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## Pyridonecarboxylic Acids as Antibacterial Agents. V.<sup>1)</sup> Synthesis of 1-Vinyl-1,4-dihydro-4-oxo-1,8- and 1,6-naphthyridine-3-carboxylic Acids<sup>2)</sup>

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A series of 7-substituted 1,4-dihydro-4-oxo-1-vinyl-1,8- and 1,6-naphthyridine-3-carboxylic acids (**5a—e** and **13a—c**) was prepared. During the preparation of **5a**, unexpected compounds (**4**, **7** and **8**) were also obtained. Structural elucidation of these compounds was achieved on the basis of chemical and spectral (ultraviolet, mass and proton nuclear magnetic resonance) data. The structure-antibacterial activity relationships are discussed.

**Keywords**—1,8-naphthyridine; 1,6-naphthyridine; imidazo[1,2,3-*ij*][1,8]naphthyridine; pyridonecarboxylic acid; synthesis; NOE experiment; antibacterial activity; structure-activity relationship

Our previous study<sup>4)</sup> on structure-antibacterial activity relationships (SARs) of a series of pyrido[2,3-*d*]pyrimidine derivatives (**I**) showed that the introduction of a vinyl group into position 8 caused an increase in activity, particularly against Gram-negative bacteria. The present study was undertaken to determine whether a similar relationship holds in case of the 1,8- and 1,6-naphthyridine analogues (**IIa** and **IIb**) having the vinyl group at position 1. Several by-products were obtained during the preparation of **IIa** and their structures were elucidated.

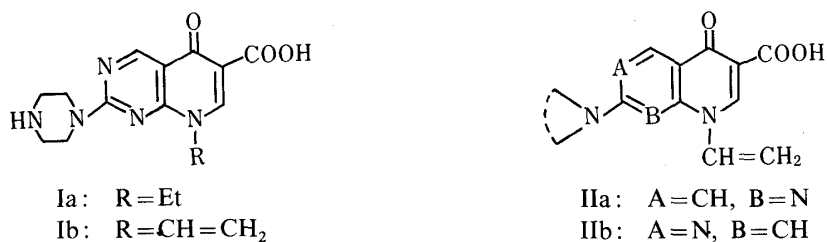


Chart 1

### Chemistry

Ethyl 7-chloro-1,4-dihydro-1-(2-hydroxyethyl)-4-oxo-1,8-naphthyridine-3-carboxylate (**1**)<sup>5)</sup> was treated with thionyl chloride to give the 1-(2-chloroethyl) analogue **2**. Displacement reaction of **2** with either an *N*-substituted piperazine or pyrrolidine in ethanol took place smoothly to give the corresponding ethyl 7-substituted 1-(2-chloroethyl)-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylates (**3a—e**) in good to excellent yields. It was expected that treatment of **3** with a strong base would permit the elimination of hydrogen chloride from the *N*-(chloroethyl) group of **3** in preference to the substitution at the methylene carbon bearing the chloro group. In fact, treatment of **3a** with 10% sodium hydroxide in ethanol produced the 7-(1-piperazinyl)-1-vinyl derivative **5a** in an excellent yield. On the other hand, when **3a** was

treated with a weak base such as potassium carbonate, ring closure of its ethylene group occurred across the N<sup>1</sup> and N<sup>8</sup> atoms with concomitant cleavage of the 7-piperazinyl group and resulted in the formation of the imidazo[1,2,3-*ij*][1,8]naphthyridine derivative **4** in a quantitative yield. The other vinyl analogues **5b–e** were similarly prepared by treatment of the corresponding *N*-(chloroethyl) compounds **3b–e** with 10% sodium hydroxide. The assigned structures of these products were confirmed by spectral analysis.

When 1-bromo-2-chloroethane was allowed to react with ethyl 7-chloro-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (**6**)<sup>5</sup> in order to prepare **2** in a one-step process, compounds **7** and **8** were unexpectedly formed in 5 and 25% yields, respectively, besides the desired compound **2** in 40% yield. The structures of these by-products were assigned on the

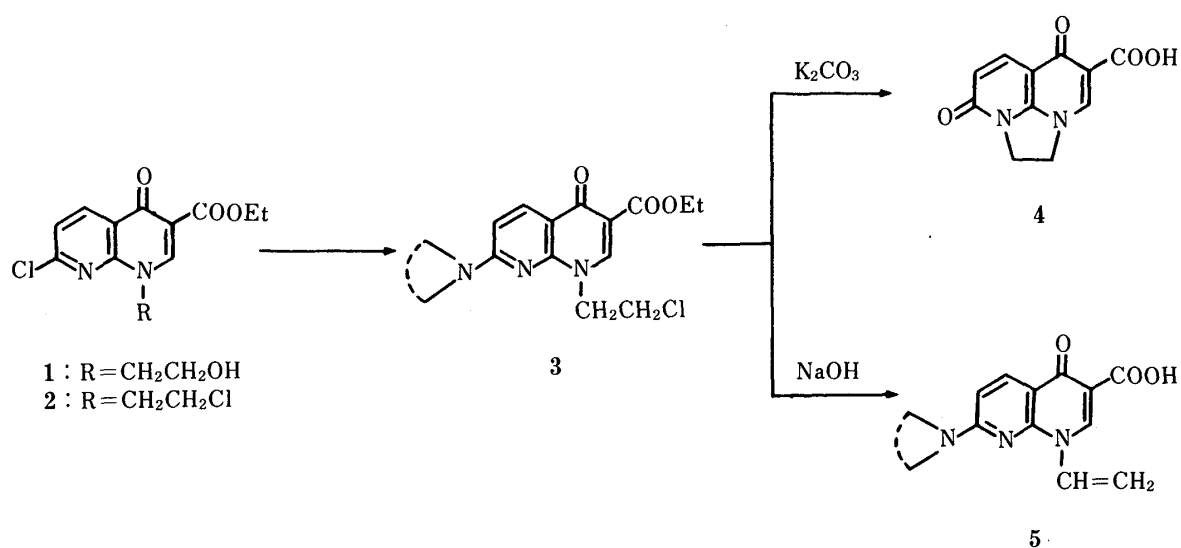


Chart 2

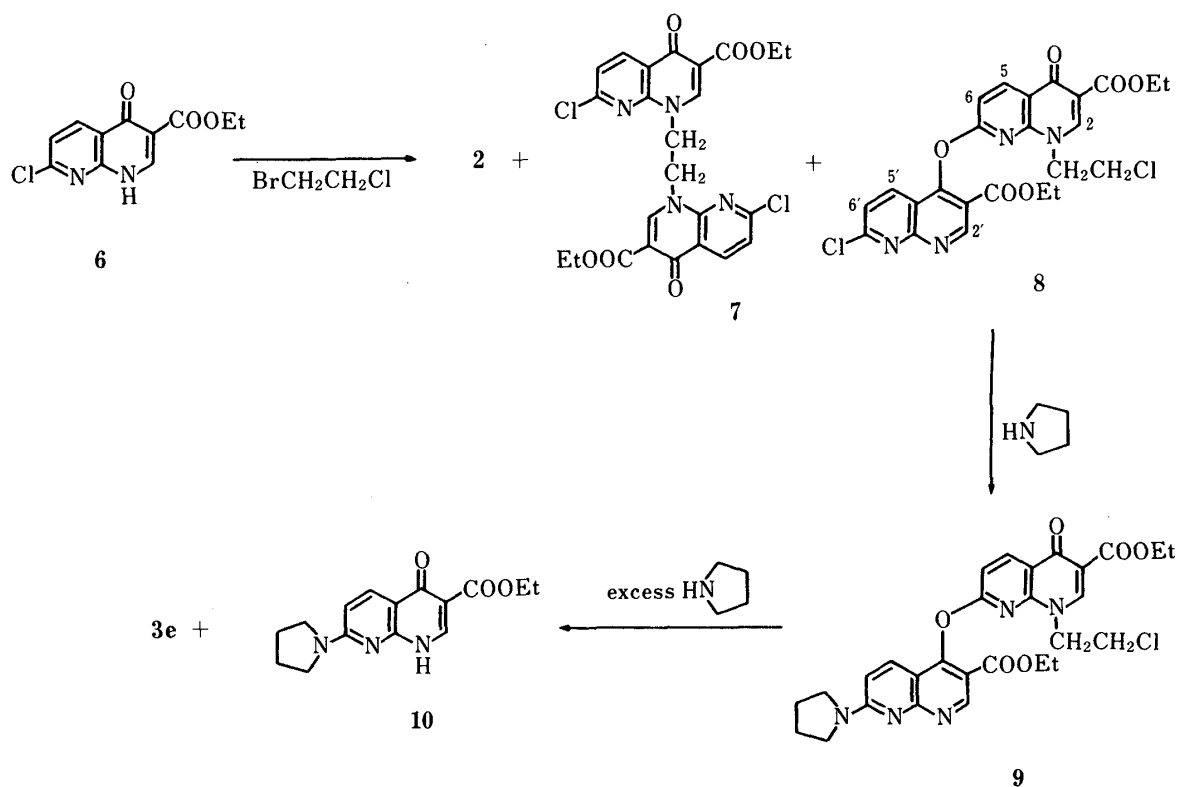
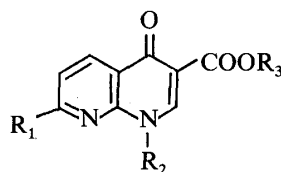


Chart 3

TABLE I. 1,7-Disubstituted 1,4-Dihydro-4-oxo-1,8-naphthyridine-3-carboxylic Acids and Their Esters



Compd. No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	Yield (%)	mp (°C) (Recrystn. solvent)	Formula	Analysis (%)			
							Calcd (Found)			
							C	H	Cl	N
3a	EtOOCN	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	85	174—175 (AcOEt)	C <sub>20</sub> H <sub>25</sub> ClN <sub>4</sub> O <sub>5</sub>	54.98 (54.83)	5.77 (5.86)	8.12 (8.21)	12.83 (12.71)
3b	MeN	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	75	125—127 (Me <sub>2</sub> CO- <i>n</i> -Hexane)	C <sub>18</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>3</sub>	57.06 (57.01)	6.11 (5.88)	9.36 (9.21)	14.79 (14.77)
3c	PhCH <sub>2</sub> N	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	92	123—124 (AcOEt- <i>n</i> -Hexane)	C <sub>24</sub> H <sub>27</sub> ClN <sub>4</sub> O <sub>3</sub>	63.36 (63.46)	5.98 (5.96)	7.79 (7.97)	12.32 (12.24)
3d	AcN	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	77	169—170 (MeCN)	C <sub>19</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>4</sub>	56.09 (56.26)	5.70 (5.93)	8.72 (8.85)	13.77 (13.88)
3e		CH <sub>2</sub> CH <sub>2</sub> Cl	Et	72	200—201 (MeCN)	C <sub>17</sub> H <sub>20</sub> ClN <sub>3</sub> O <sub>3</sub>	58.37 (58.54)	5.76 (5.79)	10.14 (10.32)	12.01 (12.18)
5a	HN	CH=CH <sub>2</sub>	H	81	266—268 (DMF)	C <sub>15</sub> H <sub>16</sub> N <sub>4</sub> O <sub>3</sub>	59.99 (60.07)	5.37 (5.53)	—	18.66 (18.62)
5b	MeN	CH=CH <sub>2</sub>	H	75	238—239 (EtOH)	C <sub>16</sub> H <sub>18</sub> N <sub>4</sub> O <sub>3</sub>	61.13 (61.36)	5.77 (5.59)	—	17.83 (18.04)
5c	PhCH <sub>2</sub> N	CH=CH <sub>2</sub>	H	85	203—205 (AcOEt)	C <sub>22</sub> H <sub>22</sub> N <sub>4</sub> O <sub>3</sub>	67.67 (67.50)	5.68 (5.51)	—	14.35 (14.15)
5d	AcN	CH=CH <sub>2</sub>	H	80	261—262 (DMF)	C <sub>17</sub> H <sub>18</sub> N <sub>4</sub> O <sub>4</sub>	59.64 (59.79)	5.30 (5.20)	—	16.37 (16.41)
5e		CH=CH <sub>2</sub>	H	69	>300 (DMF)	C <sub>15</sub> H <sub>15</sub> N <sub>3</sub> O <sub>3</sub>	63.15 (63.08)	5.30 (5.13)	—	14.73 (14.81)

basis of spectral (proton nuclear magnetic resonance (<sup>1</sup>H-NMR), ultraviolet (UV) and mass) and chemical evidence. Thus, the mass spectra (MS) and elemental analysis of **7** and **8** revealed their molecular formulae to be both C<sub>24</sub>H<sub>20</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>6</sub>. The <sup>1</sup>H-NMR spectrum of **7** (Table II) showed the presence of three aromatic protons, being indicative of a symmetrical structure for **7**. Furthermore, the symmetrical feature is strongly supported by the appearance of a fragment peak of 1/2 M<sup>+</sup> at *m/z* 265 in the MS of **7**. The UV spectrum of **7** was practically the same as that of ethyl 7-chloro-1-ethyl-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate<sup>5</sup> having a pyridone chromophore (Fig. 1). These data were fully in accord with the assigned structure **7**, 1,2-bis(7-chloro-3-ethoxycarbonyl-1,4-dihydro-4-oxo-1,8-naphthyridin-1-yl)ethane.

In the <sup>1</sup>H-NMR spectrum of **8**, an nuclear Overhauser effect (NOE) (19% enhancement of the intensity) of the signal of C<sub>2</sub>-H at δ 8.71 was observed upon irradiation at δ 4.76 (signal due to the *N*-methylene protons). The singlet at δ 9.04 was assigned to C<sub>2</sub>-H on the pyridinol ring because of its appearance at lower field than δ 8.71 for C<sub>2</sub>-H on the pyridone ring.<sup>5</sup> The presence of both pyridone and pyridinol moieties in **8** was supported by its UV spectrum (Fig. 1). Treatment of **8** with 2 mol eq of pyrrolidine in chloroform gave a 92% yield of **9**. Further treatment of **9** with an excess of pyrrolidine gave **3e** and ethyl 1,4-

TABLE II.  $^1\text{H-NMR}$  Data for Compounds 2, 7, 8 and 9

Compd. No.	Solvent	Chemical shift, $\delta$ (J, Hz)						
		C <sub>2</sub> -H	C <sub>2'</sub> -H	C <sub>5</sub> -H	C <sub>5'</sub> -H	C <sub>6</sub> -H	C <sub>6'</sub> -H	-CH <sub>2</sub> CH <sub>2</sub> -
2	CDCl <sub>3</sub>	8.66 (s)	—	8.73 (d, $J=8$ )	—	7.40 (d, $J=8$ )	—	4.00, 4.70 (each t, $J=6$ )
7	CF <sub>3</sub> COOD	9.85 (s)	—	9.04 (d, $J=8$ )	—	7.98 (d, $J=8$ )	—	4.76 (s)
8	CDCl <sub>3</sub>	8.71 (s)	9.04 (s)	8.74 (d, $J=8$ )	8.99 (d, $J=8.5$ )	7.48 (d, $J=8$ )	7.88 (d, $J=8.5$ )	4.04, 4.76 (each t, $J=5.5$ )
9	CDCl <sub>3</sub>	8.70 (s)	8.90 (s)	8.90 (d, $J=8$ )	8.45 (d, $J=8.5$ )	8.06 (d, $J=8$ )	6.52 (d, $J=8.5$ )	4.04, 4.78 (each t, $J=5$ )

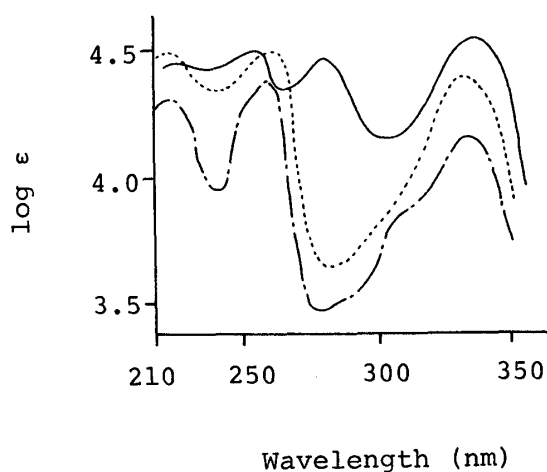


Fig. 1. UV Spectra of Compounds 7 (----) and 8 (—), and Ethyl 7-Chloro-1-ethyl-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (-·-·-)

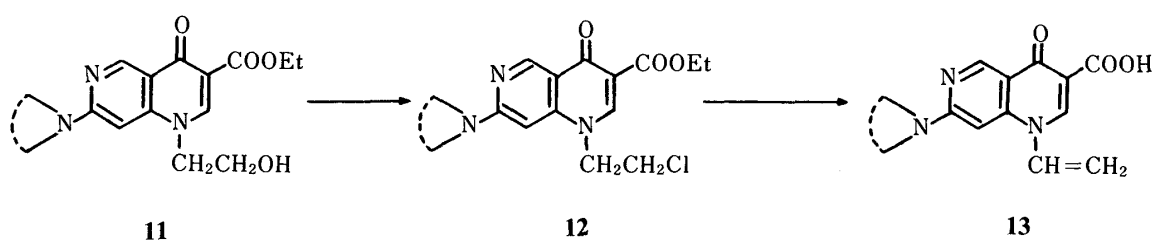


Chart 4

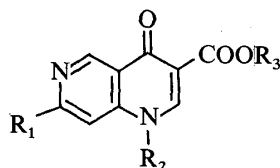
dihydro-4-oxo-7-(1-pyrrolidinyl)-1,8-naphthyridine-3-carboxylate (**10**) in 18 and 52% yields, respectively. Compounds **3e** and **10** were identical with authentic specimens. These spectral and chemical findings permit assignment of the structure of **8** as ethyl 1-(2-chloroethyl)-7-7'-chloro-3'-ethoxycarbonyl-1',8'-naphthyridine-4'-oxy-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate.

1,4-Dihydro-4-oxo-7-(4-substituted 1-piperazinyl)-1-vinyl-1,6-naphthyridine-3-carboxylic acids (**13a-c**) were analogously prepared by chlorination of 1-(2-hydroxyethyl)-1,6-naphthyridines **11a-c**<sup>5</sup> with thionyl chloride, followed by treatment of 1-(2-chloroethyl)-1,6-naphthyridines **12a-c** with 5% potassium hydroxide (Chart 4).

### Structure-Activity Relationships

The *in vitro* antibacterial activities (minimal inhibitory concentrations, MICs) of the 1-vinyl-1,8- and 1,6-naphthyridine derivatives (**5a-e** and **13a-c**) are given in Table IV, which includes the MICs of their 1-ethyl counterparts,<sup>5</sup> pipemidic acid (**Ia**) and its vinyl analogue

TABLE III. 1,7-Disubstituted 1,4-Dihydro-4-oxo-1,6-naphthyridine-3-carboxylic Acids and Their Esters



Compd. No.	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	Yield (%)	mp (°C) (Recrystn. solvent)	Formula	Analysis (%)			
							Calcd (Found)			
							C	H	Cl	N
12a	EtOOCN	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	78	205—208 (EtOH)	C <sub>20</sub> H <sub>25</sub> ClN <sub>4</sub> O <sub>5</sub>	54.98 (54.73)	5.77 (5.50)	8.12 (8.45)	12.82 (12.76)
12b	MeN	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	56	218—219 (MeCN)	C <sub>18</sub> H <sub>23</sub> ClN <sub>4</sub> O <sub>3</sub>	57.06 (57.18)	6.12 (6.17)	9.36 (9.71)	14.79 (14.84)
12c	PhCH <sub>2</sub> N	CH <sub>2</sub> CH <sub>2</sub> Cl	Et	47	210—212 (MeCN)	C <sub>24</sub> H <sub>27</sub> ClN <sub>4</sub> O <sub>3</sub>	63.36 (63.17)	5.98 (5.96)	7.79 (7.79)	12.32 (12.04)
13a	HCl·HN	CH=CH <sub>2</sub>	H	65	248—249 (H <sub>2</sub> O)	C <sub>15</sub> H <sub>16</sub> N <sub>4</sub> O <sub>3</sub> ·HCl	53.50 (53.66)	5.09 (5.13)	10.53 (10.35)	16.46 (16.57)
13b	MeN	CH=CH <sub>2</sub>	H	69	229—230 (EtOH)	C <sub>16</sub> H <sub>18</sub> N <sub>4</sub> O <sub>3</sub>	61.13 (61.15)	5.77 (5.62)	—	17.83 (17.59)
13c	PhCH <sub>2</sub> N	CH=CH <sub>2</sub>	H	96	206—208 (EtOH)	C <sub>22</sub> H <sub>22</sub> N <sub>4</sub> O <sub>3</sub>	67.67 (67.79)	5.68 (5.65)	—	14.35 (14.65)

TABLE IV. *In Vitro* Antibacterial Activity

Compound No.	Minimum inhibitory concentration (μg/ml)		
	<i>S. aureus</i> TERAJIMA	<i>E. coli</i> K-12	<i>P. aeruginosa</i> TSUCHIJIMA
5a	100 (30) <sup>a)</sup>	1 (3)	10 (10)
5b	>100 (100)	1 (1)	10 (30)
5c	10 (10)	3 (1)	>100 (>100)
5d	>100 (100)	3 (10)	100 (>100)
5e	10 (10)	1 (1)	30 (>100)
13a	>100 (>100)	3 (3)	30 (30)
13b	100 (30)	3 (3)	30 (100)
13c	10 (3)	1 (1)	100 (100)
Ib	>100 (30)	3 (1)	3 (10)

a) Figures in parentheses represent MICs of the corresponding 1-ethyl derivatives.

(Ib)<sup>4)</sup> for comparison.

The vinyl compounds **5** and **13** exhibit enhanced activity against Gram-negative bacteria (*Escherichia coli* K-12 and *Pseudomonas aeruginosa* TSUCHIJIMA) compared with the 1-ethyl congeners, whereas they tend to show lower activity against Gram-positive *Staphylococcus aureus* TERAJIMA. The 1,8-naphthyridine derivatives (**5**) are generally more active than the corresponding 1,6-naphthyridine analogues (**13**), particularly against Gram-negative bacteria. The intermediate ethyl esters (**3**, **11** and **12**) and the by-products (**4**, **7**, **8** and **9**) are practically inactive.

The present study thus demonstrates a similarity in SARs between the 1-vinylnaphthyridine derivatives and the corresponding 8-vinylpyrido[2,3-*d*]pyrimidine analogues discussed

in our previous paper.<sup>4)</sup>

### Experimental

Melting points were determined on a Yanagimoto micro melting point apparatus and are uncorrected. Infrared (IR) spectra were recorded on a Hitachi 215 spectrometer. UV spectra were measured in EtOH on a Shimadzu MPS-5000 spectrometer. NMR spectra were recorded on a Varian A-60 or HA-100D in a CDCl<sub>3</sub> solution, unless otherwise specified, with tetramethylsilane as an internal standard. MS were determined with a Hitachi RMU-6L spectrometer. Organic extracts were dried over anhydrous MgSO<sub>4</sub>.

**Ethyl 7-Chloro-1-(2-chloroethyl)-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (2)**—A mixture of ethyl 7-chloro-1,4-dihydro-1-(2-hydroxyethyl)-4-oxo-1,8-naphthyridine-3-carboxylate (**1**)<sup>5)</sup> (3.9 g, 13 mmol) and thionyl chloride (SOCl<sub>2</sub>) (1.45 ml, 20 mmol) in dry CHCl<sub>3</sub> (50 ml) was refluxed for 2 h with stirring. The mixture was cooled and poured into ice-water, and the CHCl<sub>3</sub> layer was washed successively with saturated NaHCO<sub>3</sub> solution and water. The CHCl<sub>3</sub> solution was dried, and concentrated to dryness *in vacuo*. The residue was chromatographed on silica gel using CHCl<sub>3</sub> as an eluent to give **2** (2.88 g, 70%), which was recrystallized from EtOH as pale yellow needles, mp 148–149 °C. *Anal.* Calcd for C<sub>13</sub>H<sub>12</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>3</sub>: C, 49.54; H, 3.84; Cl, 22.50; N, 8.89. Found: C, 49.56; H, 3.65; Cl, 22.80; N, 8.59. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 1670, 1640. NMR (60 MHz)  $\delta$ : 4.00 (2H, t, *J*=6 Hz, NCH<sub>2</sub>CH<sub>2</sub>Cl), 4.70 (2H, t, *J*=6 Hz, NCH<sub>2</sub>CH<sub>2</sub>Cl), 7.40 (1H, d, *J*=8 Hz, C<sub>6</sub>-H), 8.66 (1H, s, C<sub>2</sub>-H), 8.73 (1H, d, *J*=8 Hz, C<sub>5</sub>-H).

**Ethyl 7-Substituted 1-(2-Chloroethyl)-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (3a–e)**—General Procedure: A mixture of **2** (6 mmol), an appropriate amine (18 mmol), and EtOH (50 ml) was refluxed for 1–3 h. After removal of the solvent and the excess amine, the resulting residue was dissolved in CHCl<sub>3</sub>. The CHCl<sub>3</sub> solution was washed with water, dried, and concentrated to dryness *in vacuo*. The residual solid was recrystallized from an appropriate solvent to give **3a–e** (Table I).

**1,2,3,6,9,10-Hexahydro-6,9-dioxoimidazo[1,2,3-*ij*][1,8]naphthyridine-5-carboxylic Acid (4)**—To a solution of **3a** (4.37 g, 10 mmol) in EtOH (50 ml) was added aqueous 10% K<sub>2</sub>CO<sub>3</sub> (100 mmol) and the mixture was refluxed for 2 h. The EtOH was distilled off *in vacuo* and the residual aqueous solution was acidified with AcOH. The precipitate was collected, washed with water, and recrystallized from dimethylformamide (DMF) to give **4** (2.25 g, 97%) as pale yellow needles, mp > 300 °C. *Anal.* Calcd for C<sub>11</sub>H<sub>8</sub>N<sub>2</sub>O<sub>4</sub>: C, 56.90; H, 3.47; N, 12.07. Found: C, 56.86; H, 3.52; N, 12.28. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 1710, 1670. NMR (60 MHz, DMSO-*d*<sub>6</sub>)  $\delta$ : 4.2–4.8 (4H, m), 6.47 (1H, d, *J*=8 Hz, C<sub>6</sub>-H), 7.98 (1H, d, *J*=8 Hz, C<sub>5</sub>-H), 8.90 (1H, s, C<sub>2</sub>-H). EIMS *m/z*: 232 (M<sup>+</sup>), 188 (M<sup>+</sup> – CO<sub>2</sub>).

**7-Substituted 1,4-Dihydro-4-oxo-1-vinyl-1,8-naphthyridine-3-carboxylic Acids (5a–e)**—General Procedure: To a solution of **3** (10 mmol) in EtOH (50 ml) was added aqueous 10% NaOH (100 mmol) and the mixture was refluxed for 1–2 h. The EtOH was distilled off *in vacuo*. The residual aqueous solution was acidified with AcOH. The precipitate was collected, washed with water, and recrystallized from an appropriate solvent to give **5a–e** (Table I).

**Reaction of Ethyl 7-Chloro-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (6)<sup>5)</sup> with 1-Bromo-2-chloroethane**—1-Bromo-2-chloroethane (8.65 g, 30 mmol) was added to a mixture of **6** (5.05 g, 20 mmol) and anhydrous K<sub>2</sub>CO<sub>3</sub> (4.10 g, 30 mmol) in DMF (60 ml) with stirring at 60 °C. The mixture was heated at 60–70 °C for 2 h and then filtered. The filtrate was concentrated to dryness *in vacuo*. The resulting residue was extracted with CHCl<sub>3</sub>. The extract was washed with water, dried, and concentrated to dryness. The residue was chromatographed on silica gel with a CHCl<sub>3</sub>–MeOH mixture (50 : 1, v/v) to give **2** (2.50 g, 40%), ethyl 1-(2-chloroethyl)-7-7'-chloro-3'-ethoxycarbonyl-1',8'-naphthyridine-4'-oxy-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (**8**) (0.66 g, 25%) and 1,2-bis(7-chloro-3-ethoxycarbonyl-1,4-dihydro-4-oxo-1,8-naphthyridin-1-yl)ethane (**7**) (0.10 g, 5%). **7**: mp 283–286 °C (EtOH), colorless needles. *Anal.* Calcd for C<sub>24</sub>H<sub>20</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>6</sub> · 1/2 H<sub>2</sub>O: C, 53.54; H, 3.92; Cl, 13.12; N, 10.37. Found: C, 53.54; H, 3.62; Cl, 13.29; N, 10.30. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 1725, 1700. EIMS *m/z*: 530 (M<sup>+</sup>), 458 (M<sup>+</sup> – CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>), 386 (M<sup>+</sup> – 2CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>), 265 (1/2 M<sup>+</sup>). **8**: mp 246–248 °C (MeCN), colorless needles. *Anal.* Calcd for C<sub>24</sub>H<sub>20</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>6</sub>: C, 54.25; H, 3.79; Cl, 13.35; N, 10.55. Found: C, 54.09; H, 3.72; Cl, 13.45; N, 10.57. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 1730, 1700. EIMS *m/z*: 530 (M<sup>+</sup>), 458 (M<sup>+</sup> – CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>), 386 (M<sup>+</sup> – 2CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>). <sup>1</sup>H-NMR data for **7** and **8** are given in Table II.

**Ethyl 1-(2-Chloroethyl)-7-7'-(1-pyrrolidinyl)-3'-ethoxycarbonyl-1',8'-naphthyridine-4'-oxy-1,4-dihydro-4-oxo-1,8-naphthyridine-3-carboxylate (9)**—A mixture of **8** (2.65 g, 5 mmol) and pyrrolidine (0.92 ml, 11 mmol) in CHCl<sub>3</sub> (40 ml) was stirred overnight at room temperature. The mixture was washed with water, dried, and concentrated to dryness. The residual solid was recrystallized from AcOEt to give **9** (2.60 g, 92%) as colorless needles, mp 138–140 °C. *Anal.* Calcd for C<sub>28</sub>H<sub>28</sub>ClN<sub>5</sub>O<sub>6</sub> · 1/2 H<sub>2</sub>O: C, 58.48; H, 5.08; Cl, 6.17; N, 12.18. Found: C, 58.40; H, 4.78; Cl, 6.46; N, 12.23. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>: 1725, 1700. <sup>1</sup>H-NMR: See Table II.

**Reaction of 9 with an Excess of Pyrrolidine**—A mixture of **9** (1.30 g, 2.3 mmol) and pyrrolidine (0.96 ml, 11.5 mmol) in CHCl<sub>3</sub> (30 ml) was refluxed for 10 h. After the mixture had cooled, the precipitate was collected and recrystallized from DMF to give ethyl 1,4-dihydro-4-oxo-7-(1-pyrrolidinyl)-1,8-naphthyridine-3-carboxylate (**10**) (0.34 g, 52%). The filtrate was concentrated to dryness. The resulting solid was recrystallized from MeCN to give **3e** (0.14 g, 18%), which was identical with an authentic specimen (mp and IR). **10**: mp > 300 °C (DMF), pale yellow needles. *Anal.* Calcd for C<sub>15</sub>H<sub>17</sub>N<sub>3</sub>O<sub>3</sub>: C, 62.70; H, 5.96; N, 14.63. Found: C, 62.61; H, 5.87; N, 14.67. IR  $\nu_{\max}^{\text{KBr}}$  cm<sup>-1</sup>:

1680, 1620.

**Ethyl 7-Substituted 1-(2-Chloroethyl)-1,4-dihydro-4-oxo-1,6-naphthyridine-3-carboxylate (12a—c)**—General Procedure: A mixture of ethyl 7-substituted 1-(2-hydroxyethyl)-1,4-dihydro-4-oxo-1,6-naphthyridine-3-carboxylate (**11a—c**)<sup>5</sup> (2 mmol) and SOCl<sub>2</sub> (0.22 ml, 3 mmol) in dry CHCl<sub>3</sub> (30 ml) was refluxed for 1—3 h. After evaporation of the solvent and the excess SOCl<sub>2</sub>, the residue was dissolved in CHCl<sub>3</sub>. The CHCl<sub>3</sub> solution was washed with water, dried, and concentrated to dryness. The residual solid was recrystallized from an appropriate solvent to give **12a—c** (Table III).

**7-Substituted 1,4-Dihydro-4-oxo-1-vinyl-1,6-naphthyridine-3-carboxylic Acids (13a—c)**—General Procedure: A suspension of the ester **12a—c** (10 mmol) in aqueous 5% KOH (10 ml) was heated at 100 °C for 30 min, then cooled. The alkaline solution was adjusted to pH 4 with 5% HCl. The precipitate was collected, washed with water, and recrystallized from an appropriate solvent to give **13a—c** (Table III).

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#### References and Notes

- 1) Part IV: H. Egawa, T. Miyamoto, A. Minamida, Y. Nishimura, H. Okada, H. Uno and J. Matsumoto, *J. Med. Chem.*, **27**, 1543 (1984).
- 2) A part of this work was presented at the Abstracts of Papers, Vol. II, 96th Annual Meeting of the Pharmaceutical Society of Japan, Nagoya, April 7, 1976, p. 120; Dainippon Pharmaceutical Co., Ltd., German Patent Offen. 2362553 (1974) [*Chem. Abstr.*, **81**, 105562 (1974)].
- 3) Present address: *Research Laboratories, Teikoku Chemical Industry Co., Ltd., 41, 5-chome, Senzo, Itami, Hyogo 664, Japan.*
- 4) J. Matsumoto and S. Minami, *J. Med. Chem.*, **18**, 74 (1975).
- 5) T. Hirose, S. Mishio, J. Matsumoto and S. Minami, *Chem. Pharm. Bull.*, **30**, 2399 (1982).