Notes

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Studies on Metabolism of Drugs. XII.¹⁾ Quantitative Separation of Metabolites in Human and Rabbit Urine after Oral Administration of Sulfamonomethoxine and Sulfamethomidine

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Sulfamonomethoxine, N¹-(4-methoxy-6-pyrimidinyl)sulfanilamide, developed and widely used as the long-lasting sulfa drug, was synthesized by Okuda and others,³) and Oshima and others⁴) examined its antibacterial activity and toxity, reporting that it is broken down into the free compound and N⁴-acetylated compound in vivo. Nakazawa and others⁵) also reported that 60% of the drug ingested in the human body is in free form and that N⁴-actylated compound is also found. We⁶ carried out paper chromatography of human urine after ingestion of sulfamonomethoxine and found three spots which were found to be the unchanged drug, N⁴-acetylsulfamonomethoxine, and a newly discovered sulfamonomethoxine N¹-glucosiduronate.

Sulfamethomidine, N¹-(6-methoxy-2-methyl-4-pyrimidinyl)sulfanilamide was synthesized by Loop, et al., 7) which having the same pyrimidine ring, is also one of the long-lasting sulfas used widely. Its metabolites are also reported to be the free and acetylated compounds, 8) and their concentration in blood and urine has been measured. Our examination of the metabolites of sulfamethomidine 9) showed the presence of unchanged drug, N⁴-acetyl-sulfamethomidine, and newly found sulfamethomidine N¹-glucosiduronate.

Metabolites of sulfamonomethoxine and sulfamethomidine in blood and urine have been measured as the free and N⁴-acetylated compounds, but now that three substances, including unchanged drug, are known to be present, separatory determination of these metabolites was carried out in order to clarify the quantitative relation with passage of time, both in humans and in rabbits.

Experimental

Reagents—BuOH saturated with 0.5 N NH₄OH, 0.02 and 0.2 N Na₂CO₃, 1,2, and 4 N NaOH, 0.2% NaNO₂, 0.5% ammonium sulfamate solution, 0.1% Tsuda reagent (90% EtOH solution of 0.1% β -diethylaminoethyl- α -naphthylamine oxalate), Ehrlich reagent (EtOH solution containing 2% of p-dimethylaminobenzaldehyde and 1/50 volume of conc. HCl).

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- 3) N. Okuda and I. Kuniyoshi, Yakugaku Zasshi, 82, 1035 (1962); N. Okuda, I. Kuniyoshi, Y. Oshima, and S. Nagasaki, ibid., 82, 1039 (1962).
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Materials and Apparatus—Toyo Roshi No. 50 filter paper $(40 \times 26 \text{ cm})$. Paper chromatography (Toyo Kagaku Type B, with stainless steel spiral), Toshiba ultraviolet lamp Model FL-20 BLB, incubator (width \times height \times depth: $100 \times 70 \times 60 \text{ cm}$), Hitachi spectrophotometer Model 139.

Descending Elution Apparatus for PPC (Fig. 2): A wooden box $(35 \times 38 \times 33 \text{ cm})$ in inner dimensions) with a window coverd with a transparent plastic sheet on one side and a wooden board placed on the top. The box is lined with thick, absorbent cotton cloth, which is sprayed with deionized water at the time of elution so as to saturate the box with water vapor. For elution liquid, four trough made of hard vinyl plastic of 0.3 cm thickness $(5 \times 33.5 \times 1.5 \text{ cm})$ in inner dimensions) were placed in parallel on a wooden board, 29.3 cm in height from the bottom of the box. Two glass plates $(6 \times 32.5 \text{ cm})$ of 0.2 cm in thickness were placed together in one trough, to one edge (6 cm) of the glass plates in the liquid, about 1 cm of the paper to be eluted was placed between the glass plates on further end, and the paper was eluted by the descending method with the eluting solvent in the trough. The receiver was a 10-ml of conical graduate with round signed exactly at 3 ml, or a 10-ml measuring flask round graduated exactly at 1.5 ml.

Separation—This was referred to the method reported earlier. As shown in Fig. 1, 0.3 ml of the sample urine was streaked at one end of the filter paper using a pointed tip measuring pipette between P and Q, and one spot at R in order to find the position of spots. Toyo Kagaku Type B chromatography apparatus was used, BuOH saturated with $0.5 \text{ N NH}_4\text{OH}$ as the developing solvent, and developed for 16 hr in an incubator at 28°. After the paper was dried in air, the end strip (T) was cut off and sprayed with the Ehrlich reagent to know the approximate position of the spots. With this strip as a guide and with reference to the Rf of spots from ordinary urine components that fluoresce under irradiation from an ultraviolet lamp, the eluted paper was cut into two parts, AB and C, as shwon in Fig. 1.

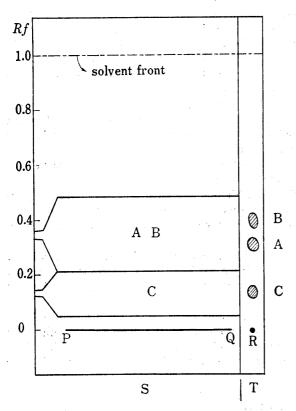


Fig. 1. Paper Chromatogram for Quantitative Separation

solvent: BuOH satd.with 0.5N NH4OH

 $T\bar{o}y\bar{o}$ filter paper No. 50 40×26.5 cm 28°, 16 hr

- -: sample (start line)
- •: reference sample
- -: Cut off line after ascending paper chromatography
- A: sulfamonomethoxine or sulfamethomidine
- B: N4-acetylsulfamonomethoxine or N4-acetylsulfamethomidine
- C : sulfamonomethoxine $N^1\!\!-\!\!\mathrm{glucuronide}$ or sulfamethomidine $N^1\!\!-\!\!\mathrm{glucuronide}$

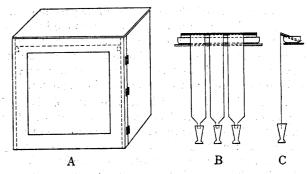


Fig. 2. Outline Views for Elution from Paper Strips

- A: front view of elution box (width \times height \times depth: $38 \times 33 \times 35$ cm)
- B: front view for elution from paper strips
- C: side view for elution from paper strips

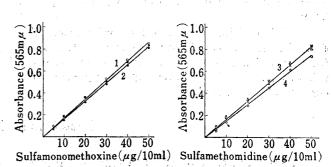


Fig. 3. Calibration Curves for Metabolites of Sulfamonomethoxine and Sulfamethomidine

- 1: —O—: N⁴-acetylsulfamonomethoxine
- ---: sulfamonomethoxine Ni-glucuronide
- 2: : sulfamonomethoxine
- 3: -x-: N4-acetylsulfamethomidine
 - $-\triangle$ —: sulfamethomidine N^1 -glucuronide
- 4: • sulfamethomidine

Right-hand side of the paper strip AB was placed between the two glass plates, placed in the trough, and developed and eluted with 0.2 n Na₂CO₃ solution, by the descending method, as shown in Fig. 2. If about 2 ml of the liquid is eluted into the 10-ml conical graduate, the substances would have been eluted completely. This eluted liquid is made up exactly to 3 ml with 0.2 n Na₂CO₃ solution, and each 1 ml of this solution is used to determine the amount of unchanged drug and its N⁴-acetylated compound.

The paper strip C is also eluted in the same manner with 0.02 N Na₂CO₃ solution and the liquid is allowed to collect in a 10-ml measuring flask. If the liquid is allowed to collect to ca. 1 ml, the substance would have been completely eluted. If the spot has been eluted with less liquid, the collected solution is made up to 1.5 ml with 0.02 N Na₂CO₃ and used for determination of N¹-glucuronide.

Method of Determination—This was effected by the diazotized method reported earlier¹⁰⁾ and is measured with spectrophotometer at $565 \text{ m}\mu$.

Calibration Curve—A solution of 100 mg of standard sample of sulfamonomethoxine and sulfamethomidine or 108.5 mg of acetylsulfamonomethoxine and 107.6 mg of acetylsulfamethomidine to correspond to 100 mg of sulfamonomethoxine and sulfamethomidine respectively dissolved in 8 ml of 0.2 N NH₄OH is diluted to 200 ml with normal urine. This solution is diluted with normal urine to make standard solution containing 0, 50, 100, 200, 300, 400, or 500 μ g/ml of the each drugs. This solution is submitted to the foregoing separation and determination procedures to prepare a calibration curve.

In the case of the glucuronides, 5.547 mg of sulfamonomethoxine N¹-glucuronide and 5.447 mg of sulfamethomidine N¹-glucuronide to correspond to 3.333 mg as sulfamonomethoxine or sulfamethomidine is diluted to 20 ml with normal urine, and these solution are prepared into standard solution containing 0, 16.66, 33.33, 66.66, 100, 133.3, or 166.6 μ g/ml. These solution are submitted to separation and determination as described above to prepare the calibration curve (Fig. 3).

Examination of the Methods of Separation and Determination 1) Sulfamonomethoxine and its N4-acetylated compound, and sulfamethomidine and its N4-acetylated compound cannot be separated completely by paper chromatography and they were eluted together. The spots which color immediately to the Tsuda reagent after diazotization are sulfamonomethoxine and sulfamethomidine. The amount of these substacted from the amount eluted from the spot that colors after hydrolysis with NaOH corresponds to the amount of the acetylated compound.

- 2) If HCl is used and heated for hydrolysis of the acetylated compound, a part of the filter paper component that is eluted undergoes hydrolysis and an aldehyde-like substance seems to be formed which is considered to condense with the amino group in the sulfa drug. The solution obtained by hydrolysis with HCl and heating, when diazotized and colored with the Tsuda reagent, gives lower content and the calibration curve does not become a straight line. If the test solution is heated in NaOH alkalinity, then acidified with HCl and diazotized, the calibration curve becomes a straight line.
- 3) The acetylated compound dissolves well in 0.2 N Na₂CO₃ without decomposition. Sulfamonomethoxine and sulfamethomidine also dissolve well in Na₂CO₃. For this reason, the components collect at Rf 0.9 by descending elution and the substances are eluted completely with a small quantity of the elution liquid.
- 4) For oral administration to rabbits, the sulfa drugs were reduced to a fine powder, suspended in 0.4% carboxymethylcellulose paste, and infused into the rabbit stomach by a rubber tubing fitted with a funnel. Male rabbits of about 3 kg in body weight are placed in a rabbit urine cage and urine was collected every 24 hr.
- 5) Direct Dilution Method: In order to calculate the total amount of the sulfa drug, the urine was diluted suitably, heated with HCl to hydrolyze the acetylated compound, and the amount determined as total sulfa drug, using normal urine as blank, instead of the above-described separation method. Since this avoid the mixing of filter paper component, calibration curve was prepared separately from which the content was calculated.

Result and Discussion

Determination of the amount of metabolites periodically after ingestion of 1 or 2g of the sulfa drug gave results summarized in Tables I and II, and cumulative values are shown in Fig. 4 and 5.

In order to calculate the recovery of the metabolites, authentic metabolites dissolved in normal human urine as shown ingredients (with nearly above measured amount) in Table III, and above separation and determination were carried out for calculate the mean values and standard errors (Table III).

In both cases, the drug was given after early breakfast. The values at 24, 48, 72, and 96 hr mean that they were measured after an overnight sleep, and these values show clearly

TABLE I.	Amount of Substances excreted in Human Urine aft	er Ingestion of
Sul	lfamonomethoxine (Calculated as Sulfamonomethoxin	ie (mg))

	Sub-	Time of urine collection (hr after ingestion)								% of total	% of			
	and	stance ^a)	2	4	8	12	24	36	48	60	72	84	96	
A 1	SM	3.4	15.0	41.4	45.5	39.9	30.4	28.8	18.9	16.1	10.0	3.5	27.8	91
	ASM	1.4	6.2	33.0	39.6	104.3	91.2	70.2	48.1	37.1	23.5	13.0	51.3	
	SMG	0.8	4.5	20.7	21.6	43.4	35.2	23.4	19.5	12.6	6.0	3.0	20.9	
B 1	SM	0.5	15.2	52.8	62.7	77.6	75.8	35.7	29.6	11.7	15.2	3.9	39.7	91
	ASM	0.8	3.2	26.4	26.6	91.3	72.3	46.2	49.6	18.9	18.4	9.9	42.3	
*	SMG	0.0	2.4	21.0	15.2	26.4	36.0	21.7	19.3	7.8	10.4	3.6	18.0	
C 2	SM	3.8	6.9	41.5	34.0	96.5	127.0	65.7	53.0	26.7	21.6	21.3	33.3	75
	ASM	1.0	10.9	45.6	65.1	142.7	106.7	105.0	89.3	50.1	42.8	20.2	45.4	
	SMG	1.5	6.3	26.3	31.1	74.6	61.8	40.6	36.7	11.7	14.1	13.5	21.3	
D 2	SM	3.0	7.1	36.8	28.4	93.0	81.7	42.8	27.6	18.0	13.4	12.5	23.8	76
	ASM	1.0	5.8	51.2	89.6	206.1	143.8	112.2	48.3	49.9	32.8	19.5	50.6	
	SMG	0.0	4.0	38.7	44.3	104.7	56.4	49.1	34.5	22.6	18.2	14.0	25.6	

a) SM: sulfamonomethoxine ASM: N4-acetylsulfamonomethoxine SMG: sulfamonomethoxine N1-glucosiduronate

TABLE II. Amount of Substances excreted in Human Urine after Ingestion of Sulfamethomidine (Calculated as Sulfamethomidine (mg))

Subject	Sub-	-		Time	of Uri	ne colle	ction (h	r after	ingesti	on)			% of total	% of
and st dose (g) st	stance ^{a)}	2	4	8	12	24	36	48	60	72	84	96		dose excreted
B 1	SD	0.5	5.5	11.5	9.0	19.2	14.4	7.0	6.4	2.9	1.0	0.3	10.1	77
	ASD	0.2	3.7	13.1	15.0	42.7	49.2	23.1	19.3	12.0	9.4	6.9	25.3	
	SDG	0.6	14.6	52.5	57.0	138.2	90.0	55.3	37.7	23.5	17.7	9.3	64.5	
E 1	SD	0.0	3.6	12.6	10.8	15.0	16.4	6.3	6.0	5.0	6.5	0.0	11.4	72
	ASD	0.6	2.8	19.6	20.0	25.0	39.7	21.0	16.0	8.0	11.7	4.0	23.4	
	SDG	0.4	11.8	56.0	59.2	123.5	82.2	46.2	41.0	22.0	18.2	8.0	65.2	
D 1	SD	1.8	3.9	9.1	6.5	20.5	10.6	5.0	4.1	1.2	2.0	0.0	7.7	84
	ASD	1.0	5.1	15.6	16.3	56.2	60.0	39.6	34.8	33.4	16.9	8.3	34.2	
	SDG	4.3	17.5	60.4	51.9	135.5	93.5	52.4	34.8	19.6	12.9	6.2	58.2	
$\mathbf{B} \; 2$	SD	2.1	11.5	37.8	20.2	39.4	41.0	18.2	14.0	6.6	7.3	5.0	12.4	82
	ASD	3.1	8.0	23.8	40.3	99.2	100.0	47.8	46.2	23.1	24.6	18.5	26.6	
	SDG	5.6	28.4	111.8	97.8	252.4	176.0	122.7	85.4	47.3	41.9	22.4	60.8	
C 2	SD	1.5	7.3	25.4	31.5	49.0	28.7	20.3	18.9	6.3	16.7	5.4	13.3	79
	ΛSD	1.1	10.9	21.2	27.0	108.5	95.7	72.8	54.3	26.6	42.2	24.6	30.7	
	SDG	2.5	25.3	82.7	75.0	215.6	149.6	119.0	81.1	37.8	67.8	29.3	56.0	
D 2	SD	1.0	3.2	15.7	14.2	39.2	28.5	12.5	11.0	3.5	6.8	1.2	7.6	90
	ASD	0.2	4.0	24.5	42.6	140.4	93.3	43.2	40.9	27.1	12.4	8.6	24.2	
	SDG	5.7	20.9	91.6	403.6	305.5	165.9	107.5	58.8	39.7	24.8	6.3	68.2	

a) SD: sulfamethomidine, ASD: N4-acetylsulfamethomidine, SDG: sulfamethomidine N1-glucosiduronate

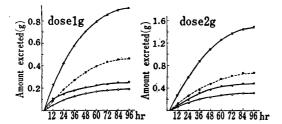


Fig. 4. Cumulative Excretion Curves of Sulfamonomethoxine Metabolites in Human Urine

---: sulfamonomethoxine

----: N4-acetylsulfamonomethoxine

-O-: sulfamonomethoxine N¹-glucosiduronate

 $-\times-$: total sulfamonomethoxine (calculated as sulfamonomethoxine)

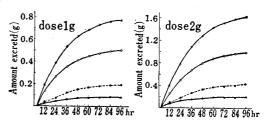


Fig. 5. Cumulative Excretion Curves of Sulfamethomidine Metabolites in Human Urine

-•-: sulfamethomidine

- N4-acetylsulfamethomidine

— : sulfamethomidine N¹-glucosiduronate

-: total sulfamethomidine (calculated as sulfamethomidine)

TABLE III. Recovery Test of Metabolites on Sulfamonomethoxine and Sulfamethomidine from the known Human Urine

Substance		n^{a}	Standard		
	Mean	Max.	Min.	n-1	Error (%)
Sulfamonomethoxine	100.3	102.1	98.2	- 5	0.69
N ⁴ -Acetylsulfamonomethoxine	100.8	103.7	98.6	5	0.91
Sulfamonomethoxine N¹-glucuronide	100.4	102.3	98.9	5	0.66
Sulfamethomidine	100.1	102.5	97.6	5	0.98
N ⁴ -Acetylsulfamethomidine	99.7	102.1	96.4	5	1.05
Sulfamethomidine N¹-glucuronide	100.5	102.6	98.6	5	0.75

Table IV. Amount of Substances excreted in Rabbit Urine after Ingestion of Sulfamonomethoxine (SM) and Sulfamethomidine (SD) (Calculated as SM or SD (mg))

Subject and	Substance a)	Time of urine (hr after inges		% of total	% of dose	
dose (g)	e electrical de la companya de la c La companya de la co	24	48	excreted	excreted	
SM	SM	30	22	25	83	
0.25	ASM	125	31	75		
SM	SM	138	23	37	86	
0.50	ASM	229	39	63	30	
SD	SD	22	2	14	68	
0.25	ASD	114	13	74		
	SDG	18	.2	12		
SD	SD	25	4	10	57	
0.5	ASD	207	23	81	J.	
	SDG	23	3	9	and the second second	

a) ASM: N⁴-acetylsulfamonomethoxine ASD: N⁴-acetylsulfamethomidine SDG: sulfamethomidine N¹-glucosiduronate

the difference in the amount of metabolism from those found during the daytime activity of up to 12 hr and at 36, 60, and 84 hr, indicating larger amount excreted during the day.

In rabbits, as shown in Table IV, the amount of the acetylated compound was much more than that found in the humans while the amount of the glucuronide was relatively smaller. In the case of human subjects, glucuronide was the most abundant among the metabolites of sulfamethomidine, and this was true in the case of rabbits, but the amount of glucuronide was small in the metabolites of sulfamonomethoxine, and the amount was not sufficient for determination in the case of rabbits.