Chem. Pharm. Bull. 28(6)1971—1972(1980)

Syntheses of Fomecins A and B, Antibiotics produced by the Basidiomycete Fomes juniperinus

Fomecins A and B, phenolic antibiotics produced by the basidiomycete *Fomes juni*perinus Schrenk, were synthesized starting from gallic acid.

Keywords—synthesis; fomecin A; fomecin B; antibiotic; phenol; Fomes juniperinus; basidicmycete

Fomecins A and B, isolated from the culture liquid of the basidiomycete Fomes juniperinus Schrenk, have been characterized as 6-hydroxymethyl-2,3,4-trihydroxybenzaldehyde (1) and 3,4,5-trihydroxyphthalaldehyde (2), respectively.¹⁾ They show weak but rather broad antibacterial activities and antiviral activities.^{1a)} Chemical characteristics of such polyhydric phenols, combined with their biological activities, led us to undertake the syntheses of these two antibiotics.

3,4,5-Trimethoxybenzylalcohol (3), prepared from gallic acid (4) by a known procedure,2) reacted with dichloromethyl methyl ether in CH₂Cl₂ at 0° (1 hr) with TiCl₄ as a catalyst to give 6-chloromethyl-2,3,4-trimethoxybenzaldehyde (5) (58%); mp 86°; MS m/e: 246 (M+2), 244 (M+), 209 (M-35); IR (Nujol) 1673 cm⁻¹ (CHO); NMR (CDCl₃, 60 MHz) δ: 10.45 (s, CHO), 6.95 (s, Ar), 5.05 (s, CH₂Cl), 4.02, 3.97, 3.90 (s, OCH₃). SeO₂ oxidation of 5 in dioxane (reflux, 2 hr) afforded 3,4,5-trimethoxyphthalaldehyde (6), mp 97—99°, in 56% yield, which was treated with BBr₃ ($-70\sim20^{\circ}$, 18 hr) in CH₂Cl₂ to give 2. The conversion of 5 into 2 was also accomplished as follows. Reaction of 5 with AgNO₃ in MeOH gave methyl ether (7), mp 77°, (92%). Treatment of 7 with BBr₃ in CH₂Cl₂ provided 6-bromomethyl-2,3,4trihydroxybenzaldehyde (8), mp ca. 150° (dec.), in 85% yield: MS m/e: 248 (M+2), 246 (M+), 167 (M-79), 166 (M-80); IR (KBr) 1630 cm⁻¹ (CHO); NMR (DMSO- d_6) δ : 10.28 (s, CHO), 6.60 (s, Ar), 4.98 (s, CH₂Br). Oxidation of 8 with SeO₂ in dioxane gave 2 in 27% yield; mp ca. 230° (dec.); MS m/e: 182 (M+), 154 (M-28), 153 (M-29); UV (EtOH) 264, 287, 343 nm; IR (KBr) 1681, 1627 cm⁻¹ (CHO); NMR (acetone- d_6) δ : 12.65 (s, OH), 10.82 (s, CHO), 10.07 (s, CHO), 9.27, 8.87 (s, OH), 7.20 (s, Ar). Melting points and the UV, IR, and NMR spectral data of 2 were consistent with those of fomecin B reported by McMorris and Anchel. 16,3)

Several attempts to hydrolyze 8 were unsuccessful. However, catalytic hydrogenation of 2 in ethanol under 3 atm hydrogen in the presence of platinum oxide afforded 1 (8%) which was separated from the starting material by countercurrent distribution; mp 160° (dec.);

¹⁾ a) M. Anchel, A. Hervey, and W.J. Robbins, Proc. Nat. Acad. Sci., 38, 655 (1952); b) T.C. McMorris and M. Anchel, Can. J. Chem., 42, 1595 (1964).

²⁾ S. Goodwin and B. Witkop, J. Am. Chem. Soc., 79, 179 (1957).

³⁾ A total synthesis of fomecin B has just been reported: S.M. Al-Mousawi, R.J.S. Green, and J.F.W. McOmie, Bull. Soc. Chim. Belg., 88, 883 (1979).

MS m/e: 184 (M+), 167 (M-17), 166 (M-18); UV (EtOH) 242, 305 nm; IR (KBr) 1637 cm⁻¹ (CHO); NMR (DMSO- d_6) δ : 10.11 (s, CHO), 6.50 (s, Ar), 4.65 (s, CH₂O). 1 was identified by mp, UV, IR, NMR, and MS with an authentic sample of fomecin A. On acetylation, 1 gave the tetraacetate (9), mp 134°, which was identical to that derived from natural fomecin A.

Acknowledgement We are grateful to Dr. M.S.R. Nair, The New York Botanical Garden, for the authentic sample of fomecin A.

Shionogi Research Laboratories Shionogi and Co., Ltd. Fukushima-ku, Osaka, 553 Japan

Received April 25, 1980

Kunio Hayashi Katsuya Tokura Kei Okabe