

Communications to the Editor

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PHOTOLYSIS OF 4-OXO-4H-BENZISOXAZOLO[2,3-a]PYRIDINES

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Photolysis of 4-oxo-4H-benzisoxazolo[2,3-a]pyridines (1a-d) and their C₁-monobromides (4a-d) afforded benzofuro[3,2-b]pyridines (2a-d) via initial N₅-O₆ bond fission. However, similar photolysis of 9-methoxy-4-oxo-4H-benzisoxazolo[2,3-a]pyridine analog (4f) gave a pyridine derivative (5).

KEYWORDS — photolysis; photo-isomerization; 4-oxo-4H-benzisoxazolo[2,3-a]pyridines; benzofuro[3,2-b]pyridines; Friedlaender reaction; ring transformation

In the preceding communication,¹⁾ we described the synthesis of benzisoxazolo[2,3-a]pyridine derivatives (1). In these studies, it was noted that 1 gradually underwent a photo-induced isomerization even in the course of repeated recrystallization from CHCl₃. Although it has been reported that only thermolysis of 9-nitrobenzisoxazolo[2,3-a]pyridinium tetrafluoroborate gave 8-nitrobenzofuro[3,2-b]pyridine in 20% yield,²⁾ any rearrangements of benzisoxazolo[2,3-a]pyridine (1) were unknown. We now report the photochemical ring transformation of 1.

Photolysis [low-pressure Hg-lamp (0.5 h) or sun light (12-24 h), CHCl₃³⁾] of 1a afforded benzofuro[3,2-b]pyridine derivative (2a).⁴⁾ Other benzofuro[3,2-b]pyridine derivatives (2) obtained by similar photolyses are shown in Table I. Photolysis of 9-methoxy-analogs (1e-f), however, gave a dirty reaction mixture from which crystalline products could not be isolated.

The ¹H-NMR (PMR) spectrum (DMSO-d₆) of 2a lacked a characteristic singlet signal attributed to C₁-H of 1a,¹⁾ and showed one proton signal at δ 12.8-13.4 which was exchangeable with D₂O. The IR spectrum (KBr) of 2a exhibited the bands (ν_{NH} 2750, ν_{C=O} 1640 cm⁻¹) attributable to the 2-pyridone moiety. Treatment of 2a with Ac₂O-pyridine gave an O-acetate (3) [mp 189-191°C, ν_{C=O}^{KBr} 1760 cm⁻¹, PMR (CDCl₃) δ 2.06 (3H, s, OCOCH₃)].

The structure of 2 was unequivocally confirmed by an alternative synthesis of 2d from ethyl 3-aminobenzofuran-2-carboxylate (6).⁵⁾ Treatment of 6 with 2-(1,2-benzisoxazol-3-yl)acetyl chloride¹⁾ afforded an amide (7) (mp 193-195°C (dec)) which was considered to be an intermediate of the Friedlaender reaction.⁶⁾ Cyclization of 7 was accomplished in the presence of NaH in DMF at 50°C to give 2d in 20% yield.

On the other hand, bromination of 1 in acetic acid at 60-80°C gave monobromide (4) as shown in Table I. With such bromination, O-acetates, 1b and 1f,

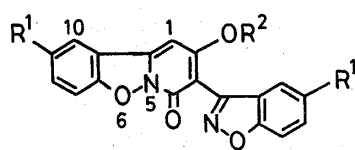
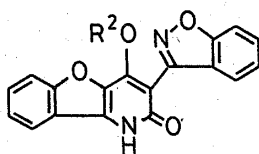
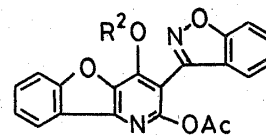
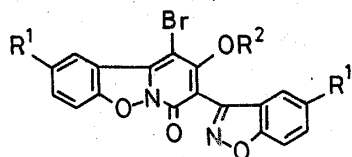
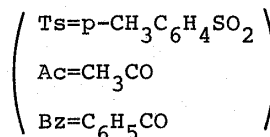
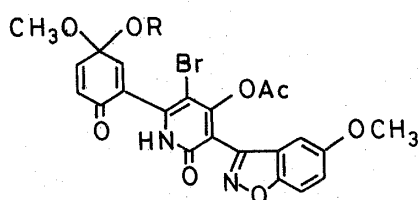
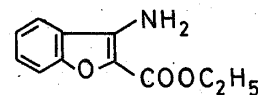
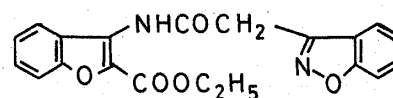
1a R¹=H, R²=Ts1b R¹=H, R²=Ac1c R¹=H, R²=Bz1d R¹=R²=H1e R¹=OCH₃, R²=Ts1f R¹=OCH₃, R²=Ac1g R¹=OCH₃, R²=H2a R²=Ts2b R²=Ac2c R²=Bz2d R²=H3 R²=Ts4a R¹=H, R²=Ts4b R¹=H, R²=Ac4c R¹=H, R²=Bz4d R¹=R²=H4e R¹=OCH₃, R²=Ts4f R¹=OCH₃, R²=Ac4g R¹=OCH₃, R²=H5a R=C₂H₅5b R=CH₃67

Table I Photolysis and Bromination of 4-Oxo-4H-benzisoxazolo[2,3-a]pyridines

Compound No.	Yield (%)			mp °C	IR(KBr) ν _{C=O} cm ⁻¹	MS (m/z) M ⁺
	<u>1</u> → <u>2</u>	<u>1</u> → <u>4</u>	<u>4</u> → <u>2</u>			
<u>2a</u>	62		43	266-269	1640	472
<u>2b</u>	90		80 ^{a)}	>300	1650, 1780	360
<u>2c</u>	47		65	254-256	1650, 1760	422
<u>2d</u>	98		74	>290	1650	318
<u>4a</u>		93		248-252	1660	550, 552
<u>4b</u>		91 ^{a)}		235-236	1670, 1770	438, 440
<u>4c</u>		93		242-245	1650, 1750	500, 502
<u>4d</u>		99		244-246	1660	396, 398
<u>4e</u>		91		248-252 (dec)	1660	610, 612
<u>4f</u>		88 ^{a)}		235-236	1660, 1760	498, 500

a) Over-all yield of photolysis followed by acetylation with Ac₂O-pyridine.

gave deacetylbromides, 4d and 4g,⁷⁾ respectively. These deacetylbromides were converted to the corresponding O-acetates, 4b and 4f, by acetylation. The substituted position of the bromine atom at C₁ was deduced from the PMR spectra of 4 which lacked C₁-H signal of the starting materials (1).

Photolysis (sun light, 12-24 h, CHCl₃^{3a)}) of the bromides, 4a, 4b, 4c and 4d, afforded 2a, 2d (accompanied by deacetylation), 2c and 2d, respectively (Table I). However, similar photolysis of 9-methoxy-analog (4f) in CHCl₃^{3a)} gave a pyridine derivative (5a)⁸⁾ (mp 190-194°C) in 28% yield. The structure of 5a was deduced from its elemental analysis, IR spectrum, mass spectrum (MS), and PMR data.⁸⁾ It was assumed that the ethoxyl group of 5a originated from EtOH in commercial chloroform. This assumption was confirmed by the following experiments. Photolysis of 4f in 5% EtOH-CHCl₃ and in EtOH gave 5a in 60% and 40% yields, respectively. Furthermore, photolysis (sun light, 4 h) of 4f in MeOH gave a methoxy-analog (5b)⁸⁾ (mp 202-204°C) in 40% yield.

In the dark condition, no isomerization of 1 and 4 was observed and the starting materials (1 and 4) were recovered unchanged. Thus, it may be considered that the ring transformation of 4-oxo-4H-benzisoxazolo[2,3-a]pyridines proceeded by initial homolytic N₅-O₆ bond fission as in the case of the well-known photolysis of isoxazole rings.

REFERENCES AND NOTES

- 1) S. Naruto, N. Nagamoto, H. Mizuta, T. Yoshida, and H. Uno, Chem. Pharm. Bull., preceding paper in this issue.
- 2) R. A. Abramovitch and M. N. Inbasekaran, J. Chem. Soc. Chem. Commun., 1978, 149. They reported that photolysis of benzisoxazolo[2,3-a]pyridinium tetrafluoroborate did not give any benzofuro[3,2-b]pyridines.
- 3) a) CHCl₃: commercial chloroform containing 0.5-1.0% of EtOH as stabilizer.
b) It was noted that 1a-d were sparingly insoluble in MeOH, EtOH and dioxane at room temperature and that photolysis (sun light, 4 h) of 1a in MeOH (concentration of 0.08 mM) afforded several products in which 2a, 2d and methyl tosylate were detected. Of these products of the above photolysis, 2d was isolated in 10% yield.
- 4) All new compounds except 4g in this paper gave satisfactory spectral (PMR, IR, and MS) and analytical data.
- 5) S. S. Sangapure and Y. S. Agasimundin, Indian J. Chem., 14B, 688 (1976).
- 6) V. P. Vaidya, S. B. Mahajan, and Y. S. Agasimundin, Indian J. Chem., 20B, 391 (1981).
- 7) This compound (4g) was used for acetylation without further purification.
- 8) 5a: IR (KBr) 3200-2400, 1770, 1640 cm⁻¹. PMR (δ in CDCl₃) quinone moiety; OCH₃ 2.95 (3H, s), OCH₂CH₃ 0.99 (3H, t), OCH₂CH₃ 3.3 (2H, m), vinyl protons 6.88 (1H, d, $J=3.0$ Hz), 6.38 (1H, dd, $J=10.5, 3.0$ Hz), 6.02 (1H, d, $J=10.5$ Hz). Pyridine moiety; OH 14.0 (1H, bs, exchangable with D₂O), COCH₃ 2.19 (3H, s). Benzisoxazole moiety; OCH₃ 3.95 (3H, s), aromatic protons 7.01 (1H, dd, $J=2.5, 0.5$ Hz), 7.18 (1H, dd, $J=9.0, 2.5$ Hz), 7.48 (1H, dd, $J=9.0, 0.5$ Hz).
5b: PMR (δ in CDCl₃) COCH₃ 2.20 (3H, s), OCH₃ 2.94 (6H, s), aromatic OCH₃ 3.91 (3H, s).

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