$(R_{\rm PA}\ 1.60)$  showed ultraviolet (UV) spectrum similar to 8-mercaptoadenosine. The reaction mixture was adjusted to pH 6 with 2N HCl and nitrogen gas was passed through the solution. The solution was again adjusted to pH 7.5 with 2N NaOH and concentrated to *ca*. 5 ml. The concentrated solution was applied to a column (2×60 nm) of Biogel P-2. Fractions of 3 ml were collected and fractions 23-37 were combined. The yield was 890 A<sub>298</sub> units 0.033 mmole, 78%. UV:  $\lambda_{\rm max}^{\rm PH}^{2}$  261, 236 nm;  $\lambda_{\rm max}^{\rm PH}^{2}$  254 nm;  $\lambda_{\rm max}^{\rm PH}^{2}$  253 nm. Paper chromatography in solvent B:  $R_{\rm PA}$ =0.40, in solvent D:  $R_{\rm PA}$ =1.74.

8-Mercaptoadensoine 3',5'-Cyclic Phosphate——8-Bromoadenosine-3',5'-cyclic phosphate<sup>7</sup>) (100 mg) was dissolved in a mixture of DMF (0.3 ml) and water (0.3 ml). Aqueous sodium hydrogen sulfide (407, 100  $\mu$ l) was added to the mixture. After 15 hr at 25°, paper electrophoresis showed almost complete conversion of the starting material. The reaction mixture was adjusted to pH 7 with 2N HCl and nitrogen gas was passed through the solution, which was adjusted to pH 7.5 with 2N NaOH. The solution was applied to a column (2×50 cm) of DEAE-cellulose (bicarbonate form). The column was eluted with triethylammonium bicarbonate buffer (pH 7.5, 0.01—0.1M, 21+21) using linear gradient technique. Fractions (10 material solution was lyophilized to give a powder. Yield was 76%. UV:  $\lambda_{max}^{pH 1}$  220, 244, 308 m $\mu$ ;  $\lambda_{max}^{ph 2}$  221, 245, 310 m $\mu$ ;  $\lambda_{max}^{ph 1}$  224, 290 nm. PPC:  $R_{pA}$  in solvent A 1.65 and in solvent B 0.65. These properties were identical to those reported by Muneyama, *et al.*<sup>14</sup>)

14) K. Muneyama, R.J. Bauer, D.S. Shuman, R. K. Robins, and L.N. Simon, Biochemistry, 10, 2390 (1971).

 $\begin{bmatrix} \text{Chem. Pharm. Bull.} \\ \mathbf{21}(2) & 445-447 & (1973) \end{bmatrix}$ 

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## Phosphorylation by Active Ester of Phosphoric Acid. I. Preparation and Reaction of Phenyl 8-Quinolyl Phosphate<sup>1)</sup>

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It is well known that active esters of phosphoric acid such as  $RO-P-OR'^{(3)}$  (diester) and O

 $RO\overset{\mu}{P}$ -NR'R''<sup>4</sup>) (phosphoramidate) are important intermediates in the synthesis of various  $\overset{\mu}{O}$ -

kinds of phosphoric acid derivatives.

In the present study, the preparation of phenyl 8-quinolyl phosphate (4) and the reactions of the active diester of phosphoric acid (4) with alcohols and amines were investigated. It was considered that the phosphate (4) would yield a reactive intermediate (4'), which in turn would further react with alcohols to give diesters of phosphoric acid as shown in Chart1. The synthesis of diester of phosphoric acid can be effected successfully by this method since it reacts exclusively with nucleophilic reagents such as alcohols and amines under mild conditions.

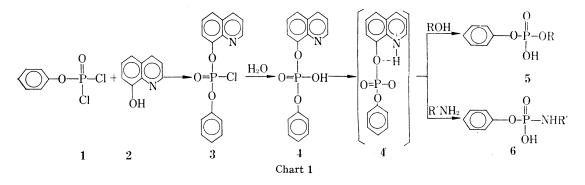
<sup>1)</sup> Part of this work was presented at the 90th Annual Meeting of the Pharmaceutical Society of Japan, Sapporo, July 1970.

<sup>2)</sup> Location: Narashino-shi, Chiba.

 <sup>3)</sup> a) W. Kampe, Chem. Ber., 98, 1031 (1965); b) W. Kampe, ibid., 98, 1038 (1965); c) K.H. Scheit and W. Kampe, ibid., 98, 1045 (1965).

<sup>4)</sup> F. Cramer, H. Schaller, and H.A. Staab, Chem. Ber., 94, 1612 (1961); H. Schaller, H.A. Staab, and F. Cramer, *ibid.*, 94, 1621 (1961); F. Cramer and H. Schaller, *ibid.*, 94, 1634 (1961); J.G. Moffatt and H.G. Khorana, J. Amer. Chem. Soc., 83, 649 (1961).

The phosphorylating reagent(4) was prepared by the sequence of reaction shown in Chart The phenyl phosphorodichloridate (1) reacted with one mole of 8-quinol (2) in dry pyridine 1. solution to give phenyl 8-quinolyl phosphorochloridate (3) which was hydrolyzed to afford phenyl 8-quinolyl phosphate (4). The phosphate (4) was isolated from the reaction mixture by extraction with carbon tetrachloride and purified by silica gel column chromatography to afford the pure phosphate (4) in 78% yield based on 1 as viscous yellowish green oil. The structure of 4 was supported by elemental analysis and infrared (IR) spectra.



Alcohols were selectively phosphorylated with 4; for example, when 4 was treated with a slight excess of ethyl alcohol in dry pyridine at 50° for 8 hr, ethyl phenyl phosphate (5b) was obtained in 72% yield as white crystal of cyclohexylammonium salt. The structure of 5b was supported by elemental analysis and IR spectrum which showed absorption bands at 1600 (C=C), 1255 (P=O) and 1025 cm<sup>-1</sup> (P–O– $C_2H_5$ ). The paper chromatography of the reaction mixture showed two spots of the phosphate (5b (Rf 0.66) and 4 (Rf 0.79)). According

Compd. No.	R	Time (hr)	Yield <sup>b)</sup> (%)	mp (°C)	Rf <sup>c)</sup>	Formula		Analysis (%)		%)
								с	H	N
5a	CH3	6	68	153—154 <sup>d)</sup>	0.60	$\mathrm{C_{13}H_{22}O_4NP}$	Calcd. Found	$\begin{array}{c} 54.40\\ 54.13\end{array}$	7.73 7.68	$\begin{array}{c} 4.78\\ 4.68\end{array}$
5b	$C_2H_5$	8	72	110—112 <sup>e)</sup>	0.66	$\mathrm{C_{14}H_{24}O_4NP}$	Calcd. Found	$\begin{array}{c} 55.86 \\ 56.92 \end{array}$	$\frac{8.04}{7.83}$	$\begin{array}{c} 4.65 \\ 4.63 \end{array}$
5c	$C_3H_7$	8	72	1155)	0.70	$\mathrm{C_{15}H_{26}O_4NP}$	Calcd. Found	<b>5</b> 7.19 56.87	$\begin{array}{c} 6.99 \\ 6.92 \end{array}$	$\begin{array}{c} 4.45 \\ 4.56 \end{array}$
5d	(CH <sub>3</sub> ) <sub>2</sub> CH	17	79	164—165 <sup>(7)</sup>	0.68	$C_{15}H_{26}O_4NP$	Calcd. Found	$\begin{array}{c} 57.19\\ 56.92 \end{array}$	$6.99 \\ 6.71$	$\begin{array}{c} 4.45 \\ 4.52 \end{array}$
5e	$C_4H_9$	10	79	112—113 <sup>h)</sup>	0.75	$\mathrm{C_{16}H_{28}O_4NP}$	Calcd. Found	$\begin{array}{c} 58.41 \\ 58.35 \end{array}$	$\begin{array}{c} 8.58\\ 8.43\end{array}$	$\begin{array}{c} 4.25 \\ 4.28 \end{array}$
5f	CH3CHCH2CH3	20	65	181—182 <sup>i)</sup>	0.72	$\mathrm{C_{16}H_{28}O_4NP}$	Calcd. Found	$58.41 \\ 57.93$	$8.58 \\ 8.49$	$\begin{array}{c} 4.25 \\ 4.09 \end{array}$
5g	$\mathrm{CH_2C_6H_5}$	20	70	143—144 <sup>j</sup> )	0.71	$\mathrm{C_{19}H_{26}O_4NP}$	Calcd. Found	$\begin{array}{c} 62.86 \\ 62.84 \end{array}$	$\begin{array}{c} 7.22 \\ 7.31 \end{array}$	$\begin{array}{c} 3.86\\ 3.95 \end{array}$
5h	$\mathrm{CH_2C_6H_4NO_2}$ (‡	») <b>2</b> 0	50	187—190 <sup>k)</sup>	0.68	$C_{19}H_{25}O_6N_2P$	Calcd. Found	$\begin{array}{c} 55.93 \\ 55.67 \end{array}$	$\begin{array}{c} 6.18 \\ 6.05 \end{array}$	$\begin{array}{c} 6.86\\ 6.83\end{array}$

TABLE I. Alkyl Phenyl Phosphates (5)<sup>a</sup>)

a) The compounds were isolated as a monocyclohexylammonium salt.

b) Yields were based on phenyl 8-quinolyl phosphate (4).

c) Paper chromatography was carried out by ascending technique using Toyo Roshi No. 50 paper. Solvent system used

was: isopropyl alcohol, conc. ammonium hydroxide, water (8:1:1 (v/v)). d) lit.3b) mp 152-153° e) lit.<sup>3b)</sup> mp 110-112° f) lit.<sup>3b)</sup> mp 115-117° g) lit.<sup>\$b)</sup> mp 166-168° h) lit.30) mp 110°

i ) lit.<sup>3b)</sup> mp 180° j ) lit.<sup>3b)</sup> mp 143-144° k) lit.<sup>5</sup> mp 188-190°

5) F. Cramer and G. Weimann, Chem. Ber., 91, 336 (1961).

to this method, various alkyl phenyl phosphates (5) were obtained in high yields. These results are summarized in Table I. Of various solvents<sup>6)</sup> examined, it was found that the yield of 5 increased when pyridine is used as the solvent.

On the other hand, when phenyl 8-quinolyl phosphate (4) was treated with amines on a similar manner to that described above, several phenyl phosphoramidates (6) were obtained in almost quantitative yields. These compounds were shown to be identical with authentic samples prepared by 2-pyridyl ester of phosphoric  $\operatorname{acid.}^{30}$ 

In conclusion, phosphorylation of alcohols by phenyl 8-quinolyl phosphate (4) affords the corresponding diesters of phosphoric acid in good yields under mild condition. Further, these phosphates were analytically pure by treating the reaction mixture with conventional organic technique.

## Experimental

Alcohols, amines and solvents were purified and dried by ordinary procedures. Phosphoryl chloride was freshly distilled before use. Paper chromatography was carried out by ascending technique using Tokyo Roshi No. 50 paper. Solvent system used was: isopropyl alcohol, concentrated ammonium hydroxide, water (8:1:1 (v/v)). Hanes-Isherwood reagent<sup>7</sup>) was used for the detection of spots.

Phenyl 8-Quinolyl Phosphate (4)——To a solution of 2.12 g (0.01 mole) of phenyl phosphorodichloridate (1) in 10 ml of dry pyridine was added, dropwise, a solution of 8-quinol (2) (1.45 g, 0.01 mole) in 10 ml of dry pyridine with stirring at 0° over a period of about 1 hr. The stirring was continued for 9 hr under cooling. The pyridinium chloride was removed by filtration. The filtrate was poured with stirring into 20 ml of pyridine containing 0.36 ml (0.02 mole) of water at temperature below 10°. After concentrating the mixture, the residue was extracted with 90 ml of carbon tetrachloride. The solution was washed with water and dried over anhydrous sodium sulfate. After removal of the solvent, the residue was passed through column of Wacogel C-200 (100—200 mesh) (10×400 mm) by using a mixture of chloroform and methanol (8: 2 v/v) for elution. The eluent was concentrated *in vacuo* to give 2.11 g (78%) of phenyl 8-quinolyl phosphate (4) as viscous green yellowish oil. *Rf* 0.79. IR  $\nu_{max}^{\text{film}}$  cm<sup>-1</sup>: 1620—1500 (C=C and C=N), 1260 (P=O) and 1220 (P=O-Caryl). Anal. Calcd for C<sub>15</sub>H<sub>12</sub>O<sub>4</sub>NP: C, 59.85; H, 4.02; N, 4.65. Found: C, 59.67; H, 4.48; N, 4.78.

Ethyl Phenyl Phosphate (5b) — A solution of 2.70 g (0.01 mole) of phenyl 8-quinolyl phosphate (4) and 2.34 ml (0.05 mole) of ethyl alcohol in 10 ml of dry pyridine was allowed to stand at 50° for 8 hr. After removal of the solvent by evaporation, the residue dissolved in 20 ml acetonitrile and then 2.14 ml of cyclohexylamine was added to the solution. The mixture was allowed to stand in a refrigerator overnight. The precipitate was collected by filtration and recrystallized from acetone-petroleum ether (5:1 v/v). The corresponding monocyclohexylammonium salt of ethyl phenyl phosphate (5b) was obtained in 2.16 g (72%) yield as white crystal, mp 110—112° (lit.<sup>3b)</sup> mp 110—112°). Rf 0.66. IR  $v_{max}^{Neio}$  cm<sup>-1</sup>: 1600 (C=C), 1255 (P=O-Calkyl). Anal. Calcd for C<sub>14</sub>H<sub>24</sub>O<sub>4</sub>NP: C, 55.86; H, 8.04; N, 4.65. Found: C, 56.92; H, 7.83; N, 4.63.

In a similar manner, the diesters 5, such as 5a, 5c, 5d, 5e, 5f, 5g and 5h were obtained in high yields as shown in Table I.

Phenyl Phosphoroanilidate (6a) — A solution of 2.70 g (0.01 mole) of the phosphate 4 and 4.72 g (0.05 mole) of aniline in 10 ml of dry pyridine was allowed to stand at 50° for 7 hr. After removal of pyridine and excess aniline by evaporation, the resulting oily residue was dissolved in 50 ml of acetone-ether (1:1 v/v) and then 2.14 ml of cyclohexylamine was added to the solution. The mixture was allowed to stand in a refrigerator overnight. The white crystalline compound undissolved in the solvent was washed with acetone and petroleum ether, to give 2.57 g (74%) of the monocyclohexylammonium salt of 6a as white crystal, mp 198° (lit.<sup>3b</sup>) mp 197°). Rf 0.67. Anal. Calcd. for  $C_{18}H_{25}O_3N_2P$ : C, 51.77; H, 6.95; N, 8.03. Found: C, 50.98; H, 7.08; N, 7.84.

Phenyl Phosphoromorpholidate (6b)——A solution of 2.70 g (0.01 mole) of 4 and 4.35 g (0.05 mole) of morpholine in 15 ml of dry pyridine was allowed to stand at 70° for 16 hr. The reaction mixture was worked up as the manner described above to afford 2.62 g (81%) of the pure monocyclohexylammonium salt of 6b as white powder, mp 192—193° (lit.<sup>3b)</sup> 193—194°). *Rf* 0.60. *Anal.* Calcd. for  $C_{16}H_{27}O_4N_2P$ : C, 56.20; H, 7.95; N, 8.18. Found: C, 56.04; H, 7.75; N, 8.46.

<sup>6)</sup> In the cases of dimethylformamide and acetonitrile, alkyl phenyl phosphates (5) were obtained in lower yields.

<sup>7)</sup> C.S. Hanes and F.A. Isherwood, Nature, 164, 1107 (1949).