

Inhibition of Guanine Deaminase with Derivatives of Imidazole

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Antitumor activity of 8-azaguanine was enhanced by simultaneous administration of 5-amino-4-imidazolecarboxamide (AICA).²⁾ This effect was suggested to be due to the inhibition of guanine deaminase (3.5.4.3) by AICA.^{2,3)} 1- β -D-Ribofuranosyl- and 5-formamide derivatives of AICA were not inhibitory.⁴⁾ In continuation of a previous report,⁴⁾ inhibitory compounds for guanine deaminase more potent than AICA were searched among derivatives of imidazole. 4-Imidazolecarboxamide was thus found as a potent inhibitor.

Material and Method

Assay of Guanine Deaminase Activity—Guanine deaminase of rat liver⁵⁾ and photometric procedure⁶⁾ were used as reported previously.⁴⁾ The reaction mixture was incubated with shaking in a water-bath at 37° for 30 min in air atmosphere. The concentration for 50% inhibition was obtained by plotting V_0/V_i against the concentration of inhibitor [I], where V_0 is the velocity without inhibitor and V_i is the velocity with inhibitor. Further, the ratio of concentration of inhibitor to substrate giving 50% inhibition, $([I]/[S])_{0.5}$, was determined.^{4,7)}

Materials—The following nine compounds were used: 5-Amino-4-imidazolecarboxamide (I), 5-iodo-4-imidazolecarboxamide (II), 5-bromo-4-imidazolecarboxamide (III), 5-chloro-4-imidazolecarboxamide (IV), 4-imidazolecarboxamide (V), imidazole (VI), 4-amino-5-carbethoxyimidazole (VII), imidazole-4,5-dicarboxylic acid (VIII), and 5-chloro-4-imidazolecarboxamidoxime (IX). Their chemical formulate are shown in Table I.

Result and Discussion

As shown in Fig. 1, the inhibitory activity against guanine deaminase of the derivatives of 5-amino-4-imidazolecarboxamide (AICA) was determined. Chloro derivative (IV) was the most inhibitory among the halogeno derivatives but it was less effective than the parent compound (I). Amino group at position 5 of I had been considered to be essential for the inhibition of the enzyme,⁴⁾ but amino group was not the only thing which inhibited the guanine deaminase. Therefore 4-imidazolecarboxamide (V) was unexpected but powerful inhibitor than I.

Furthermore, the imidazole itself (VI) was a weak inhibitor (Fig. 2). Ethylester of carboxylic acid (VII) and carboxamidoxime (IX) of I were less effective. Dicarboxylic acid of imidazole (VIII) was almost non effective.

4-Imidazolecarboxamide was found to be the most potent inhibitor among the imidazole derivatives tested for guanine deaminase, and carboxamide group and hydrogen for the sub-

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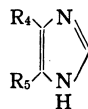
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TABLE I. Inhibition of Guanine Deaminase by Imidazole Derivatives



Compound	R ₄	R ₅	$([I]/[S])_{0.5}^{a)}$
I (AICA)	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-NH}_2 \end{array}$	-NH ₂	1.9
II	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-NH}_2 \end{array}$	-I	over 100
III	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-NH}_2 \end{array}$	-Br	10.3
IV	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-NH}_2 \end{array}$	-Cl	4.7
V	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-NH}_2 \end{array}$	-H	0.90
VI	-H	-H	10.3
VII	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-OC}_2\text{H}_5 \end{array}$	-NH ₂	4.0
VIII	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-OH} \end{array}$	$\begin{array}{c} \text{O} \\ \parallel \\ \text{-C-OH} \end{array}$	over 100
IX	$\begin{array}{c} \text{NOH} \\ \parallel \\ \text{-C-NH}_2 \end{array}$	-Cl	16.3

a) Ratio of concentration of inhibitor to substrate giving 50% inhibition.

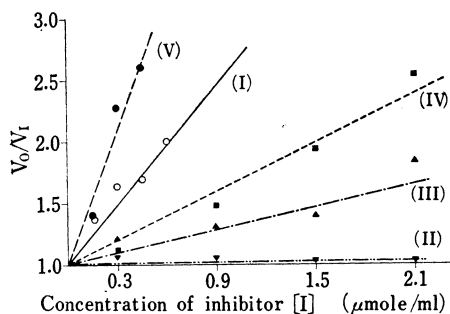


Fig. 1. Effect of the Compounds on Guanine Deaminase (1)

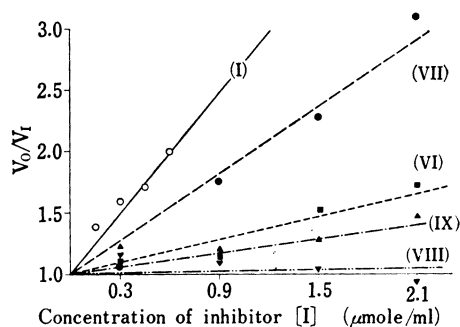


Fig. 2. Effect of the Compounds on Guanine Deaminase (2)

stituent at positions 4 and 5 of I were the most suitable groups for the inhibition of guanine deaminase.

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