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# Reactions of 3-Phenylglycidic Esters. V.<sup>1)</sup> Reaction of Methyl 3-(4-Methoxyphenyl)glycidate with 2-Nitroaniline and Synthesis of 1,5-Benzodiazepine Derivatives

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Stereochemical aspects of the oxirane-ring opening of methyl trans-3-(4-methoxyphenyl)-glycidate (1) with 2-nitroaniline (5) were investigated. ZnI<sub>2</sub> showed good catalytic activity in this reaction, giving the cis-opening product (6a) predominantly. On the other hand, the reaction on the surface of silica gel proceeded predominantly by trans-opening. Silica gel had rather little effect on the reaction of 1 with 2-nitrothiophenol or 2-nitrophenol. Some 1,5-benzodiazepine derivatives (18—21), 1-aza analogues of diltiazem, were synthesized from 6a, b for pharmacological evaluation. Some oxidative epimerization of 9a, b was observed during cyclization in boiling xylene in the presence of air. The cerebral vasodilating activity of 21a, the most potent compound in this series, was about 0.5 that of racemic diltiazem. The structure-activity relationships are discussed.

**Keywords**—methyl *trans*-3-(4-methoxyphenyl)glycidate; 2-nitroaniline; silica gel-promoted reaction; 1,5-benzodiazepine derivative; cerebral vasodilating activity; structure–activity relationship

Previous papers of this series described the results of detailed studies on the reaction of methyl 3-(4-methoxyphenyl)glycidate (1) with 2-nitrothiophenol<sup>2)</sup> or 2-nitrophenol.<sup>1)</sup> Stereoselective opening of the oxirane ring of 1 by these nucleophiles could be achieved by the choice of appropriate reaction conditions, making it possible to produce either the *threo*- or *erythro*-isomer of the hydroxy esters (2a, b). The *threo*-isomer of 2a has served as a key intermediate for diltiazem (4a), a potent antianginal agent, <sup>3c,d)</sup> and *threo*-2b was also converted to the 1-oxa analogue (4b) of diltiazem.<sup>1)</sup>

Our continued interest both in the stereochemical aspects of the ring opening of 3-arylglycidic esters by various nucleophiles and in the structure—activity relationships of diltiazem derivatives prompted us to examine the reaction of 1 with 2-nitroaniline (5). We describe here the mode of ring opening of 1 by 5 under various conditions as well as the synthesis and vasodilating activity of several 1,5-benzodiazepine derivatives (3), 1-aza analogs of diltiazem.

## Reaction of Methyl trans-3-(4-Methoxyphenyl)glycidate (1) with 2-Nitroaniline (5)

Aminolysis of 3-phenylglycidic esters is known to give *trans*-opening products resulting from  $S_N$ 2-type nucleophilic attack of amines in all cases.<sup>4)</sup> Zymalkowski and coworkers reported that treatment of 3-phenylglycidic acid derivatives with various aromatic amines also gives *erythro*-3-amino-2-hydroxy-3-phenylpropionic acid derivatives (*trans*-opening products).<sup>4a)</sup>

The results of the reaction of 1 with 5 under a variety of conditions are summarized in Table I. When 1 was allowed to react with 5 in the absence of catalyst, no reaction was

	Solvent	Additive	Reaction con-	ditions	Total yield	threo/erythro ratio	
Entry			Temperature (°C)	Time (h)	of <b>6a</b> and <b>6b</b> (%)		
1	Toluene		60	48	f)		
2	CH <sub>3</sub> CN		60	72	f)		
3	HMPA		22	72	f)		
4	Toluene	$\mathrm{MgCl}_{2}^{a,b)}$	22	48	f)		
5	CH <sub>3</sub> CN	$KOH^{a)}$	22	48	f)		
6	Toluene	$SnCl_2^{a)}$	22	48	16.8	2.0	
7	Toluene	$ZnI_2^{a}$	22	21	100	2.0	
8	Benzene	$SiO_2^{c)}$	22	17	58.4	0.5	
9	Benzene	$Al_2O_3^{d}$	22	17	17.9	0.3	
10	DMSO	NaH <sup>e)</sup>	22	0.5	g)		

TABLE I. Reaction of Methyl trans-3-(4-Methoxyphenyl)glycidate with 2-Nitroaniline

a) About 0.3 eq of catalyst was used. b)  $Sn(OCOC_7H_{15})_2$  and  $Zn(OAc)_2$  were also ineffective. c) The reaction was carried out in a silica gel chromatographic column. Stirring of the reaction mixture with a slurry of  $SiO_2$  gave a similar result. See Experimental. d) The reaction was carried out in an aluminum oxide chromatographic column. e) 1 eq of NaH was used. f) No nitro ester (6) was obtained. g) 3-(4-Methoxyphenyl)-2'-nitroglycidanilide (8) was obtained in 62.8% yield.

observed at temperatures in the range of 22—60 °C (entries 1—3). The reaction also proved to be refractory in the presence of catalysts such as MgCl<sub>2</sub>, stannous 2-ethylhexanoate, zinc acetate, and KOH, which were effective in the reaction of 1 with thiophenols.<sup>2)</sup> The reaction in the presence of SnCl<sub>2</sub>, however, gave a low yield (16.8%) of a diastereoisomeric mixture of the nitro ester (6a, b). This mixture was hydrogenated to give the amino esters (7a, b), since the separation of 6a, b was unsuccessful. The mixture of 7a, b could be readily separated by preparative thin layer chromatography (TLC) to give the *threo*-isomer (7a), mp 91—92 °C, and the *erythro*-isomer (7b), mp 120—122 °C, in yields of 40.7 and 21.7%, respectively. Thus, estimation of the *threo/erythro* ratio of 6a, b was also carried out by this procedure in other cases.

The mass spectra (MS) of 7a, **b** were superimposable on each other and showed a fragment ion characteristic of the  $\alpha$ -hydroxy-ester structure at m/e 227 (M – CH(OH)CO<sub>2</sub>Me)<sup>+</sup>.<sup>5)</sup> The stereochemistry of 7a, **b** was determined by converting them into the 1,5-benzodiazepine derivatives (10a, **b**) as shown in Chart 2 (see below). The nuclear magnetic resonance (NMR) spectral characteristics of these isomeric lactams was quite similar to those of the corresponding 1-thia<sup>2a,3)</sup> and 1-oxa analogues.<sup>1)</sup> The vicinal coupling constant between the methine protons at  $C_3$  and  $C_4$  of 10a derived from 7a was 5 Hz, while that of 10b obtained from 7b was 9 Hz. Therefore, we assigned the *cis*-lactam structure for 10a and the *trans* one for 10b, and it follows that the amino esters (7a and 7b) are the *threo* (*cis*-opening product) and *erythro*- (*trans*-opening product) isomers, respectively.

Accordingly, the reaction of 1 with 5 in the presence of SnCl<sub>2</sub> proceeded predominantly by *cis*-opening, the *threo/erythro* ratio of the product being about 2. This result is noteworthy, since no previous report has appeared on the predominant *cis*-opening of glycidic esters with amines.<sup>4)</sup> When ZnI<sub>2</sub> was used as a catalyst, the reaction was accelerated dramatically to give a mixture of **6a**, **b** in quantitative yield. The *threo/erythro* ratio of this mixture, however, was not improved (Table I, entry 7).

During the course of this study, we noticed that 1 reacts slowly with 5 on the surface of silica gel. The preparative utility of the reaction assisted by silica gel was, therefore, examined. When a solution of an equimolar mixture of 1 and 5 in benzene was allowed to stand in a silica gel column (Kieselgel 60, 230—400 mesh, Merck) at 22 °C for 17 h, a mixture of 6a, b was

Nucleophile <sup>a)</sup>	Product	Total yield of the <i>threo</i> -and <i>erythro</i> -nitro esters (%)	threo/erythro ratio
2-Nitroaniline (5)	6	58.4	$0.5^{d}$
2-Nitrothiophenol	2a	15.0	erythro <sup>b)</sup>
2-Nitrophenol	<b>2b</b>	17.2	5.75 <sup>c)</sup>

Table II. Reaction of Methyl *trans*-3-(4-Methoxyphenyl)glycidate with Nucleophiles on the Surface of Silica Gel in a Chromatographic Column at 22 °C for 17 h

obtained in 58.4% yield after elution with benzene–AcOEt.<sup>6)</sup> From this mixture, the *erythro*nitro ester (**6b**) could be isolated as yellow needles, mp 140—143 °C, by trituration with isopropanol.<sup>7)</sup> The *threo/erythro* ratio of the whole mixture in this case was estimated to be about 0.5. Therefore, the reaction assisted by silica gel proceeded predominantly by *trans*-opening. Stirring of a mixture of 1 and 5 with silica gel in toluene at room temperature gave a similar result (see Experimental).

Posner and Rogers reported that neutral alumina catalyzes stereoselective *trans*-opening of epoxides by various nucleophiles under very mild conditions.<sup>8)</sup> In the reaction of 1 with 5, however, the catalytic effect of neutral alumina (Aluminiumoxide 90, activity II—III, Merck) was rather small (Table I, entry 9).

In seeking other applications for the nucleophilic opening of 1 assisted by silica gel, we examined the reaction of 1 with 2-nitrothiophenol and 2-nitrophenol. In both cases, however, neutral silica gel showed rather little catalytic effect (Table II). The relative reactivity of nucleophiles thus decreased in the following order,  $2\text{-NO}_2\text{C}_6\text{H}_4\text{NH}_2 > 2\text{-NO}_2\text{C}_6\text{H}_4\text{SH} = 2\text{-NO}_2\text{C}_6\text{H}_4\text{OH}$ . 2-Nitrophenol gave mainly the *threo*-isomer of **2b** under these conditions in accordance with its pronounced susceptibility to *cis*-opening.<sup>1)</sup>

Finally, the reaction of 1 with sodium 2-nitroanilide in dimethyl sulfoxide (DMSO) gave the epoxy anilide (8) as a sole product in 62.8% yield. In this reaction, no nucleophilic opening of the oxirane ring by amide anion was observed.

# Synthesis of Benzodiazepine Derivatives

Alkaline hydrolysis of the amino esters (7a and 7b) gave the corresponding amino carboxylic acids (9a and 9b). Thermal cyclization of 9a, b to the lactams (10a, b) gave some anomalous results. Heating of the threo-acid (9a) in boiling xylene in the presence of air afforded, after chromatographic separation, the cis- (10a) and trans-lactam (10b) in yields of 61.8 and 14.7%. The cis-lactam (10a) was recovered completely unchanged after heating in xylene in the presence of air. When the cyclization of 9a was carried out under an argon atmosphere, formation of the anomalous product (10b) was greatly diminished (2.3%), the cis-lactam (10a) being obtained in 76.4% yield. The erythro-amino carboxylic acid (9b) behaved similarly, and the formation of the cis-lactam (10a) was observed only in the presence of air. These results are summarized in Table III. As we reported previously, <sup>2a)</sup> the thia analogues of 9a, b cyclized to the corresponding lactams (1,5-benzothiazepin-4(5H)-one) without isomerization. On the basis of the above results, formation of the anomalous products (10a from 9b and 10b from 9a) might be explained in terms of oxidative epimerization at the benzylic position of 9a, b via a plausible intermediate, 11. Cyclization of 9a by the use of 1-hydroxybenzotriazole (HOBt) and dicyclohexylcarbodiimide (DCC) gave the cis-lactam (10a) as a sole product in 94.9% yield. Direct cyclization of the erythro-amino

a) 1 eq of nucleophile was used. b) No threo-isomer was detected. c) The threo/erythro ratio was determined by separation by preparative TLC. See reference 1. d) See Table I.

Starting material	Atmosphere	Reaction time (h)	Products (10)			
			cis-Lactam: trans-Lactam	Total yield		
9a	Air	30	4:1	76.5		
9a	Argon gas	17	33:1	78.7		
9b	Air	11	1:9	79.7		
9b	Argon gas	4	0:1	88.4		

TABLE III. Cyclization of the Amino Carboxylic Acid (9) in Boiling Xylene

10b 
$$Ac_{2}O-Py$$
 or  $AcCl-Py$   $OMe$   $12b: R^{1}=R^{2}=R^{3}=Ac$   $13b: R^{1}=R^{2}=Ac, R^{3}=H$   $14b: R^{2}=Ac, R^{1}=R^{3}=H$   $OMe$   $OMe$ 

ester (7b) with dimethylsulfinyl carbanion in DMSO gave the trans-lactam (10b) in 48.7% yield.

Chart 3

Acetylation of the lactam (10b) with  $Ac_2O$  and pyridine at  $100\,^{\circ}C$  for 2h gave the triacetate (12b), while the reaction at room temperature for 3h gave the  $N^5$ , O-diacetate (13b) and the O-acetate (14b) in 60.4 and 24.7% yields, respectively. Treatment of 10b with 2.1 eq of AcCl in pyridine gave 13b in 92% yield.

As an alternative route to the O-acetate (14b), introduction of the acetoxy group at an early stage of the synthesis was attempted. The nitro ester (6b) was hydrolyzed to the corresponding acid (15b). The hydroxyl group of 15b could be acetylated selectively with  $Ac_2O$  and pyridine, giving 16b in quantitative yield. Catalytic reduction of the nitro group of 16b, however, gave the benzimidazole derivative (17b) in 55% yield  $via O \rightarrow N$  migration of the acetyl group followed by cyclization. The spectral data of 17b given in Experimental are compatible with the assigned structure.

The cis- and trans-lactams (10a, b) were alkylated with 2-(dimethylamino)ethyl chloride to give the corresponding amines (18a, b). Heating of 18a with  $Ac_2O$  gave the diacetate (19a). The corresponding trans-isomer (19b) was prepared by N-alkylation of 13b. Methylation of 18a, b with HCHO and HCO<sub>2</sub>H gave the N-methyl derivatives (20a, b), which were converted into the O-acetates (21a, b) by heating with  $Ac_2O$ . The yields, physical data, and elemental analysis data of these products are summarized in Table IV.

#### **Pharmacology**

The 1,5-benzodiazepine derivatives (18—21) were tested for cerebral vasodilating activity in anesthetized dogs after intraarterial administration by the method reported previously.<sup>1)</sup>

TABLE IV. Physical Data for 1,5-Benzodiazepine Derivatives

Compound	Salt	mp (°C)	Recryst.	Yield (%)	Formula	Analysis (%) Calcd (Found)			
						C	Н	N	Cl
18a	HClO <sub>4</sub>	208209	Α	59.4	$C_{20}H_{26}CIN_3O_7$	52.69	5.75	9.22	7.78
19a	Oxalate	147149	В	61.7	$C_{26}H_{31}N_3O_9 \cdot H_2O$	57.03	5.72 6.07	9.21	7.75)
	O manato	117 115	Б	01.7	C <sub>26</sub> 113 <sub>1</sub> 11 <sub>3</sub> O <sub>9</sub> 11 <sub>2</sub> O	(56.87	5.53	7.67 7.57	— —)
20a	HClO <sub>4</sub>	229—230	C	69.2	$C_{21}H_{28}ClN_3O_7$	53.68	6.01	8.94	7.54
		(dec.)				(53.57	6.13	8.89	7.70)
21a	HClO <sub>4</sub>	162—164	D	73.4	$C_{23}H_{30}ClN_3O_8 \cdot 1/3H_2O$		5.90	8.11	6.84
101	11010	202 205	-			(53.34	5.83	8.13	7.02)
18b	HClO <sub>4</sub>	202—205	E	73.4	$C_{20}H_{26}CIN_3O_7$	52.69	5.75	9.22	7.78
101	***		_			(52.80	5.77	9.22	7.80)
19b	HC1	261—262	D	64.9	$C_{24}H_{30}ClN_3O_5$	60.57	6.35	8.83	7.45
•••	****	(dec.)	_			(60.24	6.31	8.82	7.30)
<b>20</b> b	HClO <sub>4</sub>	215—219	E	88.7	$C_{21}H_{28}ClN_3O_7$	53.68	6.01	8.94	7.59
						(53.53	6.01	8.78	7.43)
21b	Oxalate	152—155	F	93.0	$C_{25}H_{31}N_3O_8$	59.87	6.23	8.38	
						(59.84	6.13	8.33	—)

a)  $A = H_2O$ ,  $B = EtOH - Et_2O$ , C = DMF - EtOH, D = EtOH, E = acetone - EtOH,  $F = iso - PrOH - Et_2O$ .

The results are summarized in Table V together with comparative data for racemic diltiazem  $((\pm)-4a)$ .

Activity was found to reside only in the cis-isomers, and this stereoselectivity parallels the

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TABLE V. Cerebral Vasodilating Activity of 1,5-Benzodiazepine Derivatives

		- 1		Cerebral vasodilating activity <sup>a)</sup>		
Compound	Isomer	R <sup>1</sup>	R <sup>2</sup>	vs. papaverine <sup>b)</sup>	vs. (±)-4a <sup>c)</sup>	
18a <sup>d</sup> )	cis	Н	Н	0.20	0.05	
$19a^{e)}$	cis	Ac	Ac	0.07	***********	
$20a^{d}$	cis	Me	Н	2.28	0.40	
$21a^{d}$	cis	Me	Ac	2.92	0.50	
$18b^{d}$	trans	Н	Н	0.04		
<b>19b</b> <sup>f</sup> )	trans	Ac	Ac	0.1		
$20b^{d)}$	trans	Me	Н	0.06	_	
21b <sup>e)</sup>	trans	Me	Ac	0.16	0.04	
$(\pm)$ -4a	(racemic diltiazem)			4.30	1.0	

a) Determined by measuring the increase in blood flow in the vertebral artery in anesthetized dogs after i.a. administration. See reference 1. b) Potency ratio to papaverine. c) Potency ratio to  $(\pm)$ -4a. d) Perchlorate. e) Oxalate. f) Hydrochloride.

reported observations in the 1,5-benzothiazepine  $(4a)^{3d}$  and 1,5-benzoxazepine  $(4b)^{1}$  series. As for the effect of the R<sup>1</sup> group, methyl substitution was the most favorable, while introduction of an acetyl group at this position caused a marked fall in activity. On the other hand, as in the previous cases,  $^{1,3d}$  acetylation of the 3-OH group conferred increased activity. The highest activity in this series was thus observed with the *cis*-3-acetoxy-5-methyl derivative (21a). However, it was about one-half as active as racemic diltiazem  $((\pm)$ -4a). Thus, the sulfur atom of diltiazem (4a) could not be replaced by nitrogen or oxygen<sup>1)</sup> without a decrease in potency.

## **Experimental**

Infrared (IR) spectra were taken on a Hitachi IR-215 (Hitachi) or an Analect Instruments FX-6200 FTIR spectrophotometer. NMR spectra were recorded on a JEOL PMX-60 instrument. Chemical shifts are given as  $\delta$  values from tetramethylsilane as an internal standard. The following abbreviations are used; s=singlet, d=doublet, dd=doublet doublet, brs=broad singlet, t=triplet, q=quartet, and m=multiplet. MS were recorded on a Hitachi RMU-6D spectrometer. Preparative TLC was carried out on Kieselgel GF<sub>254</sub> (Merck). Kieselgel 60 (230—400 mesh) (Merck) was used for flash column chromatography. Aluminiumoxide 90 (activity II—III) (Merck) was used for column chromatography.

Reaction of Methyl trans-3-(4-Methoxyphenyl)glycidate (1) with 2-Nitroaniline (5)—a) The Reaction in the Presence of  $SnCl_2$  (Table I, Entry 6):  $SnCl_2$  (120 mg) was added to a solution of the glycidate (1) (4.98 g, 23.9 mmol) and 2-nitroaniline (5) (3.00 g, 21.7 mmol) in toluene (60 ml) at room temperature. After being stirred at 22 °C for 48 h, the reaction mixture was separated by column chromatography on  $Al_2O_3$ . After elution of the starting materials (1 and 5) with CHCl<sub>3</sub>-hexane (1:1), the eluate with CHCl<sub>3</sub>-EtOH (10:1) gave a mixture of the *threo*- and *erythro*-nitro esters (6a and 6b) as an oil (1.26 g, 16.8%).

This mixture (1.22 g, 3.52 mmol) was hydrogenated in the presence of 10% Pd–C (200 mg) in EtOH (30 ml) under ordinary pressure and temperature for 2.5 h. After removal of Pd–C and solvent, the resulting oil (1.008 g) was separated by preparative TLC (developed twice with CHCl<sub>3</sub>–EtOH (40:1)). From the fast-moving band, the *threo*-amino ester (7a) (496 mg, 40.7%), mp 91—92 °C (from iso-PrOH), was obtained. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3500, 3390, 1740, 1730, 1610. MS m/e: 316 (M<sup>+</sup>), 227 (M–CH(OH)CO<sub>2</sub>Me)<sup>+</sup>. NMR (CDCl<sub>3</sub>)  $\delta$ : 1.19 (3H, d, J=7 Hz, iso-PrOH),

3.6—4.2 (1/2H, m, iso-PrOH), 3.75 (6H, s, OCH<sub>3</sub> and CO<sub>2</sub>CH<sub>3</sub>), 4.48 (1H, d, J = 3 Hz, C<sub>2</sub>-H), 4.83 (1H, d, J = 3 Hz, C<sub>3</sub>-H), 6.3—7.4 (8H, m, aromatic H). *Anal.* Calcd for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>O<sub>4</sub>·1/2 iso-PrOH: C, 64.05; H 7.12; N, 8.07. Found: C, 64.13; H, 6.93; N, 7.95.

The *erythro*-amino ester (**7b**) (242 mg, 21.7%), mp 120—122 °C (from iso-PrOH–iso-Pr<sub>2</sub>O), was obtained from the slower-moving band. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3420, 3380, 1740, 1620. MS m/e: 316 (M<sup>+</sup>), 227 (M—CH(OH)CO<sub>2</sub>Me)<sup>+</sup>. NMR (CDCl<sub>3</sub>)  $\delta$ : 3.72 (3H, s, OCH<sub>3</sub>), 3.76 (3H, s, CO<sub>2</sub>CH<sub>3</sub>), 4.1—4.4 (2H, m, C<sub>2</sub>-H and C<sub>3</sub>-H). This signal changed to a double doublet on addition of D<sub>2</sub>O as follows: 4.62 (1H, d, J = 4 Hz, C<sub>2</sub>-H) and 4.76 (1H, d, J = 4 Hz, C<sub>3</sub>-H), 6.5—7.4 (8H, m, aromatic H). *Anal*. Calcd for C<sub>17</sub>H<sub>20</sub>N<sub>2</sub>O<sub>4</sub>: C, 64.54; H, 6.34; N, 8.86. Found: C, 64.43; H, 6.34; N, 8.82.

Therefore, the threo/erythro ratio of the nitro esters was estimated to be 40.7/21.7 = ca. 2.0.

The ZnI<sub>2</sub>-catalyzed reaction (Table I, Entry 7) was carried out essentially in the same manner.

b) The Reaction in a Silica Gel Chromatographic Column (Table I, Entry 8): A solution of 1 (15.0 g, 72.0 mmol) and 5 (10.0 g, 72.4 mmol) in benzene (100 ml) was poured into a column of silica gel (Kieselgel 60, 230—400 mesh for flash column chromatography, Merck, 400 g). After standing at 22 °C for 17 h, the starting materials were eluted with benzene, and then a mixture of the *threo*- and *erythro*-nitro esters (**6a** and **6b**) (14.53 g, 58.4%) was obtained as an oil by elution with benzene–AcOEt (6:1). This mixture was triturated with iso-PrOH to give the crystalline *erythro*-nitro ester (**6b**). Recrystallization from iso-PrOH gave pure **6b** (8.93 g, 35.9%), mp 140—143 °C as yellow needles. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3450, 3380, 1755, 1740, 1615, 1510, 1360. MS m/e: (M+ was not seen), 257 (M – CH(OH)CO<sub>2</sub>Me)+, 224, 210. NMR (CDCl<sub>3</sub>)  $\delta$ : 2.85 (1H, d, J = 7.5 Hz, OH), 3.75 (6H, s, OCH<sub>3</sub> and CO<sub>2</sub>CH<sub>3</sub>), 4.67 (1H, dd, J = 4, 7.5 Hz, C<sub>2</sub>-H), 4.95 (1H, dd, J = 4, 7.5 Hz, C<sub>3</sub>-H), 6.4—7.5 (7H, m, aromatic H), 8.15 (1H, dd, J = 2, 8 Hz, aromatic H), 9.00 (1H, d, J = 7.5 Hz, NH). *Anal.* Calcd for C<sub>17</sub>H<sub>18</sub>N<sub>2</sub>O<sub>6</sub>: C, 58.96; H, 5.24; N, 8.09. Found: C, 59.02; H, 5.20; N, 8.11.

Hydrogenation of **6b** (1.45 g, 4.19 mmol) over 10% Pd–C in EtOH (30 ml) gave the *erythro*-amino ester (**7b**) in 73.5% yield; this product was identical with the sample previously isolated.

A small portion of the whole mixture of the *threo*- and *erythro*-nitro esters (**6a** and **6b**) obtained above was hydrogenated similarly. The *threo/erythro* ratio of the resulting mixture (**7a/7b**) was determined to be 0.5.

- c) The Reaction in the Presence of a Slurry of Silica Gel: A mixture of 1 (830 mg, 3.99 mmol), 5 (500 mg, 3.62 mmol), silica gel (Kieselgel 60 for flash chromatography, 5.5 g), and toluene (20 ml) was stirred at room temperature for 17 h. The silica gel was removed by filtration and washed with AcOEt. The filtrates and washing were concentrated, and the residual oil was separated by flash column chromatography on silica gel. After elution of the starting material with benzene, a mixture of the *threo* and *erythro*-nitro esters (6a and 6b) (605 mg, 48.2%) was obtained by elution with benzene–AcOEt (7:1); the *threo*/*erythro* ratio was estimated to be 0.5.
- d) The Reaction on Alumina (Table I, Entry 9): A solution of 1 (1.5 g, 7.20 mmol) and 5 (1.0 g, 7.24 mmol) in benzene (10 ml) was poured onto a column of Aluminiumoxide 90 (activity II—III, Merck, 60 g). After standing at room temperature for 17 h, the starting materials were washed out with benzene, and then a mixture of 6a and 6b (446 mg, 17.9% as an oil) was eluted with benzene—EtOAc (1:1); the threo/erythro ratio was estimated to be 0.3.
- e) The Reaction in the Presence of NaH in DMSO (Table I, Entry 10): A solution of 1 (1.58 g, 7.59 mmol) was added to a solution of 5 (1.0 g, 7.24 mmol) and NaH (63% dispersion in mineral oil, 276 mg, 7.24 mmol) in DMSO (5 ml) under an argon atmosphere during a period of 30 min. The reaction mixture was poured into ice-water, and the precipitated crystals were collected on a filter, washed with water, dried, and recrystallized from CHCl<sub>3</sub> to give *trans*-3-(4-methoxyphenyl)-2'-nitroglycidanilide (8) (1.43 g, 62.8%), mp 165—167 °C. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3300, 1700, 1510, 1340. MS m/e: 314 (M<sup>+</sup>), 176, 148, 121. NMR (DMSO- $d_6$ )  $\delta$ : 3.80 (3H, s, OCH<sub>3</sub>), 3.97 and 4.26 (2H, AB system dd, J = 2 Hz, C<sub>2</sub>-H and C<sub>3</sub>-H), 6.9—8.3 (8H, m, aromatic H), 10.6 (1H, br s, NH). *Anal*. Calcd for C<sub>16</sub>H<sub>14</sub>N<sub>2</sub>O<sub>5</sub>: C, 61.14; H, 4.49; N, 8.91. Found: C, 61.04; H, 4.37; N, 8.92.

Reaction of 1 with 2-Nitrothiophenol on Silica Gel (Table II)——A solution of 1 (1.3 g, 6.24 mmol) and 2-nitrothiophenol (1.0 g, 6.44 mmol) in benzene (10 ml) was poured onto the column of silica gel (for flash column chromatography, 60 g). After standing at room temperature for 17 h, the reaction mixture was worked up in the same manner as described for the reaction of 1 with 5.

The erythro-nitro ester (erythro-isomer of 2a) (340 mg, 15.0%), mp 135—136.5 °C, was obtained by elution with benzene-AcOEt (5:1), and was identical with an authentic sample.<sup>3a)</sup>

Reaction of 1 with 2-Nitrophenol on Silica Gel (Table II)—A solution of 1 (1.5 g, 7.20 mmol) and 2-nitrophenol (1.0 g, 7.19 mmol) in benzene (10 ml) was allowed to react in a column of silica gel (Kieselgel 60 for flash column chromatography, 60 g). After standing at room temperature for 17 h, the starting materials were washed out with benzene, and then a mixture of the *threo*- and *erythro*-nitro esters (2b) (430 mg, 17.2%, as an oil) was eluted with benzene–AcOEt (5:1). The *threo/erythro* ratio of the mixture was 5.75, which was determined by separation by preparative TLC as reported in our previous paper.<sup>1)</sup>

threo-3-(2-Aminoanilino)-2-hydroxy-3-(4-methoxyphenyl)propionic Acid (9a)—A solution of the threo-amino ester (7a) (2.83 g, 8.17 mmol) in a mixture of 5% aqueous NaOH (10 ml) and EtOH (10 ml) was stirred at room temperature for 2 h, then neutralized with 10% HCl. The precipitated crystals were filtered off, washed with water, dried, and recrystallized from N,N-dimethyl formamide (DMF)–EtOH to give 9a (1.995 g, 74.1%), mp 150—155 °C (dec.). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3350, 1650, 1610. MS m/e: 302 (M<sup>+</sup>), 227 (M – CH(OH)CO<sub>2</sub>H)<sup>+</sup>. NMR (DMSO- $d_6$ )  $\delta$ : 3.72 (3H, s, OCH<sub>3</sub>), 4.15 (1H, d, J=3 Hz, C<sub>2</sub>-H), 4.75 (1H, d, J=3 Hz, C<sub>3</sub>-H), 6.1—7.4 (8H, m, aromatic H). Anal. Calcd

for C<sub>16</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>·1/2H<sub>2</sub>O·1/4DMF: C, 61.04; H, 6.34; N, 9.56. Found: C, 61.31; H, 6.01; N, 9.33.

erythro-3-(2-Aminoanilino)-2-hydroxy-3-(4-methoxyphenyl)propionic Acid (9b) — The erythro-amino ester (7b) was hydrolyzed in the same manner as described for 9a. The acid (9b), mp 150—153 °C (dec.), was obtained in 67.1% yield after recrystallization from DMF-iso-PrOH-H<sub>2</sub>O. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3340, 3300, 1640, 1610. MS m/e: 302 (M<sup>+</sup>), 227 (M – CH(OH)CO<sub>2</sub>H)<sup>+</sup>. NMR (DMSO-d<sub>6</sub>) δ: 3.72 (3H, s, OCH<sub>3</sub>), 4.27 (1H, d, J = 6 Hz, C<sub>2</sub>-H), 4.64 (1H, d, J = 6 Hz, C<sub>3</sub>-H), 6.3—7.5 (8H, m, aromatic H). Anal. Calcd for C<sub>16</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>·1/2H<sub>2</sub>O: C, 61.72; H, 6.15; N, 9.00. Found: C, 61.60; H, 6.26; N, 8.94.

Cyclization of the Amino Carboxylic Acid (9)—a) The Reaction in Boiling Xylene in the Presence of Air: The *threo*-amino carboxylic acid (9a) (1.5 g, 4.55 mmol) was heated in boiling xylene (20 ml) for 30 h. After removal of the solvent, the residue was separated by flash column chromatography. *cis*-3-Hydroxy-4-(4-methoxyphenyl)-1,3,4,5-tetrahydro-2*H*-1,5-benzodiazepin-2-one (10a) (800 mg, 61.8%), mp 145—148 °C (from EtOH-iso-Pr<sub>2</sub>O), was obtained from the first eluate with benzene-AcOEt (7:1). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3370, 3330, 3240, 1700, 1680. MS m/e; 284 (M<sup>+</sup>), 237. NMR (CDCl<sub>3</sub>)  $\delta$ : 3.42 (1H, d, J=6Hz, OH), 3.74 (3H, s, OCH<sub>3</sub>), 4.16 (1H, d, J=5Hz, NH), 4.63 (1H, dd, J=5, and 6Hz, C<sub>3</sub>-H), 5.07 (1H, t, J=5Hz, C<sub>4</sub>-H), 6.7—7.5 (8H, m, aromatic H). *Anal.* Calcd for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>: C, 67.59; H, 5.67; N, 9.85. Found: C, 67.60; H, 5.62; N, 9.80.

From the second eluate with the same solvent, the *trans*-lactam (10b) (190 mg, 14.7%), mp 188—191.5 °C (EtOH), was obtained. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3490, 3420, 3360, 3340, 1670. MS m/e: 284 (M<sup>+</sup>), 237. NMR (DMSO- $d_6$ )  $\delta$ : 3.74 (3H, s, OCH<sub>3</sub>), 4.06 (1H, dd, J=6, 9 Hz, C<sub>3</sub>-H), 4.43 (1H, d, J=9 Hz, C<sub>4</sub>-H), 4.77 (1H, d, J=6 Hz, OH), 5.38 (1H, s, NH), 6.7—7.4 (8H, m, aromatic H). *Anal.* Calcd for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>: C, 67.59; H, 5.67; N, 9.85. Found: C, 67.68; H, 5.61; N, 9.84.

The cis-lactam (10a) was completely unchanged after heating in boiling xylene for 30 h in the presence of air. Similarly, the erythro-amino carboxylic acid (9b) (500 mg, 1.61 mmol) in xylene (10 ml) was heated and worked up to give the trans-lactam (10b) (329 mg, 71.8%) and the cis-lactam (10a) (36 mg, 7.9%).

b) The Reaction in Boiling Xylene under an Argon Atmosphere: The *threo*-amino carboxylic acid (9a) (500 mg, 1.52 mmol) was heated in boiling xylene (10 ml) under an argon atmosphere for 17 h. The reaction mixture was worked up in the same manner as described above to give the *cis*-lactam (10a) (330 mg, 76.4%) and the *trans*-lactam (10 mg, 2.3%), and some starting material (9a, 93 mg, 18.6%) was recovered.

Similarly, the *erythro*-amino carboxylic acid (9b) was heated under an argon atmosphere in the same manner as described above to give the *trans*-lactam (10b) in 88.4% yield as a sole product.

c) The Cyclization with HOBt-DCC: DCC (464 mg, 2.25 mmol) was added to a mixture of the *threo*-amino carboxylic acid (9a) (500 mg, 1.52 mmol) and HOBt hydrate (50 mg, 0.32 mmol) in DMF (10 ml) at room temperature. The reaction mixture was stirred at the same temperature for 17 h and concentrated under reduced pressure. Water and AcOEt were added to the residue, and the mixture was stirred at room temperature for 1 h to decompose excess DCC. The precipitated white crystals of dicyclohexyl urea were filtered off. The AcOEt solution was separated and the water layer was extracted with AcOEt. The combined AcOEt solutions were washed with sat. NaHCO<sub>3</sub> to remove HOBt and water, then dried, and concentrated. The residual solid was recrystallized from EtOH-iso-Pr<sub>2</sub>O to give the *cis*-lactam (10a) (410 mg, 94.9%), mp 145—148 °C as a sole product.

Cyclization of the erythro-Amino Ester (7b) with NaH-DMSO——A solution of 7b (3.0 g, 8.66 mmol) in DMSO (9 ml) was added to a solution of dimethylsulfinyl carbanion in DMSO, prepared from 63% NaH (760 mg, 19.9 mmol, dispersion in mineral oil) and DMSO (15 ml), under ice-cooling. The reaction mixture was warmed to 30 °C, stirred at 30 °C for 20 min, and poured into ice-water. The mixture was acidified with HOAc and extracted with CHCl<sub>3</sub>. The extracts were washed with water, dried, and evaporated. The residual solid was recrystallized from EtOH to give the trans-lactam (10b) (1.2 g, 48.7%), mp 188—191.5 °C.

Acetylation of the *trans*-Lactam (10b)—a) With Ac<sub>2</sub>O-Pyridine at 100 °C: A mixture of 10b (1.0 g, 3.52 mmol), pyridine (1 ml), and Ac<sub>2</sub>O (10 ml) was heated at 100 °C for 2 h, then concentrated under reduced pressure. The residual solid was recrystallized from EtOH to give the triacetate (12b), mp 153—155 °C (1.086 g, 73.6%). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1780, 1745, 1730, 1670, 1645, 1610. MS m/e: 410 (M<sup>+</sup>), 350, 308. NMR (CDCl<sub>3</sub>) δ: 1.80 (3H, s, NCOCH<sub>3</sub>), 2.00 (3H, s, OCOCH<sub>3</sub>), 2.74 (3H, s, NCOCH<sub>3</sub>), 5.20 (1H, d, J=11 Hz, C<sub>3</sub>-H), 6.15 (1H, d, J=11 Hz, C<sub>4</sub>-H), 6.7—7.6 (8H, m, aromatic H). *Anal*. Calcd for C<sub>22</sub>H<sub>22</sub>N<sub>2</sub>O<sub>6</sub>·1/2H<sub>2</sub>O: C, 63.00; H, 5.53; N, 6.68. Found: C, 63.69; H, 5.86; N, 6.57.

b) With Ac<sub>2</sub>O–Pyridine at Room Temperature: A mixture of **10b** (500 mg, 1.76 mmol), Ac<sub>2</sub>O (5 ml), and pyridine (0.5 ml) was stirred at room temperature for 3 h, then poured into ice-water, neutralized with NaHCO<sub>3</sub>, and extracted with EtOAc. The extracts were washed with water, dried, and concentrated. The residual oil was separated by preparative TLC (developed with CHCl<sub>3</sub>–EtOAc (3:1)). The monoacetate (**14b**) (162 mg, 24.7%), mp 188—191 °C (from EtOH), was obtained from the faster-moving band. IR  $v_{\rm max}^{\rm Nujol}$  cm<sup>-1</sup>: 3340, 1760, 1675, 1610. MS m/e: 326 (M<sup>+</sup>), 283, 266. NMR (DMSO- $d_6$ )  $\delta$ : 1.91 (3H, s, OCOCH<sub>3</sub>), 3.78 (3H, s, OCH<sub>3</sub>), 4.83 (1H, d, J=10.5 Hz, C<sub>4</sub>-H), 5.38 (1H, d, J=10.5 Hz, C<sub>3</sub>-H), 6.3—7.4 (8H, m, aromatic H), 8.36 (1H, s, NH). *Anal*. Calcd for C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>·EtOH: C, 64.50; H, 6.50; N, 7.52. Found: C, 64.56; H, 6.42; N, 7.57.

The diacetate (13b) (440 mg, 60.4%), mp 229—240 °C (from EtOH), was obtained from the slower-moving band. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3340, 1750, 1700, 1635, 1610. MS m/e: 368 (M<sup>+</sup>), 308, 283, 265. NMR (CDCl<sub>3</sub>)  $\delta$ : 1.71 (3H, s,

NCOCH<sub>3</sub>), 1.97 (3H, s, OCOCH<sub>3</sub>), 3.76 (3H, s, OCH<sub>3</sub>), 5.44 (1H, d, J=11 Hz, C<sub>3</sub>-H), 6.25 (1H, d, J=11 Hz, C<sub>4</sub>-H), 6.7—7.5 (8H, m, aromatic H), 8.66 (1H, s, NH). *Anal.* Calcd for C<sub>20</sub>H<sub>20</sub>N<sub>2</sub>O<sub>5</sub> EtOH: C, 63.75; H, 6.32; N, 6.76. Found: C, 63.78; H, 6.07; N, 6.76.

c) With AcCl-Pyridine: A solution of AcCl (290 mg, 3.69 mmol) in DMF (2 ml) was added to a solution of **10b** (500 mg, 1.76 mmol) in DMF (8 ml) and pyridine (417 mg, 5.27 mmol) at room temperature. The reaction mixture was stirred at room temperature for 1 h, poured into ice-water, and extracted with EtOAc. The extracts were washed with sat. aqueous NaCl, dried, and evaporated to give the diacetate (**13b**), mp 229—240 °C, 670 mg (91.9%).

erythro-2-Hydroxy-3-(4-methoxyphenyl)-3-(2-nitroanilino)propionic Acid (15b)—A solution of the erythronitro ester (6b) (10.0 g, 28.9 mmol) in 5% aqueous NaOH (30 ml) and EtOH (10 ml) was stirred at room temperature for 3 h, then neutralized with dil. HCl and extracted with EtOAc. The extracts were washed with water, dried, and evaporated. The residue was recrystallized from iso-PrOH to give 15b (7.64 g, 67.4%), mp 100—134 °C (dec.). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3370, 1730, 1710, 1610, 1510, 1350. MS m/e: (M<sup>+</sup> was not seen), 257 (M – CH(OH)CO<sub>2</sub>H)<sup>+</sup>. NMR (CDCl<sub>3</sub>–DMSO- $d_6$ ) δ: 3.73 (3H, s, OCH<sub>3</sub>), 4.63 (1H, d, J=3 Hz, C<sub>2</sub>-H), 5.00 (1H, dd, J=3, 8 Hz, C<sub>3</sub>-H), 6.4—7.5 (7H, m, aromatic H), 8.12 (1H, dd, J=2, 8 Hz, aromatic H), 9.00 (1H, d, J=8 Hz, NH). Anal. Calcd for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>6</sub>· iso-PrOH: C, 58.15; H, 6.16; N, 7.14. Found: C, 58.06; H, 6.11; N, 7.13.

erythro-2-Acetoxy-3-(4-methoxyphenyl)-3-(2-nitroanilino)propionic Acid (16b) — The nitro carboxylic acid (15b) (5.0 g, 12.7 mmol) was heated in Ac<sub>2</sub>O (10 ml) and pyridine (0.1 mol) on a boiling water bath for 2 h, then poured into a mixture of ice and sat. NaHCO<sub>3</sub>, and the whole was extracted with CHCl<sub>3</sub>. The extracts were washed with water, dried, and evaporated to give crude 16b (5.2 g, quantitative yield) as an oil. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3020, 1755, 1620. MS m/e: (M<sup>+</sup> was not seen), 257 (M – CH(OAc)CO<sub>2</sub>Me)<sup>+</sup>. NMR (DMSO- $d_6$ )  $\delta$ : 2.16 (3H, s, OCOCH<sub>3</sub>), 3.76 (3H, s, OCH<sub>3</sub>), 5.33 and 5.41 (2H, AB system dd, J=4 Hz, C<sub>2</sub>-H and C<sub>3</sub>-H), 6.6—7.6 (7H, m, aromatic H), 8.12 (1H, dd, J=2, 8 Hz aromatic H).

Hydrogenation of 16b—The crude acetoxy compound (16b) obtained above was hydrogenated in EtOH (30 ml) in the presence of 10% Pd–C (500 mg) under ordinary pressure and temperature for 18 h. After removal of the Pd–C and solvent, the residue was triturated with EtOAc to give *erythro*-2-hydroxy-3-(2-methyl-1*H*-benzimidazol-1-yl)-3-(4-methoxyphenyl)propionic acid (17b) (2.35 g, 55.0%) as colorless prisms of mp 211—212 °C (from DMF–EtOH). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3340, 1660, 1610. MS m/e: 326 (M<sup>+</sup>), 251 (M–CH(OH)CO<sub>2</sub>H)<sup>+</sup>, 131, 121. NMR (DMSO- $d_6$ ) δ: 2.56 (3H, s, CH<sub>3</sub>), 3.73 (3H, s, OCH<sub>3</sub>), 5.04 (1H, d, J=7.5 Hz, C<sub>2</sub>-H), 5.85 (1H,d, J=7.5 Hz, C<sub>3</sub>-H), 6.8—7.7 (8H, m, aromatic H). *Anal.* Calcd for C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>·1/2H<sub>2</sub>O: C, 64.47; H, 5.71; N, 8.35. Found: C, 64.49; H, 5.56; N, 8.42.

cis-1-[2-(Dimethylamino)ethyl]-3-hydroxy-4-(4-methoxyphenyl)-1,3,4,5-tetrahydro-2H-1,5-benzodiazepin-2-one (18a)—A mixture of the cis-lactam (10a) (510 mg, 1.79 mmol), 2-(dimethylamino)ethyl chloride hydrochloride (285 mg, 1.98 mmol), and  $K_2CO_3$  (powdered, 571 mg, 4.13 mmol) in acetone (30 ml) was heated under reflux for 17 h, then allowed to cool. Inorganic compounds were removed by filtration. The filtrates were evaporated under reduced pressure, and the residue was converted into the perchlorate, which was recrystallized from water to give 18a perchlorate (485 mg, 59.4%), mp 208—209 °C. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3300, 1650. MS m/e: 355 (M<sup>+</sup>), 284, 266. NMR (DMSO- $d_6$ )  $\delta$ : 2.80 (6H, s, NCH<sub>3</sub>), 3.0—4.2 (4H, m, NCH<sub>2</sub>), 3.70 (3H, s, OCH<sub>3</sub>), 4.35 (1H, d, J=6 Hz,  $C_3$ -H), 4.80 (1H, d, J=5, 6 Hz,  $C_4$ -H), 5.64 (1H, d, J=5 Hz, NH), 6.75 and 7.35 (4H, AB system dd, J=8 Hz, p-substituted aromatic H), 7.15 (4H, br s, aromatic H).

The *trans*-isomer (18b) perchlorate, mp 202—205 °C, was obtained by alkylation of 10b in the same manner as described above in 73.4% yield. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3430, 3270, 1670. MS m/e: 355 (M<sup>+</sup>), 284, 266. NMR (DMSO- $d_6$ )  $\delta$ : 2.85 (6H, s, NCH<sub>3</sub>), 3.0—4.2 (4H, m, NCH<sub>2</sub>-), 3.70 (3H, s, OCH<sub>3</sub>), 4.14 (1H, d,  $J=10\,\text{Hz}$ , C<sub>3</sub>-H), 4.38 (1H, d,  $J=10\,\text{Hz}$ , C<sub>4</sub>-H), 5.32 (1H, s, NH), 6.7—7.4 (8H, m, aromatic H).

cis-3-Acetoxy-5-acetyl-1-[2-(dimethylamino)ethyl]-4-(4-methoxyphenyl)-1,3,4,5-tetrahydro-2*H*-1,5-benzodiazepin-2-one (19a)—A mixture of 18a (free base, 525 mg, 1.48 mmol), Ac<sub>2</sub>O (5 ml), and pyridine (0.1 ml) was heated at 100 °C for 2 h. After removal of the Ac<sub>2</sub>O, HOAc, and pyridine, the residue was converted into the oxalate, which was recrystallized from EtOH–Et<sub>2</sub>O to give 19a oxalate (500 mg, 61.7%), mp 147—149 °C. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1750, 1660. MS m/e: 439 (M<sup>+</sup>), 379. NMR (DMSO- $d_6$ )  $\delta$ : 1.42 (3H, s, NCOCH<sub>3</sub>), 1.83 (3H, s, OCOCH<sub>3</sub>), 2.66 (6H, s, NCH<sub>3</sub>), 2.8—4.3 (4H, m, NCH<sub>2</sub>), 5.28 (1H, d, J=4Hz, C<sub>3</sub>-H), 6.28 (1H, d, J=4Hz, C<sub>4</sub>-H), 6.7—7.7 (8H, m, aromatic H).

trans:-3-Acetoxy-5-acetyl-1-[2-(dimethylamino)ethyl]-4-(4-methoxyphenyl)-1,3,4,5-tetrahydro-2H-1,5-benzodiazepin-2-one (19b)—A mixture of the trans-diacetyl lactam (13b) (670 mg, 1.62 mmol), 2-(dimethylamino)ethyl chloride hydrochloride (314 mg, 2.18 mmol), and  $K_2CO_3$  (powdered, 628 mg, 4.55 mmol) in acetone (40 ml) was heated under reflux for 17 h, then allowed to cool. Inorganic compounds were filtered off, the filtrates were concentrated and the residue was converted into the hydrochloride. After recrystallization from EtOH, 19b·hydrochloride (500 mg, 64.9%), mp 261—262 °C (dec.), was obtained. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1760, 1690, 1670. MS m/e: 439 (M<sup>+</sup>), 379. NMR (CDCl<sub>3</sub>)  $\delta$ : 1.72 (3H, s, NCOCH<sub>3</sub>), 1.92 (3H, s, OCOCH<sub>3</sub>), 2.80 (6H, s, NCH<sub>3</sub>), 3.1—4.5 (4H, m, -CH<sub>2</sub>N), 3.73 (3H, s, OCH<sub>3</sub>), 5.12 (1H, d, J=11 Hz,  $C_3$ -H), 5.95 (1H, d, J=11 Hz,  $C_4$ -H), 6.7—8.0 (8H, m, aromatic H).

cis-1-[2-(Dimethylamino)ethyl]-3-hydroxy-4-(4-methoxyphenyl)-5-methyl-1,3,4,5-tetrahydro-2H-1,5-

benzodiazepin-2-one (20a)——A mixture of 18a (free base, 2.455 g, 6.91 mmol), 37% HCHO (4 ml), and HCOOH (20 ml) was heated at 50 °C for 3 h and concentrated under reduced pressure. Sat. aqueous NaHCO<sub>3</sub> was added to the residue, and the mixture was extracted with EtOAc. The extracts were washed with sat. NaCl, dried, and concentrated. The residue was dissolved in 5% aqueous NaOH (10 ml) and EtOH (10 ml) and the solution was stirred at room temperature for 1 h to hydrolyze the 3-formyloxy group, then neutralized with 10% HCl, and extracted with EtOAc. The extracts were washed with water, dried, and evaporated to give an oil which was converted to the perchlorate. Recrystallization from DMF–EtOH gave 20a perchlorate (2.247 g, 69.2%), mp 229—230 °C (dec.). IR  $\nu_{\rm max}^{\rm Nujol}$  cm<sup>-1</sup>: 3400, 1650, 1610. MS m/e: 369 (M<sup>+</sup>), 340, 280. NMR (DMSO- $d_6$ )  $\delta$ : 2.55 (3H, s, NCH<sub>3</sub>) (free base in CDCl<sub>3</sub>, 2.62 (3H, s)), 2.88 (6H, s, NCH<sub>3</sub>), 3.0—4.4 (4H, m, NCH<sub>2</sub>), 3.78 (3H, s, OCH<sub>3</sub>), 4.27 (2H, s, C<sub>3</sub>- and C<sub>4</sub>-H), 6.8—7.5 (8H, m, aromatic H).

The *trans*-isomer (**20b**), mp 215—219 °C (perchlorate), was prepared similarly (Table IV). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3470, 1680, 1610. MS m/e: 369 (M<sup>+</sup>), 340, 280. NMR (DMSO- $d_6$ )  $\delta$ : 2.55 (3H, s, NCH<sub>3</sub>) (free base in CDCl<sub>3</sub>, 2.29 (3H, s)), 2.85 (6H, s, NCH<sub>3</sub>), 3.0—4.3 (6H, m, NCH<sub>2</sub>, C<sub>3</sub>- and C<sub>4</sub>-H), 3.74 (3H, s, OCH<sub>3</sub>), 6.85—7.5 (8H, m, aromatic H).

cis-3-Acetoxy-1-[2-(dimethylamino)ethyl]-4-(4-methoxyphenyl)-5-methyl-1,3,4,5-tetrahydro-2*H*-1,5-benzodiazepin-2-one (21a)——A mixture of 20a (free base, 710 mg, 1.92 mmol), Ac<sub>2</sub>O (5 ml), and pyridine (0.2 ml) was heated at 100 °C for 2 h, and then Ac<sub>2</sub>O, AcOH, and pyridine were removed. The residue was dissolved in EtOH and converted into the perchlorate, which was recrystallized from EtOH to give 21a perchlorate (730 mg, 73.4%), mp 162—164 °C. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1740, 1680, 1610. MS m/e: 411 (M<sup>+</sup>), 340, 280. NMR (DMSO- $d_6$ ) δ: 1.81 (3H, s, OCOCH<sub>3</sub>), 2.48 (3H, s, NCH<sub>3</sub>), 2.81 (6H, s, N-CH<sub>3</sub>), 3.2—4.3 (4H, m, NCH<sub>2</sub>), 3.76 (3H, s, OCH<sub>3</sub>), 4.44 (1H, d, J=6 Hz, C<sub>4</sub>-H), 5.20 (1H, d, J=6 Hz, C<sub>3</sub>-H), 6.8—7.5 (8H, m, aromatic H).

The *trans*-isomer (21b), mp 152—155 °C (oxalate), was prepared similarly (Table IV). IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1750, 1690, 1610. MS m/e: 411 (M<sup>+</sup>), 340, 280. NMR (D<sub>2</sub>O)  $\delta$ : 1.94 (3H, s, OCOCH<sub>3</sub>), 2.53 (3H, s, NCH<sub>3</sub>), 2.96 (6H, s, NCH<sub>3</sub>), 3.2—4.2 (4H, m, NCH<sub>2</sub>), 3.73 (3H, s, OCH<sub>3</sub>), 4.70 (1H, d, J=10 Hz, C<sub>4</sub>-H), 5.32 (1H, d, J=10 Hz, C<sub>3</sub>-H), 6.8—7.7 (8H, m, aromatic H).

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## References and Notes

- 1) Part IV: T. Hashiyama, A. Watanabe, H. Inoue, M. Konda, M. Takeda, S. Murata, and T. Nagao, *Chem. Pharm. Bull.*, 33, 634 (1985).
- 2) a) T. Hashiyama, H. Inoue, M. Konda, and M. Takeda, J. Chem. Soc., Perkin Trans. 1, 1984, 1725; b) T. Hashiyama, H. Inoue, and M. Takeda, ibid., 1985, 421; c) T. Hashiyama, H. Inoue, M. Konda, and M. Takeda, Chem. Pharm. Bull., 33, 1256 (1985).
- 3) a) H. Kugita, H. Inoue, M. Ikezaki, and S. Takeo, Chem. Pharm. Bull., 18, 2028 (1970); b) H. Kugita, H. Inoue, M. Ikezaki, M. Konda, and S. Takeo, ibid., 18, 2284 (1970); c) H. Inoue, S. Takeo, M. Kawazu, and H. Kugita, Yakugaku Zasshi, 93, 729 (1973); d) T. Nagao, M. Sato, H. Nakajima, and A. Kiyomoto, Chem. Pharm. Bull., 21, 92 (1973).
- 4) a) A. Elker, J. Lehmann, and F. Zymalkowski, Arch. Pharm., 312, 26 (1979); b) E. Kamandi, A. W. Frahm, and F. Zymalkowski, ibid., 307, 871 (1974); c) Idem, ibid., 308, 135 (1975); d) W. Sucrow, M. Slopianka, and H. J. Vetter, Chem. Ber., 111, 791 (1978); e) E. C. Taylor, C. A. Maryanoff, and J. S. Skotnicki, J. Org. Chem., 45, 2512 (1980); f) C. Sabate-Alduy, J. Bastide, P. Berqot, and J. Lematre, Bull. Soc. Chim., Fr., 1974, 1942.
- 5) This fragment pattern is quite similar to those observed with the thia and oxa analogues (2a, b) and rules out reversed positions for the amino and hydroxyl groups. See references 1 and 2b.
- 6) During such a short period as that required for separation by column chromatography on silica gel or alumina, no appreciable reaction of 1 with 5 was observed. This rules out significant changes of the yields and isomer ratios of the product during work-up in entries 6 and 7.
- 7) The threo-nitro ester could not be isolated in a pure state.
- 8) a) G. H. Posner and D. Z. Rogers, J. Am. Chem. Soc., 99, 8208 and 8214 (1977); b) G. H. Posner and J. R. Lever, J. Org. Chem., 49, 2029 (1984).