## Communications to the Editor

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CASUARIIN, STACHYURIN AND STRICTININ, NEW ELLAGITANNINS FROM

CASUARINA STRICTA AND STACHYURUS PRAECOX

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Three new ellagitannins, casuariin (5), stachyurin (6) and strictinin (7) have been isolated from Casuarina stricta, and their structures elucidated. Of these, 5 and 6 were found to be C-glucosidic tannins. All of the ellagitannins (1-7) isolated from C. stricta have been found also in Stachyurus praecox.

KEYWORDS —— Casuarina stricta; Casuarinaceae; Stachyurus praecox; Stachyuraceae; tannin; ellagitannin; C-glucosidic tannin; isolation; structure elucidation; <sup>1</sup>H and <sup>13</sup>C NMR spectra

Plants of Casuarina species have been used as the astringents in Pacific islands, and are regarded as rich tannin resources. Besides casuarinin (1), casuarictin (2), pedunculagin (3) and tellimagrandin-I (4), 1) three new ellagitannins, which we have named casuariin (5), stachyurin (6) and strictinin (7), have been isolated from the leaves of Casuarina stricta Ait. (Casuarinaceae). Five of these seven tannins, 1-3, 6 and 7 have also been isolated from the leaves of Stachyurus praecox Sieb. et Zucc. (Stachyuraceae) whose fruit was used in Japan as a substitute for Chinese gall. The other two tannins, 4 and 5 were also detected in S. praecox by high-performance liquid chromatography (normal and reversed phase).

The aqueous acetone homogenate of the leaves was partitioned between AcOEt and water. The water layer yielded casuariin and stachyurin by droplet countercurrent chromatography (n-BuOH-n-PrOH-H $_2$ O 2:1:3, ascending method), followed by column chromatography on Sephadex LH-20 and cellulose. Strictinin was isolated from the AcOEt layer in an analogous way.

Casuariin (5) was obtained as a pale yellow amorphous powder,  ${\rm C_{34}^{H}_{24}^{O}_{22}\cdot ^{5H}_{2}^{O}}$ ,  ${\rm [\alpha]_D}$  +162° (c=0.5, MeOH), UV  $\lambda_{\rm max}^{\rm EtOH}$  nm (log  $\epsilon$ ) 213 (4.70), 233 (4.74), 257 (infl.) (4.50). The  $^1$ H nuclear magnetic resonance (NMR) analysis (Table I) indicated that 5 could be degalloylated casuarinin, and this structure has been confirmed by direct comparison with a specimen  $^{1}$ ) prepared by hydrolysis of  $^{1}$  with tannase.

Stachyurin (6) was isolated as an off-white amorphous powder,  $C_{41}^{H}28^{O}26^{.5}H_{2}^{O}$ , [ $\alpha$ ]  $_{D}$  +39° (c=0.4, MeOH), UV  $\lambda_{\rm max}^{\rm MeOH}$  nm (log  $\epsilon$ ) 221 (4.88), 267 (sh.) (4.53). The absence of the hemiacetal carbon signal of sugar at the region of  $\delta$  90-100 in the

 $^{1}$ H NMR Spectra of 1, 2, 5, 6 and 7 (90 MHz, acetone-d<sub>6</sub>)  $^{a)}$ 

	HHDPb)	Galloyl	Glucose $(J, Hz)$						
	(lH each, s)	(2H, s)	H-1	H-2	H-3	H-4	H <b>-</b> 5	H-6	н-6'
1	6.78, 6.56	7.12	5.64d	4.67dd				4.18dd	4.06d
	6.50		(5)	(2,5)	5	.45-5.33		(3,13)	(13)
5	6.67, 6.53		5.66d	4.75dd	5.44t	5.02dd	4.11dd	4.65dd	3.84d
	6.43		(5)	(5,3)	(3)	(3,9)	(9,3)	(3,12)	(12)
6 <sup>℃)</sup>	6.82, 6.55	7.14	4.93d	4.86t	4.98t	5.62dd	5.36dd	4.84dd	4.02d
	6.50		(2)	(2)	(2)	(2,9)	(9,3)	(3,13)	(13)
7	6.72, 6.61	7.21	5.76d	3.73dd	3.84t	4.91t	4.11dd	5.22dd	3.79d
			(7)	(7,9)	(9)	(9)	(9,6)	(6,14)	(14)
2 .	6.68, 6.55	7.18	6.21d	5.18dd	5.48t	5.18t	4.50dd	5.30dd	3.88d
	6.47, 6.38		(8)	(8,9)	(9)	(9)	(9,6)	(6,14)	(14)

a) Chemical shifts are given in  $\delta$ -values (ppm) from TMS as internal standard. Assignments were aided by appropriate double resonance experiments. b) Hexahydroxydiphenoyl. c) Recorded on 200 MHz instrument.

$$gall(Me)_3 = -CO - OMe$$
OMe
OMe

gall (Me)3

gall (Me)<sub>3</sub> Me

8

9

OH

OAc

	$R^1 = R^2$	R <sup>3</sup> R <sup>4</sup>
2	O-gall H	(S)-HHDP
3	н,он	(S)-HHDP
4	н, он	gall gall
7	O-gall H	н н
10	н, он	н н

 $^{13}\mathrm{C}$  NMR spectrum, and the  $^{1}\mathrm{H}$  NMR spectrum (Table I) suggest that stachyurin is a C-glucosidic ellagitannin which is closely related to 1. The distinguishing feature of the  $^{1}$ H NMR spectrum of 6 is the doublet (J=2 Hz) due to H-1, which resonates at a higher field ( $\delta$  4.93) than that in 1 ( $\delta$  5.64, d, J=5 Hz). Methylation of  $\bf 6$  with diazomethane yielded a pentadecamethyl derivative ( $\bf 8$ ),  ${\rm C}_{56158}{\rm C}_{26}$ ,  $M^{+}$  m/z 1146, which gave a monoacetate (9),  $C_{58}H_{60}O_{27}$ ,  $M^{+}$  m/z 1188. The H NMR spectrum of  $\bf 9$  exhibited the H-1 signal as a broad singlet at  $\delta$  6.18, which shifted downfield by ca. 1 ppm from that of 8. These data indicate that the hydroxyl group at C-1 in 6 is free, and also that the dihedral angle between H-1 and H-2 is approximately 90°. Stachyurin is therefore presumed to be an epimer at C-1 of casuarinin (1), and this assumption was confirmed as follows. epimerized in boiling water as in the epimerization<sup>2)</sup> of castalagin, and the product which was isolated by column chromatography on cellulose from a mixture with unchanged 1 (ca. 1:1) was identified as 6. Based on these data, stachyurin is represented by the formula 6.

Strictinin (7) forms an off-white amorphous powder,  $C_{27}H_{22}O_{18}\cdot 3H_2O$ ,  $[\alpha]_D$  -3° (c=0.4, MeOH), UV  $\lambda_{\rm max}^{\rm MeOH}$  nm (log  $\epsilon$ ) 218 (4.78), 267 (4.44). Comparison of the  $1_{
m H~NMR}$  spectrum with that of casuarictin (2) (Table I) indicates that 7 is an ellagitannin in which the hydroxyl groups at C-1, 4 and 6 of the glucopyranose are esterified by a hexahydroxydiphenoyl group and a galloyl group. sis of 7 with tannase yielded 4,6-0-(S)-hexahydroxydiphenoyl-glucose (10),  $^{3,4}$ ) [ $\alpha$ ]  $_{D}$  +44° (c=0.5, EtOH) and gallic acid. The  $\beta$ -linkage of the galloyl group at C-1 of 7 was deduced from the coupling constant (J=7 Hz), based on the  $\mathcal{C}1$  conformation of the glucopyranose indicated by the <sup>1</sup>H NMR spectrum. Therefore, the structure of strictinin is 1-0-galloy1-4, 6-0-(S)-hexahydroxydiphenoy $1-\beta-D$ glucose (7).

A possible biogenetic precursor of 1, 5 and 6 may be coexisiting pedunculagin (3), as they could be formed from 3 through the cleavage of the glucopyranose ring upon the formation of the C-glucosidic linkage as illustrated in Chart 2.

It is remarkable that these seven tannins with their distinct structures constitute tannin resources produced by two plants which stand far apart from each other taxonomically.

Chart 2

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