## Communications to the Editor

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## A FIRST AND STEREOSELECTIVE SYNTHESIS OF (±)-RADDEANAMINE

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Raddeanamine (]), a unique spirobenzylisoquinoline alkaloid, was stereoselectively synthesized from the corresponding protoberberine (9) via 8-methyl-8,14-cycloberbine ([2]) through regionselective  $C_8$ -N bond cleavage and stereoselective hydroxylation at  $C_8$ .

KEYWORDS—— raddeanamine; ochotensimine; spirobenzylisoquinoline; protoberberine; 8,14-cycloberbine; stereoselective synthesis; photo-isomerization

Raddeanamine (1), 1) isolated from *Corydalis ochotensis* var. *raddeana*, has been shown to be an unusual spirobenzylisoquinoline alkaloid possessing a tertiary hydroxy group *cis* to the nitrogen and a methyl group in the five-membered ring. Most of the spirobenzylisoquinoline alkaloids have no extra carbon on their five-membered ring. 2) Synthesis of the related alkaloids ochotensimine (2) and ochotensine (3), having one extra carbon, has been accomplished through the Wittig reaction of the corresponding ketone (4). 3,4)

A synthesis of raddeanamine (]) seems to be easily achieved by methylation of the ketone (4). From analogy with hydride reduction of such ketones,  $^{5}$  however, a methylation reagent is presumed to approach the carbonyl group of 4 from the side of the nitrogen resulting in a *trans* alcohol instead of ]. In fact, on treatment with methyllithium, the ketone (5)  $^{6}$  afforded the *trans* alcohol (6) [93%;  $\delta$  1.53 (3H, s, C-Me)] as a sole product.  $^{7}$ 

Therefore, another method should be developed for a stereoselective synthesis of a spirobenzylisoquinoline having a cis tertiary alcohol. This communication describes a first and stereoselective synthesis of raddeanamine ([]) from the corresponding protoberberine (9) via the 8-methyl-8,14-cycloberbine ([2]) by regioselective  $C_8$ -N bond cleavage and stereoselective hydroxylation at  $C_8$ .

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Treatment of the protoberberine (9)  $^{8)}$  with methylmagnesium iodide in ether afforded the 8-methyldihydroprotoberberine ([0) (91%), which was oxidized with m-chloroperbenzoic acid in methylene chloride to furnish the phenolbetaine ([]) [81%; m/z 365 ( $M^{+}$ );  $\delta$  9.24 (1H, s,  $C_1$ -H), 2.76 (3H, s, C-Me)]. This phenolbetaine was irradiated  $^{9)}$  in methanol with a high-pressure mercury lamp through a Pyrex filter to afford the 8-methyl-8,14-cycloberbine ([2) [57%; mp 183-185°C; m/z 365 ( $M^{+}$ );  $\nu$  1715;  $\delta$  1.55 (3H, s, C-Me)]. Reduction of the cycloberbine ([2) with sodium borohydride in methanol (method A) or lithium tri-tert-butoxyaluminum hydride in benzene-tetrahydrofuran (THF) (10:1) (method B) gave the alcohol ([3) as a mixture of two diastereoisomers, which were difficult to separate.  $^{10}$ 

Previously we found that nucleophilic substitution of 8-alkyl-8,14-cycloberbines (7) effected regionselective  $C_8$ -N bond cleavage to give exclusively spirobenzylisoquinolines (8),  $^{11)}$  while the reaction of 8-unsubstituted cycloberbines led to benzindenoazepines through  $C_{14}$ -N bond cleavage.  $^{12)}$  Thus, an intramolecular nucleophilic opening of the aziridine ring in the 8-methylcycloberbine (13) is expected to afford a spirobenzylisoquinoline possessing a cis alcohol by a regioand stereo-selective control.

Treatment of the alcohol ([3]) with 37% aq. formaldehyde solution in methanol afforded the oxazolidine ([4q and [4b]) in 79% (from [2 via metjod A: [4q:[4b=7.8:1) or 61% overall yield (from [2 via method B: [4q:[4b=1:7.8) [[4q: mp 191.5-192.5°C; m/z 397 ( $M^+$ );  $\nu$  3250;  $\delta$  6.28 (1H, s,  $C_1$ -H), 5.07 (1H, s,  $C_8$ -H), 4.49, 4.13 (2H, AB-q, J=7, NCH<sub>2</sub>O); [4b: mp 196.5-197°C; m/z 397 ( $M^+$ );  $\nu$  3550;  $\delta$  6.28 (1H, s  $C_1$ -H),

5.42 (1H, s,  $C_{8}$ -H), 4.62, 4.56 (2H, AB-q, J=7, NCH $_{2}$ O)]. Chlorination of the alcohol (14g or 14b) with thionyl chloride in methylene chloride followed by reduction with tri-n-butyltin hydride in THF in the presence of azobisisobutyronitrile yielded the deoxygenated product ([5) [mp 138-139°C; m/z 381 (M<sup>+</sup>);  $\delta$  4.55, 4.49 (2H, AB-q, J=7.5, NCH<sub>2</sub>O), 3.42, 3.31 (2H, AB-q, J=18.5, C<sub>8</sub>-H)] in 52 or 46% yield, respectively.

Finally, the oxazolidine ([5]) was treated with sodium cyanoborohydride in methanol in the presence of hydrochloric  $acid^{14}$  to provide (±)-raddeanamine (]) [94%; mp 144.5-146°C; m/z 383 (M<sup>+</sup>); v 3250;  $\delta$  6.88, 6.73 (2H, AB-q, J=8,  $C_{11}^{-}$  and  $C_{12}^{-H}$ ), 6.56 (lH, s,  $C_4^{-H}$ ), 6.19 (lH, s,  $C_1^{-H}$ ), 5.96 (2H, s, OCH $_2^{0}$ ), 3.83, 3.43 (each 3H, s, OMe x 2), 3.46, 3.15 (2H, AB-q, J=16,  $C_{8}-H$ ), 2.58 (3H, s, NMe), 1.24 (3H, s, C-Me)]. The synthetic product was proved by spectral comparisons to be identical with natural raddeanamine.

Thus, we have accomplished a first and stereoselective synthesis of  $(\pm)$ raddeanamine and the present conversion of the oxazolidine (15) to 1 confirmed unambiguously the cis relationship between the hydroxy group and the N-methyl group in raddeanamine. The present synthesis of 1 also amounts to a formal synthesis of ochotensimine (2), since | has been transformed to 2 by dehydration. 1)

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## REFERENCES AND NOTES

- 1) T. Kametani, M. Takemura, M. Ihara, and K. Fukumoto, Heterocycles, 4, 723 (1976); idem, J. Chem. Soc. Perkin Trans. 1, 1977, 390.

  2) R.M. Preisner and M. Shamma, J. Nat. Prod., 43, 305 (1980).

  3) H. Irie, T. Kishimoto, and S. Uyeo, J. Chem. Soc. (C), 1968, 3051.

  4) S. McLean, M.-S. Lin, and J. Whelan, Tetrahedron Lett., 1968, 2425; idem, Can. J. Chem., 48, 948 (1970).

  5) D.B. MacLean, R.A. Bell, J.K. Saunders, C.-Y. Chen, and R.H.F. Manske, Can. J. Chem., 47, 3593 (1969); T. Kishimoto and S. Uyeo, J. Chem. Soc. (C), 1971, 1644; H. Irie, S. Tani, and H. Yamane, J. Chem. Soc. Perkin Trans. 1, 1972, 2986; M. Hanaoka, S. Yasuda, Y. Hirai, K. Nagami, and T. Imanishi, Heterocycles, 14, 1455 (1980). 14, 1455 (1980).
- 6) The ketone (5) was prepared from its N-demethyl derivative 15) by the Eschweiler-Clarke reaction.
- 7) The C-methyl signal in the NMR spectrum of 6 showed a normal chemical shift without anisotropic effect due to the aromatic ring A.
- without anisotropic effect due to the aromatic ring A.
  8) M. Hanaoka, K. Nagami, Y. Hirai, S. Sakurai, and S. Yasuda, Chem. Pharm. Bull., 33, 2273 (1985).
  9) M. Hanaoka, S. Yasuda, K. Nagami, K. Okajima, and T. Imanishi, Tetrahedron Lett., 1979, 3749; M. Hanaoka, C. Mukai, K. Nagami, K. Okajima, and S. Yasuda, Chem. Pharm. Bull., 32, 2230 (1984).
  10) Reduction of such ketones by method A and B has been found to give predominantly cis alcohols and trans alcohols, respectively.11,16)
  11) M. Hanaoka, S. Sakurai, Y. Sato, and C. Mukai, Heterocycles, 19, 2263 (1982).
  12) M. Hanaoka, S.K. Kim, M. Inoue, K. Nagami, Y. Shimada, and S. Yasuda, Chem. Pharm. Bull., 33, 1434 (1985) and references cited therein.
  13) N. Murugesan, G. Blaskó, R.D. Minard, and M. Shamma, Tetrahedron Lett., 22, 3131 (1981).

- 3131 (1981).
- S. McLean and D. Dime, Can. J. Chem., 55, 924 (1977); D. Dime and S. McLean, ibid., 57, 1569 (1979).
   M. Hanaoka, M. Inoue, M. Takahashi, and S. Yasuda, Heterocycles, 19, 31 (1982); idem, Chem. Pharm. Bull., 32, 4431 (1984).
   M. Hanaoka, S. Sakurai, T. Ohshima, S. Yasuda, and C. Mukai, Chem. Pharm. Bull., 2440 (1982)
- <u>30</u>, 3449 (1982).

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