

Communications to the Editor

[Chem. Pharm. Bull.]
33(9)4102-4105(1985)

REACTION OF CONJUGATED ENONES WITH
tert-BUTYLDIMETHYLSILYL TRIFLATE AND INTRAMOLECULAR ANNELETION

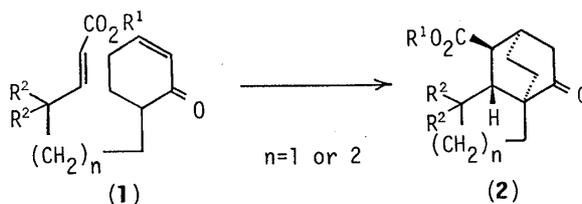
Masataka Ihara,^a Yohhei Ishida,^a Keiichiro Fukumoto,^{*,a} and
Tetsuji Kametani^b

Pharmaceutical Institute, Tohoku University,^a Aobayama, Sendai 980, Japan and
Institute of Medicinal Chemistry, Hoshi University,^b Ebara 2-4-41,
Shinagawa-ku, Tokyo 142, Japan

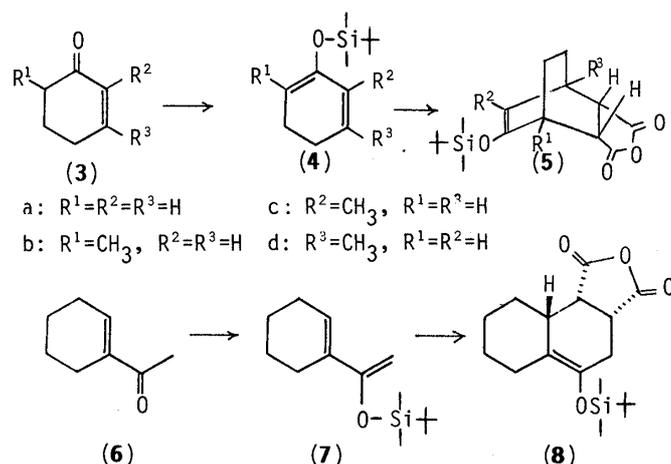
Reaction of 2-cyclohexen-1-ones (**3a-c**) with *tert*-butyldimethylsilyl triflate and triethylamine gave selectively cross-conjugated dienol ethers (**4a-c**) except 3-methyl-2-cyclohexen-1-one (**3d**). Reaction of 6-(5-ethoxycarbonyl-3,3-dimethyl-4-pentenyl-1)-2-cyclohexen-1-one (**10**) under the same reaction conditions gave rise to annelation affording to tricyclo[5.2.2.0^{1,5}]undecenes (**11**) along with the through-conjugated dienol ether (**12**).

KEYWORDS — intramolecular double Michael reaction; *tert*-butyldimethylsilyl triflate; cross-conjugated dienol silyl ether; tricyclo[5.2.2.0^{1,5}]undecene

In the course of our study on the intramolecular double Michael reaction,^{1,2)} we were interested in the generation of the cross-conjugated silyl dienol ethers from conjugated enones. It is known that such dienes are useful substrates for the Diels-Alder reaction.³⁾ The formation of cross-conjugated dienolate anions by α' -deprotonation of enones with lithium dialkylamides under kinetically controlled conditions, followed by silylation, is a useful method for the regioselective formation of 2-silyloxy-1,3-dienes from conjugated enones.⁴⁻⁶⁾ When the reaction was applied to the α,β -unsaturated enone esters (**1**), the double Michael reaction occurred spontaneously affording, in highly stereoselective manner, the tricyclic compound (**2**).¹⁾ We further studied the reaction of conjugated enones with *tert*-butyldimethylsilyl triflate in the presence of triethylamine^{7,8)} and here report the findings.

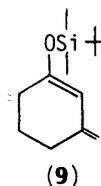


Reaction of 2-cyclohexen-1-ones (**3a-c**) and 1-acetyl-1-cyclohexene (**6**) with 1.1 mole equivalent of *tert*-butyldimethylsilyl triflate and 1.5 mole equivalent of triethylamine in dichloromethane for 5 min at ambient temperature produced quantitatively 2-silyloxy-1,3-dienes (**4a-c**)⁹⁾ and (**7**),⁹⁾ respectively. The structures of the products were determined by ¹H-NMR spectroscopy⁵⁻⁷⁾ and the Diels-Alder reaction with maleic anhydride. The cycloaddition was carried out in dichloromethane for 16~24 h at ambient temperature to give the adducts (**5a-c**)⁹⁾ and (**8**)⁹⁾ in 74%, 77%, 83%, and 82% overall yields from the corresponding enones, respectively. The endo structures of the the products, each obtained as a single isomer, were deduced on consideration of the reaction mechanism. The formation of the silyl enol ethers (**5a-c**) by the cycloaddition suggested the structure of the cross-conjugated dienol ether, since the through-conjugated dienol ether would give the bicyclo[2.2.2]octane having the silyloxy group at the angular position. On the other hand, 3-methyl-2-cyclohexen-1-one (**3d**) gave a mixture of **4d** and **9**^{7,9)} in a ratio of 5 : 6 under the same reaction conditions as above. Treatment of the mixture with maleic anhydride in dichloromethane for 20 h at room temperature afforded the single adduct (**5d**)⁹⁾ in 40% overall yield from **3d**.



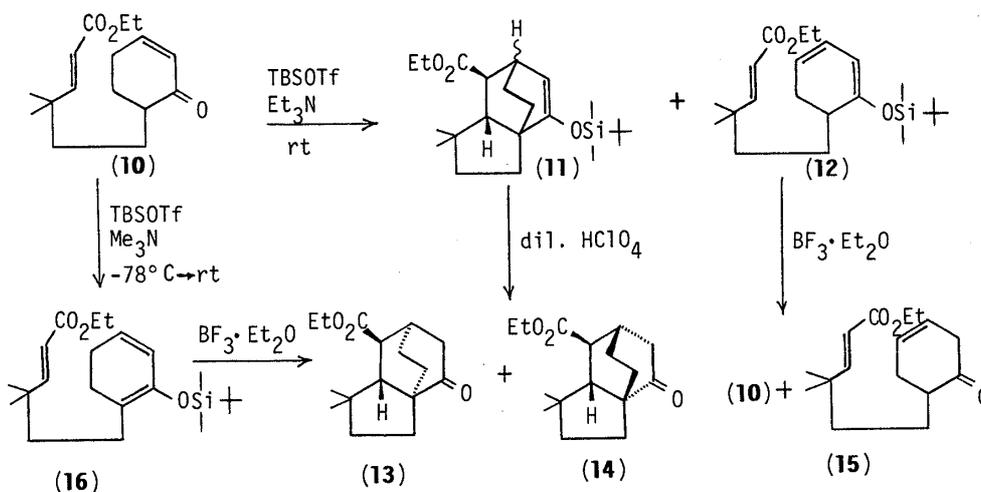
The above method offers a convenient way to synthesize the rather stable cross-conjugated silyl dienol ether by the simple manipulation performed under mild and not so strict anhydrous conditions as in the case of lithium dialkylamide and the silyl chloride.

Surprisingly, when the reaction was applied to 6-(5-ethoxycarbonyl-3,3-dimethyl-4-pentenyl-1)-2-cyclohexen-1-one (**10**),²⁾ two diastereoisomers of the tricyclic compound (**11**) were created. Namely, the reaction of **10** with 1.1 mole equivalent of *tert*-butyldimethylsilyl triflate and 1.5 mole equivalent of triethylamine in dichloromethane for 5 min at ambient temperature formed three products. After medium pressure chromatography on silica gel, tricyclo[5.2.2.0^{1,5}]-undecene (**11**)⁹⁾ was obtained in 37% yield as a mixture of two stereoisomers in the ratio of about 3 : 2 along with the through-conjugated silyl dienol ether (**12**)⁹⁾ in 45% yield. Treatment of the mixture (**11**) with 10% perchloric acid in tetrahydrofuran for 1 h at ambient temperature, followed by chromatographic separation, gave two ketones (**13**)²⁾ and (**14**)⁹⁾ in 55% and 33% yields, respectively. On the reaction of boron trifluoride etherate with **12** in dichloromethane for 1 h at room



temperature, the conjugated enone (10) and the deconjugated enone (15)⁹⁾ were obtained in 54% and 45% yields, respectively.

The formation of the tricyclo[5.2.2.0^{1,5}]undecenes (11) could be regarded as an intramolecular double Michael reaction catalyzed by *tert*-butyldimethylsilyl triflate. The intramolecular Diels-Alder reaction of the cross-conjugated dienol ether could be ruled out for the following reason. The reaction of 10 with *tert*-butyldimethylsilyl triflate in the presence of excess trimethylamine in dichloromethane at -78°C for 6 h and at room temperature for 1 h gave the cross-conjugated dienol ether (16).⁹⁾ No formation of 11 and 12 was detected on the basis of ¹H-NMR analysis. The silyl ether (16) was isolated in 37% yield along with the starting enone (10) and converted into the mixture of 13 and 14 in low yield when treated with boron trifluoride etherate.



REFERENCES AND NOTES

- 1) M. Ihara, M. Toyota, K. Fukumoto, and T. Kametani, *Tetrahedron Lett.*, **25**, 2167 and 3235 (1984); *idem, ibid.*, **26**, 1537 (1985).
- 2) M. Ihara, Y. Ishida, M. Abe, M. Toyota, K. Fukumoto, and T. Kametani, *Chem. Lett.*, in the press.
- 3) S. Danishefsky, *Acc. Chem. Res.*, **14**, 400 (1981); T. Ibuka, Y. Mori, and Y. Inubushi, *Tetrahedron Lett.*, **1976**, 3169; M. E. Jung and C. A. McCombs, *Tetrahedron Lett.*, **1976**, 2935; G. M. Rubottom and D. S. Krueger, *Tetrahedron Lett.*, **1977**, 611.
- 4) R. A. Lee, C. McAndrews, K. M. Patel, and W. Reusch, *Tetrahedron Lett.*, **1973**, 965; C. Girard and J. M. Conia, *Tetrahedron Lett.*, **1974**, 3327; K. B. White and W. Reusch, *Tetrahedron*, **24**, 2439 (1978); M. E. Jung, C. A. McCombs, Y. Takeda, and Y. -G. Pan, *J. Am. Chem. Soc.*, **103**, 6677 (1981).
- 5) G. M. Rubottom and J. M. Gruber, *J. Org. Chem.*, **42**, 1051 (1977).
- 6) Selective formation of through-conjugated dienolate: M. E. Krafft and R. A. Holton, *J. Am. Chem. Soc.*, **106**, 7619 (1984); M. Kawanishi, Y. Itoh, T. Hieda, S. Kozima, T. Hitomi, and K. Kobayashi, *Chem. Lett.*, **1985**, 647.
- 7) H. Emde, A. Gotz, K. Hofmann, and G. Simchen, *Justus Liebigs Ann. Chem.*,

1981, 1643.

- 8) L. N. Mänder and S. P. Sethi, *Tetrahedron Lett.*, **25**, 5953 (1984).
- 9) All new compounds gave satisfactory IR (CHCl_3), $^1\text{H-NMR}$ (CDCl_3), MS and high-resolution MS data. Characteristic data are recorded as follows. **4a**, $^1\text{H-NMR}$ δ : 0.10(6H, s), 1.00(9H, s), 2.10 - 2.30(4H, m), 4.80(1H, m, $\text{C}_1\text{-H}$), 5.70 - 5.90(2H, m, $\text{C}_3\text{-}$ and $\text{C}_4\text{-H}$). **4b**, $^1\text{H-NMR}$ δ : 0.10(6H, s), 0.90(6H, s), 1.65(3H, s), 2.10(4H, br s), 5.60(2H, m, C_3 and $\text{C}_4\text{-H}$), **4c**, $^1\text{H-NMR}$ δ : 0.10(6H, s), 0.90(9H, s), 1.08(3H, br s), 2.10(4H, br s), 4.80(1H, m, $\text{C}_1\text{-H}$), 5.55(1H, m, $\text{C}_4\text{-H}$). **4d**, $^1\text{H-NMR}$ δ : 0.10(6H, s), 0.85(9H, s), 1.80(3H, br s), 4.66(1H, m), 5.40(1H, m). **9**, $^1\text{H-NMR}$ δ : 0.15(6H, s), 0.91(9H, s), 4.50(2H, s), 5.44(1H, s). **7**, $^1\text{H-NMR}$ δ : 0.13(6H, s), 0.93(9H, s), 1.43 - 1.83(4H, m), 1.96 - 2.26(4H, m), 4.05 and 4.20(each 1H, each s, $=\text{CH}_2$), 6.10(1H, m, $\text{C}_2\text{-H}$). **5a**, IR ν_{max} : 1870, 1780(C=O), 1635 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 4.96(1H, dd, $J=1$ and 7 Hz). **5b**, mp 122 - 126°C; IR ν_{max} : 1860, 1780(C=O), 1625 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 4.90(1H, d, $J=7$ Hz). **5c**, mp 86.5 - 87.5°C; IR ν_{max} : 1865, 1785(C=O), 1675 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 1.60(3H, s). **5d**, mp 112 - 113.5°C; IR ν_{max} : 1865, 1780(C=O), 1635 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 1.45(3H, s), 4.60(1H, d, $J=1$ Hz). **8**, mp 89 - 91°C; IR ν_{max} : 1860, 1782(C=O), 1685 cm^{-1} (C=C). **11**, IR ν_{max} : 1720(C=O), 1625 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 0.10(6H, s), 0.90(9H, s), 3.80 - 4.20(2H, m), 4.70 and 4.90(1H, each d, each $J=6$ Hz). **12**, IR ν_{max} : 1700(C=O), 1640 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 0.10(6H, s), 0.86(9H, s), 0.96(6H, s), 1.15(3H, t, $J=7$ Hz), 4.00(2H, q, $J=7$ Hz), 4.85(1H, d, $J=6$ Hz), 5.10-5.70(2H, m), 5.46(1H, d, $J=16$ Hz), 6.67(1H, d, $J=16$ Hz). **14**, IR ν_{max} : 1725, 1720 cm^{-1} (C=O); $^1\text{H-NMR}$ δ : 0.66(3H, s), 0.94(3H, s), 1.27(3H, t, $J=7$ Hz), 4.16(2H, q, $J=7$ Hz). **15**, IR ν_{max} : 1708 - 1715 cm^{-1} (C=O); $^1\text{H-NMR}$ δ : 1.10(6H, s), 1.30(3H, t, $J=6.5$ Hz), 2.80(2H, br s), 4.05(2H, q, $J=6.5$ Hz), 5.57(1H, d, $J=15$ Hz), 5.70(2H, br s), 6.73(1H, d, $J=15$ Hz). **16**, IR ν_{max} : 1710(C=O), 1645 cm^{-1} (C=C); $^1\text{H-NMR}$ δ : 0.10(6H, s), 0.96(9H, s), 1.10(6H, s), 1.32(3H, t, $J=7$ Hz), 4.13(2H, q, $J=7$ Hz), 5.60(1H, d, $J=16$ Hz), 5.62(2H, br s), 6.80(1H, d, $J=16$ Hz).

(Received July 26, 1985)