Chem. Pharm. Bull. 33(2) 865-868 (1985)

## Synthesis of Deuterium-Labelled 16α-Hydroxy-4-androstene-3,17dione and 16α-Hydroxydehydroepiandrosterone

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(Received June 4, 1984)

 $16\alpha$ -Hydroxy-4-androstene-3,17-dione- $d_5$  (4) and  $3\beta$ ,16 $\alpha$ -dihydroxy-5-androsten-17-one- $d_6$  (10) were synthesized. Treatment of  $16\alpha$ -bromo-5-androstene-3,17-dione (2) with deuterium oxide and methanol-OD gave the 4-en-3-oxo derivative-d 3. The bromide 3 was converted into the  $16\alpha$ -hydroxide- $d_5$  4 via controlled alkaline hydrolysis using sodium hydroxide-OD in deuterium oxide-pyridine.  $3\beta$ -Hydroxy-5-androsten-17-one- $d_5$  (8), which was obtained from 17,17-ethylenedioxy-5-androsten-3-one (5) via treatment with potassium tert-butoxide in tert-butanol-OD, followed by lithium aluminum tri-tert-butoxydeuteride reduction and then acid hydrolysis, was derivatized to the  $16\alpha$ -bromide 9 by treatment with cupric bromide. The bromide 9 was similarly hydrolyzed to yield the  $16\alpha$ -hydroxide- $d_6$  10.

**Keywords**—controlled alkaline hydrolysis; deuterium labelling;  $16\alpha$ -bromodehydro-epiandrosterone- $d_5$ ;  $16\alpha$ -bromoandrostenedione- $d_4$ ;  $16\alpha$ -hydroxydehydroepiandrosterone- $d_6$ ;  $16\alpha$ -hydroxyandrostenedione- $d_5$ 

We recently discovered a controlled stereospecific alkaline hydrolysis of 16-bromo-17-oxo androgens<sup>1)</sup> and also developed a hydrolytic method for the synthesis of several  $16\alpha$ -hydroxy-17-oxo steroids.<sup>2)</sup> A stereospecific deuterium incorporation at the  $16\beta$ -position of the  $16\alpha$ -hydroxides by hydrolysis using a medium containing deuterium oxide was also reported by us.<sup>2a,d)</sup> In view of the quantitative importance<sup>3)</sup> of estriol in pregnancy, it is of interest to know the concentrations of  $16\alpha$ -hydroxylated  $C_{19}$  steroid precursors in the feto-placental unit.

We report here an efficient synthesis of deuterium-labelled  $16\alpha$ -hydroxy-4-androstene-3,17-dione (4) and  $16\alpha$ -hydroxydehydroepiandrosterone<sup>4)</sup> (10) with good isotopic purity. These compounds are required as carriers and internal standards for quantitative evaluation of the steroids by combined gas chromatography-mass spectrometry (GC-MS). Labelling experiments were undertaken with the intention of placing deuterium atoms at positions 2, 3, 4, 6 and 16 of the steroids, involving the controlled alkaline hydrolysis of the  $16\alpha$ -bromo-17-ones 3 and 9 as a key reaction.

## **Results and Discussion**

When  $16\alpha$ -bromo-5- androstene-3,17-dione (2), obtained by 8 N chromic acid oxidation<sup>5)</sup> of the corresponding  $3\beta$ -hydroxide 1, was heated under reflux in deuterium oxide and diglyme according to Malhotra and Ringold,<sup>6)</sup> a debrominated and isomerized product, 4-androstene-3,17-dione- $d_6$  ( $d_3$  13%,  $d_4$  17%,  $d_5$  20%,  $d_6$  50%), was unexpectedly obtained in high yield. On the other hand, the use of methanol-OD instead of diglyme in the reaction gave a deuterated and isomerized product,  $16\alpha$ -bromo-4-en-3-one derivative 3. The mechanism involved in the efficient debromination of the  $16\alpha$ -bromide 2 under the former conditions is not clear at present. However, hydrobromic acid liberated by partial decomposition of the bromide 2-4, 4- $d_2$ , initially produced,<sup>6)</sup> should catalyze the isomerization of the 5-en-3-one to the 4-en-3-

one. Previous studies<sup>6,7)</sup> have demonstrated that deuterium atoms are incorporated at positions 2, 4 and 6 of the product 3 by the above reaction. The labelled positions were further checked by proton nuclear magnetic resonance (<sup>1</sup>H-NMR) analysis showing that hydrogens at the C-2 and -6 positions and hydrogen at C-4 had been exchanged for deuterium to the extents of about 50 and 80%, respectively.

HO 1 2 
$$(D) R_2$$
 $R = R_1 = R_1 = R_2 = R_1$ 
 $R = R_2 = R_3 = R_4 = R_2 = R_3$ 
 $R = R_1 = R_2 = R_3 = R_4 = R_4 = R_4 = R_5 =$ 

Treatment of deuterio  $16\alpha$ -bromo-17-one 3 with  $1.2\,\mathrm{eq}$  of sodium hydroxide-OD in deuterium oxide and pyridine<sup>2a)</sup> gave the corresponding  $16\alpha$ -hydroxide 4 ( $d_3$  16%,  $d_4$  27%,  $d_5$  52%,  $d_6$  5%) with more than 98% deuterium at C-16 $\beta$  in quantitative yield. Equilibration of the unsaturated ketones 3 and 4 in the basic medium might cause enrichment of the isotope contents at C-2, -4 and -6.8)

In order to obtain the  $16\alpha$ -hydroxy-17-one- $d_6$  10 having a 5-en-3 $\beta$ -ol system, we initially synthesized dehydroepiandrosterone- $d_5$  (8) from 17,17-ethylenedioxy-5-androsten-3-one (5). Treatment of compound 5 with potassium tert-butoxide in tert-butanol-OD, followed by acetic acid protonation<sup>9)</sup> gave the deuterio derivative 6 along with the 4-en-3-oxo isomer. The deuterated mixture was subjected to reduction with lithium aluminum tri-tert-butoxydeuteride. The product,  $3\beta$ -hydroxy-17,17-ethyleneacetal- $d_5$  7, was then treated with acid (without isolation) to give compound 8. The yield (30%) was much improved, compared to that previously reported (4%) which was obtained using 4-en-3-one derivatives as substrates. Deuterium was also efficiently incorporated at C-2, -3 $\alpha$  and -4 as expected ( $d_2$  6%,  $d_3$  10%,  $d_4$  25%,  $d_5$  59%).

Bromination of the 17-oxo derivative 8 with  $3 eq^{2a}$  of cupric bromide gave the  $16\alpha$ -bromo-17-one 9 in quantitative yield. Treatment of the bromide 9 with sodium hydroxide-OD in deuterium oxide and pyridine<sup>2a)</sup> afforded the  $16\alpha$ -hydroxide <sup>10</sup>  $(d_3 5\%, d_4 11\%, d_5 26\%, d_6 58\%)$  in high yield.

The structures of deuterio compounds 3, 4, 9 and 10 were confirmed by the  $^{1}$ H-NMR spectra. This synthesis offers the advantage of permitting a high deuterium content without contamination by the natural form ( $d_0$ -species). The deuterio androgens should be suitable as internal standards for mass fragmentography. A quantitative GC-MS analysis of the androgens in biological fluid is under way to further investigate their physiological importance.

## **Experimental**

Melting points were measured on a Yanagimoto melting-point apparatus and are uncorrected. Infrared (IR) spectra were recorded on a Shimadzu IR 400 spectrometer as KBr pellets. NMR spectra were obtained with a JEOL PMX 60 spectrometer at 60 MHz with tetramethylsilane as an internal standard. Mass spectra were measured on a Hitachi RMU-7 spectrometer.

**16α-Bromo-5-androstene-3,17-dione (2)**—A solution of  $16\alpha$ -bromo-3 $\beta$ -hydroxy-5-androsten-17-one (1) (1.0 g) in acetone (80 ml) was treated with 8 N chromic acid<sup>5)</sup> (4 ml) at 0 °C for 5 min. The reaction mixture was poured into ice-water (500 ml). The precipitate was collected by filtration and crystallized from acetone to give **2** (630 mg, 63%) as colorless needles, mp 147—148 °C. IR  $_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 1745 and 1710.  $^{\rm 1}$ H-NMR (CDCl<sub>3</sub>):  $\delta$  0.95 (3H, s, 18-CH<sub>3</sub>), 1.21 (3H, s, 19-CH<sub>3</sub>), 4.56 (1H, t, J = 8 Hz, 16 $\beta$ -H), 5.40 (1H, m, 6-H). *Anal.* Calcd for C<sub>19</sub>H<sub>25</sub>BrO<sub>2</sub>: C, 62.47; H, 6.90; Br, 21.87. Found: C, 62.45; H, 6.66; Br, 21.99.

Treatment of Compound 2 with  $D_2O$ —A) A solution of 2 (288 mg) in  $D_2O$  (0.7 ml) and diglyme (7 ml) was heated under reflux for 4.5 d. Water was added and the precipitate was collected by filtration and crystallized from acetone to give 4-androstene-3,17-dione- $d_6$  (130 mg, 45%) as colorless needles, mp 169—171 °C (lit. 10) 168—170 °C). MS:  $d_3$  13%,  $d_4$  17%,  $d_5$  20%,  $d_6$  50%.

B) A solution of 2 (312 mg) in  $D_2O$  (0.8 ml) and MeOD (10 ml) was heated under reflux for 4d. Evaporation of the solvent under reduced pressure gave a solid, which was crystallized from acetone– $H_2O$  to give  $16\alpha$ -bromo-4-androstene-3,17-dione-d (3) (242 mg, 78%) as colorless needles, mp 170—171 °C (lit. 11) 172—174 °C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  0.98 (3H, s, 18-CH<sub>3</sub>), 1.23 (3H, s, 19-CH<sub>3</sub>), 2.30—2.70 (ca. 5H), 5.76 (0.2H, s, 4-H).

16α-Hydroxy-4-androstene-3,17-dione-2,2,4,6,16β- or 2,4,6,6,16β- $d_5$  (4)—Sodium metal (18 mg) was carefully added to a mixture of D<sub>2</sub>O (1.5 ml) and pyridine (5.5 ml) and then 3 (242 mg) was dissolved in the solution. The mixture was allowed to stand at room temperature for 2 h. After the same work-up as reported previously,<sup>2d)</sup> the crude product (200 mg) was obtained. Crystallization of the product from acetone yielded 4 (185 mg, 93%) as colorless needles, mp 187—188 °C (lit.<sup>2a)</sup> 188—190 °C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  1.00 (3H, s, 18-CH<sub>3</sub>), 1.21 (3H, s, 19-CH<sub>3</sub>), 5.72 (0.2H, s, 4-H). MS:  $d_3$  16%,  $d_4$  27%,  $d_5$  52%,  $d_6$  5%.

17,17-Ethylenedioxy-5-androsten-3-one (5)—Compound 5 was synthesized according to Williams  $et\ al.^{12)}$  mp 142—146 °C (lit. 12) 141—146 °C).

 $3\beta$ -Hydroxy-5-androsten-17-one-2,2,3 $\alpha$ ,4,4- $d_5$ (8)—Under a nitrogen atmosphere, tert-BuOH (12 ml) was added to 5 (454 mg) and tert-BuOK (1.15 g), and the yellow solution was allowed to stand at room temperature for 2 h. Cold AcOD (22 ml, 10%) was rapidly added and the mixture was poured into ice-water. After extraction with ether (300 ml  $\times$  2), the organic layer was washed with 5% NaHCO<sub>3</sub> solution and water, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to give an oily residue. Thin-layer chromatographic analysis (hexane-AcOEt, 3:1) indicated the residue to be a mixture of 5-en- and 4-en-3-oxo steroids (ca. 3:1).

LiAlD<sub>4</sub> (181 mg) was placed in dry tetrahydrofuran (THF) (14 ml) and the suspension was stirred in an ice bath for 10 min. While the slurry was being stirred, tert-BuOH (0.9 ml) was added dropwise, followed by the dropwise addition of a solution of the oily residue obtained above in THF (6 ml). The entire mixture was stirred for 30 min at 0 °C and for 2 h at room temperature. A few drops of water were added to decompose the excess LiAl (tert-BuO)<sub>3</sub>D and the mixture was acidified with 10% HCl solution (10 ml) and then allowed to stand for 5 h at room temperature. The 17-oxo products were recovered by extraction with AcOEt and subjected to silica gel column chromatography (hexane–AcOEt) to give a solid, which was crystallized from acetone–H<sub>2</sub>O to yield 8 (120 mg, 30%), mp 150—153 °C (lit. 13) 153 °C). 1H-NMR (CDCl<sub>3</sub>):  $\delta$  0.90 (3H, s, 18-CH<sub>3</sub>), 1.03 (3H, s, 19-CH<sub>3</sub>), 5.43 (1H, m, 6-H). MS:  $d_2$  6%,  $d_3$  10%,  $d_4$  25%,  $d_5$  59%.

16α-Bromo-3β-hydroxy-5-androsten-17-one-2,2,3α,4,4- $d_5$  (9)—A solution of 8 (100 mg) in dry MeOH (10 ml) was heated under reflux for 12 h. After the same work-up as reported previously, <sup>14</sup> a crude product was obtained. Crystallization of the product from MeOH gave 9 (120 mg, 94%), mp 176—178 °C (lit. <sup>14</sup>) 175—176 °C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$ 0.93 (3H, s, 18-CH<sub>3</sub>), 1.00 (3H, s, 19-CH<sub>3</sub>), 4.40 (1H, m, 16β-H), 5.40 (1H, m, 6-H).

3 $\beta$ ,16 $\alpha$ -Dihydroxy-5-androsten-17-one-2,2,3 $\alpha$ ,4,4,16 $\beta$ - $d_6$  (10)—Compound 9 (100 mg) was hydrolyzed with NaOD in D<sub>2</sub>O and pyridine essentially as described for the synthesis of 4 to afford 10 (76 mg, 92%) as colorless needles, mp 187—189 °C (MeOH) (lit.<sup>2a)</sup> 188—190 °C). <sup>1</sup>H-NMR (CDCl<sub>3</sub>): δ 0.98 (3H, s, 18-CH<sub>3</sub>), 1.00 (3H, s, 19-CH<sub>3</sub>), 5.41 (1H, m, 6-H). MS:  $d_3$  5%,  $d_4$  11%,  $d_5$  26%,  $d_6$  58%.

Acknowledgment We are grateful to Professor T. Nambara and Dr. K. Shimada of Tohoku University for mass analysis.

## References and Notes

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