

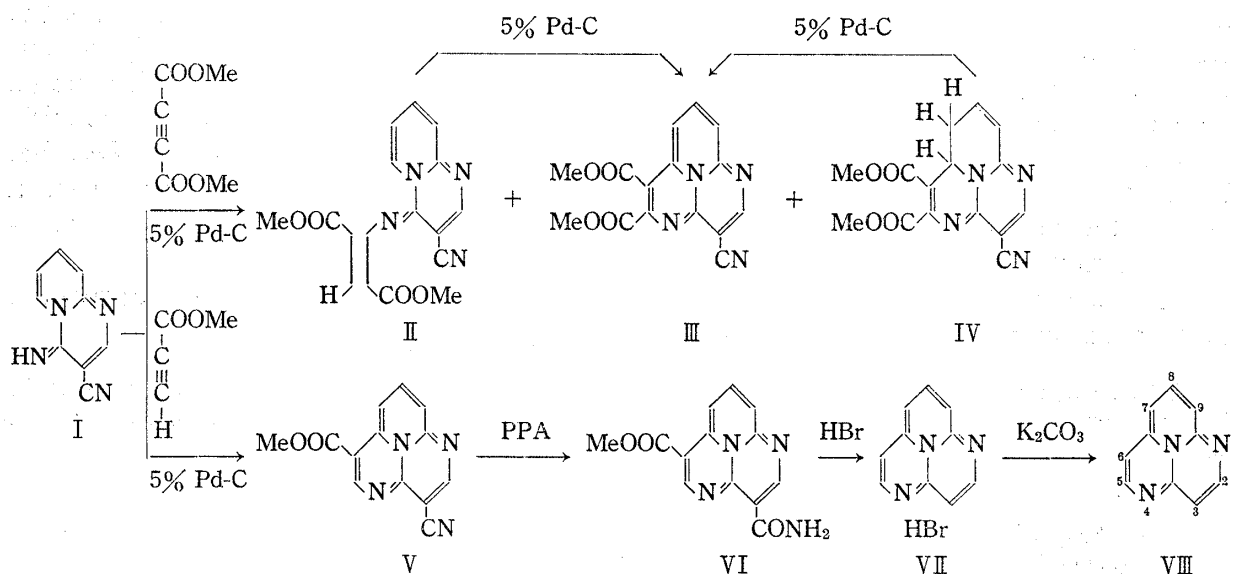
### Studies on Quinolizine Derivatives. XIX.<sup>1)</sup> The Synthesis of Diazacycl[3.3.3]azine Derivatives. (13)

By the reaction of 4-imino-4H-pyrido[1,2-*a*]pyrimidine derivative (I), and dimethyl acetylenedicarboxylate or methyl acetylenecarboxylate, with 5% Pd-C, 1,4-diazacycl[3.3.3]azine derivatives (III, V) were obtained respectively. Moreover, the degradation of V was attempted to give a new parent compound, namely 1,4-diazacycl[3.3.3]azine (VIII). VIII may be an antiaromatic compound by the PMR spectrum.

**Keywords**—cycl[3.3.3]azines;  $4n$  ( $n=3$ ) $\pi$ -electron system; nonbenzenoid aromatic compound; palladium carbon as catalyst; antiaromatic compound; 1,4-diazacycl[3.3.3]azine; paramagnetic ring current

The cycl[3.3.3]azines which are a well-known  $4n$  ( $n=3$ )  $\pi$ -electron system of a series of nonbenzenoid aromatic compounds are particularly interesting in both chemical and physical properties.<sup>2)</sup> Previously we reported a synthetic method of 2-methylthio-1,4-diazacycl[3.3.3]azine derivative in spite of poor yield from 2-methylthio-4-imino-4H-pyrido[1,2-*a*]pyrimidine derivative and dimethyl acetylenedicarboxylate (DMAD).<sup>3)</sup>

In this communication, we wish to report an improved synthetic method of 1,4-diazacycl[3.3.3]azine derivatives (III, V) with palladium carbon (Pd-C) as catalyst, and moreover the degradation of V to obtain a parent compound; 1,4-diazacycl[3.3.3]azine (VIII) which, compared with cycl[3.3.3]azines in chemical shifts of proton magnetic resonance (PMR), may be an antiaromatic compound.



Thus, a xylene solution of 4-imino-4H-pyrido[1,2-*a*]pyrimidine derivative (I), DMAD, and 5% Pd-C was refluxed under nitrogen ( $N_2$ ) atmosphere for 5 hr. After evaporation under reduced pressure, the residue was chromatographed on  $Al_2O_3$  to give 3-cyano-4-(1,2-

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dimethoxycarbonylvinyl)imino-4H-pyrido[1,2-*a*]pyrimidine (II), mp 185—186°, mass spectrum (MS) *m/e*: 312 (M<sup>+</sup>) from benzene elution, dimethyl 3-cyano-1,4-diazacycl[3.3.3]azine-5,6-dicarboxylate (III), mp 238—239°, MS *m/e*: 310 (M<sup>+</sup>) from benzene: acetone=20:1 elution, and dimethyl 3-cyano-6a, 7-dihydro-1,4-diazacycl[3.3.3]azine-5,6-dicarboxylate (IV), mp 220—221°, MS *m/e*: 312 (M<sup>+</sup>) from benzene: acetone=20:1 elution respectively. By the treatment of II or IV with 5% Pd-C, III was obtained, too.

Furthermore, a xylene solution of I, methyl acetylenecarboxylate, and 5% Pd-C was heated in sealed tube substituted with N<sub>2</sub> atmosphere at 230—250° for 10 hr to afford methyl 3-cyano-1,4-diazacycl[3.3.3]azine-6-carboxylate (V) after Al<sub>2</sub>O<sub>3</sub> column chromatography, mp 297—298°, PMR spectrum (CDCl<sub>3</sub>) ppm: 2.87 (3H, singlet, OMe), 6.87 (1H, doublet, *J*=8 Hz, C<sub>7</sub>-H or C<sub>9</sub>-H), 7.42 (1H, singlet, C<sub>2</sub>-H or C<sub>5</sub>-H), 7.66 (1H, triplet, *J*=8 Hz, C<sub>8</sub>-H), 7.95 (1H, singlet, C<sub>2</sub>-H or C<sub>5</sub>-H), 8.07 (1H, doublet, *J*=8 Hz, C<sub>7</sub>-H or C<sub>9</sub>-H), MS *m/e*: 252 (M<sup>+</sup>).

Then, the degradation of 1,4-diazacycl[3.3.3]azine derivative (V) was examined to obtain a parent compound, namely 1,4-diazacycl[3.3.3]azine (VIII). Thus, a mixture of V and polyphosphoric acid was heated at 100° for 10 hr to give methyl 3-carbamoyl-1,4-diazacycl[3.3.3]azine-6-carboxylate (VI) by alkalizing the aqueous solution with K<sub>2</sub>CO<sub>3</sub>, mp 275—276°, MS *m/e*: 270 (M<sup>+</sup>).

VI was refluxed in 48% HBr solution for 5 hr to obtain 1,4-diazacycl[3.3.3]azine hydrobromide (VII) as stable salts after evaporation under reduced pressure, mp >300°, MS *m/e*: 169 (M<sup>+</sup>-HBr), PMR spectrum (DMSO-*d*<sub>6</sub>) ppm: 5.30 (1H, doublet, *J*=7 Hz, C<sub>3</sub>-H or C<sub>6</sub>-H), 5.85 (1H, doublet, *J*=7 Hz, C<sub>3</sub>-H or C<sub>6</sub>-H), 6.13 (1H, doublet, *J*=8 Hz, C<sub>7</sub>-H or C<sub>9</sub>-H), 6.21 (1H, doublet, *J*=8 Hz, C<sub>7</sub>-H or C<sub>9</sub>-H), 6.61 (1H, doublet, *J*=7 Hz, C<sub>2</sub>-H or C<sub>5</sub>-H), 7.05 (1H, doublet, *J*=7 Hz, C<sub>2</sub>-H or C<sub>5</sub>-H), 7.28 (1H, triplet, *J*=8 Hz, C<sub>8</sub>-H).

A parent compound, namely 1,4-diazacycl[3.3.3]azine (VIII), was obtained by the treatment of VII with K<sub>2</sub>CO<sub>3</sub>, but VII could not be recrystallized due to its instability. So, the structure of VIII was only confirmed by the PMR spectrum (CDCl<sub>3</sub>) ppm: 4.55 (1H, doublet, *J*=7 Hz, C<sub>3</sub>-H or C<sub>6</sub>-H), 4.61 (1H, doublet, *J*=7 Hz, C<sub>3</sub>-H or C<sub>6</sub>-H), 4.63 (1H, doublet, *J*=8 Hz, C<sub>7</sub>-H or C<sub>9</sub>-H), 5.25 (1H, doublet, *J*=8 Hz, C<sub>7</sub>-H or C<sub>9</sub>-H), 5.90 (1H, doublet, *J*=7 Hz, C<sub>2</sub>-H or C<sub>5</sub>-H), 6.15 (1H, triplet, *J*=8 Hz, C<sub>8</sub>-H), 6.34 (1H, doublet, *J*=7 Hz, C<sub>2</sub>-H or C<sub>5</sub>-H). By comparing with other cycl[3.3.3]azines which we had summarized in table of the chemical shifts of PMR in our previous paper,<sup>4)</sup> we think of presence of a paramagnetic ring current in 1,4-diazacycl[3.3.3]azine (VIII).

The structure of the products were confirmed by the satisfactory infrared, ultraviolet, PMR, mass spectra, and elemental analyses.

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