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Synthesis of [1,2,3]Thiadiazolo[5,4-d]pyrimidines

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Treatment of 4,6-dimethyl-v-triazolo[4,5-d]pyrimidine-5,7(4H,6H)-dione (I) with phosphorus pentasulfide in pyridine afforded 5,7-dimethyl[1,2,3]thiadiazolo[5,4-d]pyrimidine-4(5H)-one-6(7H)-thione (II) through the v-triazole-thiadiazole rearrangement. The reaction of (II) with thionyl chloride gave 5,7-dimethyl[1,2,3]thiadiazolo[5,4-d]-pyrimidine-4,6(5H,7H)-dione (III).

 $\label{eq:Keywords} \textbf{Keywords} -- 4,6-\text{dimethyl-}v\text{-triazolo}[4,5-d] pyrimidine-5,7(4H,6H)-\text{dione}; \quad \text{phosphorus pentasulfide-pyridine}; \quad v\text{-triazole-thiadiazole rearrangement}; \quad 5,7-\text{dimethyl}[1,2,3]-\text{thiadiazolo}[5,4-d] pyrimidine-4(5H)-\text{one-}6(7H)-\text{thione}; \quad \text{thionyl chloride}; \quad 5,7-\text{dimethyl-}[1,2,3]-\text{thiadiazolo}[5,4-d] pyrimidine-4,6(5H,7H)-\text{dione}$

A recent paper²⁾ from our laboratory described a facile synthesis of [1,2,3]thiadiazolo-[4,5-d]pyrimidine derivatives, a new class of heterocyclic ring system, by the reaction of 6-hydrazinouracils with thionyl chloride. During the course of study on the thermal and photochemical reactions of this ring system,³⁾ it became desirable to synthesize 5,7-dimethyl-[1,2,3]thiadiazolo-[5,4-d]pyrimidine-4(5H)-one-6(7H)-thione (II) and 5,7-dimethyl-[1,2,3]thiadiazolo-[5,4-d]pyrimidine-4,6(5H,7H)-dione (III), which are isomeric with [1,2,3]thiadiazolo-[4,5-d]pyrimidines.

Refluxing of 4,6-dimethyl-v-triazolo[4,5-d]pyrimidine-5,7(4H,6H)-dione (I)⁴⁾ with an excess of phosphorus pentasulfide in pyridine for 5 hr afforded II in a moderate yield. The structure of II was ascertained by the following evidences. The characteristic secondary amino absorption band at 3100 cm⁻¹ of I was disappeared and a new carbonyl absorption band came out at 1700 cm⁻¹ in the infrared (IR) spectrum (Nujol). The mass spectrum revealed a parent ion at m/e 214 and the elemental analysis indicated good agreement with the proposed structure.

The conversion of I into II is best explained by assuming the initial formation of the 5,7-dithione intermediate (A), v-triazole-thiadiazole rearrangement of (A) to (B), followed by hydrolysis of the imino group to II (Chart 1). The rearrangement of this type has previously been documented.^{5a-a)}

Replacement of the sulfur of II by oxygen to give III was simply carried out in a high yield by refluxing of II in excess thionyl chloride for $30 \text{ min.}^{6)}$ The formation of III was indicated by the satisfactory elemental analysis and spectral data. Namely, the mass spectrum showed a parent ion at m/e 198 and the IR spectrum (Nujol) revealed two carbonyl

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⁶⁾ Replacement of a sulfur by oxygen with thionyl chloride has recently been reported in certain isothiazolo-[3,4-d]pyrimidine: Y. Furukawa, O. Miyashita, and S. Shima, *Chem. Pharm. Bull.* (Tokyo), 24, 970 (1976).

absorption bands at 1685 and 1720 cm⁻¹. Moreover, the nuclear magnetic resonance (NMR) spectrum (DMSO- d_6) showed singlets at δ 3.33 (N⁵-CH₃) and δ 3.56 (N⁷-CH₃).

Experimental7)

5,7-Dimethyl[1,2,3]thiadiazolo[5,4-d]pyrimidine-4(5H)-one-6(7H)-thione (II)—A mixture of 4,6-dimethyl-v-triazolo[4,5-d]pyrimidine-5,7(4H,6H)-dione (I) 4) (7.5 g, 0.04 mol) and phosphorus pentasulfide (26.5 g, 0.12 mol) in pyridine (100 ml) was refluxed for 5 hr. The reaction mixture was evaporated in vacuo and the residue was treated with hot water (50 ml). After cooling, the precipitates were collected and recrystallized from ethyl acetate to give 4.8 g (55%) of pure product (II) as pale yellow scales, mp 209—210° (dec.). IR $v_{\rm max}^{\rm Nuloi}$ cm⁻¹: 1700 (C=O). Anal. Calcd. for C₆H₆N₄OS₂: C, 33.63; H, 2.83; N, 26.15. Found: C, 33.77; H, 2.89; N, 26.42.

5,7-Dimethyl[1,2,3]thiadiazolo[5,4-d]pyrimidine-4,6(5H,7H)-dione (III)—A mixture of II (3.57 g, 0.017 mol) and thionyl chloride (20 ml) was refluxed for 30 min. The reaction mixture was evaporated in vacuo and the residue was triturated with 5% aqueous ammonia to give a solid. Recrystallization from ethanol gave 3.2 g (94%) of pure product (III) as colorless needles, mp 164—165°. IR v_{\max}^{Nulol} cm⁻¹: 1685 (C=O), 1720 (C=O). NMR (DMSO- d_6) δ : 3.33 (3H, s, N⁵-CH₃), 3.56 (3H, s, N⁷-CH₃). UV $\lambda_{\max}^{\text{BioH}}$ nm (log ε): 230 (shoulder) (3.94), 275 (3.92). Anal. Calcd. for $C_6H_6N_4O_2S$: C, 36.35; H, 3.06; N, 28.27. Found: C, 36.18; H, 3.06; N, 28.38.

⁷⁾ Melting points were taken on a Yanagimoto melting point apparatus and are uncorrected. IR spectra were recorded on a Japan Spectroscopic Co., Ltd. spectrophotometer, Model IR-E from samples mulled in Nujol. NMR spectrum was determined at 60 MHz with a Varian T-60 spectrometer using tetramethylsilane as the internal reference. Ultraviolet (UV) spectrum was recorded on a Hitachi 124 spectrophotometer.