

Studies on the Constituents of Asclepiadaceae Plants. XXXIX.¹⁾ Component of *Marsdenia tomentosa* DECNE: Structure of Deacetyldehydrotomentodin, 20-*O*-Acetylpenupogenin, Deacetylkidjolanin, and Kidjolanin

HIDEO SETO, TAMIIHIKO SASAKI, KOJI HAYASHI, and HIROSHI MITSUHASHI

Faculty of Pharmaceutical Sciences, Hokkaido University²⁾

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Four new polyoxypregnane derivatives, deacetyldehydrotomentodin (12 β -*O*-cinnamoylutendin) (III), 20-*O*-acetylpenupogenin (12 β -*O*-cinnamoyl-20-*O*-acetylsarcostin) (V), deacetylkidjolanin (12 β -*O*-tigloylsarcostin) (VIII), and kidjolanin (12 β -*O*-tigloyl-20-*O*-acetylsarcostin) (IX), were isolated from the stem of *Marsdenia tomentosa* DECNE. Deacetyldehydrotomentodin is the first example of a monoester possessing the utendin skeleton to be isolated from the Asclepiadaceae plants.

Our previous papers reported the isolation and characterization of tomentosin,³⁾ tomentin,⁴⁾ dehydrotomentin,⁴⁾ tomentonin,⁵⁾ tomentodin,⁵⁾ dehydrotomentosin,⁵⁾ deacetyltomentosin,⁶⁾ and tomentidin,⁶⁾ the new polyoxypregnane derivatives possessing tomentogenin or utendin skeleton from the stem of *Marsdenia tomentosa* DECNE. In this paper, we report some later findings on the components of this plant.

The aglycone mixture, obtained by a mild acid hydrolysis of the crude glycoside,⁷⁾ was chromatographed over silica gel and separated into several main fractions. Preparative thin-layer chromatography (TLC) of the most polar fraction yielded six crystalline substances, tentatively named compounds I, J, K, L, M, and N.

Among these, compounds I (I) and J (II) were identical with kidjolanin^{7a)} and penupogenin,^{7a,8)} respectively, from the comparison of spectral data and mixed mp with authentic samples. Compound K (III) mp 210–215°, $[\alpha]_D^{25} +7^\circ$ ($c=0.8$, CHCl₃), with a molecular formula of C₃₀H₄₀O₆ from its elemental analysis and mass spectrum of m/e 496 (M⁺), whose

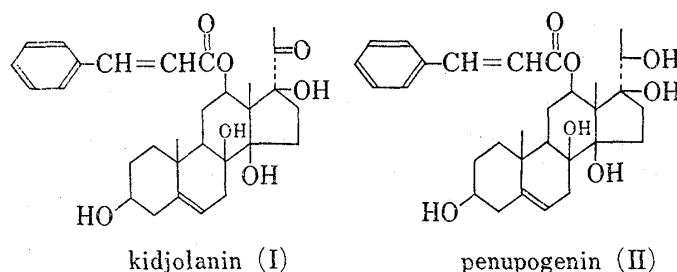


Fig. 1

infrared (IR) spectrum showed absorptions for hydroxyl groups at 3450 and 1070 cm⁻¹, and an α,β -unsaturated ester at 1710, 1690, 1635, and 1150 cm⁻¹. The nuclear magnetic resonance (NMR) spectrum of III showed signals for two tertiary methyl groups at δ 1.00 (s) and 1.32 (s), one secondary methyl group at 1.12 (d, $J=6$ Hz), three hydroxy-methines at 3.50 (m),

- 1) Part XXXVIII: H. Bando, T. Yamagishi, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **24**, 1842 (1976).
- 2) Location: *Kita-12-jo, Nishi-5-chome, Kita-ku, Sapporo, 060, Japan.*
- 3) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **23**, 1552 (1975).
- 4) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **23**, 2397 (1975).
- 5) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **24**, 443 (1976).
- 6) H. Seto, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **24**, 1552 (1976).
- 7) a) H. Mitsuhashi, I. Takemori, Y. Shimizu, T. Nomura, and E. Yamada, *Chem. Pharm. Bull.* (Tokyo), **10**, 804 (1962); b) H. Mitsuhashi, T. Sato, T. Nomura, and I. Takemori, *ibid.*, **13**, 267 (1965); c) M. Fukuoka, and H. Mitsuhashi, *ibid.*, **16**, 1634 (1968); d) T. Sasaki, K. Hayashi, and H. Mitsuhashi, *ibid.*, **20**, 628 (1972).
- 8) H. Mitsuhashi and Y. Shimizu, *Chem. Pharm. Bull.* (Tokyo), **8**, 565 (1960).

3.62 (q, $J=6$ Hz), and 4.70 (d.d, $J=6, 11$ Hz), and eight olefinic protons at 5.36 (1H, t, $J=3.5$ Hz), 6.40 (1H, d, $J=16$ Hz), 7.40 (5H, m), and 7.70 (1H, d, $J=16$ Hz). The mass spectrum of III showed the presence of a cinnamoyl group at m/e 348 ($M^+ - C_9H_8O_2$), 148, 131, and 103 which was supported by ultraviolet (UV) absorptions at 217 ($\log \epsilon$ 4.30), 223 (4.28) and 279 nm (4.15).

Hydrolysis of III with 5% methanolic potassium hydroxide afforded utendin⁹⁾ (IV) as a neutral product. These facts suggest that III is a monoester of utendin with cinnamic acid, and mass spectral peak of III at m/e 451 ($M^+ - 45$)¹⁰⁾ definitely indicated that cinnamate moiety was not at C-20 of utendin.

The NMR spin decoupling experiments were carried out to confirm the position of the ester linkage of III. Irradiation of 21-Me group protons at δ 1.12 collapsed the quartet at δ 3.62 to a singlet but the double-doublet at δ 4.70, while that of the hydroxy-methine at δ 3.62 collapsed 21-Me group protons to a singlet, so that the hydroxy-methines at 3.62 and 4.70 (d.d, $J=6, 11$ Hz) correspond to 20- and 12 α -H,¹¹⁾ respectively. From these evidences, the structure of compound K (III) was determined as 12 β -O-cinnamoylutendin and was named deacetyldehydrotomentodin, which is the first example of a monoester possessing an utendin skeleton to be isolated from a plant of the Asclepiadaceae family. Compound L (V), mp

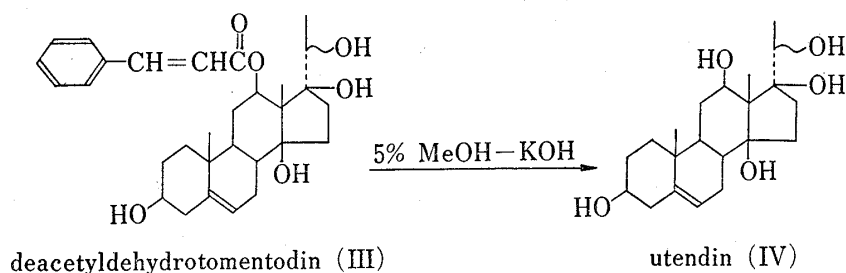


Fig. 2

142—145°, $[\alpha]_D^{25} +52^\circ$ ($c=0.4, CHCl_3$), with a molecular formula of $C_{32}H_{42}O_8$ from its elemental analysis, and mass spectrum of m/e 494 ($M^+ - AcOH$), whose IR spectrum showed absorptions for hydroxyl groups at 3400 and 1050 cm^{-1} , a saturated ester at 1730 and 1260 cm^{-1} , and an α, β -unsaturated ester at 1710, 1680, 1635, and 1170 cm^{-1} . The NMR spectrum of V showed signals for two tertiary methyl groups at δ 1.14 (s) and 1.46 (s), one secondary methyl group at 1.20 (d, $J=6$ Hz), one acetyl group at 1.95 (s), three hydroxy-methines at 3.60 (m), 4.62 (q, $J=6$ Hz), and 4.68 (d.d, $J=6, 11$ Hz), and eight olefinic protons at 5.40 (1H, t, $J=3.5$), 6.34 (1H, d, $J=16$ Hz), 7.50 (5H, m), and 7.66 (1H, d, $J=16$ Hz). The mass spectral peaks of V at m/e ($M^+ - C_9H_8O_2$), 131 (C_9H_7O), and 103 (C_8H_6), as well as m/e 494 ($M^+ - AcOH$) and 43 ($COCH_3$), showed the presence of a cinnamoyl group and an acetyl group, respectively. The presence of cinnamoyl group was supported by UV absorptions at 217 ($\log \epsilon$ 4.29), 223 (4.22), and 278 nm (4.38).

Hydrolysis of V with 5% methanolic potassium hydroxide afforded sarcostin¹²⁾ (VI) as a neutral product. These facts suggest that V is a diester of sarcostin (VI) with acetic acid and cinnamic acid. Acetylation of V with acetic anhydride-pyridine afforded a monoacetate (VII), which was identical with 3 $\beta, 20$ -O,O-diacetylpenupogenin (3 $\beta, 20$ -O,O-diacetyl-12 β -O-cinnamoylsarcostin). From these evidences, the structure of compound L (V) was determined as 12 β -O-cinnamoyl-20-O-acetylsarcostin and was named 20-O-acetylpenupogenin, which

9) E. Abisch, Ch. Tamm, and T. Reichstein, *Helv. Chim. Acta*, **42**, 1014, (1959).

10) M. Fukuoka, K. Hayashi, and H. Mitsuhashi, *Chem. Pharm. Bull.* (Tokyo), **19**, 1469 (1971).

11) N.S. Bhacca and D.H. Williams, "Application of NMR Spectroscopy in Organic Chemistry—Illustrations from Steroid Field," Holden-Day, Inc., San Francisco, 1964.

12) J.W. Cornforth and J.C. Earl, *J. Chem. Soc.*, **1939**, 737.

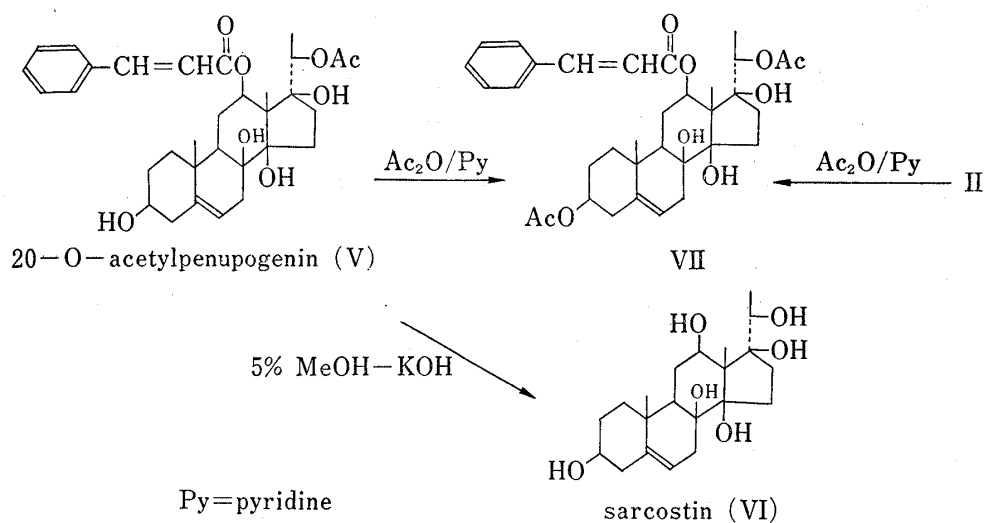


Fig. 3

corresponds to the acylgenin of amplexoside A isolated from *Asclepias amplexicaulis* by Piatak.¹³ Compound M (VIII), mp 205–209°, $[\alpha]_D^{25} +9.5^\circ$ ($c=0.85$, CHCl_3), with molecular formula of $\text{C}_{26}\text{H}_{40}\text{O}_7$ from its elemental analysis and mass spectrum of m/e 464 (M^+), whose IR spectrum of VIII showed absorptions for hydroxyl groups at 3400 and 1035 cm^{-1} , and an α,β -unsaturated ester at 1710, 1690, 1640, and 1140 cm^{-1} , which was supported by UV absorption at 215 nm ($\log \epsilon$ 3.9). The NMR spectrum of VIII showed signals for two tertiary methyl groups at δ 1.12 (s) and 1.46 (s), one secondary methyl group at 1.20 (d, $J=6$ Hz), two vinyl methyl groups at 1.82 (d, $J=6$ Hz), and 1.83 (s), three hydroxy methines at 3.58 (q, $J=6$ Hz), 3.60 (m), 4.64 (d.d, $J=6, 11$ Hz), and two olefinic protons at 5.40 (t, $J=3.5$ Hz), and 6.86 (d, $J=6$ Hz).

Hydrolysis of VIII with methanolic potassium hydroxide afforded sarcostin (VI) as a neutral product. Prominent mass spectral peaks of VIII indicative of tiglate functional group were observed at m/e 83 ($\text{C}_5\text{H}_7\text{O}$) and 55 (C_4H_7). Further evidences were secured from the mass spectral peaks of VIII, since there were a faint parent ion at m/e 464 and other fragments at 446 ($\text{M}^+ - \text{H}_2\text{O}$), 419 ($\text{M}^+ - \text{CHOH} \cdot \text{Me}$),¹⁰ 374 ($\text{M}^+ - 5\text{H}_2\text{O}$), 364 ($\text{M}^+ - \text{tiglic acid}$), 346 ($\text{M}^+ - \text{tiglic acid} - \text{H}_2\text{O}$), and 83 (tigloyl cation). These evidences indicated that VIII is a monoester of sarcostin (VI) with tiglic acid. The mass spectral peak at m/e 419, and the chemical shift and coupling constant¹¹ of the NMR spectrum at δ 4.64 (d.d, $J=6, 11$ Hz) definitely indicated that tiglate moiety was not at C-20 but at C-12 β of sarcostin.

In order to confirm the position of ester linkage of VIII, the NMR spin decoupling experiments were carried out. Irradiation of 21-Me group protons (δ 1.20, 3H, d, $J=6$ Hz) collapsed the quartet at δ 3.58 to a singlet but not the double-doublet at 4.64 and that of the hydroxy-methine at δ 3.58 collapsed 21-Me to a singlet, so that the hydroxy-methines at 3.58 and 4.64 correspond to 20- and 12 α -H,¹¹ respectively. From these results, the structure of compound M (VIII) was determined as 12 β -O-tigloyl sarcostin and was named deacetylkidjolidinin.

Compound N (IX), mp 250–255°, $[\alpha]_D^{25} +43.4^\circ$ ($c=1.0$, CHCl_3), with a molecular formula of $\text{C}_{28}\text{H}_{42}\text{O}_8$ from its elemental analysis and mass spectrum of m/e 506 (M^+), whose IR spectrum showed absorptions for hydroxyl groups at 3500, 3430, and 1070 cm^{-1} , a saturated ester at 1710 and 1280 cm^{-1} , and an α,β -unsaturated ester at 1700, 1645 and 1145 cm^{-1} . The NMR spectrum of IX showed signals for two tertiary methyl groups at δ 1.15 (s) and 1.44 (s), one secondary methyl group at 1.23 (d, $J=6$ Hz), two vinyl-methyl groups at 1.85 (s) and 1.86 (d, $J=6$ Hz), one acetyl group at 1.95 (s), three hydroxy-methines at 3.52 (m), 4.63 (d.d, $J=$

13) A.M. Ahsan, D.M. Piatak, and P.D. Sorensen, *Experientia*, **29**, 788 (1973).

6, 11 Hz), 4.65 (q, $J=6$ Hz), and two olefinic protons at 5.35 (t, $J=3.5$ Hz) and 6.85 (q, $J=6$ Hz).

Hydrolysis of IX with methanolic potassium hydroxide afforded sarcostin (VI) as a neutral product. The mass spectrum of IX showed the presence of a tigloyl group at m/e 406 ($M^+ - C_5H_8O_2$), 83 (C_5H_7O), and 55 (C_4H_7), and an acetyl group at m/e 446 ($M^+ - AcOH$) and 43 ($COCH_3$). The presence of tigloyl group was supported by UV absorption at 214 nm ($\log \epsilon$ 4.14). These facts indicate that IX is a diester of sarcostin (VI) with acetic acid and tiglic acid. Acetylation of IX with acetic anhydride-pyridine afforded a monoacetate (X), which was identical with deacetylkidjolanin diacetate. From these results, the structure of compound N (IX) was determined as 12 β -*O*-tigloyl-20-*O*-acetylsarcostin and was named kidjolanin.

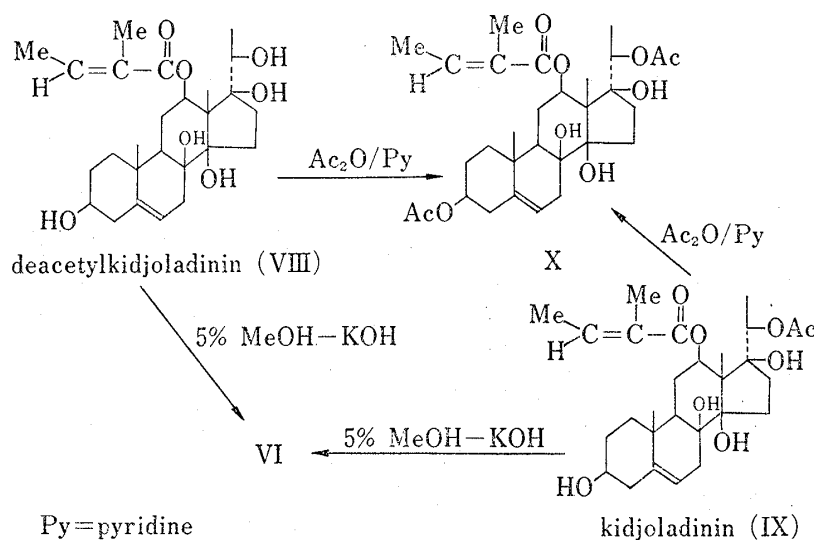


Fig. 4

Experimental

Melting points were determined on a Kofler hot stage and are uncorrected. Optical rotations were measured in $CHCl_3$ solution on a Hitachi S115-4 polarimeter. NMR spectra were determined on a JEOL PS-100 spectrometer operating at 100 MHz with tetramethylsilane (TMS) as an internal standard. Mass spectra were determined on a Hitachi RMU-7 mass spectrometer, IR spectra in Nujol mull on a Hitachi 215 spectrometer, and UV spectra in EtOH solution on a Hitachi EPS-3T spectrometer. TLC was performed on silica gel HF₂₅₄ (Merck, Type 60), and silica gel 0.05–0.2 mm (Merck, 70–325 mesh ASTM) was used for column chromatography.

Deacetyldehydrotomentodin (III)—From 15 g of the ester-type aglycone mixture obtained by the same procedure as reported previously,³⁾ 74 mg of deacetyldehydrotomentodin (III) was obtained by silica gel column chromatography and preparative TLC ($CHCl_3$:MeOH=19:1, cyclohexane:EtOAc=1:1). III was recrystallized from acetone-hexane to prisms, mp 210–215°, $[\alpha]_D^{20} +7^\circ$ ($c=0.8$, $CHCl_3$). Mass Spectrum m/e : 496 (M^+), 478 ($M^+ - H_2O$), 460 ($M^+ - 2H_2O$), 451 ($M^+ - CHOH \cdot Me$), 442 ($M^+ - 3H_2O$), 433 ($M^+ - CHOH \cdot Me - H_2O$), 415 ($M^+ - CHOH \cdot Me - 2H_2O$), 397 ($M^+ - CHOH \cdot Me - 3H_2O$), 348 ($M^+ - cinnamic acid$), 330 ($M^+ - cinnamic acid - H_2O$), 312 ($M^+ - cinnamic acid - 2H_2O$), 303 ($M^+ - cinnamic acid - CHOH \cdot Me$), 294 ($M^+ - cinnamic acid - 3H_2O$), 285 ($M^+ - cinnamic acid - CHOH \cdot Me - H_2O$), 267 ($M^+ - cinnamic acid - CHOH \cdot Me - 2H_2O$), 249 ($M^+ - cinnamic acid - CHOH \cdot Me - 3H_2O$), 148, 147, 131 (base peak), 105, 103. IR ν_{max}^{Nujol} cm^{-1} : 3450, 1710, 1690, 1635, 1575, 1280, 1150, 1070, 1050. UV λ_{max}^{EtOH} nm ($\log \epsilon$): 217 (4.30), 223 (4.28), 279 (4.15). NMR $\delta_{ppm}^{CDCl_3}$: 1.00 (3H, s, 19-Me), 1.12 (3H, d, $J=6$ Hz, 21-Me), 1.32 (3H, s, 18-Me), 3.50 (1H, m, 3 α -H), 3.62 (1H, q, $J=6$ Hz, 20-H), 4.70 (1H, d.d, $J=6, 11$ Hz, 12 α -H), 5.36 (1H, t, $J=3.5$ Hz, Δ^5 -olefinic proton), 6.40 (1H, d, $J=16$ Hz), 7.40 (5H, m, aromatic protons), 7.70 (1H, d, $J=16$ Hz). Anal. Calcd. for $C_{30}H_{40}O_6$: C, 72.55; H, 8.12. Found: C, 72.63; H, 7.95.

Alkaline Hydrolysis of Deacetyldehydrotomentodin (III)—A solution of 20 mg of deacetyldehydrotomentodin (III) in 1 ml of 5% MeOH-KOH was refluxed for 30 min and the reaction mixture was purified directly by preparative TLC (MeOH: $CHCl_3$ =9:1). Recrystallization from acetone gave 10 mg of utendin

(IV) as prisms, mp 240—243°. Mass Spectrum m/e : 366 (M^+), 348 ($M^+ - H_2O$), 330 ($M^+ - 2H_2O$), 321 ($M^+ - CHO\cdot H \cdot Me$), 312 ($M^+ - 3H_2O$), 303 ($M^+ - CHO\cdot H \cdot Me - H_2O$), 285 ($M^+ - CHO\cdot H \cdot Me - 2H_2O$), 267 ($M^+ - CHO\cdot H \cdot Me - 3H_2O$), 260, 234, 159, 145. IR ν_{max}^{Nujol} cm^{-1} : 3400, 1070, 1050, 1035.

20-O-Acetylpenupogenin (V)—From the same column chromatographic fraction which contained III, 55 mg of 20-O-acetylpenupogenin (V) was obtained by preparative TLC (ether, $CHCl_3$: MeOH=99:1). V was recrystallized from acetone-hexane to fine needles, mp 142—145°, $[\alpha]_D^{18} + 52^\circ$ ($c=0.4$, $CHCl_3$). Mass Spectrum m/e : 494 ($M^+ - AcOH$), 476 ($M^+ - AcOH - H_2O$), 467 ($M^+ - CHO\cdot Ac \cdot Me$), 458 ($M^+ - AcOH - 2H_2O$), 449 ($M^+ - CHO\cdot Ac \cdot Me - H_2O$), 440 ($M^+ - AcOH - 3H_2O$), 431 ($M^+ - CHO\cdot Ac \cdot Me - 2H_2O$), 406 ($M^+ - cinnamic acid$), 346 ($M^+ - cinnamic acid - AcOH$), 328 ($M^+ - cinnamic acid - AcOH - H_2O$), 319 ($M^+ - cinnamic acid - CHO\cdot Ac \cdot Me$), 301 ($M^+ - cinnamic acid - CHO\cdot Ac \cdot Me - H_2O$), 283 ($M^+ - cinnamic acid - CHO\cdot Ac \cdot Me - 2H_2O$), 265 ($M^+ - cinnamic acid - CHO\cdot Ac \cdot Me - 3H_2O$), 148, 147, 131 (base peak), 103, 43. IR ν_{max}^{Nujol} cm^{-1} : 3400, 1730, 1710, 1680, 1635, 1260, 1170, 1050. UV λ_{max}^{EtOH} nm ($\log \epsilon$): 217 (4.29), 223 (4.22), 278 (4.38). NMR $\delta_{ppm}^{CDCl_3}$: 1.14 (3H, s, 19-Me), 1.20 (3H, d, $J=6$ Hz, 21-Me), 1.46 (3H, s, 18-Me), 1.95 (3H, s, OAc), 3.60 (1H, m, 3α -H), 4.62 (1H, q, $J=6$ Hz, 20-H), 4.68 (1H, d.d, $J=6, 11$ Hz, 12α -H), 5.40 (1H, t, $J=3.5$ Hz, Δ^5 -olefinic proton), 6.34 (1H, d, $J=16$ Hz), 7.50 (5H, m, aromatic protons), 7.66 (1H, d, $J=16$ Hz), *Anal.* Calcd. for $C_{32}H_{42}O_8$: C, 69.29; H, 7.63. Found: C, 69.07; H, 7.85.

Alkaline Hydrolysis of 20-O-Acetylpenupogenin (V)—A solution of 15 mg of 20-O-acetylpenupogenin (V) in 1 ml of 5% MeOH-KOH was allowed to stand for 24 hr at room temperature and the reaction mixture was purified directly by preparative TLC (MeOH: $CHCl_3$ =1:9). Recrystallization from MeOH-acetone gave 6 mg of sarcostin (VI) as prisms, mp 245—250°. Mass Spectrum m/e : 382 (M^+), 364 ($M^+ - H_2O$), 346 ($M^+ - 2H_2O$), 337 ($M^+ - CHO\cdot H \cdot Me$), 328 ($M^+ - 3H_2O$), 319 ($M^+ - CHO\cdot H \cdot Me - H_2O$), 310 ($M^+ - 4H_2O$), 301 ($M^+ - CHO\cdot H \cdot Me - 2H_2O$), 283 ($M^+ - CHO\cdot H \cdot Me - 3H_2O$), 265 ($M^+ - CHO\cdot H \cdot Me - 4H_2O$), 244 ($M^+ - 138$),^{10,14} 226 ($M^+ - 138 - H_2O$), 161, 138, 114, 105, 43. IR ν_{max}^{Nujol} cm^{-1} : 3470, 3380, 3250, 175, 1045, 1040.

Acetylation of 20-O-Acetylpenupogenin (V)—A solution of 20 mg of 20-O-acetylpenupogenin (V) in 1 ml of Ac_2O and 1 ml of pyridine was allowed to stand for 24 hr at room temperature, and poured into ice water. A white powder that appeared was collected and recrystallized from acetone-hexane to afford 18 mg of an acetate (VII) as needles, mp 137—139°. Mass Spectrum m/e : 536 ($M^+ - AcOH$), 518 ($M^+ - AcOH - H_2O$), 500 ($M^+ - AcOH - 2H_2O$), 476 ($M^+ - 2 \times AcOH$), 448 ($M^+ - cinnamic acid$), 388 ($M^+ - cinnamic acid - AcOH$), 370 ($M^+ - cinnamic acid - AcOH - H_2O$), 352 ($M^+ - cinnamic acid - AcOH - 2H_2O$), 328 ($M^+ - cinnamic acid - 2 \times AcOH$), 310 ($M^+ - cinnamic acid - 2 \times AcOH - H_2O$), 292 ($M^+ - cinnamic acid - 2 \times AcOH - 2H_2O$), 148, 147, 131 (base peak), 120, 103, 43. IR ν_{max}^{Nujol} cm^{-1} : 3450, 1725, 1705, 1690, 1635, 1245, 1160, 1070, 1030. NMR $\delta_{ppm}^{CDCl_3}$: 1.16 (3H, s, 19-Me), 1.20 (3H, d, $J=6$ Hz, 21-Me), 1.46 (3H, s, 18-Me), 2.02 (3H, s, OAc), 2.18 (3H, s, OAc), 4.60 (3H, m, 3α -H + 12α -H + 20-H), 5.36 (1H, t, $J=3.5$ Hz, Δ^5 -olefinic proton), 6.24 (1H, d, $J=16$ Hz), 7.40 (5H, m, aromatic protons), 7.56 (1H, d, $J=16$ Hz). *Anal.* Calcd. for $C_{34}H_{44}O_9$: C, 68.43; H, 7.43. Found: C, 68.28; H, 7.34.

Acetylation of Penupogenin (II)—A solution of 55 mg of penupogenin (II) in 1 ml of Ac_2O and 1 ml of pyridine was allowed to stand for 19 hr at room temperature and worked up in the same manner as in the acetylation of V to produce 50 mg of an amorphous product, which was recrystallized from acetone-hexane to needles, mp 136—138.5°, and mixed mp with VII 135—139°. All spectral data were identical with those of VII.

Deacetylkidjolidinin (VIII)—From the same column chromatographic and thin-layer chromatographic fraction which contained V, 37 mg of deacetylkidjolidinin (VIII) was obtained by preparative TLC (ether, $CHCl_3$: MeOH=99:1). VIII was recrystallized from acetone-hexane to prisms, mp 205—209°, $[\alpha]_D^{19} + 9.5^\circ$ ($c=0.85$, $CHCl_3$). Mass Spectrum m/e : 464 (M^+), 446 ($M^+ - H_2O$), 428 ($M^+ - 2H_2O$), 419 ($M^+ - CHO\cdot H \cdot Me$), 413 ($M^+ - 2H_2O - Me$), 410 ($M^+ - 3H_2O$), 401 ($M^+ - CHO\cdot H \cdot Me - H_2O$), 392 ($M^+ - 4H_2O$), 383 ($M^+ - CHO\cdot H \cdot Me - 2H_2O$), 374 ($M^+ - 5H_2O$), 365 ($M^+ - CHO\cdot H \cdot Me - 3H_2O$), 364 ($M^+ - tiglic acid$), 346 ($M^+ - tiglic acid - H_2O$), 328 ($M^+ - tiglic acid - 2H_2O$), 310 ($M^+ - tiglic acid - 3H_2O$), 301 ($M^+ - tiglic acid - CHO\cdot H \cdot Me - H_2O$), 277, 240, 208, 83 (base peak), 55. IR ν_{max}^{Nujol} cm^{-1} : 3400, 1710, 1690, 1640, 1280, 1140, 1070, 1035. UV λ_{max}^{EtOH} 215 nm ($\log \epsilon$ 3.93). NMR $\delta_{ppm}^{CDCl_3}$: 1.12 (3H, s, 19-Me), 1.20 (3H, d, $J=6$ Hz, 21-Me), 1.46 (3H, s, 18-Me), 1.82 (3H, d, $J=6$ Hz, vinyl-Me), 1.83 (3H, s, vinyl-Me), 3.58 (1H, q, $J=6$ Hz, 20-H), 3.60 (1H, m, 3α -H), 4.64 (1H, d.d, $J=6, 11$ Hz, 12α -H), 5.40 (1H, d, $J=3.5$ Hz, Δ^5 -olefinic proton), 6.86 (1H, d, $J=6$ Hz). *Anal.* Calcd. for $C_{26}H_{40}O_7$: C, 67.21; H, 8.68. Found: C, 67.09; H, 8.75.

Alkaline Hydrolysis of Deacetylkidjolidinin (VIII)—A solution of 7 mg of deacetylkidjolidinin (VIII) in 0.5 ml of 5% MeOH-KOH was allowed to stand for 18 hr at room temperature and worked up in the same manner as in the hydrolysis of V to afford 3 mg of sarcostin (VI) as prisms.

Acetylation of Deacetylkidjolidinin (VIII)—A solution of 20 mg of deacetylkidjolidinin (VIII) in 1 ml of Ac_2O and 1 ml of pyridine was allowed to stand for 23 hr at room temperature and worked up in the same manner as in the acetylation of V to afford 20 mg of a diacetate (X), which was recrystallized from acetone-hexane to plates, mp 231—236°. Mass Spectrum m/e : 488 ($M^+ - AcOH$), 470 ($M^+ - AcOH - H_2O$), 452 ($M^+ - AcOH - 2H_2O$), 448 ($M^+ - tiglic acid$), 434 ($M^+ - AcOH - 3H_2O$), 430 ($M^+ - tiglic acid - H_2O$), 428 ($M^+ - 2 \times$

AcOH), 410 ($M^+ - 2 \times \text{AcOH} - \text{H}_2\text{O}$), 392 ($M^+ - 2 \times \text{AcOH} - 2\text{H}_2\text{O}$), 388 ($M^+ - \text{tiglic acid} - \text{AcOH}$), 370 ($M^+ - \text{tiglic acid} - \text{AcOH} - \text{H}_2\text{O}$), 352 ($M^+ - \text{tiglic acid} - \text{AcOH} - 2\text{H}_2\text{O}$), 328 ($M^+ - \text{tiglic acid} - 2 \times \text{AcOH}$), 310 ($M^+ - \text{tiglic acid} - 2 \times \text{AcOH} - \text{H}_2\text{O}$), 292 ($M^+ - \text{tiglic acid} - 2 \times \text{AcOH} - 2\text{H}_2\text{O}$), 274 ($M^+ - \text{tiglic acid} - 2 \times \text{AcOH} - 3\text{H}_2\text{O}$), 120, 105, 83 (base peak), 55, 43. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3450, 1730, 1705, 1690, 1640, 1260, 1240, 1150, 1070, 1030. NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 1.14 (3H, s, 19-Me), 1.20 (3H, d, $J=6$ Hz, 21-Me), 1.42 (3H, s, 18-Me), 1.82 (3H, d, $J=6$ Hz, vinyl-Me), 1.83 (3H, s, vinyl-Me), 1.94 (3H, s, OAc), 2.01 (3H, s, OAc), 4.60 (1H, m, $3\alpha\text{-H}$), 4.64 (1H, d, $J=6, 11$ Hz, $12\alpha\text{-H}$), 4.66 (1H, q, $J=6$ Hz, 20-H), 5.38 (1H, t, $J=3.5$ Hz, $\Delta^5\text{-olefinic proton}$), 6.82 (1H, d, $J=6$ Hz).

Kidjoladinin (IX)—From the same column chromatographic and thin-layer chromatographic fractions which contained V, 52 mg of kidjoladinin (IX) was obtained by preparative TLC (ether, CHCl_3 : MeOH=99:1). Recrystallization of IX from acetone-hexane gave needles, mp $250-252^\circ$, $[\alpha]_D^{25} +43.4^\circ$ ($c=1.0$, CHCl_3). Mass Spectrum m/e : 506 (M^+), 488 ($M^+ - \text{H}_2\text{O}$), 470 ($M^+ - 2\text{H}_2\text{O}$), 452 ($M^+ - 3\text{H}_2\text{O}$), 446 ($M^+ - \text{AcOH}$), 428 ($M^+ - \text{AcOH} - \text{H}_2\text{O}$), 413 ($M^+ - \text{AcOH} - \text{H}_2\text{O} - \text{Me}$), 410 ($M^+ - \text{AcOH} - 2\text{H}_2\text{O}$), 406 ($M^+ - \text{tiglic acid}$), 395 ($M^+ - \text{AcOH} - 2\text{H}_2\text{O} - \text{Me}$), 388 ($M^+ - \text{tiglic acid} - \text{H}_2\text{O}$), 370 ($M^+ - \text{tiglic acid} - 2\text{H}_2\text{O}$), 368 ($M^+ - 138$), 352 ($M^+ - \text{tiglic acid} - 3\text{H}_2\text{O}$), 350 ($M^+ - 138 - \text{H}_2\text{O}$), 346 ($M^+ - \text{tiglic acid} - \text{AcOH}$), 332 ($M^+ - 138 - 2\text{H}_2\text{O}$), 328 ($M^+ - \text{tiglic acid} - \text{AcOH} - \text{H}_2\text{O}$), 314 ($M^+ - 138 - 3\text{H}_2\text{O}$), 310 ($M^+ - \text{tiglic acid} - \text{AcOH} - 2\text{H}_2\text{O}$), 190 ($M^+ - \text{tiglic acid} - 138 - 2\text{H}_2\text{O}$), 120, 105, 83 (base peak), 55, 43. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm^{-1} : 3500, 3430, 1710, 1700, 1650, 1645, 1280, 1145, 1070, 1050. UV $\lambda_{\text{max}}^{\text{EtOH}}$: 214 nm ($\log \epsilon$ 4.14). NMR $\delta_{\text{ppm}}^{\text{CDCl}_3}$: 1.15 (3H, s, 19-Me), 1.23 (3H, d, $J=6$ Hz, 21-Me), 1.44 (3H, s, 18-Me), 1.85 (3H, s, vinyl-Me), 1.86 (3H, d, $J=6$ Hz, vinyl-Me), 1.95 (3H, s, OAc), 3.52 (1H, m, $3\alpha\text{-H}$), 4.63 (1H, d, $J=6, 11$ Hz, $12\alpha\text{-H}$), 4.65 (1H, q, $J=6$ Hz, 20-H), 5.35 (1H, t, $J=3.5$ Hz, $\Delta^5\text{-olefinic proton}$), 6.85 (1H, d, $J=6$ Hz). Anal. Calcd. for $\text{C}_{28}\text{H}_{42}\text{O}_8$: C, 66.38; H, 8.36. Found: C, 66.64; H, 8.44.

Alkaline Hydrolysis of Kidjoladinin (IX)—A solution of 10 mg of kidjoladinin (IX) in 0.5 ml of 5% MeOH-KOH was allowed to stand for 24 hr at room temperature and worked up in the same manner as in the hydrolysis of V to afford 4 mg of sarcostin (VI) as prisms.

Acetylation of Kidjoladinin (IX)—A solution of 25 mg of kidjoladinin (IX) in 1 ml of Ac_2O and 1 ml of pyridine was allowed to stand for 24 hr at room temperature and worked up in the same manner as in the acetylation of V to afford 23 mg of an amorphous product, which was recrystallized from acetone-hexane to plates, mp $230-233^\circ$ and mixed mp with X, $230-235^\circ$. All spectral data were identical with those of X.

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