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Further Studies on Dammarane-Saponins of Ginseng Roots

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A new dammarane-saponin named ginsenoside-Ra₃ (3) was isolated from both White and Red Ginseng in yields of 0.005%. The structure of 3 was established to be (20S)-protopanaxadiol $3-O-(\beta-D-glucopyranosyl(1\rightarrow 2)-\beta-D-glucopyranosido)-20-<math>O-\beta-D-xylopyranosyl(1\rightarrow 3)-\beta-D-glucopyranosyl(1\rightarrow 6)-\beta-D-glucopyranoside. Further, notoginsenoside-R4 (14), previously isolated from Sanchi Ginseng, roots of$ *Panax notoginseng*, was also isolated from Red Ginseng in a yield of <math>0.002%.

Keywords—Ginseng root; Araliaceae; dammarane-saponin; ginsenoside-Ra₃; noto-ginsenoside-R4; ¹³C NMR of oligoglycoside

The isolation and structure determination of two new dammarane-saponins, ginsenosides- Ra_1 (1)^{1,2)} and $-Ra_2$ (2),¹⁾ from both White and Red Ginseng³⁾ have been reported. The present paper deals with the further isolation and structure determination of two minor saponins.

The ginsenoside-Ra fraction reported in previous papers^{1,3)} was subjected to repeated column chromatography on reversed-phase highly porous polymer and on silica gel to afford a new saponin, named ginsenoside-Ra₃ (3), in a yield of 0.005% along with 1 and 2. On mineral acid hydrolysis, 3 gave glucose and xylose. In the ¹³C nuclear magnetic resonance (NMR) spectrum of 3, all of the carbon signals due to the aglycone moiety appeared at almost the same positions as those of ginsenoside-Rb₁ (4), indicating that 3 should be a glycoside of 20 (S)-protopanaxadiol (5) at both the 3- and 20-hydroxyl groups, like 1 and 2. Inspection of the anomeric carbon signals revealed the presence of five monosaccharide units in this saponin. The field desorption mass spectrum (FD-MS) of 3 exhibited a molecular cluster ion at m/z 1263 $(M+Na)^+$ and fragment peaks corresponding to stepwise elimination of the sugar units at m/z 1131 $(M + Na - xylosyl)^+$, 1101 $(M + Na - glucosyl)^+$, 969 (M + Na - xylosylglucosyl)⁺, 939 (M+Na-glucosyl-glucosyl)⁺, and 807 (M+Na-xylosyl-glucosyl-glucosyl)⁺. It has been reported that a glycosyl linkage at the C-20-tert-hydroxyl group of dammarane-saponins is very unstable, and in the electron impact mass spectra (EI-MS) of the acetates or trimethylsilyl (TMSi) ethers, no fragment ions having an intact O-glycosyl group at the C-20 position can be observed.⁴⁾ The EI-MS of the acetate of 3 exhibited a pair of ions at m/z 1042 and 1043 due to the elimination of the 20-O-glycosyl moiety and ions at m/z 259, 331, 547, and 619, characteristic of terminal xylosyl, terminal glucosyl, xylosyl-glucosyl, and glucosyl-glucosyl, respectively.

The mild hydrolysis of 3 with 50% acetic acid gave a prosapogenin mixture and an oligosaccharide (6), the former of which, on methylation followed by successive hydrolysis,

reduction, and acetylation, afforded the two partially methylated alditol acetates, 1,5-di-O-acetyl-2,3,4,6-tetra-O-methylhexitol (7) and 1,2,5-tri-O-acetyl-3,4,6-tri-O-methylhexitol (8) detected by gas chromatography-mass spectrometry (GC-MS) (alditol acetate analysis 5). The oligosaccharide (6) was subjected to alditol acetate analysis in the same way as above to afford 1,5-di-O-acetyl-2,3,4-tri-O-methylpentitol (9), 1,3,5-tri-O-acetyl-2,4,6-tri-O-methylhexitol (10), and 1,5,6-tri-O-acetyl-2,3,4-tri-O-methylhexitol (11). This evidence coupled with the results of MS analysis indicated that 6 should be formulated as either xylosyl(1 \rightarrow 3)-glucosyl(1 \rightarrow 6)glucose or xylosyl(1 \rightarrow 6)glucosyl(1 \rightarrow 3)glucose.

Kochetkov et al.⁶⁾ reported that the EI-MS of TMSi derivatives of oligosaccharides having a 1,6-linked biosyl unit exhibit characteristic fragment ions, i.e., m/z 583 for hexosyl(1 \rightarrow 6)hexose and m/z 481 for pentosyl(1 \rightarrow 6)hexose. The EI-MS of the TMSi ether of 6 showed the ion at m/z 859 (12), which is characteristic of xylosyl(1 \rightarrow 3)glucosyl(1 \rightarrow 6)glucose (see Fig. 2). Further, the above sugar sequence of 6 was supported by the fragment ion at m/z 513 (13) in the EI-MS of the permethyl ether of 6 (see Fig. 2).⁷⁾

A comparison of the ¹³C NMR spectrum of 3 with that of 4 revealed an additional set of signals due to a terminal β -xylosyl unit in the spectrum of 3. Further, on going from 4 to 3, a carbon signal at 78.0 ppm due to C-3 of one of the β -glucopyranosyl units was displaced to low-field by 9.5 ppm,⁸⁾ while other signals of the sugar moiety of 3 remained almost unshifted. It follows that 3 can be represented as (20S)-protopanaxadiol 3-O- $(\beta$ -D-glucopyranosyl- $(1\rightarrow 2)$ - β -D-glucopyranosido)-20-O- β -D-xylopyranosyl $(1\rightarrow 3)$ - β -D-glucopyranosyl $(1\rightarrow 6)$ - β -D-glucopyranoside.

 $1: R_1 = O-glc^{\frac{2}{2}}glc$

 $R_2 = O - glc \frac{6}{1} ara(py) \frac{4}{1} xyl$

2: $R_1 = O-glc^{\frac{2}{2}}glc$

 $R_2 = O - glc \frac{6}{} ara(fu) \frac{2}{} xyl$

3: $R_1 = O - glc^{\frac{2}{1}}glc$

 $R_2 = O - glc \frac{6}{} \frac{1}{} glc \frac{3}{} \frac{1}{} xyl$

4: $R_1 = O - glc^{\frac{2}{3}}glc$ 5: $R_1 = OH$ $R_2 = O - glc^{\frac{6}{1}}glc$ $R_2 = OH$

14: $R_1 = O - glc^{\frac{2}{1}}glc$

 $R_2 = O - glc \frac{6}{} glc \frac{6}{} glc \frac{6}{}$

glc: β -D-glucopyranosyl ara(py): α -L-arabinopyranosyl ara(fu): α -L-arabinofuranosyl xyl: β -D-xylopyranosyl

Fig. 1

Fig. 2

TABLE I. ¹³ C NMR Chemica	l Shifts: Aglycone	Moiety (in	$C_{\varepsilon}D_{\varepsilon}N$
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	5	4	3		5	4	3
C- 1	39.5	39.1	39.3	C-16	26.8	26.6	26.6
C- 2	28.2	26.6	26.6	C-17	54.7	51.6	51.6
C- 3	77.9	89.3	89.0	C-18	$16.2^{a)}$	$16.2^{a)}$	16.2^{a}
C- 4	39.5	39.6	39.7	C-19	$15.8^{a)}$	$15.9^{a)}$	$16.0^{a)}$
C- 5	56.3	56.3	56.5	C-20	72.9	83.5	83.5
C- 6	18.7	18.6	18.3	C-21	26.9	22.6	22.8
C- 7	35.2	35.1	35.1	C-22	35.8	36.1	36.1
C- 8	40.0	39.9	40.0	C-23	22.9	23.1	23.2
C- 9	50.4	50.1	50.2	C-24	126.2	125.8	126.0
C-10	37.3	36.8	36.9	C-25	130.6	131.0	130.9
C-11	32.0	30.8	30.8	C-26	25.8	25.8	25.8
C-12	70.9	70.1	70.1	C-27	17.6^{a}	$17.9^{a)}$	$17.9^{a)}$
C-13	48.5	49.3	49.5	C-28	28.6	28.0	28.0
C-14	51.6	51.3	51.4	C-29	$16.4^{a)}$	$16.5^{a)}$	$16.5^{a)}$
C-15	31.8	30.8	30.8	C-30	17.0^{a}	$17.3^{a)}$	$17.4^{a)}$
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a) Assignments in any column may be reversed, though those given here are preferred.

TABLE II. ¹³C NMR Chemical Shifts: Sugar Moiety

		4	3			4	3
3-glc	1	105.0	104.9	20-glc	1	97.9	98.0
(Inner)	2' ~	82.9	83.5	(Inner)	2	74.9	$74.8^{c)}$
	3	$77.2^{a)}$	$78.0^{a)}$		3	$78.0^{a)}$	$78.0^{a)}$
	4	71.5	71.7^{d}		4	71.5	71.7^{d}
	5	$78.0^{a)}$	$78.0^{a)}$		5	76.7	77.0
	6	62.6	62.7		6	71.5	69.6
3-glc	1	105.6	$105.9^{b)}$	20-glc	1	105.0	104.9
(Terminal)	2	76.7	77.0		2	74.9	74.1 ^{c)}
	3	$78.8^{a)}$	$79.2^{a)}$		3	$78.0^{a)}$	87.5
	4	71.5	71.7^{d}		4	71.5	71.3^{d}
	5	$78.0^{a)}$	$78.0^{a)}$		5	$78.0^{a)}$	$78.0^{a)}$
	6	62.6	62.7		6	62.6	62.4
				20-xyl	1		$106.3^{b)}$
				•	2		$75.3^{c)}$
					3		77.0
					4		70.8
					5		67.3

a—d) Assignments in any column may be reversed, though those given here are preferred. glc, β -D-glucopyranosyl; xyl, β -D-xylopyranosyl.

A saponin fraction which showed a lower Rf value than 1, 2, and 3 on silica gel thin layer chromatography (TLC) was purified by reversed-phase chromatography, affording a minor saponin (14) (yield: 0.002%), which was identical with notoginsenoside-R4, previously isolated from Sanchi Ginseng, roots of *Panax notoginseng*. 9)

Experimental

The 13 C NMR spectra were taken in pyridine- d_5 on a JEOL PFT-100 spectrometer (25.15 MHz) and the chemical shifts are expressed on the δ scale from an internal standard, tetramethylsilane (TMS). The EI-MS were recorded on a JEOL JMS-DX300 mass spectrometer at 70 eV and the FD-MS were obtained with a JEOL JMS-

DX300 machine with an emitter heating current of 22—30 mA. Identifications of the known saponin and the resulting monosaccharides after hydrolysis, and acetylation and trimethylsilylation for EI-MS were carried out as described in previous papers.^{1,10)}

Isolation of 3 and 14—The ginsenoside-Ra fraction of White or Red Ginseng (see previous papers)^{1,3)} was subjected to repeated column chromatography on reversed-phase highly porous polymer (MCI CHP20P, Mitsubishi Chemical Ind., Ltd.) (solvent: 70% aqueous MeOH), affording 1, 2, and the saponin mixture. This mixture was purified by silica gel chromatography (solvent: 1-BuOH-AcOEt-H₂O (4:1:2, upper phase)) to give 3: yeild, 0.005% each from both White and Red Ginseng.

Ginsenoside-Ra₃ (3): white powder (reprecipitated from EtOH–AcOEt), $[\alpha]_D^{17} + 9.8^{\circ}$ (c = 0.43, MeOH). Anal. Calcd for $C_{59}H_{100}O_{27} \cdot 4H_2O$: C, 53.95; H, 8.29. Found: C, 53.72; H, 8.15. 3-Acetate: EI-MS % (m/z); 1043 (1), 1042 (2), 619 (1), 547 (1), 331 (24), 259 (20), 169 (100).

The saponin fraction of Red Ginseng was chromatographed on a column of silica gel (gradient elution with $CHCl_3-MeOH-H_2O$ (50:10:1 (homogeneous) \rightarrow 7:3:0.5 \rightarrow 13:7:2 (lower phase))) to provide eight fractions (Fr.), tentatively designated as Frs. I—VIII in increasing order of polarity (see previous paper³)). Fr. VII was subjected to repeated column chromatography; first on silica gel (solvent: $CHCl_3-MeOH-H_2O$ (13:7:2, lower phase)), then on highly porous polymer (solvent: 70% aqueous MeOH), affording 14 (yield; 0.002%), which was identical with an authentic sample in TLC behavior on Silica gel $100F_{254}$ (Merck) (solvents: 1-BuOH-AcOEt- H_2O (4:1:2, upper phase) and $CHCl_3-MeOH-H_2O$ (13:7:2, lower phase)) and in HPTLC Rp-18F_{254s} (Merck) (solvent: 80% aqueous MeOH), as well as in EI-MS of the acetate and TMSi ether and ^{13}C NMR spectroscopy.

Partial Hydrolysis of 3 with 50% AcOH — A solution of 3 (10 mg) in 50% AcOH was heated at 70 °C for 4 h. The reaction mixture was diluted with H_2O and extracted with 1-BuOH (saturated with H_2O). The BuOH layers were concentrated to dryness, giving a prosapogenin mixture, while the aqueous layer was deionized on Amberlite IR-45 (OH-form), affording an oligosaccharide (6) after freeze-drying.

TMSi Ether of 6: EI-MS % (m/z); 859 $(xyl(TMSi)_3-glc(TMSi)_3-O-CH_2CH=O^+TMSi, 0.4)$, 829 $(glc(TMSi)_4-glc(TMSi)_3^+, 0.2)$, 727 $(xyl(TMSi)_3-glc(TMSi)_3^+, 0.3)$, 451 $(glc(TMSi)_4^+, 3)$, 349 $(xyl(TMSi)_3^+, 21)$, 204 (100).

Permethylation Followed by Alditol Acetate Analysis of Prosapogenin Mixture and 6—According to Hakomori's method,¹¹⁾ the prosapogenin mixture and 6 were methylated with NaH and dimethylsulfoxide (DMSO), and CH₃I, respectively. Each reaction product was purified by column chromatography on silica gel (solvent: CHCl₃–MeOH (70:1)) to afford the corresponding permethyl ether of prosapogenin mixture and 6.

Permethyl Ether of 6: EI-MS % (m/z); 614 $(M^+, weak)$, 582 $(M^+ - MeOH, weak)$, 569 $(M^+ - CH_2OMe, weak)$, 513 $(xyl(Me)_3-glc(Me)_3-O-CH_2CH=O^+CH(OMe)_2, 1)$, 423 $(glc(Me)_3-glc(Me)_4^+, 0.7)$, 219 $(glc(Me)_4^+, 6)$, 175 $(xyl(Me)_3^+, 33)$, 75 (100).

A solution (0.5 ml) of the resulting permethylated prosapogenin mixture and 6 in 93% AcOH containing 0.5 N H_2SO_4 was heated in a sealed tube at 76 °C for 3 h and the reaction mixture was applied to a column (5 mm × 5 cm) of Amberlite IR-45 (AcOH-form). The column was eluted with H_2O (20 ml) and then MeOH (10 ml). The H_2O and MeOH eluates were combined and evaporated *in vacuo*, and the residue was reduced with 1% NaBH₄ aqueous solution (1 ml) at room temperature for 2 h. The reaction was stopped by adding a few drops of AcOH and the reaction mixture was concentrated to dryness *in vacuo*. The residue was heated with Ac_2O (0.5 ml) in a sealed microtube at 100 °C for 4 h. After removal of the solvent by evaporation, a solution of the residue in CHCl₃ was washed with H_2O , and the CHCl₃ layer was concentrated to dryness. The resulting alditol acetates were subjected to GC-MS. GC-MS conditions: a) 3% OV-225 on Gas Chrom Q; glass column, $2 \text{ mm} \times 2 \text{ m}$; column temperature, 200 °C; injection temperature, 240 °C; carrier gas, He (40 ml/min). t_R (min): 7 (8.1), 8 (13.7), 9 (5.7), 10 (15.7), 11 (18.8). b) 2% OV-17 on Gas Chrom Q; glass column, $2 \text{ mm} \times 1 \text{ m}$; column temperature, 180 °C; injection temperature, 240 °C; carrier gas, He (40 ml/min). t_R (min): 7 (5.0), 8 (8.3), 9 (3.2), 10 (9.9), 11 (11.0).

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