(Chem. Pharm. Bull.) 30(7)2579-2582(1982)

## Synthesis of $\alpha$ -Amino-cycloheptatriene-1-acetic Acids and Their 7-Acylaminocephalosporin Derivatives

Taiichiro Watanabe, Shinichi Sugawara, and Tetsuo Miyadera\*

Chemical Research Laboratories, Sankyo Co., Ltd., 1-2-58 Hiromachi, Shinagawa-ku, Tokyo 140, Japan

(Received December 15, 1981)

tert-Butyl  $\alpha$ -amino-2,4,6-cycloheptatriene-1-acetate (6) was prepared by treatment of tert-butyl N-benzylideneglycinate (5) with n-butyllithium followed by reaction with tropylium tetrafluoroborate. Heating 6 in xylene resulted in isomerization to tert-butyl  $\alpha$ -amino-1,3,6-cycloheptatriene-1-acetate (7). Both the glycine derivatives were utilized for the acylation of a 7-amino-3-deacetoxycephalosporanic acid (7-ADCA) derivative (10) to obtain the orally active acyl derivative (14).

**Keywords**—tropylium salt; cycloheptatriene; glycine derivative; 7-ADCA derivative;  $\beta$ -lactam; thermal isomerization; orally active cephalosporin.

Although a wide variety of parenterally administrable cephalosporins has been developed in the past decade, surprisingly few cephalosporin compounds have been reported as orally useful. The 7-acyl moieties of orally active cephalosporins are limited to phenylglycyl or related groups, and the compounds are characterized by similarity in antibacterial spectrum as well as chemical structure. The purpose of the present work was to synthesize 7-( $\alpha$ -aminocycloheptatriene-1-acetamido)cephalosporin derivatives whose acyl moiety is similar to those of orally absorbed cephalexin and cephradine (1 and 2). This paper reports the synthesis of two  $\alpha$ -amino-cycloheptatriene-1-acetic acids (3<sup>1)</sup> and 4) and their cephalosporin derivatives for testing to determine their antibacterial activities and bioavailabilities.

Previous papers<sup>2)</sup> described the usefulness of tert-butyl Nbenzylideneglycinate (5) for the preparation of amino acid This synthetic derivatives for cephalosporin synthesis. method was successfully applied in the present work for the synthesis of the  $\alpha$ -amino-cycloheptatriene-1-acetic acid derivatives. The glycinate 5 was, after being treated with nbutyllithium in tetrahydrofuran (THF), allowed to react with tropylium tetrafluoroborate<sup>3)</sup> followed by treatment with Girard reagent T to yield tert-butyl α-amino-2,4,6-cycloheptatriene-1-acetate (6). The structure of 6 was confirmed by the nuclear magnetic resonance (NMR) spectrum which indicated the presence of six olefinic protons and two methine protons, as well as amino and tert-butyl groups. The 2,4,6-cycloheptatriene-1-acetate derivative 6 was heated under reflux in xylene for 26 h, resulting in isomerization to the α-amino-1,3,6cycloheptatriene-1-acetic acid derivative (7).4) Treatment of 7 with trifluoroacetic acid at room temperature afforded the amino acid (4) as the trifluoroacetic acid salt. Prior to the use of this product as an acylating agent, the amino group of 4 was protected by blocking with 2-(tert-butoxycarbonyloxyimino)-2-phenylacetonitrile(boc-on) to give 8. The initially formed cycloheptatriene-1-acetic acid derivative 6 was also treated with trifluoroacetic acid followed by protection of the

Chart 1

10 :  $R^1 = H$ ,  $R^2 = t - Bu$ 

11: 
$$R^1 = \begin{pmatrix} H \\ CHCO - \\ NHCOO_{t}-Bu \end{pmatrix}$$
  $R^2 = t-Bu$ 

12 : 
$$R^1 = \begin{pmatrix} H \\ CHCO- \\ NH_2 \end{pmatrix}$$

13 : 
$$R^1 =$$

$$\begin{array}{c} CHCO - \\ NHCOOt - Bu \end{array}$$

$$R^2 = t - Bu$$

14 : 
$$R^1 = \bigcirc$$
 CHCO-  $R^2 = H$ 

Chart 3

amino group to yield 9. The two α-tert-butoxycarbonylaminocycloheptatriene-1-acetic acid derivatives thus formed were utilized for the acylation of a 7-aminocephalosporin derivative (10). The acylation of 10 was carried out after the two acids (3 and 4) had been converted to mixed anhydrides by treatment with isobutyl chloroformate and triethylamine. The resultant glycine products (11 and 13) consisting of two isomers with respect to the 2-position of the acyl moiety were treated with trifluoroacetic acid to yield the 4-carboxylic acids 12 and 14, respectively.

For comparison with cephalexin, 14 was orally administered to mice (50 mg/kg) to examine the urinary and fecal recoveries of 14. The 1,3,6-cycloheptatriene-1-acetamido derivative was largely recovered from the urine in an amount (79%) comparable to that of cephalexin (68%), although it is only approximately half as active against gramnegative bacteria as the latter. The 2,4,6-cycloheptatriene-1-acetamidocephalosporin compound 12 displayed almost no antibacterial activity against gram-negative bacteria; this result suggests the importance of the position of the double bond.

## Experimental

Solutions were concentrated below  $30^{\circ}\text{C}$  in rotary evaporators under reduced pressure. The silica gel plates used for preparative thick layer chromatography were obtained from E. Merck, Darmstadt, West Germany. NMR spectra were recorded on a Varian A-60, a Hitachi R-24 or a Varian HA-100 spectrometer and signals are given in  $\delta$  units downfield from tetramethylsilane as an internal standard. Infrared (IR) spectra were measured on a Nihon-Bunko Jasco IR-A machine. A Nihon-Denshi JMS-01-SG spectrometer was used to obtain mass spectra.

tert-Butyl  $\alpha$ -Amino-2,4,6-cycloheptatriene-1-acetate (6)——A solution of n-butyllithium in n-hexane (19.0 ml, 1.63 mmol/ml) was added to a stirred solution of tert-butyl N-benzylideneglycinate (5, 6.60 g) in THF (100 ml) with cooling at  $-78^{\circ}$ C under nitrogen. The mixture was stirred for 30 min, then tropylium tetrafluoroborate (5.35 g) was added to the above solution and the whole was stirred for 30 min and gradually warmed to room temperature. After removal of the solvent, the residue was dissolved in MeOH (150 ml) containing Girard reagent T (6.0 g) and the mixture was stirred for 2 h at room temperature. The solvent was removed, the residue was shaken with AcOEt and  $\rm H_2O$ , and the insoluble material was removed by filtration. The organic layer was washed with  $\rm H_2O$ , dried over MgSO<sub>4</sub> and concentrated to give a residue. The crude poduct was chromatographed on a column of silica gel, eluting with benzene-AcOEt (10: 1) to give 6 (5.0 g) as an oil. NMR (CDCl<sub>3</sub>)  $\delta$ : 1.44 (9H, s, t-Bu), 1.65 (2H, s, NH<sub>2</sub>), 1.94 (1H, td, J=6.0, 7.0 Hz, 1-H), 3.63 (1H, d, J=7.0 Hz, CHCOO), 5.39 (2H, dd, J=10.0, 6.0 Hz, 2-H, 7-H), 6.45—6.10 (2H, m, 3-H, 6-H), 6.68 (2H, t, J=3.0 Hz, 4-H, 5-H).

tert-Butyl α-Amino-1,3,6-cycloheptatriene-1-acetate (7)—A solution of the 2,4,6-cycloheptatriene isomer (6, 1.1 g) in xylene (70 ml) was refluxed for 26 h under nitrogen. The solvent was evaporated off and the residue was purified by preparative thin-layer chromatography (TLC) (benzene-AcOEt=2:1) to give 7 (376 mg) as an oil. NMR (CDCl<sub>3</sub>) δ: 1.42 (9H, s, t-Bu), 2.16 (1H, ddd, J=13.0, 6.5, 6.5 Hz, 5-H), 2.35 (1H, ddd, J=13.0, 7.0, 7.0 Hz, 5-H), 3.91 (2H, s, NH<sub>2</sub>), 4.12 (1H, s, CHCOO), 5.40 (1H, ddd, J=9.5, 7.0, 6.5 Hz, 4-H or 6-H), 5.46 (1H, ddd, J=9.5, 7.0, 6.5 Hz, 6-H or 4-H), 6.15 (1H, dd, J=9.5, 5.5 Hz, 3-H), 6.16 (1H, dd, J=9.5, 1.0 Hz, 7-H), 6.61 (1H, dd, J=5.5, 1.0 Hz, 2-H).

a-tert-Butoxycarbonylamino-1,3,6-cycloheptatriene-1-acetic Acid (8)——A solution of the tert-butyl ester (7, 376 mg) in CF<sub>3</sub>COOH (5 ml) was stirred for 2.5 h at room temperature. After removal of the CF<sub>3</sub>COOH, the resultant crude amino acid (4) was dissolved in dioxane (6 ml)-H<sub>2</sub>O (6 ml), then triethylamine (630  $\mu$ l) and boc-on (462 mg) were added. The mixture was stirred for 4 h at room temperature, diluted with AcOEt and extracted with aq. NaHCO<sub>3</sub>. After being washed with AcOEt, the aq. layer was acidified with 5% aq. citric acid and extracted with AcOEt. The extracts were combined, washed with H<sub>2</sub>O, dried over MgSO<sub>4</sub> and concentrated to give a residue. The crude product was purified by preparative TLC (AcOEt–MeOH=4: 1) to give 8 (336 mg) as a powder. NMR (CDCl<sub>3</sub>)  $\delta$ : 1.35 (9H, s, t-Bu), 2.20 (2H, m, 2×5-H), 4.80 (1H, br, CHCOO), 5.40 (2H, m, 4-H, 6-H), 6.16 (1H, dd, J=9.0, 5.0 Hz, 3-H), 6.20 (1H, d, J=9.0 Hz, 7-H), 6.68 (1H, d, J=5.0 Hz, 2-H), 10.78 (1H, s, COOH).

a-tert-Butoxycarbonylamino-2,4,6-cycloheptatriene-1-acetic Acid (9)—A solution of the tert-butyl ester (6, 464 mg) in CF<sub>3</sub>COOH (3 ml) was stirred for 2 h at room temperature. The CF<sub>3</sub>COOH was evaporated off and the resulting amino acid (3) was treated with boc-on (620 mg) and triethylamine (740  $\mu$ l) as described for the preparation of 8. The reaction mixture was worked up as usual to give 9 (394 mg) as crystals. NMR (CDCl<sub>3</sub>)  $\delta$ : 1.44 (9H, s, t-Bu), 2.10 (1H, bq, J=7.0 Hz, 1-H), 4.67 (1H, br, CHCOO), 5.37 (2H, m, 2-H, 7-H), 6.05—6.43 (2H, m, 3-H, 6-H), 6.07 (2H, bt, J=3.0 Hz, 4-H, 5-H), 10.68 (1H, s, COOH).

tert-Butyl 7β-(RS-α-tert-Butoxycarbonylamino-1,3,6-cycloheptatriene-1-acetamido)-3-methyl-3-cephem-4-carboxylate (13)——Triethylamine (48 μl), N,N-diethylaniline (3 drops) and isobutyl chloroformate (45 μl) were added to a stirred solution of the acetic acid derivative (8, 86 mg) in THF (5 ml) with cooling at -8— $-10^{\circ}$ C. The mixture was stirred for 3 h, then a solution of tert-butyl 7β-amino-3-methyl-3-cephem-4-carboxylate (10, 92 mg) in THF (1 ml) was added and the whole was stirred for 4 h. After standing overnight at  $-20^{\circ}$ C, the reaction mixture was diluted with AcOEt, washed successively with  $H_2$ O, 5% HCl, 5% aq. NaHCO<sub>3</sub> and  $H_2$ O, and dried over MgSO<sub>4</sub>. The solvent was evaporated off and the residue was purified by preparative TLC (benzene-AcOEt=4: 1) to give 13 (91 mg). NMR (CDCl<sub>3</sub>) δ: 1.37 (9H, s, t-Bu), 1.46 (1H, s, t-Bu), 2.03 (3H, s, Me), 2.20 (2H, m, 2×5-H), 3.05, 3.48 (2H, AB-q, J=19 Hz, 2×2-H), 4.92 (1H, d, J=5.0 Hz, 6-H), 5.6 (1H, dd, J=5.0, 9.0 Hz, 7-H), 4.92 (1H, d, J=8—9 Hz, CHCONH), 5.2—5.8 (2H, m, olefinic 4-H and 6-H), 6.0—6.4 (2H, m, olefinic 3-H and 7-H), 6.78 (1H, d, J=5.0 Hz, olefinic 2-H), 7.18 1H, d, J=9.0 Hz, CONHCH).

 $7\beta$ -(RS-α-Amino-1,3,6-cycloheptatriene-1-acetamido)-3-methyl-3-cephem-4-carboxylic Acid CF<sub>3</sub>COOH (14)—A solution of the *tert*-butyl ester (13, 91 mg) in CF<sub>3</sub>COOH (2 ml) was stirred for 2 h at room temperature. After removal of the CF<sub>3</sub>COOH, the residue was dissolved in a small amount of AcOEt, and ether was added to give 14 (35 mg) as a powder. NMR (CD<sub>3</sub>OD) δ: 2.09 (1H, br s, Me), 2.33 (2H, m, 2×5-H), 3.16, 3.56 (2H, AB-q, J=18.0 Hz, 2×2-H), 4.80 (1H, br s, CHNH<sub>2</sub>), 5.28—5.83 (4H, m, 6-H, 7-H, olefinic 4-H and 6-H), 6.03—6.48 (2H, m, olefinic 3-H and 7-H), 7.00 (1H, br d, J=5.0 Hz, olefinic 2 H). IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 1765 ( $\beta$ -lactam).

## References and Notes

- 1) The Squibb group has claimed that the glycine derivative can be obtained in three steps from tropylium tetrafluoroborate and dimethyl formamidomalonate: J. Bernstein, P.A. Diassi, and F.L. Weisenborn, Japanese patent 47-175, Jan. 7, 1972.
- 2) T. Watanabe, Y. Kawano, T. Tanaka, T. Hashimoto, M. Nagano, and T. Miyadera, Tetrahedron Lett., 1977, 3053; T. Hashimoto, T. Watanabe, Y. Kawano, T. Tanaka, and T. Miyadera, Chem. Pharm. Bull., 28, 2980 (1980); three other groups have reported amino acid syntheses utilizing the benzylideneglycinate: G. Stork, A.Y.W. Leong, and A.M. Touzin, J. Org. Chem., 41, 3491 (1976); T. Oguri, T. Shioiri, and S. Yamada, Chem. Pharm. Bull., 25, 2287 (1977); P. Bey and J.P. Vevert, Tetrahedron Lett., 1977, 1455.
- 3) Watanabe and Soma have shown that tropylium tetrafluoroborate reacts with enamines to give 2,4,6-cycloheptatrien-7-yl derivatives: T. Watanabe and N. Soma, *Chem. Pharm. Bull.*, 18, 1595 (1970). This type of derivative has also been obtained from the reaction of the tropylium ion with ylides: G. Cavicchio, M. D'Antonio, G. Gaudiano, V. Marchetti, and P.P. Ponti, *Tetrahedron Lett.*, 1977, 3493.
- 4) [1—5]Transannular shift of hydrogen in cycloheptatrienes has been investigated by ter Borg and coworkers: A.P. ter Borg, H. Kloosterziel, and N. van Meurs, *Proc. Chem. Soc. London*, 359 (1962); A.P. ter Borg, H. Kloosterziel, and N. van Meurs, *Rec. Trav. Chim. Pays-Bas*, 82, 717 (1963); A.P. ter Borg and H. Kloosterziel, *ibid.*, 82, 741 (1963).