

## Communications to the Editor

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A NEW METHOD FOR THE PREPARATION OF  
3,4-DIHYDRO AND 1,2,3,4-TETRAHYDRO- $\beta$ -CARBOLINES

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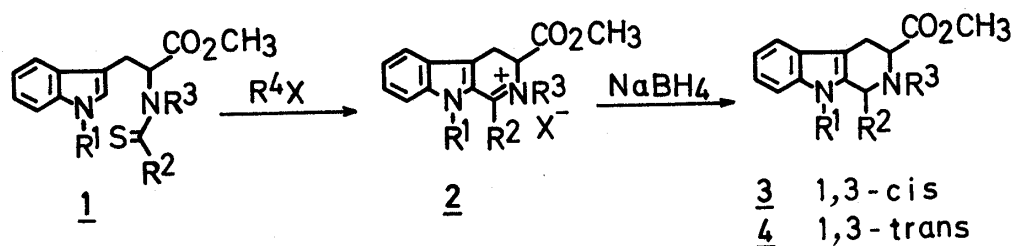
The  $\text{NaBH}_4$  reduction of 1,3-disubstituted 3,4-dihydro- $\beta$ -carbolines (2) gave cis (3)- or trans -1,2,3,4-tetrahydro- $\beta$ -carbolines (4) with satisfactory stereoselectivity. The synthesis of optically active 2 and 3 is also described.

KEYWORDS —  $\text{N}_b$ -thioacyl tryptophan ; 3,4-dihydro- $\beta$ -carboline-3-carboxylate 1-substituted ; 1,2,3,4-tetrahydro- $\beta$ -carboline-3-carboxylate 1-substituted ; Bischler-Napieralsky reaction ; Pictet-Spengler reaction

Recently<sup>1)</sup> we found that treatment of  $\text{N}_b$ -thioacyl tryptophan derivatives (1) with an alkylating or acylating reagent in an aprotic solvent gave rise to the cyclization reaction to result in the formation of a variety of 3, 4-dihydro- $\beta$ -carbolines (2) which are hardly accessible by the classical Bischler-Napieralsky reaction.<sup>2)</sup>

In continuation of the above study, this paper describes the reduction of 2 with  $\text{NaBH}_4$  to yield 1,2,3,4-tetrahydro- $\beta$ -carbolines (3 and/or 4), and also the application of this cyclization-reduction method to the synthesis of optically active 1,2,3,4-tetrahydro- $\beta$ -carbolines (D- and L-3) from D- and L- $\text{N}_b$ -thioacyl tryptophans. The results are summarized in Table I and II.

The reduction of 2 with  $\text{NaBH}_4$  proceeded almost stereoselectively to give the 1,3-cis isomer (3) as a major product. Only  $\text{N}_b$ -benzyl derivative (2h) afforded the 1,3-trans isomer (4h) predominantly. The stereochemistry of the reduction products<sup>3-6)</sup> was deduced from the spectral ( $^1\text{H}$  and  $^{13}\text{C}$  NMR) and physical (mp and Rf values) properties, and also from chemical correlation with known compounds. Particularly informative was a characteristic ABX pattern for C-3 methine and C-4 methylene protons in the  $^1\text{H}$  NMR spectra ( $J_{\text{AB}}=14-16$  Hz,  $J_{\text{AX}}=10-12$  Hz,  $J_{\text{BX}}=3-5$  Hz) which was observed with all the 1,3-cis isomers (3a-g) in agreement with the results reported by several authors.<sup>3a,b)</sup>



- a)  $\text{R}^1=\text{R}^3=\text{H}$  ,  $\text{R}^2=\text{CH}_3$                       e)  $\text{R}^1=\text{CH}_3$  ,  $\text{R}^2=\text{C}_2\text{H}_5$  ,  $\text{R}^3=\text{H}$   
 b)  $\text{R}^1=\text{R}^3=\text{H}$  ,  $\text{R}^2=\text{C}_6\text{H}_5$                       f)  $\text{R}^1=\text{CH}_3$  ,  $\text{R}^2=\text{C}_6\text{H}_5$  ,  $\text{R}^3=\text{H}$   
 c)  $\text{R}^1=\text{R}^3=\text{H}$  ,  $\text{R}^2=\text{C}_6\text{H}_{11}$                       g)  $\text{R}^1=\text{CH}_3$  ,  $\text{R}^2=\text{C}_6\text{H}_{11}$  ,  $\text{R}^3=\text{H}$   
 d)  $\text{R}^1=\text{R}^3=\text{H}$  ,  $\text{R}^2=(\text{CH}_2)_3\text{C}$                       h)  $\text{R}^1=\text{H}$  ,  $\text{R}^2=\text{CH}_3$  ,  $\text{R}^3=\text{C}_6\text{H}_5\text{CH}_2$

Chart 1

Comparison of the diastereoisomeric ratios with those of the Pictet-Spengler (P.S.) reaction products is of interest. A great difference is observed with  $\text{N}_a$ -methyl derivatives on entries 5, 6, and 7; in contrast to 1,3-trans predominance in the P.S. reaction, mainly the 1,3-cis isomers (3e, f, g) were obtained by our method. On the other hand, the result with  $\text{N}_b$ -benzyl derivative (entry 8), is parallel with those of the P.S. reaction of various  $\text{N}_b$ -benzyl tryptophan derivatives with an aldehyde or its congeners.<sup>7)</sup>

Table I. The Reduction of DL-2 with  $\text{NaBH}_4$ <sup>a)</sup>

Entry	DL-2	Yields (%) of reduction products		P.S. reaction
		<u>3</u> (mp, °C)	<u>4</u> (mp, °C)	Ratio ( <u>3</u> : <u>4</u> ) <sup>e)</sup>
1	a	93 (130-131)	- <sup>c)</sup>	91 : 9 <sup>f)</sup>
2	b	95 (209-210)	- <sup>c)</sup>	40 : 60
3	c	85 <sup>d)</sup> (150-152)	10 <sup>d)</sup> (144-147)	40 : 60
4	d	95 (116-117)	-	
5	e <sup>b)</sup>	94 <sup>d)</sup> (70-73)	-	0 : 100
6	f <sup>b)</sup>	85 (130-132)	3	0 : 100
7	g <sup>b)</sup>	61 (128-129)	6 (148-150)	0 : 100
8	h	10 (Foam)	85 (144-145)	<u>3</u> << <u>4</u> <sup>g)</sup>

- a) Carried out at  $-20 \sim -70^\circ\text{C}$  in MeOH.  
 b) Prepared from 1e, f, g by treatment with  $\text{CH}_3\text{I}$  in  $\text{CH}_3\text{CN}$  at  $50^\circ\text{C}$  in 88% (2e), 79% (2f), and 33% (2g) yield, respectively.  
 c) A trace of 4 was detected in the reaction mixture by TLC.  
 d)  $\text{NaBH}_3\text{CN}$  was used.  $\text{NaBH}_4$  was less effective in stereoselectivity (3c/4c=4:1).  
 e) Ratios reported in Lit. 3e), unless otherwise stated.  
 f) The P.S. reaction of H-Trp-OH and acetaldehyde. Lit. 9a).  
 g)  $\text{N}_b$ -Benzyl H-Trp-OCH<sub>3</sub> and acetaldehyde gave 48% of 4h and 9% of 3h in our hands.

The cyclizations of D- and L-N<sub>b</sub>-methylthiocarbonyl tryptophan methyl esters (D- and L-1a) (entries 9 - 16) were examined with several different reagents as shown in Table II. All the reagents employed were found to be effective for our purpose, and the corresponding salts (X=I, Br, or Cl) of the optically active 1-methyl-3,4-dihydro- $\beta$ -carboline derivatives (D- and L-2a)<sup>8)</sup> were obtained in good yields. Each salt was stereoselectively converted into the 1,3-cis isomer (D- or L-3a) with NaBH<sub>4</sub>. Specific rotation values of the products (entries 9, 10, 11, 12, 15, and 16) coincided with one another and also with that of the authentic sample.<sup>9)</sup> This indicates that the reduction, and thus the cyclization proceeded without racemization. As for D- and L-phenylthiocarbonyl derivatives (1b) on entries 17 and 18, the cyclization products (D- and L-2b) could not be obtained in crystalline form, so we converted the crude products into the tetrahydro- $\beta$ -carbolines (D- and L-3b) without purification. Although not determined precisely, optical purity<sup>10)</sup> of the products (D- and L-3b) thus obtained is presumed to be considerably high, since the reaction conditions were also as mild as those used for the preparation of optically active 1-methyl derivatives (2a and 3a).

Table II. The Cyclization and the Reduction of the Optically Active Isomers

Entry	<u>1</u>	R <sup>4</sup> X	Cyclization ( <u>1</u> → <u>2</u> )			Reduction	
			Conditions <sup>a)</sup>	<u>2</u> (%) <sup>b)</sup>	( $\alpha$ ) <sub>D</sub> <sup>20</sup> <sup>c)</sup>	<u>3</u> (%) <sup>d)</sup>	( $\alpha$ ) <sub>D</sub> <sup>20</sup> <sup>d)</sup>
9	L-a	CH <sub>3</sub> I	CH <sub>3</sub> COCH <sub>3</sub> , RT, 24h	90	+221°	90	-83.2°
10	D-a	CH <sub>3</sub> I	CH <sub>3</sub> COCH <sub>3</sub> , RT, 24h	87	-222°	90	+82.4°
11	L-a	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Br	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	74	+263°	91	-81.6°
12	D-a	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Br	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	75	-264°	91	+81.0°
13	L-a	CH <sub>2</sub> =CHCH <sub>2</sub> Br	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 48h	84	+260°		
14	L-a	CH <sub>3</sub> COCl	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	72	+299°		
15	L-a	C <sub>6</sub> H <sub>5</sub> COCl	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	83	+305°	90	-80.6°
16	D-a	C <sub>6</sub> H <sub>5</sub> COCl	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	82	-308°	92	+80.2°
17	L-b	CH <sub>3</sub> I	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	Not purified		78 <sup>e)</sup>	-91.8 <sup>f)</sup>
18	D-b	CH <sub>3</sub> I	CH <sub>2</sub> Cl <sub>2</sub> , Refl, 24h	Not purified		63 <sup>e)</sup>	+89.8 <sup>f)</sup>

a) Solvent, temp, time. b) Yields of isolated products as HX salts corresponding to R<sup>4</sup>X used. c) ( $\alpha$ )<sub>D</sub><sup>20</sup> (c=1.0, MeOH). d) Yields and ( $\alpha$ )<sub>D</sub><sup>20</sup> (c=1.0, MeOH) of isolated products as HCl salts. The authentic L-isomer<sup>9)</sup>; ( $\alpha$ )<sub>D</sub><sup>20</sup> -82.4° (c=1.0, MeOH). e) Yields from 1. f) Measured in 1N-HCl.

Cyclization (L-1a  $\rightarrow$  L-2a) (entry 11).<sup>11)</sup> ----- A solution of L-1a (2.76g, 10 mmol) and benzyl bromide (1.71 g, 10 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (40 ml) was refluxed for 24 h under an argon atmosphere without light, and then concentrated in vacuo. The resulting crystals were collected, rinsed with CH<sub>2</sub>Cl<sub>2</sub>, and dried to give 2.40 g (74%) of L-2a (X=Br) as pale yellow needles; mp 196-198°C (dec.), ( $\alpha$ )<sub>D</sub><sup>20</sup> +263° (c=1.0, MeOH).

Reduction (L-2a  $\rightarrow$  L-3a). ----- NaBH<sub>4</sub> (271 mg, 7.1 mmol) was added to a cooled (-70°C) solution of L-2a (2.1 g, 6.5 mmol) in MeOH (150 ml). The solution was stirred for 1 h at the same temperature, and the reaction was quenched by addition of acetone (1 ml). After concentration, the resulting oil was dissolved in CH<sub>2</sub>Cl<sub>2</sub>, washed with water, and dried over MgSO<sub>4</sub>. Removal of the solvent and treatment of the residue with HCl-Et<sub>2</sub>O gave 1.81 g of L-3a (HCl salt): mp 245-248°C (dec.), ( $\alpha$ )<sub>D</sub><sup>20</sup> -79.6° (c=1.0, MeOH). Recrystallization from MeOH gave 1.66 g (91%) of pure L-3a (HCl salt) as colorless prisms; mp 252-254°C (dec.), ( $\alpha$ )<sub>D</sub><sup>20</sup> -81.6° (c=1.0, MeOH).

## REFERENCES AND NOTES

- 1) A. Ishida, T. Nakamura, K. Irie, and T. Oh-ishi, Chem. Pharm. Bull., **30**, 4226 (1982). There are some ERRATA in Chart on p. 4227: g) R<sup>1</sup>=Y=H, R<sup>2</sup>=CH<sub>3</sub>; h) R<sup>1</sup>=Y=H, R<sup>2</sup>=C<sub>6</sub>H<sub>5</sub>; i) R<sup>1</sup>=Y=H, R<sup>2</sup>=C<sub>6</sub>H<sub>11</sub>, should be corrected to g) R<sup>1</sup>=CH<sub>3</sub>, R<sup>2</sup>=Y=H; h) R<sup>1</sup>=C<sub>6</sub>H<sub>5</sub>, R<sup>2</sup>=Y=H; i) R<sup>1</sup>=C<sub>6</sub>H<sub>11</sub>, R<sup>2</sup>=Y=H.
- 2) a) D. G. Harvey, E. J. Miller, and W. Robson, J. Chem. Soc., **1941**, 153. b) I. Murakoshi, Yakugaku Zasshi, **77**, 550 (1957). c) H. R. Snyder and F. X. Werber, J. Am. Chem. Soc., **72**, 2962 (1950).
- 3) 3a,b,c : Known compounds; a) K. T. D. De Silva, D. King, and G. N. Smith, Chem. Commun., **1971**, 908; b) F. Hamaguchi, T. Nagasaka, and S. Ohki, Yakugaku Zasshi, **94**, 351 (1974); c) J. Sandrin, D. Soerens, and J. M. Cook, Heterocycles **4**, 1249 (1976); d) D. Soerens, J. Sandrin, F. Ungemach, P. Mokry, G. S. Wu, E. Yamanaka, L. Hutchins, M. DiPierro, and J. M. Cook, J. Org. Chem., **44**, 535 (1979); e) F. Ungemach, D. Soerens, R. Weber, M. DiPierro, O. Campos, P. Mokry, J. M. Cook, and J. V. Silverton, J. Am. Chem. Soc., **102**, 6976 (1980).
- 4) 3e,f,g : The <sup>13</sup>C NMR signals for C-1 and C-3 appeared at lower-field (52.8 and 56.0 ppm in 3e; 57.8 and 56.4 ppm in 3f; 56.5 and 55.9 ppm in 3g) than those reported for the corresponding trans isomers (Lit. 3c, e).
- 5) 3h and 4h: Debenzylation by hydrogenolysis with Pd-C gave 3a (85%) and 4a (91%), respectively.
- 6) 3d: The cis-relationship is deduced only from analogy with other 1,3-cis homologues; <sup>13</sup>C NMR (CDCl<sub>3</sub>) signals for C-1 and C-3 appeared at 62.2 and 56.2 ppm, in the range predictable from the spectra of the analogous cis isomers. In the <sup>1</sup>H NMR (CDCl<sub>3</sub>) spectrum, an ABX pattern for C-3 and C-4 protons with a long-range coupling with a C-1 proton was observed at 3.65 ppm (1H, dd, J=3.7 and 11.0 Hz), 3.13 ppm (1H, ddd, J=14.5, 3.7, and 1.7 Hz), and 2.75 ppm (1H, ddd, J=14.5, 11.0, and 1.7 Hz).
- 7) a) F. Ungemach, M. DiPierro, R. Weber, and J. M. Cook, J. Org. Chem., **46**, 164 (1981); b) R. T. Brown and C. L. Chapple, Chem. Commun., **1973**, 886; c) G. Massiot and T. Mulamba, Chem. Commun., **1983**, 1147.
- 8) T. Kametani, N. Takagi, N. Kanaya, and T. Honda, Heterocycles, **19**, 535 (1982).
- 9) Esterification of L-1-methyl-1,2,3,4-tetrahydro- $\beta$ -carboline-3-carboxylic acid<sup>9a)</sup> with SOCl<sub>2</sub>-MeOH gave authentic L-3a. Yamada et al. similarly prepared the optically active Et-ester<sup>9b)</sup>: a) A. Brossi, A. Focella, and S. Teitel, J. Med. Chem., **16**, 418 (1973); b) H. Akimoto, K. Okamura, M. Yui, T. Shioiri, M. Kuramoto, Y. Kikugawa, and S. Yamada, Chem. Pharm. Bull., **22**, 2614 (1974).
- 10) Optically active 1-phenyl derivatives (D,L-2b and 3b) are unknown in literature.
- 11) CH<sub>3</sub>I in acetone at room temperature (entry 9) gave a better result (see Lit. 1).

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