.7	9.1	26	1.4	2.2	
70	5.0	1.9	415	35.3	31.5	1518	4	3.2	2.0	6.16	8.9	174	4.5	4.6	5.1	516	4.16.4	36	9.3	9.9	41	8.0	1.7		
80	31.4	31.2	2153	31.9	31.7	5107	0.7	2.8	34.8	512.0	20	2	7.1	0.6	105	6.1	3.2	46	3.9	3.9	51	5.5	8.6		
90	26.1	30.4	816.5	52.8	2.3	3177	2.7	13.9	21	7	4.1	3.3	07	7.1	5.9	10412									

TABLE III. Comparison of the Observed and Calculated Intensity Differences
Used for the Establishment of Absolute Configuration

h	k	l	$F_c^2(h\bar{k}l)/F_c^2(hkl)$	$I_o(h\bar{k}l)/I_o(hkl)$	h	k	l	$F_c^2(h\bar{k}l)/F_c^2(hkl)$	$I_o(h\bar{k}l)/I_o(hkl)$
11	3	4	0.627	< 1	8	9	3	2.215	> 1
13	8	1	1.150	> 1	13	2	3	0.824	< 1
11	9	1	0.805	< 1	15	5	1	0.781	< 1
11	2	2	2.057	> 1	11	9	1	0.805	< 1
11	1	3	0.858	< 1	11	10	1	0.776	< 1
13	2	3	0.824	< 1	10	5	1	1.078	> 1
13	1	3	0.891	< 1	3	20	2	0.950	< 1

TABLE IV. Bond Angles

C(1)-O(2)-C(21)	116°	C(10)-C(12)-C(14)	109°
C(13)-O(5)-C(19)	100	O(5)-C(13)-C(11)	109
C(16)-O(6)-C(19)	102	O(5)-C(13)-C(14)	108
O(2)-C(1)-C(2)	118	O(5)-C(13)-C(15)	102
O(2)-C(1)-C(5)	124	C(11)-C(13)-C(14)	111
C(2)-C(1)-C(5)	124	C(11)-C(13)-C(15)	113
C(1)-C(2)-C(3)	120	C(14)-C(13)-C(15)	112
C(2)-C(3)-C(4)	120	C(12)-C(14)-C(13)	112
C(3)-C(4)-C(6)	121	C(12)-C(14)-C(16)	113
C(1)-C(5)-C(6)	116	C(13)-C(14)-C(16)	108
C(6)-C(5)-C(7)	117	C(13)-C(15)-C(17)	97
C(1)-C(5)-C(7)	126	O(6)-C(16)-C(14)	105
C(4)-C(6)-C(5)	118	O(6)-C(16)-C(18)	102
C(4)-C(6)-C(8)	121	C(14)-C(16)-C(18)	112
C(5)-C(6)-C(8)	120	C(15)-C(17)-C(18)	107
O(3)-C(7)-C(5)	119	C(15)-C(17)-C(19)	105
O(3)-C(7)-C(9)	121	C(18)-C(17)-C(19)	102
C(5)-C(7)-C(9)	120	C(16)-C(18)-C(17)	99
C(6)-C(8)-C(10)	124	O(5)-C(19)-(O(6))	107
C(7)-C(9)-C(10)	119	O(5)-C(19)-C(17)	106
C(7)-C(9)-C(11)	120	O(5)-C(19)-C(20)	107
C(10)-C(9)-C(11)	121	O(6)-C(19)-C(17)	108
C(8)-C(10)-C(9)	119	O(6)-C(19)-C(20)	110
C(8)-C(10)-C(12)	121	C(17)-C(19)-C(20)	118
C(9)-C(10)-C(12)	120	O(1)-C(21)-O(2)	126
O(4)-C(11)-C(9)	120	O(1)-C(21)-C(22)	128
O(4)-C(11)-C(13)	121	O(2)-C(21)-C(22)	107
C(9)-C(11)-C(13)	119	Br-C(22)-C(21)	113

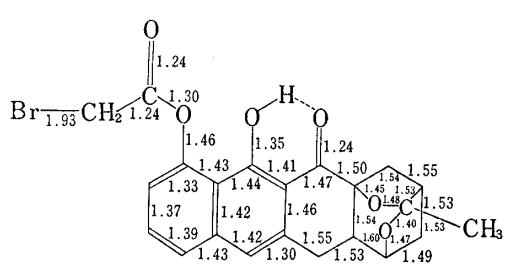


Fig. 3. Bond Distances

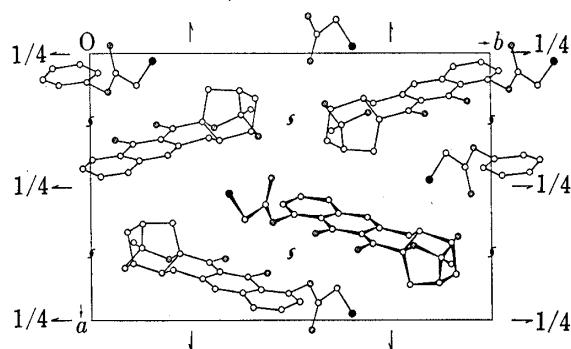


Fig. 4. Packing of the Structure Seen Down the c Axis

In Fig. 4 is shown a view of the structure along the *c* axis, together with the packing of the molecules in a unit cell. Since the molecule has no polar groups available for intermolecular hydrogen bonds, the association of the molecules seems to be mainly due to the van der Waals interactions.

Acknowledgement We thank Dr. Sueo Tatsuoka, General Manager of Research and Development Division, Takeda Chemical Industries, Ltd., for permission of this investigation.