

Communications to the Editor

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REGIOSELECTIVE MONO-OXIDATION OF NON-PROTECTED GLYCOSIDES
BY THE BIS-TRIBUTYL TIN OXIDE-BROMINE METHOD¹⁾

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Treatment of methyl glycosides with an excess of bis-tributyltin oxide and molecular sieves 3A in chloroform followed by brominolysis resulted in mono-oxidation to yield oxo-glycosides regioselectively. Thus Me β -D-Glc and Me β -D-Xyl gave the 3-oxo derivatives in more than 90% yield, and Me α -D-Glc, Me α -D-Gal, Me α -D-Xyl, and Me β -L-Ara gave the 4-oxo derivatives in 70-80% yield.

KEYWORDS—methyl glycoside; oxidation; brominolysis; bis-tributyltin oxide; oxo-glycoside; glycopyranosid-3-ulose; glycopyranosid-4-ulose

The O-Sn linkage formed from hydroxyl groups by the action of bis-tributyltin oxide or dibutyltin oxide is very sensitive to brominolysis and yields a carbonyl compound at the speed of titration, thus providing mild oxidation of a hydroxyl group to the corresponding carbonyl compound.²⁾ This method, however, is seldom applied to compounds carrying more than three hydroxyl groups. Only two such examples are reported,^{3,4)} in which dibutyltin oxide is used to regioselectively activate a particular hydroxyl group, aided by the formation of a cyclic stannylene derivative. The success of regioselective acylation⁵⁾ and alkylation⁶⁾ of various non-protected glycosides through tin intermediates suggested that the method was worth testing for use with various non-protected carbohydrates.

This communication describes the result when the bis-tributyltin oxide method is applied to glycosides where three or more contiguous hydroxyl groups are present in the molecule.

After several fruitless attempts to apply the previously reported procedure,^{2a,b)} we found that the following modification is suitable for oxidizing non-protected glycosides; i. e., stannylation of glycosides in refluxing chloroform with an excess of the reagent and molecular sieves 3A followed by *in situ* brominolysis, then chromatography without extraction procedure (see General Procedure). Other solvents such as tetrahydrofuran resulted in over-oxidation.

By this modified method, Me β -D-Glc⁷⁾ **1** was oxidized to the 3-oxo derivative **2**, mp 130-134°C, in 96% yield. The product **2** exhibited a single OMe peak in ¹H-NMR spectrum (δ 3.62).⁸⁾ The homogeneity of the compound was confirmed by GLC of its trimethylsilyl (TMS) derivative. NaBH₄ reduction of **2** in methanol gave back **1** almost quantitatively. The ¹³C-NMR spectrum (Table I) clearly indicated that the compound is the 3-oxo derivative. The same 3-oxo-glucoside (lit. mp 127-128°C) was previously prepared in low yield by dichromate oxidation of **1** in the presence of oxalic acid.⁹⁾

The β -D-p-nitrophenyl derivative **3** was also smoothly oxidized (>90%) to the corresponding 3-oxo derivative **4**, mp 190-194°C, although 4 eq of (Bu₃Sn)₂O was necessary for complete reaction in this case.

Me α -D-Glc **5** was similarly oxidized to give exclusively a compound with an OMe peak at δ 3.52 whose intensity ratio suggested that the product was more than 75% homogeneous. NaBH₄ reduction

of this product in methanol gave Me α -D-Glc **5** and Me α -D-Gal **7** in a ratio of ca. 7:3 (GLC of TMS derivative) suggesting that the oxidation product is the 4-oxo derivative **6**. The ^{13}C -NMR spectrum confirmed this identification. Me α -D-Gal **7** gave the same product **6** in ca. 70% yield (as judged from the OMe peak at δ 3.52 and confirmed by TLC, GLC of the TMS derivative, and ^{13}C -NMR) by similar oxidation.

Me β -D-Xyl **8** was oxidized to the 3-oxo derivative¹⁰⁾ **9** in more than 90% yield. The product **9** (^{13}C -NMR, see Table I), on reduction with NaBH_4 , regenerated **8**, while on catalytic hydrogenation over Pt in AcOH gave Me β -D-Rib **10** and Me β -D-Xyl **8** in a ratio of 9:1 as confirmed by TLC and GLC (TMS derivative).

On similar oxidation, Me α -D-Xyl **11** and Me β -L-Ara **12** gave the same product **13** (>80% yield). However, their ^1H - and ^{13}C -NMR were complex, suggesting that the product is a mixture of at least three compounds. This was eventually found to be identical with the oxidation product of Me β -L-

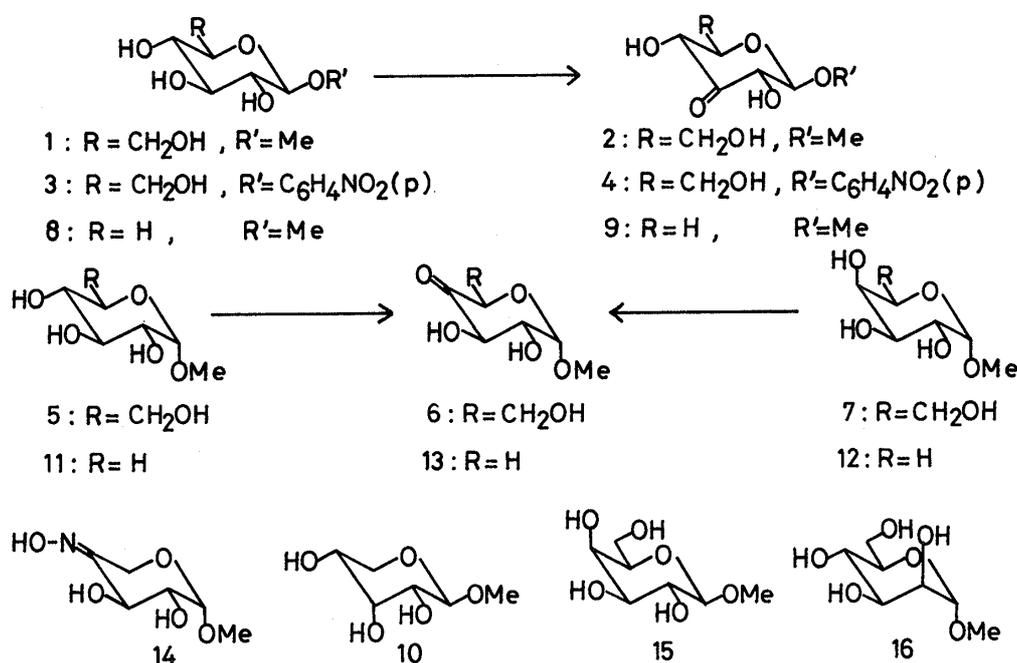


Table I. ^{13}C -NMR Spectra of Oxo-Glycosides in Pyridine- d_5
 (Parenthical values indicate the shifts from the corresponding Me β - or α -D-Glc or D-Xyl).

Compds	C-1	C-2	C-3	C-4	C-5	C-6	OMe
2	106.4 (+0.9)	78.2 ^{a)} (+2.7)	207.4 --	73.5 (+2.1)	78.0 ^{a)} (-0.3)	62.0 (-0.6)	56.9 (+0.2)
4	102.4	77.7	206.4	73.5	79.0	61.9	--
6	101.0 (-0.3)	74.6 (+0.9)	77.4 (+2.1)	205.8 --	76.1 (+2.1)	60.5 (-2.1)	55.8 (+0.8)
9	107.5 (+1.4)	78.2 (+3.5)	206.9 --	73.2 (+2.2)	66.9 (+0)	--	56.8 (+0.2)

2: methyl β -D-ribo-hexopyranosid-3-ulose, **6**: methyl α -D-xylo-hexopyranosid-4-ulose;

9: methyl β -D-erythro-pentopyranosid-3-ulose.

a) Assignment may be interchanged.

Ara **12** by dibutyltin oxide and bromine, where the product was shown to be in equilibrium between the monomer **13** and dimers.⁴⁾ The GLC of the above product (TMS derivative) verified that this is actually the case. The structure of the product was finally confirmed by converting it to the single oxime **14**, mp 138-139°C, which was identical with the authentic specimen⁴⁾ in all respects. Thus, regioselective C-4 oxidation was established for **11** and **12**.

The oxidation of Me β -D-Gal **15** and Me α -D-Man **16** was not highly regioselective. They give a mixture of two regio-isomeric oxo-derivatives (probably 4-oxo and 3-oxo derivatives from **15**, and 2-oxo and 4-oxo derivatives from **16**).

In conclusion, most of glycosides are smoothly oxidized regioselectively by the bis-tributyltin oxide-bromine method to yield the mono-oxo derivative in high yield without protection of the other hydroxyl groups. Oxidation takes place at C-3 for the glycosides which have an equatorial glycosidic linkage (usually β -glycosides) and at C-4 for those which have an axial glycosidic linkage (usually α -glycosides). At the present we can not explain the reason for this. The above obtained oxo-glycosides should be useful intermediates for synthesis in the carbohydrate field, particularly for the synthesis of amino, branched, and the other unusual sugars.

GENERAL PROCEDURE— Dried methyl glycoside (0.5-1 g), $(n\text{-Bu}_3\text{Sn})_2\text{O}$ (ca. 2 eq), and an excess of molecular sieves 3A in chloroform (20-40 ml) were heated under reflux until the glycoside dissolved completely (2-3 h required), then cooled. To this mixture, bromine (ca. 2 eq required) was added at 0°C with stirring until the solution was faintly colored (5-8 min), then the mixture was poured onto a column of silica gel. The column was washed thoroughly with chloroform to remove tin compound(s), then eluted with ethyl acetate to yield the oxo-derivative (70-96%) which is pure enough in ^{13}C -NMR (in most cases) and can be used without further purification. Further elution of the column with methanol gave a small amount of the starting material (if present).

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- 7) Abbreviations: Me (methyl), Glc (glucoside), Gal (galactoside), Xyl (xyloside), Ara (arabinoside), Rib (riboside), and Man (mannoside).
- 8) ^1H - and ^{13}C -NMR spectra were taken in pyridine- d_5 solution.
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