

Studies on Organic Sulfur Compounds. X.¹⁾ The Reactions of Alkoxy-carbonyl Isothiocyanates with *prim-α*-Acetylenic Alcohols²⁾

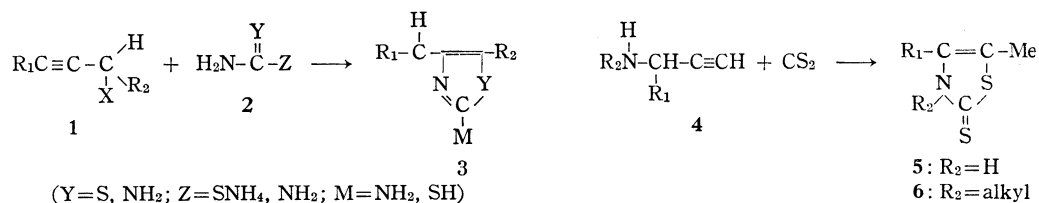
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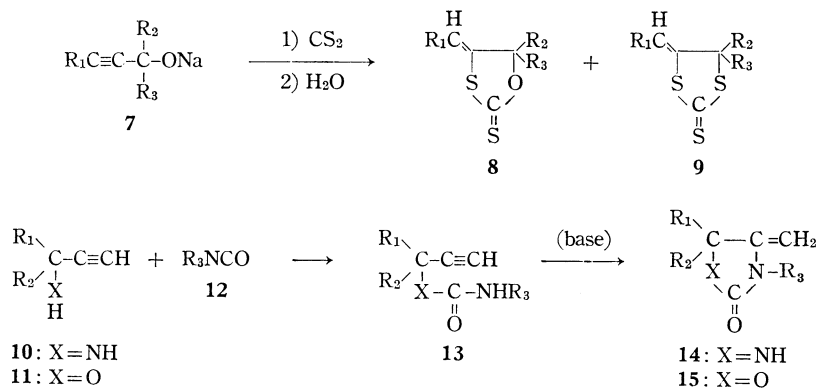
The reactions of alkoxy-carbonyl isothiocyanates and *prim-α*-acetylenic alcohols afford N-alkoxy-carbonyl-O-acetylenyl thiocarbamates, N-alkoxy-carbonyl-S-allenyl thiolcarbamates and 4-alkylidene-2-alkoxy-carbonylimino-1,3-oxathiolanes. However, the reaction patterns are dependent on the substituents on the 3-positions of *α*-acetylenic alcohols: 3-Phenyl-2-propyn-1-ol (**40**) react smoothly with the isothiocyanates to give 1:1 adducts which cyclize immediately to 4-benzylidene-1,3-oxathiolanes (**42**). 2-Butyn-1-ol (**36**) reacts slowly with the isothiocyanates to afford N-alkoxy-carbonyl-O-2-butyn-1-yl thiocarbamates (**37**), but these compounds (**37**) cyclize hardly to 4-ethylidene-1,3-oxathiolanes (**39**) even in the presence of a base. In the case of 2-propyn-1-ol (**19**), N-alkoxy-carbonyl-O-2-propyn-1-yl thiocarbamates (**24**), N-alkoxy-carbonyl-S-allenyl thiolcarbamates (**25**) and 2-alkoxy-carbonylimino-4-methylidene-1,3-oxathiolanes (**26**) are obtained. Furthermore, the cyclization mechanisms of *α*-acetylenyl thiocarbamates (**45**) were confirmed by using N(D)-O-(2-propyn-1-yl)-N-isopropoxycarbonyl thiocarbamate (**43**).

There are many reports concerning the intramolecular cyclization reactions of compounds having both a carbon-carbon triple bond and a nucleophilic radical. 2-Aminothiazole (**3**: Y=S; M=NH₂), 2-thiazolethiol (**3**: Y=S; M=SH), and 2-aminodiazole (**3**: Y=NH; M=NH₂) derivatives are prepared by treatment of *α*-haloacetylenic compounds (**1**) with thiourea, ammonium dithiocarbamate and guanidine, respectively.^{4,5)} 4-Methyl-2-thiazolethiols (**5**), and 4-methylthiazoline-2-thiones (**6**) are also obtained by the reactions of propargyl amines (**4**)^{6,7)} and carbon disulfide. Tomita, Nagano, and Oka^{8,9)} have reported that in the reactions of sodium *α*-acetylenic alcoholates (**7**) and carbon disulfide, two five

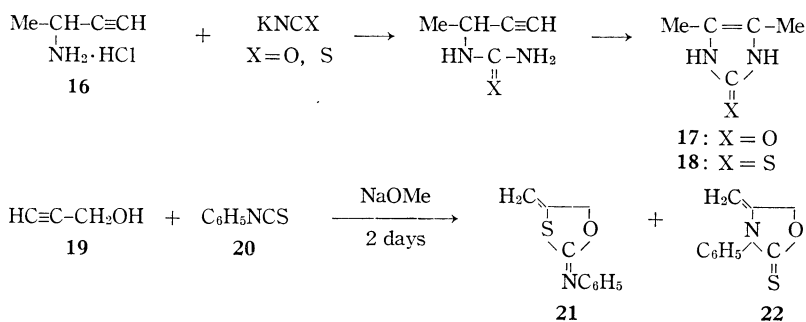


membered hetero compounds [4-alkylidene-1,3-oxathiolane-2-thiones (**8**) and -1,3-dithiolane-2-thiones (**9**)] are obtained. It is known that in the presence of a basic catalyst, propargyl amines (**10**) and propargyl alcohols (**11**) react with isocyanates (**12**) to give 4-methylidene-

- 1) Part IX: M. Nagano, T. Matsui, J. Tobitsuka, and K. Oyamada, *Chem. Pharm. Bull.* (Tokyo), **20**, 2626 (1972).
- 2) This was presented at the 5th Symposium of Organic Sulfur Chemistry, Kiryu, February 1971.
- 3) Location: *Hiromachi, Shinagawa-ku, Tokyo*.
- 4) Y. Yura, *Chem. Pharm. Bull.* (Tokyo), **10**, 372 (1962).
- 5) Y. Yura, *Chem. Pharm. Bull.* (Tokyo), **10**, 376 (1962).
- 6) Y. Yura, *Chem. Pharm. Bull.* (Tokyo), **10**, 1087 (1962).
- 7) J.W. Batty and B.C.L. Weedon, *J. Chem. Soc.*, **1944**, 746.
- 8) K. Tomita, M. Nagano, and H. Oka, *Chem. Pharm. Bull.* (Tokyo), **16**, 914 (1968).
- 9) K. Tomita and M. Nagano, *Chem. Pharm. Bull.* (Tokyo), **16**, 1324 (1968).



2-imidazolidinones (**14**) and -2-oxazolidinones (**15**), respectively.¹⁰⁻¹² Yura⁶) has reported that 4,5-dimethyl-2-imidazoline (**17**) or -2-imidazoethiol (**18**) is prepared by treatment of 3-amino-1-butyne hydrochloride (**16**) with potassium cyanate or potassium thiocyanate, respectively. Furthermore, it has been reported by Shachat, *et al.*¹⁰) that propargyl alcohol (**19**) reacts with phenyl isothiocyanate (**20**) in the presence of sodium methylate to afford two cyclic compounds (**21** and **22**). From these facts, α -acetylenic alcohols would be con-



sidered to react with alkoxycarbonyl isothiocyanates to afford the corresponding thiocarbamates or their cyclization products. Nevertheless, we found no reports of such a reaction. The present paper describes on the reactions of some prim- α -acetylenic alcohols and some alkoxycarbonyl isothiocyanates (**23**).

Alkoxycarbonyl isothiocyanates (**23**) are easily decomposed by bases such as alcoholates, *tert.* amines and inorganic salts.^{13,14}) Therefore, the mixture of 2-propyn-1-ol(propargyl alcohol; 1.0 mole) and ethoxycarbonyl isothiocyanate (**23-b**; 1.1 mole) was heated at 70—80° on an oil bath for 5 hr without a solvent, and the reaction mixture was chromatographed over silica gel to give N-ethoxycarbonyl-O-(2-propyn-1-yl)thiocarbamate(**24-b**) in 12.3%, N-ethoxycarbonyl-S-(1,2-propadien-1-yl)thiolcarbamate(**25-b**) in 10.2% and 2-ethoxycarbonylimino-4-methylidene-1,3-oxathiolane(**26-b**) in 48.2% yield. Furthermore, the compound (**24-b**) transformed to **25-b** and **26-b** by the same thermal conditions as above. However, a similar phenomenon has been already observed in the rearrangement of α -acetylenyl xanthates to the corresponding allenyl thiolcarbonates.¹⁵) The structures of **24-b** and **25-b** were

10) N. Shachat and J.J. Bagnell, Jr., *J. Org. Chem.*, **28**, 991 (1963).

11) K. Sisido, K. Hukuoka, M. Tuda, and H. Nozaki, *J. Org. Chem.*, **27**, 2663 (1962).

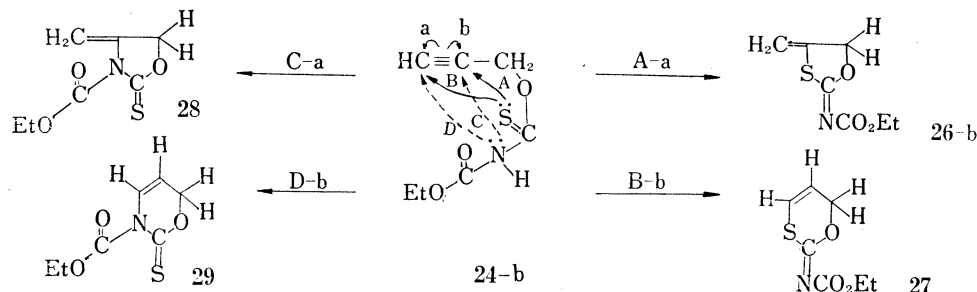
12) N.R. Easton, D.R. Cassady, and R.D. Dillard, *J. Org. Chem.*, **27**, 2927 (1962).

13) A. Takamizawa, K. Hirai, and K. Matsui, *Bull. Chem. Soc. Japan*, **36**, 1214 (1962).

14) J. Goerdeler, *Q. Reports on Sulfur Chemistry*, **5**, 1619 (1970).

15) K. Tomita and M. Nagano, *Chem. Pharm. Bull.* (Tokyo), **16**, 1911 (1968).

determined on the basis of elemental analyses and spectral data. The characteristic infrared (IR) absorption bands of the former compound (**24-b**) were observed at 3400 and 1775 cm^{-1} for a secondary amide $[-\text{C}(\text{O})\text{NH}-]$, at 3300 and 2120 cm^{-1} for an ethynyl group ($\text{HC}\equiv\text{C}-$); the nuclear magnetic resonance (NMR) spectrum showed a triplet at 8.69 τ for three methyl protons ($-\text{CH}_2\text{CH}_3$, $J=7.0$ Hz), a triplet at 7.43 τ for one ethynyl proton ($\text{HC}\equiv\text{C}-$, $J=2.2$ Hz), a quartet at 5.74 τ for two methylene protons ($-\text{CH}_2\text{CH}_3$, $J=7.0$ Hz) and a doublet at 4.85 τ for two methylene protons ($\text{HC}\equiv\text{C}-\text{CH}_2$, $J=2.2$ Hz) and approximately at 1.62 τ for one secondary amide proton ($>\text{NH}$). The characteristic IR absorption bands of the latter compound (**25-b**), which would be formed by the [3,3]-sigmatropic rearrangement of **24-b**, were observed at 3400, 1755, and 1685 cm^{-1} for a secondary amide $[-\text{C}(\text{O})\text{NHC}(\text{O})-]$, and at 1955 cm^{-1} for an allene group ($-\text{C}=\text{C}=\text{C}-$), and the NMR spectrum showed a triplet at 8.67 τ for three methyl protons ($-\text{CH}_3\text{CH}_3$, $J=7.0$ Hz), a quartet at 5.72 τ for two methylene protons ($-\text{CH}_2\text{CH}_3$, $J=7.0$ Hz), a doublet at 4.92 τ for two vinylidene protons ($=\text{C}-\text{CH}_2$, $J=6.5$ Hz), a triplet at 2.90 τ for one vinylidene proton ($-\text{HC}=\text{C}=\text{C}$, $J=6.5$ Hz), and approximately at 2.05 τ for one secondary amide proton $[-\text{C}(\text{O})\text{NHC}(\text{O})-]$. However, it is not clear why the coupling constants of the protons of the above allene group (**25-b**) are larger than those of general allene compounds.¹⁵⁾ Compound (**26-b**) was indicated as being a 1:1 addition product of **23-b** and **19** by elemental analysis and mass spectrum (MS): $M^+=187$, and the characteristic IR absorption bands were at 1678 cm^{-1} for a carbonyl group ($>\text{C}=\text{O}$), at 1635 cm^{-1} for a carbon-carbon double bond ($>\text{C}=\text{C}<$), and at 1580 cm^{-1} for a conjugated carbon-nitrogen double bond $[\text{N}=\text{C}(\text{O})-]$. From these data, four possible structures (**26-b**, **27**, **28**, and **29**), which would be formed by four intramolecular cyclization processes of **24-b**, were considered



for this compound. The NMR spectrum showed a triplet at 8.67 τ for three methyl protons ($-\text{CH}_2\text{CH}_3$, $J=7.0$ Hz), a quartet at 6.77 τ for two methylene protons ($-\text{CH}_2\text{CH}_3$, $J=7.0$ Hz), four peaks (a triplet and a doublet) centered at 4.92 τ for one unequivalent methylene proton ($\frac{\text{Hb}}{\text{Ha}}>\text{C}=\text{C}$, $J_1=2.5$ Hz, $J_2=2.3$ Hz), and four peaks (a triplet and a doublet centered at 4.63 τ for one another methylene proton ($\frac{\text{Hb}}{\text{Ha}}>\text{C}=\text{C}$, $J_1=2.5$ Hz, $J_2=2.3$ Hz). The NMR spectrum was more similar to that of the compound (**21**) than that of the compound (**22**) (see Fig. 1). For further confirmation, this compound (**26-b**) was hydrolyzed with one normal hydrogen chloride to give O-acetyl-N-ethoxycarbonyl-carbamate (**30**) which was also obtained by oxydation of O-acetyl-N-ethoxycarbonyl-thiocarbamate (**32**). Compound (**21**) was treated with normal hydrogen chloride to afford 4-methylidene-1,3-oxathiolane-2-one (**33**) which was also prepared by oxidation of 4-methylidene-1,3-oxathiolane-2-thione (**34**)⁹⁾ with mercuric acetate, but in the case of compound (**22**) the *exo* double bond rearranged to form 5-methyl-1-phenyl-oxazoline-2-thione (**35**). From these data, the structure of **26-b** was consequently confirmed as being 2-ethoxycarbonylimino-4-methylidene-1,3-oxathiolane. The yields of the products in the reactions of alkoxy carbonyl isothiocyanates (**23**) with 2-propyn-1-ol (**19**) are summarized in Table I.

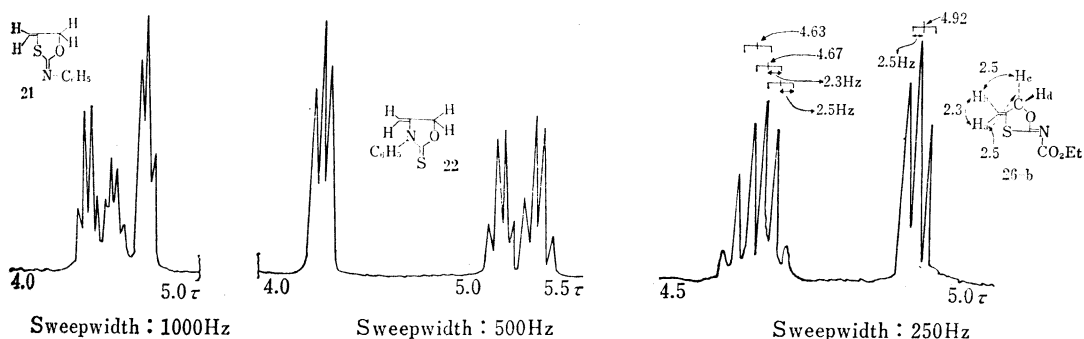
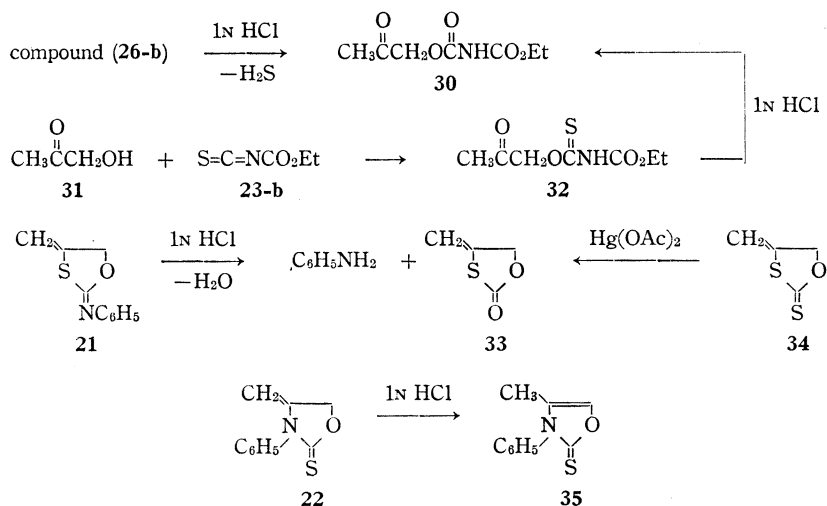
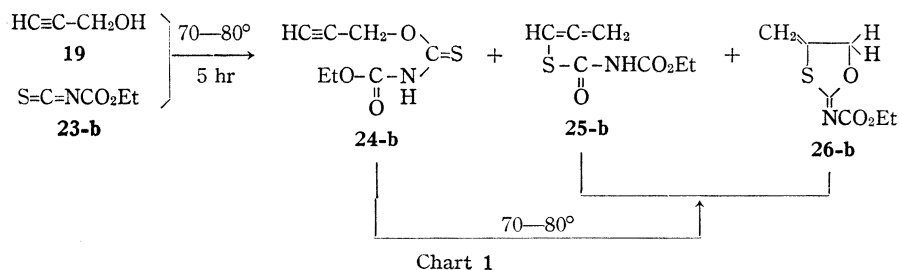


Fig. 1. NMR Spectra (CDCl_3) of 4-Methylidene-2-phenylimino-1,3-oxathiolane (**21**), 4-Methylidene-3-phenyl-oxazolidine-2-thione (**22**), and 2-Ethoxycarbonylimino-4-methylidene-1,3-oxathiolane (**26-b**).



A mixture of 2-butyne-1-ol (**36**) and ethoxycarbonyl isothiocyanate (**23-b**) in carbon tetrachloride was refluxed for 3 hr, and the resulting residue was chromatographed to give only N-ethoxycarbonyl-O-(2-butyne-1-yl)thiocarbamate (**37-b**) in 76.3% yield. The yield of **37-b** under this reaction condition was larger than when no solvent was employed. The adduct (**37-b**) was heated at 120–130° for 3 hr to afford N-ethoxycarbonyl-S-(2,3-butadiene-2-yl)thiolcarbamate (**38**) in 21.4% yield. However, when **37-b** was treated with triethylamine in ethyl alcohol, 2-ethoxycarbonylimino-4-ethylidene-1,3-oxathiolane (**39-b**) was obtained in 55.7% yield. Other alkoxy carbonyl isothiocyanates were reacted with **36** under

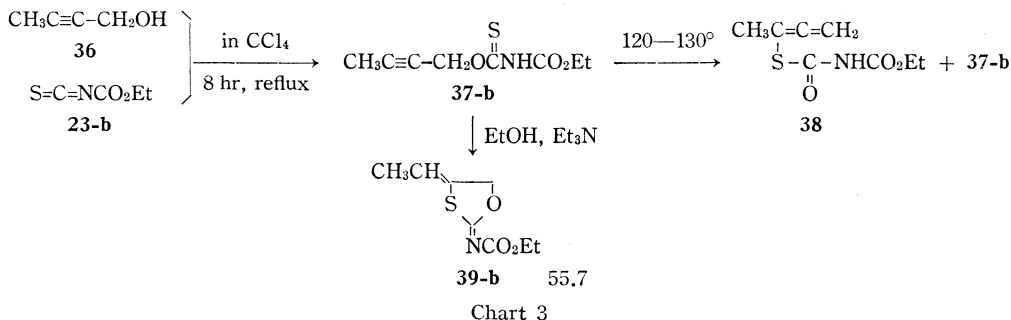
TABLE I. Yields of Isolated Products in the Reactions of Alkoxy carbonyl Isothiocyanates (23) with 2-Propyn-1-ol (19)

$S=C=NCO_2R$ 23	$HC\equiv C-CH_2-O$ $ROC-N-\overset{\overset{H}{\parallel}}{C}=S$ 24	$HC=C=CH_2$ $\overset{\overset{H}{\parallel}}{S}-C-NHCO_2R$ 25	$CH_2=C$ $\overset{\overset{H}{\parallel}}{S}-O$ $\overset{\overset{H}{\parallel}}{N}CO_2R$ 26
23-a: R=Me	24-a: 14.9 (%)	25-a: 10.3 (%)	26-a: 55.4 (%)
23-b: R=Et	24-b: 12.3 (%)	25-b: 10.2 (%)	26-b: 48.2 (%)
23-c: R= <i>n</i> -Pr	24-c: 10.2 (%)	25-c: 9.4% (%)	26-c: 49.3 (%)
23-d: R=iso-Pr	24-d: 9.8 (%)	25-d: 8.2 (%)	26-d: 44.1 (%)
23-e: R= <i>n</i> -Bu	24-e: 12.1 (%)	25-e: 5.3 (%)	26-e: 41.8 (%)
23-f: R=iso-Bu	24-f: 14.9 (%)	not isolated	26-f: 32.1 (%)

TABLE II. Yields of 2-Butyn-1-yl-N-alkoxy carbonyl-thiocarbamates (37) and Their Cyclization Products (39)

$CH_3C\equiv C-CH_2-\overset{\overset{S}{\parallel}}{O}CNHCO_2R$ 37	$CH_3CH=C$ $\overset{\overset{H}{\parallel}}{S}-O$ $\overset{\overset{H}{\parallel}}{N}CO_2R$ 39
R=Me (37-a): 82.9 (%)	R=Me (39-a): 60.2 (%)
R=Et (37-b): 76.3 (%)	R=Et (39-b): 55.7 (%)
R= <i>n</i> -Pr (37-c): 58.4 (%)	R= <i>n</i> -PR (39-c): 42.3 (%)
R=iso-Pr (37-d): 60.2 (%)	R=iso-Pr (39-d): 42.6 (%)
R= <i>n</i> -Bu (37-e): 21.9 (%)	R= <i>n</i> -Bu (39-e): 22.5 (%)
R=iso-Bu (37-f): 22.4 (%)	R=iso-Bu (39-f): 23.3 (%)

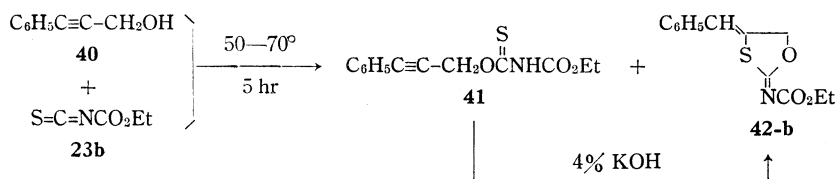
similar reaction conditions as above, and then the resulting addition compounds (37) were treated with triethyl amine to afford the corresponding cyclization compounds (39). The yields of 37 and 39 are summarized in Table II.



3-Phenyl-2-propyn-1-ol (40) was reacted with ethoxycarbonyl isothiocyanate (23-b) at 50–70° for 5 hr with no solvent, and then the reaction mixture was chromatographed over silica gel to give N-ethoxycarbonyl-O-(3-phenyl-2-propyn-1-yl)thiocarbamate (41) in 5.7% and 4-benzylidene-2-ethoxycarbonylimino-1,3-oxathiolane(42-b) in 71.5% yield. However, the adduct (41) was so unstable that on treatment with 4% potassium hydroxide at room temperature, it cyclized immediately to form 42-b. The reaction mixtures of other alkoxy carbonyl isothiocyanates and 40 were treated with 4% potassium hydroxide under the same

TABLE III. Yields of 4-Benzylidene-2-alkoxycarbonylimino-1,3-oxathiolanes (**42**)

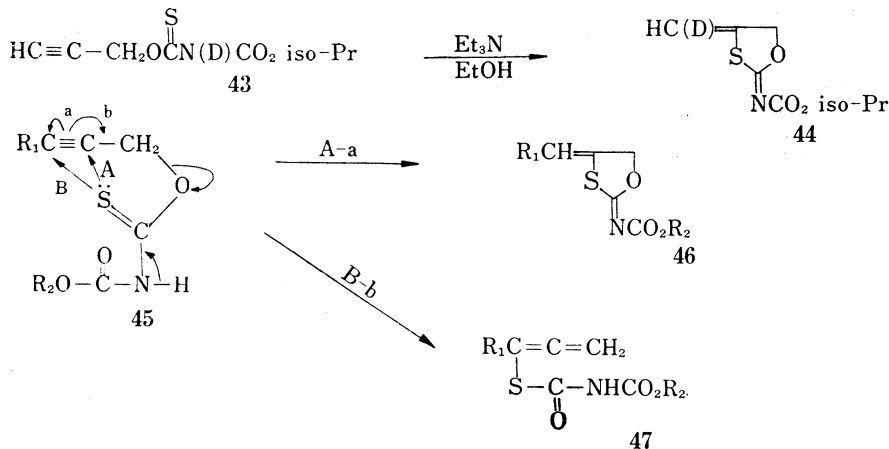
$\begin{array}{c} \text{S}=\text{C}=\text{NCO}_2\text{R} \\ \mathbf{23} \\ + \\ \text{C}_6\text{H}_5\text{C}\equiv\text{C}-\text{CH}_2\text{OH} \\ \mathbf{40} \end{array}$	$\xrightarrow{4\% \text{ KOH}}$	$\begin{array}{c} \text{C}_6\text{H}_5\text{CH}=\text{C} \\ \diagup \quad \diagdown \\ \text{S} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{NCO}_2\text{R} \\ \mathbf{42} \end{array}$
R=Me (23-a)		42-a : 90.3 (%)
R=Et (23-b)		42-b : 86.9 (%)
R= <i>n</i> -Pr (23-c)		42-c : 72.3 (%)
R=iso-Pr (23-d)		42-d : 72.4 (%)
R= <i>n</i> -Bu (23-e)		42-e : 68.5 (%)
R=iso-Bu (23-f)		42-f : 69.0 (%)



reaction conditions as above. The yields of the corresponding cyclic compounds (**42**) are summarized in Table III.

The reaction patterns of alkoxycarbonyl isothiocyanates (**23**) with prim- α -acetylenic alcohols were dependent on the substituent groups at the 3-position of the alcohols. 3-Phenyl-2-propyn-1-ol (**40**) reacted smoothly with alkoxycarbonyl isothiocyanates (**23**) to give 1:1 adducts which cyclized readily to **42** in the presence of a base. On the contrary, 2-butyln-1-ol (**36**) reacted slowly with the isothiocyanates (**23**) to afford 1:1 adducts (**37**), and these adducts hardly underwent intramolecular cyclization. However, in the reactions of 2-propyn-1-ol (**19**) with the isothiocyanates, the corresponding 1:1 adducts (**24**), the [3,3]-sigmatropic rearrangement products (**25**) and the five membered cyclic compounds (**26**) were obtained. On the other hand, the larger the alkyl groups of alkoxycarbonyl isothiocyanates (**23**) were, the lower the yields of reaction products were.

For further confirmation of the mechanism of the formation of the 4-alkylidene-2-ethoxycarbonylimino-1,3-oxathiolanes (**46**), N(D)-O-(2-propyn-1-yl)isopropoxycarbonyl-thiocarba-



mate (**43**), which was prepared by the treatment of O-(2-propyn-1-yl)isopropoxycarbonylthiocarbamate (**24-d**) with heavy water (D_2O), was heated in triethyl amine to afford 2-isopropoxycarbonylimino-4-D-methylidene-1,3-oxathiolane (**44**). The structures of the compounds (**43** and **44**) were confirmed on the basis of elemental analyses and spectral data. From above experimental data, it became clear that O-acetylenyl-N-alkoxycarbonylthiocarbamates (**45**) should cyclize intramolecularly to give the products (**46**) *via* course A-a, and on the other hand, the allenes (**47**) would be formed by [3,3]-sigmatropic rearrangement of the compound (**45**) *via* course B-b (Chart 5).

Experimental¹⁶⁾

General Method for Reactions of Alkoxycarbonyl Isothiocyanates (23) and 2-Propyn-1-ol (19)—The mixtures of 0.022 mole of alkoxycarbonyl isothiocyanates (**23**) and 0.02 mole of 2-propyn-1-ol (**19**) were stirred at room temperature for 1 hr, was heated at 70–80° in nitrogen atmosphere for 5 hr on an oil bath. The reaction mixtures, without further treatment with a base or an acid, was eluted with benzene–AcOEt on silica gel. The isolated products were recrystallized from a suitable solvent.

Reaction of Methoxycarbonyl Isothiocyanate (23-a) and 2-Propyn-1-ol (19)—2.85 g of **23-a** was reacted with 1.12 g of **19** by the general method to afford three products (**24-a**, **25-a** and **26-a**). O-2-Propynyl-N-methoxycarbonylthiocarbamate (**24-a**), 0.57 g, colorless needles from ether–pet. ether, mp 53–55°. *Anal.* Calcd. for $C_6H_7O_3NS$: C, 41.62; H, 4.08; N, 8.09; S, 18.48. Found: C, 41.27; H, 4.59; N, 7.99; S, 18.46. IR ν_{max}^{Nujol} cm^{-1} : 3320 and 2120 ($HC\equiv C-$), 3220 and 1772 [$-C(O)NHC(S)-$]. NMR ($CDCl_3$) τ ($J=Hz$): 7.40 (1H, t, $J=2.8$), 6.18 (3H, s), 4.84 (2H, d, $J=2.8$), *ca.* 1.61 (1H, broad). N-Methoxycarbonyl-S-1,2-propadien-1-yl-thiolcarbamate (**25-a**), 0.36 g, colorless needles from ether–pet. ether, mp 106–108°. *Anal.* Calcd. for $C_6H_7O_3NS$: C, 41.62; H, 4.08; N, 8.09; S, 18.48. Found: C, 41.41; H, 4.12; N, 7.76; S, 18.56. IR ν_{max}^{Nujol} cm^{-1} : 2200 ($-C=C-C-$), 3175, 1754 and 1655 [$-C(O)NHC(O)-$]. NMR ($CDCl_3$) τ ($J=Hz$): 7.17 (3H, s), 4.90 (2H, d, $J=6.5$), 3.68 (1H, t, $J=6.5$), *ca.* 1.71 (1H, broad). 2-Methoxycarbonylimino-4-methylidene-1,3-oxathiolane (**26-a**), 1.91 g, colorless needles from benzene–*n*-hexane, mp 89–90°. *Anal.* Calcd. for $C_6H_7O_3NS$: C, 41.62; H, 4.08; N, 8.09; S, 18.48. Found: C, 41.76; H, 4.11; N, 7.96; S, 18.60. IR ν_{max}^{Nujol} cm^{-1} : 1688 ($>C=O$), 1630 ($>C=C<$), 1572 ($>C=N-$). NMR ($CDCl_3$) τ ($J=Hz$): 6.15 (3H, s), 4.88 for two equivalent ring methylene protons ($-CH_2-$; t, $J=2.5$), 4.88 for one nonequivalent exo methylene proton ($\begin{smallmatrix} H_a \\ H_b \end{smallmatrix}$); t, d, $J_1=2.5$, $J_2=2.3$), 4.65 for another nonequivalent exo methylene proton ($\begin{smallmatrix} H_a \\ H_b \end{smallmatrix}$); t, d, $J_1=2.5$, $J_2=2.3$).

Reaction of Ethoxycarbonyl Isothiocyanate (23-b) and 2-Propyn-1-ol (19)—2.89 g of **23-b** was reacted with 1.12 g of **19** by the general method to afford three products (**24-b**, **25-b** and **26-b**): O-2-Propynyl-N-ethoxycarbonylthiocarbamate (**24-b**), 0.46 g, pale yellow needles from ether–pet. ether, mp 50–52°. *Anal.* Calcd. for $C_7H_9O_3NS$: C, 44.92; H, 4.84; N, 7.48; S, 17.12. Found: C, 44.62; H, 4.95; N, 7.36; S, 16.82. IR $\nu_{max}^{CHCl_3}$ cm^{-1} : 3300 and 2120 ($HC\equiv C-$), 3400 and 1770 [$-C(O)NHC(S)-$]. NMR ($CDCl_3$) τ ($J=Hz$): 8.69 (3H, t, $J=7.0$), 7.43 (1H, t, $J=2.2$), 5.74 (2H, q, $J=7.0$), 4.85 (2H, d, $J=2.2$), *ca.* 1.62 (1H, broad). UV λ_{max}^{EtOH} $m\mu$ (ϵ): 259 (7500). N-Ethoxycarbonyl-S-1,2-propadien-1-yl-thiolcarbamate (**25-b**), 0.382 g, colorless needles from ether–pet. ether, mp 78–80°. *Anal.* Calcd. for $C_7H_9O_3NS$: C, 44.92; H, 4.84; N, 7.48; S, 17.12. Found: C, 44.93; H, 5.00; N, 7.43; S, 16.97. IR $\nu_{max}^{CHCl_3}$ cm^{-1} : 1955 ($-C=C-C-$), 3400, 1755 and 1675 [$-C(O)NHC(O)-$]. NMR ($CDCl_3$) τ ($J=Hz$): 8.67 (3H, t, $J=7.0$), 5.72 (2H, q, $J=7.0$), 4.92 (2H, d, $J=7.0$), 2.90 (1H, t, $J=7.0$), *ca.* 2.05 (1H, broad). UV λ_{max}^{EtOH} $m\mu$ (ϵ): 238 (6500). 2-Ethoxycarbonylimino-4-methylidene-1,3-oxathiolane (**26-b**), 1.80 g, colorless needles from *n*-hexane, mp 108–109°. *Anal.* Calcd. for $C_7H_9O_3NS$: C, 44.92; H, 4.84; N, 7.48; S, 17.12. Found: C, 45.07; H, 4.93; N, 7.46; S, 17.04. Mass Spectrum m/e : $M^+=187$. IR ν_{max}^{Nujol} cm^{-1} : 1677 ($>C=O$), 1637 ($>C=C<$), 1580 ($>C=N-$). NMR ($CDCl_3$) τ ($J=Hz$): 8.67 (3H, t, $J=7.0$), 6.77 (2H, q, $J=7.0$), 4.92 (2H, t, $J=2.5$), 4.63 for one nonequivalent exo methylene proton ($\begin{smallmatrix} H_a \\ H_b \end{smallmatrix}$); t, d, $J_1=2.5$, $J_2=2.3$), 4.67 for another nonequivalent exo methylene proton ($\begin{smallmatrix} H_a \\ H_b \end{smallmatrix}$); t, d, $J_1=2.5$, $J_2=2.3$). UV λ_{max}^{EtOH} $m\mu$ (ϵ): 259.5 (6700).

Hydrolysis of 2-Ethoxycarbonylimino-4-methylidene-1,3-oxathiolane (26-b)—A suspension of 1.87 g of **26-b** in 20 ml of 1N HCl was refluxed for 1 hr, neutralized with sat. $NaHCO_3$, and extracted with $CHCl_3$. $CHCl_3$ layer was washed with H_2O , and dried over anhyd. Na_2SO_4 . Solvent was removed under reduced pressure, and then the residual solid was recrystallized from benzene–*n*-hexane to afford O-acetylenyl-N-ethoxycarbonylcarbamate (**30**) as colorless needles of mp 85–87°, 1.30 g (68.5%). Mass Spectrum m/e :

16) All melting and boiling points were uncorrected. NMR spectra were obtained in the specified solvents on a Varian A-60 and Varian HA-100 spectrometer with tetramethylsilane as an internal standard. Mass spectra were determined on a JEOL JMS-OISG spectrometer.

M^+ = 189. *Anal. Calcd.* for $C_7H_{11}O_3N$: C, 44.44; H, 5.86; N, 7.41. Found: C, 44.45; H, 5.62; N, 7.51. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3170 [$-C(O)NHC(O)-$], 1810, 1728 and 1715 for three carbonyl groups. NMR ($CDCl_3$) τ ($J = \text{Hz}$): 8.68 (3H, t, $J = 7.0$), 7.82 (3H, s), 5.71 (2H, q, $J = 7.0$), 5.28 (2H, s), *ca.* 2.64 (1H, broad).

Preparation of O-Acetyl-N-ethoxycarbonyl-carbamate (30)—a) A solution of 1.48 g of acetol (31),¹⁷ and 2.90 g of ethoxycarbonyl isothiocyanate (23-b) in 60 ml of AcOEt was refluxed for 1 hr, the solvent was removed under reduced pressure, and the resulting residue was eluted with benzene on silica gel. The isolated solid was recrystallized from *n*-hexane to afford 1.04 g of O-acetyl-N-ethoxycarbonyl-thiocarbamate (32) as pale yellow needles of mp 68–69°. *Anal. Calcd.* for $C_7H_{13}O_4NS$: C, 40.58; H, 6.32; N, 6.76; S, 15.44. Found: C, 40.67; H, 6.41; N, 6.80; S, 15.16. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3260 (NH), 1776 and 1730 for two carbonyl groups. NMR ($CDCl_3$) τ ($J = \text{Hz}$): 8.66 (3H, t, $J = 7.0$), 7.75 (3H, s), 5.71 (2H, q, $J = 7.0$), 4.85 (2H, s), *ca.* 1.95 (1H, broad). b) A suspension of 0.2 g of 32 in 10 ml of 1N HCl was refluxed for 2 hr, neutralized with sat. $NaHCO_3$ aqueous, and extracted with $CHCl_3$. $CHCl_3$ layer was washed with H_2O , and dried over anhyd. Na_2SO_4 . The solvent was removed, and the residue was recrystallized from *n*-hexane-benzene to afford 0.1 g of O-acetyl-N-ethoxycarbonyl-carbamate (30).

Reaction of 4-Methylidene-2-phenylimino-1,3-oxathiolane (21)¹⁰ with 1N HCl—A suspension of 0.8 g of 21 in 10 ml of 1N HCl was refluxed for 30 minutes, neutralized with sat. $NaHCO_3$ aqueous, and extracted with ether. Ether layer was washed with H_2O , and dried over anhyd. Na_2SO_4 . After removal of ether, the residual oil was distilled to afford 0.2 g of 4-methylidene-1,3-oxathiolane-2-one (33)* as a pale yellow oil of bp 80–90° (0.3 mmHg). *Anal. Calcd.* for $C_4H_4O_2S$: C, 41.39; H, 3.47; S, 27.57. Found: C, 41.26; H, 3.45; S, 27.78. IR ν_{\max}^{liq} cm^{-1} : 1750 ($>C=O$), 1630 for an exo carbon-carbon double bond ($\begin{matrix} H \\ | \\ C=C \end{matrix}$). NMR ($CDCl_3$) τ ($J = \text{Hz}$): 4.99 (2H, t, $J = 2.1$), 4.74 for one nonequivalent exo methylene proton ($\begin{matrix} H \\ | \\ H-C \end{matrix}$) τ ; t, d, $J_1 = J_2 = 2.1$), 4.65 for another nonequivalent exo methylene proton ($\begin{matrix} H_a \\ | \\ H_b-C \end{matrix}$) τ ; t, d, $J_1 = J_2 = 2.1$). *0.5 g of the compound (33) was also obtained by treatment of 1.32 g of 4-methylidene-1,3-oxathiolane-2-thione (34)⁸ with mercury acetate (4.0 g) in $CHCl_3$.

Reaction of 4-Methylidene-3-phenyl-oxazolidine-2-thione (22) and 1N HCl—A suspension of 0.19 g of 22 in 10 ml of 1N HCl was refluxed for 30 min, was neutralized with sat. $NaHCO_3$ aqueous solution. The precipitated solid was collected by filtration and recrystallized from benzene to afford 0.15 g of 4-methylidene-2-thione (35) as colorless needles of mp 99–100°. *Anal. Calcd.* for $C_{10}H_9ONS$: C, 62.82; H, 4.75; N, 7.33; S, 16.74. Found: C, 62.65; H, 4.77; N, 7.17; S, 16.74. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 1660 for a carbon-carbon double bond. NMR ($CDCl_3$) τ ($J = \text{Hz}$): 8.01 (3H, d, $J = 2.0$), 2.78 (1H, q, $J = 2.0$), 2.83–2.28 for five aromatic protons (m).

Reaction of *n*-Propoxycarbonyl Isothiocyanate (23-c) and 2-Propyn-1-ol (19)—2.9 g of 23-c was reacted with 1.12 g of 19 by the general method to afford three products (24-c, 25-c and 26-c). O-2-Propynyl-N-*n*-propoxycarbonyl-thiocarbamate (24-b), 0.41 g, colorless needles from benzene-*n*-hexane, mp 48–50°. *Anal. Calcd.* for $C_8H_{11}O_3NS$: C, 47.76; H, 5.51; N, 6.96; S, 15.90. Found: C, 47.50; H, 5.60; N, 6.61; S, 16.06. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3300 and 2100 ($HC \equiv C-$), 3230 and 1755 [$-C(O)NHC(S)-$]. NMR ($CDCl_3$) τ ($J = \text{Hz}$): 9.04 (3H, t, $J = 7.0$), 8.31 (2H, q, t, $J_1 = J_2 = 7.0$), 7.43 (1H, t, $J = 2.8$), 5.84 (2H, t, $J = 7.0$), 4.85 (2H, d, $J = 2.8$), *ca.* 1.50 (1H, broad). S-1,2-Propadien-1-yl-N-*n*-propoxycarbonyl-thiolcarbamate (25-c), 0.38 g, colorless needles from benzene, mp 65–66°. *Anal. Calcd.* for $C_8H_{11}O_3NS$: C, 47.76; H, 5.51; N, 6.96; S, 15.90. Found: C, 47.51; H, 5.67; N, 6.79; S, 16.17. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3200, 1755 and 1655 [$-C(O)NHC(O)-$], 1945 ($-C=C-$). NMR ($CDCl_3$) τ ($J = \text{Hz}$): 9.04 (3H, t, $J = 7.0$), 8.28 (2H, q, t, $J_1 = J_2 = 7.0$), 5.80 (2H, t, $J = 7.0$), 4.94 (2H, d, $J = 6.5$), 3.70 (1H, t, $J = 6.5$), *ca.* 5.70 for one secondary amide proton (broad). 4-Methylidene-2-*n*-propoxycarbonylimino-1,3-oxathiolane (26-c), 2.0 g, colorless needles from *n*-hexane-benzene, mp 74–75°. *Anal. Calcd.* for $C_8H_{11}O_3NS$: C, 47.76; H, 5.51; N, 6.96; S, 15.90. Found: C, 47.50; H, 5.68; N, 6.97; S, 16.05. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 1695 ($>C=O$), 1633 ($>C=C-$), 1579 ($>C=N-$). NMR ($CDCl_3$) τ ($J = \text{Hz}$): 9.15 (3H, t, $J = 7.0$), 8.27 (2H, q, t, $J_1 = J_2 = 7.0$), 5.84 (2H, t, $J = 7.0$), 4.93 (2H, t, $J = 2.5$) 4.67 for one nonequivalent exo methylene proton ($\begin{matrix} H_a \\ | \\ H_d \end{matrix}$) τ ; t, d, $J_1 = 2.5$, $J_2 = 2.3$), 4.61 for another nonequivalent exo methylene proton ($\begin{matrix} H_a \\ | \\ H_b \end{matrix}$) τ ; t, d, $J_1 = 2.5$, $J_2 = 2.3$).

Reaction of Isopropoxycarbonyl Isothiocyanate (23-d) and 2-Propyn-1-ol (19)—2.9 g of 23-d was reacted with 1.12 g of 19 by the general method to afford three products (24-d, 25-d and 26-d). O-2-Propynyl-N-isopropoxycarbonyl-thiocarbamate (24-d) 0.4 g, colorless needles from *n*-hexane-benzene, mp 82–84°. *Anal. Calcd.* for $C_8H_{11}O_3NS$: C, 47.76; H, 5.51; N, 6.96; S, 15.90. Found: C, 47.56; H, 5.37; N, 6.46; S, 16.35. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3300 and 2120 ($HC \equiv C-$), 3220 and 1758 [$-C(S)NHC(O)-$]. NMR ($CDCl_3$) τ ($J = \text{Hz}$): 8.68 (6H, d, $J = 6.8$), 7.42 (1H, t, $J = 2.5$), 5.01 (1H, q, q, $J_1 = J_2 = 6.8$), 4.82 (2H, d, $J = 2.5$), *ca.* 1.66 (1H, broad). S-1,2-Propadien-1-yl-N-isopropoxycarbonyl-thiolcarbamate (25-b), 0.33 g, colorless needles from *n*-hexane-benzene, mp 43–46°. *Anal. Calcd.* for $C_8H_{11}O_3NS$: C, 47.76; H, 5.51; N, 6.96;

17) P.A. Leven and A. Walti, *Org. Syn.*, Collect. II, p 5.

S, 15.90. Found: C, 47.58; H, 5.39; N, 6.85; S, 16.23. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3230, 1750 and 1665 [$-\text{C}(\text{O})\text{NHC}(\text{O})-$], 1945 ($-\text{C}=\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.68 (6H, d, $J=7.0$), 4.06 (1H, q, q, $J_1=J_2=7.0$), 4.04 (2H, d, $J=6.5$), 3.65 (1H, t, $J=6.5$), ca. 1.72 (1H, broad). 4-Methylidene-2-isopropoxycarbonylimino-1,3-oxathiolane (26-d), 1.79 g, colorless needles from *n*-hexane-benzene, mp 64–65°. Anal. Calcd. for $\text{C}_9\text{H}_{11}\text{O}_3\text{NS}$: C, 47.76; H, 5.51; N, 6.96; S, 15.90. Found: C, 47.57; H, 5.61; N, 6.83; S, 16.16. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 1680 ($>\text{C}=\text{O}$), 1632 ($>\text{C}=\text{C}<$), 1572 ($>\text{C}=\text{N}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.67 (6H, d, $J=7.0$), 4.99 (2H, t, $J=2.5$), 4.67 for one nonequivalent exo methylene proton ($\begin{smallmatrix} \text{H}^{\text{a}} \\ \text{H}^{\text{b}} \end{smallmatrix}$) τ ; t, d, $J_1=2.5$, $J_2=2.3$), 4.62 for another nonequivalent methylene proton ($\begin{smallmatrix} \text{H}^{\text{a}} \\ \text{H}^{\text{b}} \end{smallmatrix}$) τ ; t, d, $J_1=2.5$, $J_2=2.3$).

Reaction of *n*-Butoxycarbonyl Isothiocyanate (23-e) and 2-Propyn-1-ol (19)—3.5 g of 23-e was reacted with 1.12 g of 19 by the general method to afford three products (24-e, 25-e and 26-e). O-2-Propynyl-N-*n*-butoxycarbonyl-thiocarbamate (24-e), 0.52 g, pale yellow oil, $n_D^{20} = 1.5194$. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.23; H, 6.08; N, 6.51; S, 14.86. Found: C, 50.61; H, 6.31; N, 6.81; S, 14.81. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3300 and 2100 ($\text{HC}\equiv\text{C}-$), 1780 ($>\text{C}=\text{O}$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.06 (3H, t, $J=6.5$), 8.90–7.95 (4H, m), 5.79 (2H, t, $J=6.5$), 7.46 (1H, t, $J=2.5$), 4.84 (2H, d, $J=2.5$), ca. 1.60 (1H, broad). S-1,2-Propadien-1-yl-N-*n*-butoxycarbonyl-thiocarbamate (25-e), 0.29 g, colorless needles from *n*-hexane, mp 50–51°. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.23; H, 6.08; N, 6.51; S, 14.86. Found: C, 49.87; H, 6.10; N, 6.27; S, 14.81. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3200, 1752 and 1660 [$-\text{C}(\text{O})\text{NHC}(\text{O})-$], 1940 ($-\text{C}=\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.06 (3H, t, $J=6.5$), 0.93–7.99 (4H, m), 5.76 (2H, t, $J=6.5$), 4.90 (2H, d, $J=6.5$), 3.68 (1H, t, $J=6.5$), ca. 1.50 (1H, broad). 2-*n*-Butoxycarbonylimino-4-methylidene-1,3-oxathiolane (26-e), 1.78 g, colorless needles from *n*-hexane-benzene, mp 49–50°. Anal. Calcd. for $\text{C}_9\text{H}_{11}\text{O}_3\text{NS}$: C, 50.23; H, 6.08; N, 6.51; S, 14.86. Found: C, 50.06; H, 6.16; N, 6.45; S, 15.01. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 1685 ($>\text{C}=\text{O}$), 1632 ($>\text{C}=\text{C}<$), 1570 ($>\text{C}=\text{N}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.06 (3H, t, $J=6.5$), 8.92–7.98 (4H, m), 5.80 (2H, t, $J=6.5$), 4.88 (2H, t, $J=2.5$), 4.65 for one nonequivalent exo methylene ($\begin{smallmatrix} \text{H}^{\text{a}} \\ \text{H}^{\text{b}} \end{smallmatrix}$) τ ; t, d, $J_1=2.5$, $J_2=2.3$), 4.60 for another nonequivalent exo methylene proton ($\begin{smallmatrix} \text{H}^{\text{a}} \\ \text{H}^{\text{b}} \end{smallmatrix}$) τ ; t, d, $J_1=2.5$, $J_2=2.3$).

Reaction of Isobutoxycarbonyl Isothiocyanate (23-f) and 2-Propyn-1-ol (19)—3.5 g of 23-f was reacted with 1.12 g of 19 by the general method to afford two products (24-f and 26-f): O-2-Propynyl-N-isobutoxycarbonyl-thiocarbamate (24-f), 0.64 g, colorless needles from *n*-hexane-benzene, mp 52–54°. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.23; H, 6.09; N, 6.51; S, 14.86. Found: C, 50.08; H, 6.28; N, 6.20; S, 14.89. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 2230 and 2120 ($\text{HC}\equiv\text{C}-$), 3230 and 1770 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$]. NMR (CDCl_3) τ ($J=\text{Hz}$): 9.04 (6H, t, $J=7.0$), 8.02 (1H, q, q, t, $J_1=J_2=J_3=7.0$), 7.44 (1H, t, $J=3.0$), 6.00 (2H, d, $J=7.0$), 4.83 (2H, d, $J=3.0$), ca. 1.72 (1H, broad). 2-Isobutoxycarbonylimino-4-methylidene-1,3-oxathiolane (26-f), 1.38 g, colorless needles from *n*-hexane-benzene, mp 53–54°. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.23; H, 6.09; N, 6.51; S, 14.86. Found: C, 49.89; H, 6.17; N, 6.24; S, 14.89. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 1685 ($>\text{C}=\text{O}$), 1632 ($>\text{C}=\text{C}<$), 1570 ($>\text{C}=\text{N}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.02 (6H, d, $J=7.0$), 7.98 (1H, q, q, t, $J_1=J_2=J_3=7.0$), 6.01 (2H, d, $J=7.0$), 4.89 (2H, t, $J=2.5$), 4.66 for one nonequivalent exo methylene proton ($\begin{smallmatrix} \text{H}^{\text{a}} \\ \text{H}^{\text{b}} \end{smallmatrix}$) τ ; t, d, $J_1=2.5$, $J_2=2.3$), 4.60 for another nonequivalent exo methylene proton ($\begin{smallmatrix} \text{H}^{\text{a}} \\ \text{H}^{\text{b}} \end{smallmatrix}$) τ ; t, d, $J_1=2.5$, $J_2=2.3$).

General Method for Preparation of N-Alkoxy carbonyl-O-2-butynyl-thiocarbamates (37)—A solution of 0.01 mole of 2-butyn-1-ol (36), 0.011 mole of alkoxy carbonyl isothiocyanates (23) in 50 ml of tetra chloromethane was refluxed for 8 hr, and the solvent was evaporated. The resulting solid was recrystallized from a suitable solvent.

Reaction of Methoxycarbonyl Isothiocyanate (23-a) and 2-Butyn-1-ol (36)—1.3 g of 23-a was reacted with 0.7 g of 36 by the general method, and the resulting solid was recrystallized from *n*-hexane to afford 1.55 g of O-2-butynyl-N-methoxycarbonyl-thiocarbamate (37-a) as colorless needles of mp 90–92°. Mass Spectrum m/e : $M^+ = 187$. Anal. Calcd. for $\text{C}_7\text{H}_9\text{O}_3\text{NS}$: C, 44.91; H, 4.85; N, 7.48; S, 17.13. Found: C, 45.00; H, 5.00; N, 7.23; S, 16.98. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3190 and 1758 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2220 ($-\text{C}\equiv\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.10 (3H, t, $J=2.5$), 6.18 (3H, s), 4.88 (2H, q, $J=2.5$), ca. 1.44 (1H, broad).

Reaction of Ethoxycarbonyl Isothiocyanate (23-b) and 2-Butyn-1-ol (36)—1.45 g of 23-b was reacted with 0.7 g of 36 by the general method, and the resulting solid was recrystallized from isopropyl ether to afford 1.75 g O-2-butynyl-N-ethoxycarbonyl-thiocarbamate (37-b) as colorless needles of mp 85.87°. Mass Spectrum m/e : $M^+ = 201$. Anal. Calcd. for $\text{C}_8\text{H}_{11}\text{O}_3\text{NS}$: C, 47.76; H, 5.51; N, 6.96; S, 15.65. Found: C, 47.65; H, 5.64; N, 7.00; S, 15.77. IR $\nu_{\max}^{\text{Nujol}}$ cm^{-1} : 3200 and 1775 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2210 ($-\text{C}\equiv\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.70 (3H, t, $J=7.0$), 8.12 (3H, t, $J=2.8$), 5.70 (2H, q, $J=7.0$), 4.90 (1H, q, $J=2.8$), ca. 1.62 (1H, broad). *1.45 g (0.011 mole) of 23-b was reacted with 0.7 g (0.01 mole) of 36 at 50–70° in nitrogen atmosphere to give 1.16 g (58.0%) of 37-b.

Reaction of *n*-Propoxycarbonyl Isothiocyanate (23-c) and 2-Butyn-1-ol (36)—1.6 g of 23-c was reacted with 0.7 g of 36 by the general method. The resulting solid was recrystallized from *n*-hexane-benzene to afford 1.26 g of O-2-butynyl-N-isopropoxycarbonyl-thiocarbamate (37-c) as colorless needles of mp 70–71°. Mass Spectrum m/e : $M^+ = 215$. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.21; H, 6.09; N, 6.51; S, 14.90.

Found: C, 50.03; H, 6.16; N, 6.61; S, 14.87. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3210 and 1766 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2230 ($-\text{C}\equiv\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.03 (3H, t, $J=6.8$), 8.29 (2H, q, t, $J_1=J_2=6.8$), 8.10 (3H, t, $J=2.8$), 5.84 (2H, t, $J=6.8$), 4.99 (2H, q, $J=2.8$), ca. 1.59 (1H, broad).

Reaction of Isopropoxycarbonyl Isothiocyanate (23-d) and 2-Butyn-1-ol (36)—1.6 g of 23-d was reacted with 0.7 g of 36 by the general method, and then the resulting solid was recrystallized from *n*-hexane–benzene to afford 1.39 g of O-2-butynyl-N-isopropoxycarbonyl-thiocarbamate (37-d) as colorless needles of mp 88–90°. Mass Spectrum m/e : $M^+=215$. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.21; H, 6.09; N, 6.91; S, 14.90. Found: C, 50.29; H, 6.35; N, 6.52; S, 14.55. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3200 and 1760 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2230 ($-\text{C}\equiv\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.71 (6H, d, $J=7.0$), 8.10 (3H, t, $J=2.8$), 5.00 (1H, q, q, $J_1=J_2=7.0$), 4.88 (2H, q, $J=2.8$), ca. 1.65 (1H, broad).

Reaction of *n*-Butoxycarbonyl Isothiocyanate (23-e) and 2-Butyn-1-ol (36)—1.75 g of 23-e was reacted with 0.7 g of 36 by the general method, and then the resulting solid was recrystallized from *n*-hexane to afford 0.5 g of N-*n*-butoxycarbonyl-O-2-butynyl-thiocarbamate (37-e) as colorless needles from *n*-hexane of mp 43–44°. Mass Spectrum m/e : $M^+=229$. Anal. Calcd. for $\text{C}_{10}\text{H}_{15}\text{O}_3\text{NS}$: C, 52.38; H, 6.59; N, 6.11; S, 13.98. Found: C, 52.18; H, 6.67; N, 6.20; S, 13.97. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3200 and 1772 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2230 ($-\text{C}\equiv\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.07 (3H, t, $J=6.5$), 8.91–8.00 (4H, m), 8.13 (3H, t, $J=2.8$), 5.82 (2H, t, $J=6.5$), 4.90 (2H, q, $J=2.8$), ca. 1.75 (1H, broad).

Reaction of Isobutoxycarbonyl Isothiocyanate (23-f) and 2-Butyn-1-ol (36)—1.75 g of 23-f was reacted with 0.7 g of 36 by the general method, and then the resulting solid was recrystallized from *n*-hexane to afford 0.51 g of O-2-butynyl-N-isobutoxycarbonyl-thiocarbamate as colorless needles of mp 50–51°. Mass Spectrum m/e : $M^+=229$. Anal. Calcd. for $\text{C}_{10}\text{H}_{15}\text{O}_3\text{NS}$: C, 52.38; H, 6.59; N, 6.11; S, 13.98. Found: C, 52.08; H, 6.51; N, 5.89; S, 13.98. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3210 and 1765 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2230 ($-\text{C}\equiv\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.02 (6H, d, $J=7.0$), 8.11 (3H, t, $J=3.0$), 8.08 (1H, t, d, d, $J_1=J_2=J_3=7.0$), 6.08 (2H, d, $J=7.0$), 4.94 (2H, q, $J=3.0$), ca. 1.62 (1H, broad).

General Cyclization Method of N-Alkoxycarbonyl-O-2-butynyl-thiocarbamates (37)—A solution of 0.01 mole of 37 in 50 ml of EtOH involving 0.01 mole of Et_3N was refluxed for 1 hr, and EtOH was removed under reduced pressure. The residual solid was recrystallized from a suitable solvent.

Synthesis of 4-Ethylidene-2-methoxycarbonylimino-1,3-oxathiolane (39-a)—1.87 g of O-2-butynyl-N-methoxycarbonyl-thiocarbamate (37-a) was treated with Et_3N in EtOH according to the general method to afford 1.12 g of 39-a as colorless prisms (*n*-hexane–benzene) of mp 102–104°. Anal. Calcd. for $\text{C}_7\text{H}_9\text{O}_3\text{NS}$: C, 44.91; H, 4.85; N, 7.48; S, 17.13. Found: C, 44.70; H, 5.02; N, 7.21; S, 17.23. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1698 (>C=O), 1640 (>C=C<), 1585 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.29 (3H, d, t, $J_1=7.0$, $J_2=2.0$), 6.14 (3H, s), 4.95 (2H, d, q, $J_1=2.2$, $J_2=2.0$), 4.25 (1H, q, t, $J_1=7.0$, $J_2=2.2$).

Synthesis of 4-Ethylidene-2-ethoxycarbonylimino-1,3-oxathiolane (39-b)*—2.01 g of O-2-butynyl-N-ethoxycarbonyl-thiocarbamate (37-b) was treated with Et_3N according to the general method to afford 1.12 g of 39-b as colorless prisms (isopropyl ether) of mp 104–105°. Anal. Calcd. for $\text{C}_8\text{H}_{11}\text{O}_3\text{NS}$: C, 47.74; H, 5.50; N, 6.96; S, 15.93. Found: C, 47.60; H, 5.41; N, 7.03; S, 15.80. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1685 (>C=O), 1640 (>C=C<), 1590 (>C=N-). UV $\lambda_{\text{max}}^{\text{EtOH}}$ $m\mu$ (ϵ): 266 (6400). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.66 (3H, t, $J=7.0$), 8.20 (3H, d, t, $J_1=7.0$, $J_2=2.0$), 5.73 (2H, q, $J=7.0$), 4.96 (2H, d, q, $J_1=2.2$, $J_2=2.0$), 4.26 (1H, q, t, $J_1=7.0$, $J_2=2.2$). *2.01 g of 37-b was heated at 120–130° for 3 hr on an oil bath, and then the reaction mixture was eluted with benzene on silica gel column to afford 0.82 g of 37-b and 0.43 g (21.4%) of S-2,3-butadiene-2-yl-N-ethoxycarbonyl-thiocarbamate (38) as colorless needles of mp 74–75°. Anal. Calcd. for $\text{C}_8\text{H}_{11}\text{O}_3\text{NS}$: C, 47.76; H, 5.51; N, 6.96; S, 15.91. Found: C, 47.80; H, 5.60; N, 6.63; S, 16.17. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3180, 1742 and 1656 [$-\text{C}(\text{O})\text{NHC}(\text{O})-$], 1945 ($-\text{C}=\text{C}-$). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.68 (3H, t, $J=7.0$), 7.90 (3H, t, $J=3.5$), 5.73 (2H, q, $J=7.0$), 5.62 (2H, q, $J=3.5$), ca. 1.97 (1H, broad).

Synthesis of 4-Ethylidene-2-*n*-propoxycarbonylimino-1,3-oxathiolane (39-c)—2.16 g of O-2-butynyl-N-*n*-propoxycarbonyl-thiocarbamate (27-c) was treated with Et_3N according to the general method to afford 0.91 g of 39-c as colorless needles (*n*-hexane–benzene) of mp 73–75°. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.21; H, 6.09; N, 6.51; S, 14.90. Found: C, 50.06; H, 6.43; N, 6.27; S, 15.24. *IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1695 (>C=O), 1590 (>C=N-). *The absorption bands of a carbon-carbon double bond (>C=C<) are not clear. NMR (CDCl_3) τ ($J=\text{Hz}$): 9.08 (3H, t, $J=7.0$), 8.25 (2H, q, t, $J_1=J_2=7.0$), 8.20 (3H, d, t, $J_1=7.0$, $J_2=1.8$), 5.83 (2H, t, $J=7.0$), 4.95 (2H, d, q, $J_1=2.0$, $J_2=1.8$), 4.25 (1H, q, t, $J_1=7.0$, $J_2=2.0$).

Synthesis of 4-Ethylidene-2-isopropoxycarbonylimino-1,3-oxathiolane (39-d)—2.16 g of O-2-butynyl-N-isopropoxycarbonyl-thiocarbamate (27-d) was treated with Et_3N according to the general method to afford 0.92 g of 39-d as colorless needles (*n*-hexane–benzene) of mp 74–76°. Anal. Calcd. for $\text{C}_9\text{H}_{13}\text{O}_3\text{NS}$: C, 50.21; H, 6.09; N, 6.51; S, 14.90. Found: C, 49.88; H, 6.04; N, 6.59; S, 14.86. *IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1690 (>C=O), 1590 ($-\text{C}=\text{N-}$). *The absorption bands of a carbon-carbon double bond ($-\text{C}=\text{C}-$) are not clear. NMR (CDCl_3) τ ($J=\text{Hz}$): 8.67 (6H, d, $J=6.4$), 8.20 (3H, d, t, $J_1=7.0$, $J_2=1.9$), 4.98 (2H, d, q, $J_1=2.0$, $J_2=1.9$), 4.26 for (1H, q, t, $J_1=7.0$, $J_2=2.0$).

Synthesis of 2-*n*-Butoxycarbonylimino-4-ethylidene-1,3-oxathiolane (39-e)—2.3 g of N-*n*-Butoxycarbonyl-O-2-butynyl-thiocarbamate (37-e) was treated with Et_3N according to the general method to afford 0.515 g of 39-e as colorless needles (pet. ether) of mp 35–37°. Anal. Calcd. for $\text{C}_{10}\text{H}_{15}\text{O}_3\text{NS}$: C, 52.38; H, 6.59; N, 6.11; S, 13.98. Found: C, 52.18; H, 6.78; N, 6.12; S, 14.13. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1685 (>C=O), 1640

(>C=C-), 1560 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.09 (3H, t, $J=6.0$), 8.95–8.01 (4H, m), 8.22 (3H, d, t, $J_1=7.0$, $J_2=1.9$), 5.81 (2H, t, $J=6.0$), 4.97 (2H, q, t, $J_1=7.0$, $J_2=2.0$), 4.27 (1H, q, t, $J_1=7.0$, $J_2=2.0$).

Synthesis of 2-Isobutoxycarbonylimino-4-ethylidene-1,3-oxathiolane (39-f)—2.3 g of *N*-isobutoxycarbonyl-O-2-butynyl-thiocarbamate (37-f) was treated with Et_3N according to the general method to afford 0.535 g of 39-f as colorless needles (pet. ether) of mp 41–43°. *Anal.* Calcd. for $\text{C}_{10}\text{H}_{11}\text{O}_3\text{NS}$: C, 52.38; H, 6.59; N, 6.11; S, 13.98. Found: C, 52.03; H, 6.50; N, 6.17; S, 13.95. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1684 (>C=O), 1632 (>C=C<), 1583 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.03 (6H, d, $J=6.5$), 8.21 (3H, d, t, $J_1=7.0$, $J_2=1.9$), 7.92 (1H, q, q, t, $J_1=J_2=J_3=6.5$), 6.01 (2H, d, $J=6.5$), 4.97 (2H, d, t, $J_1=2.0$, $J_2=1.9$), 4.24 (1H, q, t, $J_1=7.0$, $J_2=2.0$).

Reaction of Ethoxycarbonyl Isothiocyanate (23-b) and 3-Phenyl-2-propyn-1-ol (40)—A mixture of 1.45 g of 23-b and 1.32 g of 40 was heated at 50–70° for 5 hr on an oil bath. The reaction mixture was eluted with benzene on silica gel column to give two products (41 and 42-b): O-3-Phenyl-2-propynyl-*N*-ethoxycarbonyl-thiocarbamate (41), 1.88 g, n_D^{20} : 1.5890, pale yellow oil. *Anal.* Calcd. for $\text{C}_{13}\text{H}_{13}\text{O}_3\text{NS}$: C, 59.31; H, 4.98; N, 5.32; S, 12.16. Found: C, 59.53; H, 5.26; N, 5.17; S, 11.63. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3270 and 1770 [$-\text{C}(\text{S})\text{NHC}(\text{O})-$], 2230 (>C=C-). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.69 (3H, t, $J=7.0$), 5.74 (2H, q, $J=7.0$), 4.60 (2H, s), 2.82–2.35 for five aromatic protons (m), *ca.* 1.64 (1H, broad). UV $\lambda_{\text{max}}^{\text{EtOH}}$ $m\mu$ (ϵ): 246 (20200), 256 (21000). 2-Ethoxycarbonylimino-4-benzylidene-1,3-oxathiolane (42-b), 0.15 g, colorless needles from *n*-hexane–benzene, mp 96–97°. Mass Spectrum m/e : M^+ =263. *Anal.* Calcd. for $\text{C}_{13}\text{H}_{13}\text{O}_3\text{NS}$: C, 59.31; H, 4.98; N, 5.32; S, 12.16. Found: C, 59.26; H, 4.95; N, 5.42; S, 12.15. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1680 (>C=O), 1650 (>C=C<), 1575 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 8.60 (3H, t, $J=7.0$), 5.74 (2H, q, $J=7.0$), 4.75 (2H, d, $J=2.0$), 3.35 (1H, t, $J=2.0$), 2.82–2.47 for five aromatic protons (m). UV $\lambda_{\text{max}}^{\text{EtOH}}$ $m\mu$ (ϵ): 271 (15300).

Cyclization Reaction of 41—To benzene (200 ml) solution of 2.63 g of 41, 4% KOH aqueous (10 ml) was added at room temperature, and the mixture was stirred for 30 minutes at room temperature. The organic layer was washed with H_2O , dried over anhyd. Na_2SO_4 , and then the solvent was removed under reduced pressure. The resulting solid was recrystallized from isopropyl ether to afford 2.17 g of 42-b.

General Method for Preparation of 2-Alkoxy carbonylimino-4-benzylidene-1,3-oxathiolanes (42)—The mixture of 0.011 mole of alkoxy carbonyl isothiocyanates (23) and 0.01 mole of 3-phenyl-2-propyn-1-ol (40) was heated at 50–70° in nitrogen atmosphere for 5 hr, and then to the reaction mixture 200 ml of benzene and 4% KOH aqueous (10 ml) were added and stirred at room temperature for 30 minutes. The organic layer was washed with H_2O , and dried over anhyd. Na_2SO_4 . After removal of benzene under reduced pressure, the resulting solid was recrystallized from a suitable solvent.

Synthesis of 4-Benzylidene-2-methoxycarbonylimino-1,3-oxathiolane (42-a)—1.31 g of methoxycarbonyl isothiocyanate (23-a) was reacted with 1.32 g of 40, and the reaction mixture was treated with 4% KOH by the general method to give 2.25 g of 42-a, colorless needles from *n*-hexane–benzene, mp 106–108°. *Anal.* Calcd. for $\text{C}_{12}\text{H}_{11}\text{O}_3\text{NS}$: C, 57.83; H, 4.45; N, 5.62; S, 12.84. Found: C, 57.63; H, 4.74; N, 5.71; S, 13.13. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1690 (>C=O), 1640 (>C=C-), 1575 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 6.15 (3H, s), 4.74 (2H, d, $J=2.3$), 3.32 (1H, t, $J=2.3$), 2.81–2.44 for five aromatic protons (m).

Synthesis of 4-Benzylidene-2-ethoxycarbonylimino-1,3-oxathiolane (42-b)—The reaction of ethoxycarbonyl isothiocyanate (23-b, 1.45 g) with 1.32 g of 40 afforded 2.29 g of 42-b.

Synthesis of 4-Benzylidene-2-*n*-propoxycarbonylimino-1,3-oxathiolane (42-c)—The reaction of *n*-propoxycarbonyl isothiocyanate (23-c; 1.59 g) with 1.32 g of 40 afforded 2.06 g of 42-c, colorless needles from *n*-hexane–benzene, mp 86–88°. *Anal.* Calcd. for $\text{C}_{14}\text{H}_{15}\text{O}_3\text{NS}$: C, 60.64; H, 5.45; N, 5.05; S, 11.54. Found: C, 60.62; H, 5.77; N, 5.02; S, 11.65. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1695 (>C=O), 1635 (>C=C<), 1588 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.02 (3H, t, $J=7.0$), 6.25 (2H, q, t, $J_1=J_2=7.0$), 5.82 (2H, t, $J=7.0$), 4.75 (2H, d, $J=2.3$), 3.33 (1H, t, $J=2.3$), 2.87–2.45 for five aromatic protons (m).

Synthesis of 4-Benzylidene-2-isopropoxycarbonylimino-1,3-oxathiolane (42-d)—The reaction of isopropoxycarbonyl isothiocyanate (23-d, 1.59 g) with 1.32 g of 40 afforded 2.01 g of 42-d, colorless needles from *n*-hexane–benzene, mp 93–94°. *Anal.* Calcd. for $\text{C}_{14}\text{H}_{15}\text{O}_3\text{NS}$: C, 60.64; H, 5.45; N, 5.05; S, 11.54. Found: C, 60.70; H, 5.46; N, 5.09; S, 11.82. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1675 (>C=O), 1635 (>C=C<), 1580 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 2.67 (6H, d, $J=6.5$), 5.03 (1H, q, q, $J_1=J_2=6.5$), 4.75 (2H, d, $J=2.1$), 3.33 (1H, t, $J=2.1$), 2.86–2.47 for five aromatic protons (m).

Synthesis of 4-Benzylidene-2-*n*-butoxycarbonylimino-1,3-oxathiolane (42-e)—The reaction of *n*-butoxycarbonyl isothiocyanate (23-e, 1.73 g) with 1.32 g of 40 afforded 1.99 g of 42-e, colorless needles from isopropyl ether, mp 58–59°. *Anal.* Calcd. for $\text{C}_{15}\text{H}_{17}\text{O}_3\text{NS}$: C, 61.85; H, 5.88; N, 4.81; S, 10.98. Found: C, 61.64; H, 6.00; N, 4.70; S, 10.97. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1692 (>C=O), 1630 (>C=C<), 1580 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.06 (3H, t, $J=7.0$), 8.95–7.98 (4H, m), 5.78 (2H, t, $J=7.0$), 4.75 (2H, d, $J=2.1$), 3.34 (1H, t, $J=2.1$), 2.80–2.44 for five aromatic protons (m).

Synthesis of 4-Benzylidene-2-isobutoxycarbonylimino-1,3-oxathiolane (42-f)—The reaction of isobutoxycarbonyl isothiocyanate (23-f, 1.73 g) with 1.32 g of 40 afforded 2.02 g of 42-f, colorless needles from *n*-hexane–benzene, mp 79–81°. *Anal.* Calcd. for $\text{C}_{15}\text{H}_{17}\text{O}_3\text{NS}$: C, 61.85; H, 5.88; N, 4.81; S, 10.98. Found: C, 61.77; H, 5.91; N, 4.56; S, 10.98. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 1680 (>C=O), 1640 (>C=C<), 1590 (>C=N-). NMR (CDCl_3) τ ($J=\text{Hz}$): 9.03 (6H, d, $J=7.0$), 8.02 (1H, q, q, t, $J_1=J_2=J_3=7.0$), 5.99 (2H, d, $J=7.0$), 4.74 (2H, d, $J=2.3$), 3.32 (1H, t, $J=2.3$), 2.84–2.49 for five aromatic protons (m).

Synthesis of O-2-Propynyl-N-isopropoxycarbonyl-N(D)-thiocarbamate (43)—4.02 g (0.02 mole) of **24-d** was suspended in D₂O (50 ml), and stirred for 5 hr at room temperature. After removal of D₂O under reduced pressure, the residual solid was recrystallized from isopropyl ether to afford 3.2 g of **43** as colorless needles of mp 82—83°. *Anal.* Calcd. for C₈H₁₂O₃NS: C, 47.50; H, 5.98; N, 6.92; S, 15.85. Found: C, 47.33; H, 5.62; N, 6.92; S, 15.84. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3250, 2100 for a ethynyl group (HC≡C-), 2320 for Nitrogen-Deuterium bond (>N-D: strong), 1755 (>C=O). NMR (CCl₄) τ (*J*=Hz): 8.70 (6H, d, *J*=6.7), 7.52 for one ethynyl proton (HC≡C-; t, *J*=2.5), 5.00 (1H, q, q, *J*₁=*J*₂=6.7), 4.90 for two methylene protons (-C≡C-CH₂-O-; d, *J*=2.5).

Cyclization Reaction of 43 to C(D)-4-Methylidene-2-isopropoxycarbonylimino-1,3-oxathiolane (44)—A solution of 2.02 g of **43** in 30 ml of triethyl amine was refluxed for 30 minutes, and triethyl amine was removed under reduced pressure. The resulting solid was recrystallized from isopropyl ether to afford 0.8 g of **44** as colorless needles of mp 64—65°. *Anal.* Calcd. for C₈H₁₂O₃NS: C, 47.50; H, 5.58; N, 6.92; S, 15.85. Found: C, 47.78; H, 5.98; N, 6.86; S, 15.85. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 1675 (>C=O), 1570 (>C=N-). NMR (CDCl₃) τ (*J*=Hz): 8.68 (6H, d, *J*=7.0), 4.98 (1H, q, q, *J*₁=*J*₂=7.0), 4.92 for two ring methylene protons (d, *J*=2.3), 4.67 for one exo methylene proton ($\frac{\text{H}}{\text{D}}$ >C=C-; t, *J*=2.3).

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