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Chemistry of Leucomycins. XI. Chemical Transformation of a Basic Macrolide to a Neutral Macrolide

The modification of the 16-membered macrolide antibiotics has been under study in our laboratory for several years. 1a,b) In this course of the study, the aglycone moiety2) of leucomycin-A₃, leuconolide-A₃ 5,18-hemiacetal has been succesfully isolated by applying the modified Polonovski reaction3) to leucomycin-A3 N-oxide. While, a new neutral macrolide was obtained from the same reaction product and it's structural determination was mainly investigated by means of proton and carbon-13 nuclear magnetic resonance (NMR) spectro-

scopy comparing with that of leucomycin-A₃ (I).

The oxidation of I with m-chloroperbenzoic acid in chloroform gave leucomycin-A₃ N-oxide (II), $[\alpha]_{\rm p}^{\rm is}$ -19.0 (c=0.5, ethanol) in a high yield (90%). Compound (II) was refluxed with acetic anhydride in chloroform for 1.5 hr and followed the hydrolysis with sodium bicarbonate in order to chemically modify the NMe₂→O group at position-3' on mycaminose moiety. The product was purified by silica gel column chromatography, obtaining a neutral macrolide, compound (III) as a main product (yield 35%) and aglycone moiety, leuconolide-A₃ 5,18-hemiacetal (IV) (yield 10%) as shown in Chart 1. Compound (III) was crystallized from ether to give a colorless needles (mp 129—130°, $[\alpha]_{D}^{20}$ —104.2° (c=0.5, ethanol), UV $\lambda_{max}^{\text{ethanol}}$ nm (ε): 232 (10760), C₄₂H₆₄O₁₇). The acetylation of III with acetic anhydride and pyridine gave monoacetate (V) ($[\alpha]_{D}^{20}$ -106.4° (c=0.5, ethanol), $C_{44}H_{66}O_{18}$). The mass spectrum of V exhibited the fragmentation peak at m/e 451 due to the aglycone and that at m/e 229 due to isovaleryl mycarose moiety, suggesting that only the mycaminose moiety is subject to modification.

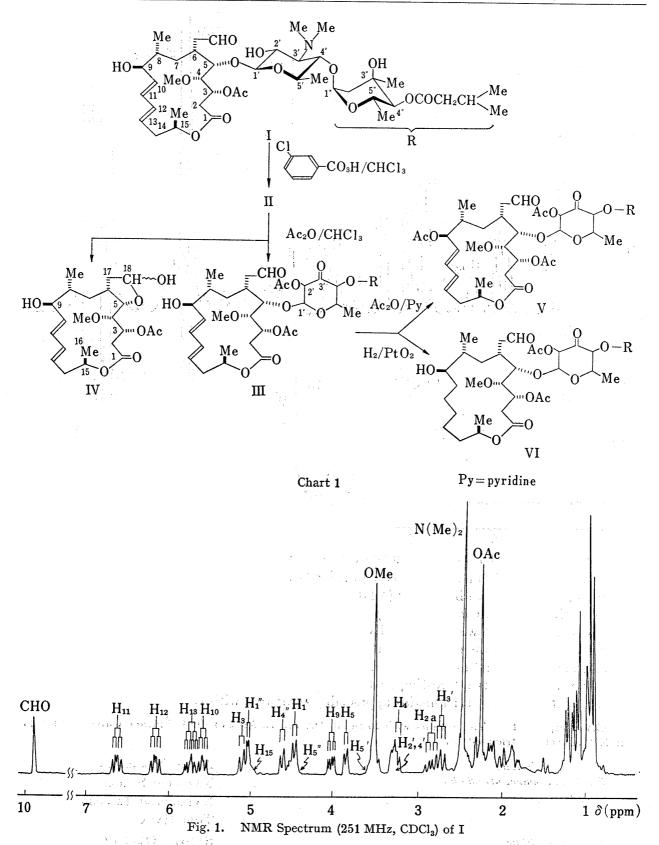
Concerning the structural information for III, NMR spectrometric investigation of I was carried out. The assignment of each proton on I was mainly performed by interpreting the NMR spectra of the α,β -isovaleryl mycarose, α,β -methyl-mycaminoside⁴⁾ and IV. The NMR spectrum (251 MHz) (Fig. 1) of I revealed the signals for olefinic proton at δ 5.5—6.8, β -anomeric on mycaminose at δ 4.3 ($J_{1',2'}=7.4$ Hz), α -anomeric on mycarose at δ 5.02 ($J_{1',2'}=$ 2.8 Hz), and for position-2',3',4' and 5' on mycaminose at δ 2.7-3.6 region.

In the NMR spectrum (Fig. 2) of III, the signals of each proton based on aglycone and mycarose moiety unchanged comparing with those of I, but the signal of NMe2 group observed at δ 2.56 in I disappeared, and an OAc group appeared at δ 2.1. Further, the each proton corresponding to position-1',2',3', and 4' on mycaminose disappeared in III, and the anomeric proton at position-1' observed at δ 4.3 in I and the proton of the basis of the acetylated hydroxyl group at position-2' appeared at δ 4.9 and δ 5.1 ($J_{1'.2'}$ =8.0 Hz) as a doublet, respectively, shifting to lower field. In order to assign the three protons observing at near δ 4.0, III was catalytic hydrogenated over PtO₂, forming tetrahydro compound (VI) ([α]_D²² -118.2°

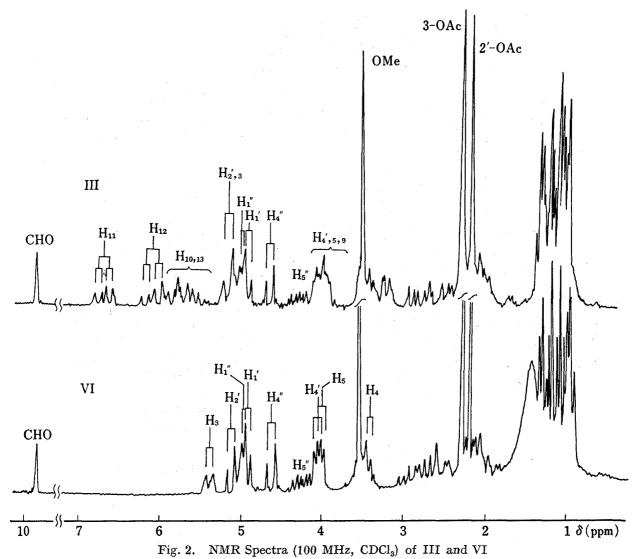
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(c=0.5, ethanol)). In the NMR spectrum of VI, the proton of the basis of the OH group at position-9 observed at δ 4.0 as a double of doublet in I shifted to a high field region, and the remaining two protons of double of doublet at near δ 4.0 could be assigned as the protons of position-4' on mycaminose and 5 on lactone ring, respectively. Also, from the comparison of the ¹³C-NMR spectrum of III with that of I, two signals based on carbonyl group at δ 170



and δ 197 were freshly observed in III, and since these show singlet by a off-resonance, the both could be assigned to be a ester (δ 170) and ketone (δ 197) carbonyl carbons. From the result described above, compound (III) was confirmed to be 2'-O-acetyl 3'-desdimethylamino 3'-oxo-leucomycin-A₃ in which the >CHNMe₂ group at position-3' and the OH at position-2' on mycaminose in I are converted into a ketone and O-acetyl group respectively.

Thus, this is the first report on the chemical conversion of a basic 16-membered macrolide to a neutral one. It is interesting in the point of structure-activity relationship that the antimicrobial activity of III remains appreciably compared with I.

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