

## Communications to the Editor

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INTRAMOLECULAR PHOTO[2+2]CYCLOADDITION OF 2-ALLYLOXY- AND  
2-ALLYLAMINOCYCLOHEX-2-ENONES: FORMATION OF 2-OXA- AND  
2-AZA-BICYCLO[2.1.1]HEXANE SYSTEMS

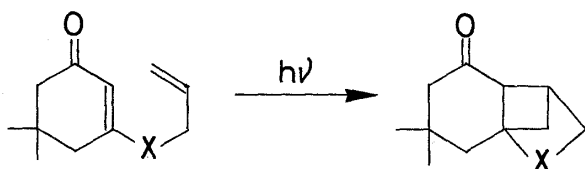
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Photoirradiation of 2-allyloxy- and 2-(allylacetamido)-cyclohex-2-enones gave 2-oxa- and 2-aza-bicyclo[2.1.1]hexane systems as the major products.

KEYWORDS — photocycloaddition; 2-allyloxycyclohex-2-enone; 2-allylamino-cyclohex-2-enone; 2-oxabicyclo[2.1.1]hexane; 2-azabicyclo[2.1.1]hexane; long-range coupling; N-acylenamine

We have previously reported that irradiation of 3-allyloxy- and 3-allylamino-cyclohex-2-enones (1) gives 2-oxa- and 2-aza-bicyclo[2.1.1]hexane systems (2).<sup>1)</sup> Since then several papers have appeared reporting similar reactions on the related systems.<sup>2)</sup> We have now investigated the photochemical behavior of 2-allyloxy- (4) and 2-allylamino-cyclohex-2-enones (6) and (11).



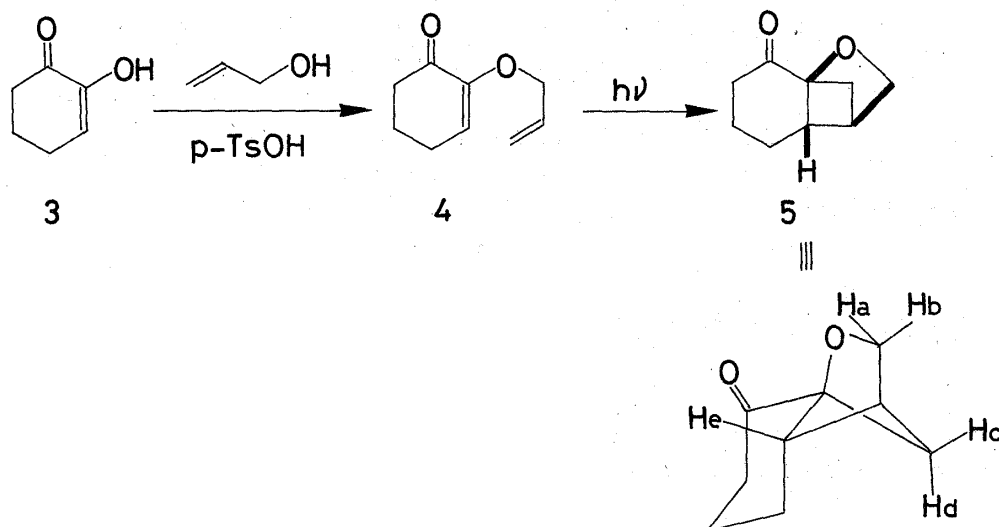
1

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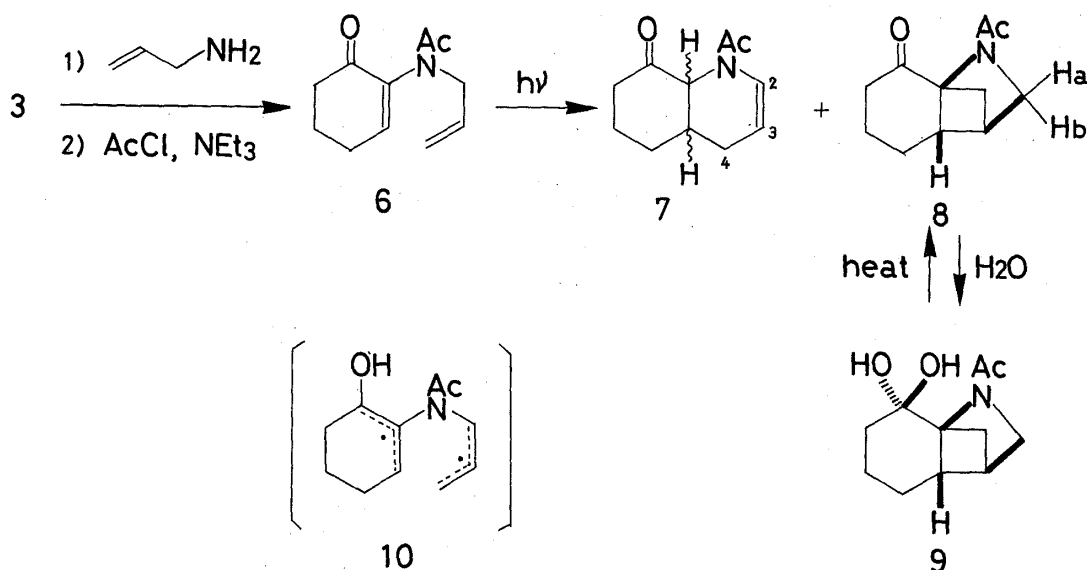
X=O, NMe, NAc etc.

Compound 4 was prepared in 74% yield by refluxing 1,2-cyclohexanedione (3) with allylalcohol in benzene in the presence of *p*-toluenesulfonic acid. Irradiation<sup>3)</sup> of 4 in degassed benzene in the presence of acetophenone gave a 2-oxabicyclo[2.1.1]hexane 5, mp 64-66°C, as the major product (40-47%). The elemental analysis and molecular weight [ $m/z$  152 ( $M^+$ )] of 5 indicate that it is an isomer of 4. The infrared (IR) ( $1705\text{ cm}^{-1}$  in  $\text{CHCl}_3$ ), ultraviolet (UV), and nuclear magnetic resonance (NMR) spectra show no unsaturation. The distinguishing features of the NMR spectrum (90 MHz) of 5 are Ha and Hb, which occurred at  $\delta$  3.82 and 3.93 as an AB quartet ( $J=6\text{ Hz}$ ). The structure of 5 was deduced from analysis of the NMR spectrum using a shift reagent [ $\text{Eu}(\text{FOD})_3$ ] and spin decoupling experi-

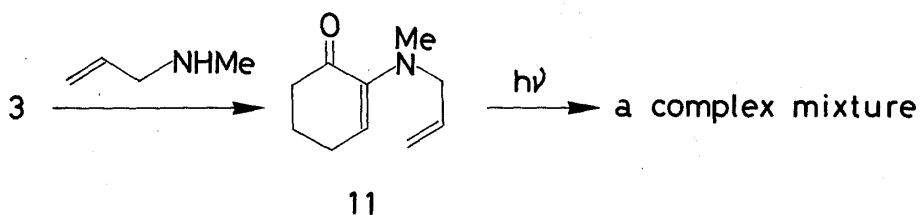
ments. The stereochemistry of 5 was defined by a large long-range coupling ( $J=8$  Hz) between Hc and He: such coupling would occur only if these protons were in W-configuration in the bicyclo[2.1.1]hexane.<sup>4)</sup> Final confirmation of structure 5 was given by an X-ray structure analysis.<sup>5)</sup>



Compound 6 was prepared in 46% yield by refluxing 3 with allylamine in benzene followed by treatment with acetyl chloride in the presence of triethylamine. Irradiation of 6 in benzene in the presence of acetophenone gave two products which could be separated by column chromatography on silica gel using AcOEt:n-hexane (1:1). The minor product [ca. 9%, an oil] was assigned 7 from the spectral evidence. The elemental analysis and mass spectrum of 7 indicate the molecular formula  $C_{11}H_{15}NO_2$ . Its IR spectrum ( $CHCl_3$ ) shows two strong carbonyl absorptions at 1720 and 1635  $cm^{-1}$  and the UV (EtOH) spectrum is that of a typical six-membered N-acylenamine with  $\lambda_{max}$  231 nm ( $\log \epsilon$  3.67).<sup>6)</sup> The NMR spectrum ( $CDCl_3$ ) reveals a singlet (3H,  $NCOCH_3$ ) at  $\delta$  2.21, a doublet of triplets (1H, H-3,  $J=8$  and 4 Hz) at  $\delta$  4.82, a doublet (1H, H-8a,  $J=6$  Hz) at  $\delta$  5.21, and a broad doublet (1H, H-2,  $J=8$  Hz) at  $\delta$  6.56. The remaining signal is a multiplet between  $\delta$  1.4 and 2.8 (9H). The stereochemistry is not clear at this time. The formation of 7 may proceed via initial hydrogen abstraction by the carbonyl oxygen<sup>7)</sup> to give a biradical 10 which may cyclize. The major product was isolated as its hydrate 9 in ca. 43% yield. When 9 was dried at 70°C *in vacuo* for 20 h, the ketone 8, mp 111-114°C, was obtained as a pure form but easily absorbed water to revert to the hydrate form 9. The IR spectrum ( $CHCl_3$ ) of 8 shows two carbonyl absorption at 1705 and 1640  $cm^{-1}$ , and its UV and NMR spectra show no unsaturation. The NMR spectrum of 8 indicates an AB quartet at  $\delta$  3.42 and 3.61 ( $J=8$  Hz) due to Ha and Hb, and a singlet (3H,  $NCOCH_3$ ) at  $\delta$  1.99. Final confirmation of structure 8 was made by an X-ray structure analysis of its hydrate 9.<sup>5)</sup> The hydrate 9 is stabilized by the formation of hydrogen bond between one of two hydroxyl groups and the acetyl carbonyl oxygen.



Irradiation of compound 11, prepared from 3 and *N*-methylallylamine, afforded a very complex mixture, and no further investigation was carried out.



## REFERENCES AND NOTES

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