Chem. Pharm. Bull. 34(1) 71-76 (1986)

## Synthesis of a Griseofulvin Analogue

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(Received May 29, 1985)

Ring contraction of 6-ethoxycarbonyl-4,5,7-trihydroxycoumarin (7) to a 3(2H)-benzo-furanone (13) was achieved. Compound 13 was utilized in a synthesis of a griseofulvin analogue (1b) with an ethoxycarbonyl group at the 5 position. However, 1b was found to be inactive in a test of fungicidal activity.

**Keywords**—coumarin; 3(2H)-benzofuranone; griseofulvin; ring contraction; alkoxy-carbonylation; spiro-annelation; fungicidal activity

Griseofulvin (1a) is now produced industrially by fermentation, but its total synthesis has been studied by several workers. In those studies, 3(2H)-benzofuranone derivatives were used as the common synthetic intermediates: the 3(2H)-benzofuranone (3a) was used in the method of Stork and Tomasz,<sup>1)</sup> the 2-methoxycarbonyl analogue (4a) in the method of Brossi *et al.*,<sup>2)</sup> and the 2-acetyl analogue (5a) in the method of Danishefsky and Walker.<sup>3)</sup>

Our earlier paper<sup>4)</sup> described the self-condensation of diethyl acetonedicarboxylate (6) to give 6-ethoxycarbonyl-4,5,7-trihydroxycoumarin (7) in a quantitative yield. We thought that if the ring contraction of the 4-hydroxycoumarine (7) to 3(2H)-benzofuranones (3b and 4b) could be achieved, the griseofulvin analogue (1b) with an ethoxycarbonyl group in the benzene ring of 1a could be conveniently prepared. Moreover, we were interested in determining the influence of an electron-withdrawing group such as an ethoxycarbonyl group in the benzene ring of 1a on the fungicidal activity.

3-Halogenocoumarins are well-known to afford the ring-contracted coumarilic acids (2-carboxybenzofurans) on treatment with alcoholic potassium hydroxide. On the other hand, 3-bromocoumarin, on treatment with sodium methoxide in methanol, gave methyl coumarilate (2-methoxycarbonylbenzofuran) in a high yield. However, ring contraction of 4-hydroxycoumarins to 3(2H)-benzofuranones has not been reported. This paper describes the ring contraction of the 4-hydroxycoumarin (7) to 3(2H)-benzofuranones, and their application to the synthesis of a dl-griseofulvin analogue (1b).

1a:  $R^1 = H$ ,  $R^2 = Me$ ,  $R^3 = H$ 

**1b**:  $R^1 = COOEt$ ,  $R^2 = Me$ ,  $R^3 = H$ **2a**:  $R^1$ ,  $R^2 = H$ ,  $R^3 = Me$  3a: R=H 3b: R=COOEt OMe OH R<sup>1</sup> OH R<sup>2</sup>

4a:  $R^1 = H$ ,  $R^2 = COOMe$ 

4b:  $R^1 = COOEt$ ,  $R^2 = COOMe$ 

5a:  $R^1 = H, R^2 = COMe$ 

Chart 1

### **Results and Discussion**

Conversion of 7 to 3(2H)-benzofuranones (3b, 4b), which are key intermediates for the synthesis of the dl-griseofulvin analogue (1b), was undertaken (Chart 2).

Thus, methylation of 7 to give the corresponding trimethoxycoumarin (8) was attempted in the usual ways, using dimethyl sulfate, methyl iodide, or diazomethane; however, a satisfactory result could not be obtained. On the other hand, phase-transfer reaction of 7 with dimethyl sulfate in the presence of benzyltrimethylammonium chloride (BTEAC) gave 8 in 55% yield. Treatment of 8 with bromine at 0°C gave the 3-bromo derivative (9). Next, ring contraction of the coumarin (9) to the 2-methoxycarbonylbenzofuranone (10) was examined by treating 9 with excess sodium methoxide in refluxing methanol for 5 h. However, not 10 but the methyl cinnamate (11) was obtained in 43% yield. Heating of 11 with methanolic potassium hydroxide gave the  $\alpha$ -bromoacetophenone (12), which was cyclized to the benzofuranone (13) by heating with sodium acetate in ethanol in 71% yield. Based on these findings, conversion of 9 to 13 was also performed by a one-pot reaction in 75% yield. Treatment of 13 with chlorine at 0°C gave 3b, which corresponds to the intermediate of Stork's method.

Since direct ring-contraction of 9 to 10 was unsuccessful, an attempt was made to prepare 2-alkoxycarbonylbenzofuranones (10 or 4b) by the alkoxycarbonylation of the benzofuranones (13 or 3b). The reaction of 13 with diethyl carbonate in the presence of sodium hydride<sup>8)</sup> gave a mixture of many products, from which 10 could not be isolated. In the reaction of 13 with magnesium methylcarbonate (MMC)-diazomethane,<sup>9)</sup> the dimer of 13 was obtained and the starting material was recovered. The reaction of 13 with ethyl chloroformate in the presence of lithium diisopropylamide (LDA)<sup>10)</sup> gave the 3-ethoxycarbonyloxybenzofuran (14) in 84% yield.

Consequently we attempted the reaction of 3b with methyl cyanoformate (with a soft leaving group) in the presence of LDA, 11) to give the 2-methoxycarbonylbenzofuranone (4b), which corresponds to the intermediate of Brossi's method, in 92% yield.

The synthesis of the *dl*-griseofulvin analogue (**1b**) was attempted according to Brossi *et al.*<sup>2)</sup> (Chart 3). The Michael condensation of **4b** with 3-penten-2-one in the presence of benzyltrimethylammonium hydroxide (triton-B) was attempted, but the reaction gave a mixture of many products. Potassium fluoride in the presence of a catalytic amount of 18-crown-6-ether was reported to be an effective basic catalyst in the Michael reaction. We used potassium fluoride and 18-crown-6-ether in the same way, and obtained the Michael adduct, the 2-methoxycarbonyl-2-(4-oxo-pentanyl)benzofuranone (**15**) in 98% yield as a 1:1 mixture of the diastereomers. Spiro-annelation of **15** to **16** was attempted under the conditions used by Brossi *et al.*<sup>2)</sup> However, structurally unknown compounds, probably formed by bimolecular condensation of **15**, were obtained. McMillan *et al.*<sup>13)</sup> and Dawkins and Mulholland<sup>14)</sup> similarly found that the spiro-annelation of 4,6-dimethoxy-2-methoxy-carbonyl-2-(4-oxopentan-2-yl)-3(2*H*)-benzofuranone was unsuccessful, providing the dibenzofuranone derivative.

At this point, we decided to synthesize *dl*-1b by using 3b according to Stork and Tomasz. <sup>1a)</sup> Their method has the advantage that the double Michael reaction of 3a with the 2-propenyl ketone (17) is stereospecific, affording *dl*-griseofulvin (1a) with no epigriseofulvin (2a). However, the preparation of the 2-propenyl ketone (17) is troublesome and the yield of *dl*-1a is very poor (5%). Considering that the reason for the poor yield of *dl*-1a may have been insufficient base, we carried out the reaction of 3b with 17 in diglyme, using 1.2 mol eq of potassium *tert*-butoxide with respect to 3b, and *dl*-1b was obtained in 25% yield. When the same reaction was undertaken in dimethoxyethane (DME), the 2-(1-methoxy-3-oxo-4-hexenylidene)benzofuranone (18) was obtained in 32% yield with no formation of *dl*-1b. It is

unclear whether the sterochemistry of the side chain of **18** is *E*- or *Z*-form. Stirring of **18** with potassium *tert*-butoxide in the presence of 18-crown-6-ether in benzene gave *dl*-**1b** in 90% yield with no formation of the epi analogue.

The stereochemistry of dl-1b was determined by comparison of its nuclear magnetic resonance (NMR) spectrum with those reported for griseofulvin (1a) and epigriseofulvin

Chart 2

4b

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(2a). The signal due to the 6'-methyl group of 1b appeared at  $\delta$  0.97 ppm, which is closer to that of griseofulvin ( $\delta$  0.98 ppm) than to that of epigriseofulvin ( $\delta$  0.90 ppm); moreover, the signals due to the 5'-methylene and 6'-methine of 1b were analogous to those of griseofulvin and different from those of epigriseofulvin.

Compound *dl-1b* was tested for growth-inhibitory activity on various eumycetes (tricophyton, microporum, and archroderma genera) by the agar dilution method. However, the activity was considerably inferior to that of griseofulvin or clotrimazol.

Chart 3

## Experimental

All melting points are uncorrected. NMR spectra were taken with a Hitachi R-24 spectrometer (60 MHz), with tetramethylsilane as an internal standard. Infrared (IR) spectra were recorded on a Nippon Bunko A-102 spectrometer. Mass spectrum (MS) were obtained on a Shimadzu LKB-9000 spectrometer.

**6-Ethoxycarbonyl-4,5,7-trimethoxycoumarin (8)**—Dimethyl sulfate (57 g, 450 mmol), benzyltrimethylammonium chloride (3 g, 13.2 mmol), and CH<sub>2</sub>Cl<sub>2</sub> (135 ml) were successively added to a solution of 6-ethoxycarbonyl-4,5,7-trihydroxycoumarin<sup>4)</sup> (7) (30 g, 110 mmol) in 10% NaOH solution (135 ml), and the mixrure was vigorously stirred at room temperature for 2 d. The CH<sub>2</sub>Cl<sub>2</sub> layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The extracts and the CH<sub>2</sub>Cl<sub>2</sub> layer were collected, dried over MgSO<sub>4</sub>, and concentrated. Crystallization of the residue from MeOH gave 8 (19 g, 55%), mp 186—188 °C. *Anal.* Calcd for C<sub>15</sub>H<sub>16</sub>O<sub>7</sub>: C, 58.44; H, 5.23. Found: C, 58.07; H, 5.16. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1725 (CO), 1710 (CO). NMR (CDCl<sub>3</sub>) δ: 1.38 (3H, t, J = 7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 3.87, 3.92, 4.02 (each 3H, each s, each OMe), 4.42 (2H, q, J = 7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 5.60 (1H, s, 3-H), 6.69 (1H, s, 8-H). MS m/z: 308 (M<sup>+</sup>).

**3-Bromo-6-ethoxycarbonyl-4,5,7-trimethoxycoumarin** (9)—A solution of bromine (4.25 g, 26.6 mmol) in CHCl<sub>3</sub> (30 ml) was added dropwise to a mixture of **8** (7.25 g, 23.5 mmol), CaCO<sub>3</sub> (7 g, 7 mmol), and CHCl<sub>3</sub> (60 ml) at 0 °C. The mixture was stirred for 2 h at 0 °C and the precipitate was filtered off. The filtrate was washed with 10% HCl, dried over MgSO<sub>4</sub>, and concentrated to give **9** (6.9 g, 76%), mp 145—147 °C (from MeOH). *Anal.* Calcd for  $C_{15}H_{15}BrO_7$ : C, 46.52; H, 3.91. Found: C, 46.18; H, 3.75. IR  $v_{max}^{Nujol}$  cm<sup>-1</sup>: 1750 (CO), 1725 (CO). NMR (CDCl<sub>3</sub>)  $\delta$ : 1.33 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 3.89, 3.99, 4.03 (each 3H, each s, each OMe), 4.43 (2H, q, J=7H, OCH<sub>2</sub>CH<sub>3</sub>), 7.22

(1H, s, 8-H). MS m/z: 388 (M<sup>+</sup> +2), 386 (M<sup>+</sup>).

Methyl α-Bromo-3-ethoxycarbonyl-6-hydroxy- $\beta$ ,2,4-trimethoxycinnamate (11) — A mixture of 9 (3.1 g, 8 mmol), MeONa (43.5 mmol), and abs. MeOH (70 ml) was refluxed for 5 h, poured into ice-water, made acidic with 10% HCl, and extracted with AcOEt. The AcOEt layer was dried over MgSO<sub>4</sub> and concentrated. Crystallization of the residue from MeOH gave 11 (1.4 g, 43%), mp 142—144 °C. *Anal.* Calcd for C<sub>16</sub>H<sub>19</sub>BrO<sub>8</sub>: C, 45.84; H, 4.57. Found: C, 45.68; H, 4.54. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3230 (OH), 1725 (CO), 1655 (CO). NMR (Me<sub>2</sub>CO-d<sub>6</sub>) δ: 1.32 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 3.58 (6H, s, OMe × 2), 3.70, 3.78 (each 3H, each s, each OMe), 4.33 (2H, q, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 6.39 (1H, s, 5-H), 8.85 (1H, s, OH). MS m/z: 420 (M<sup>+</sup> + 2), 418 (M<sup>+</sup>).

Ethyl 3-(Bromoacetyl)-4-hydroxy-2,6-dimethoxybenzoate (12)——A solution of 11 (1 g, 2.4 mmol) and KOH (0.4 g, 7.1 mmol) in 95% MeOH (5 ml) was refluxed for 2 h, poured into ice-water, made acidic with 10% HCl, and extracted with AcOEt. The AcOEt layer was dried over MgSO<sub>4</sub> and concentrated to give 12 (0.7 g, 87%), mp 110—111 °C (from Et<sub>2</sub>O). *Anal.* Calcd for C<sub>13</sub>H<sub>15</sub>BrO<sub>6</sub>: C, 44.98; H, 4.36. Found: C, 44.85; H, 4.27. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1710 (CO), 1635 (CO). NMR (CDCl<sub>3</sub>) δ: 1.27 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 3.78, 3.87 (each 3H, each s, each OMe), 4.30 (2H, q, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 4.55 (2H, s, CH<sub>2</sub>Br), 6.18 (1H, s, 5-H). MS m/z: 348 (M<sup>+</sup> + 2), 346 (M<sup>+</sup>).

**5-Ethoxycarbonyl-4,6-dimethoxy-3(2H)-benzofuranone (13)**—Method A: A solution of **12** (2 g, 5.8 mmol) and AcONa ·  $3H_2O$  (1.8 g, 13.2 mmol) in 95% EtOH (20 ml) was refluxed for 1 h, poured into ice-water, and extracted with AcOEt. The AcOEt layer was washed with 10% HCl, dried over MgSO<sub>4</sub>, and concentrated. Crystallization of the residue from Et<sub>2</sub>O gave **13** (1.1 g, 71%), mp 102 °C. *Anal.* Calcd for  $C_{13}H_{14}O_6$ : C, 58.65; H, 5.30. Found: C, 58.37; H, 5.30. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1725 (CO), 1690 (CO). NMR (CDCl<sub>3</sub>)  $\delta$ : 1.40 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 3.96, 4.20 (each 3H, each s, each OMe), 4.41 (2H, q, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 4.65 (2H, s, 2-H<sub>2</sub>), 6.37 (1H, s, 7-H). MS m/z: 266 (M<sup>+</sup>).

Method B: A solution of 9 (9.75 g, 25 mmol) and KOH (6.2 g, 110 mmol) in 95% EtOH (120 ml) was refluxed for 2 h and cooled to 0 °C, then AcOH (6.6 g, 110 mmol) was added dropwise. The solution was refluxed for 1 h and the solvent was removed. The residue was extracted with AcOEt, then the AcOEt layer was dried over MgSO<sub>4</sub> and concentrated to give 13 (5 g, 75%), after recrystallization from Et<sub>2</sub>O.

7-Chloro-5-ethoxycarbonyl-4,6-dimethoxy-3(2H)-benzofuranone (3b) — A solution of chlorine (6.6 mmol) in dry CHCl<sub>3</sub> was added dropwise to a solution of 13 (1.71 g, 6.43 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (30 ml) at 0 °C, then the mixture was washed successively with 10% HCl, 10% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, and H<sub>2</sub>O, dried over MgSO<sub>4</sub>, and concentrated. Chromatography of the residue on silica gel with AcOEt–petr. ether (1:9) gave 3b (1.37 g, 71%), mp 111—113 °C. Anal. Calcd for C<sub>13</sub>H<sub>13</sub>ClO<sub>6</sub>: C, 51.92; H, 4.36. Found: C, 52.03; H, 4.30. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1720 (CO), 1710 (CO). NMR (CDCl<sub>3</sub>)  $\delta$ : 1.42 (3H, t, J=7Hz, OCH<sub>2</sub>CH<sub>3</sub>), 4.05, 4.08 (each 3H, each s, each OMe), 4.44 (2H, q, J=7Hz, OCH<sub>2</sub>CH<sub>3</sub>), 4.77 (2H, s, 2-H<sub>2</sub>). MS m/z: 302 (M<sup>+</sup>+2), 300 (M<sup>+</sup>).

**5-Ethoxycarbonyl-3-ethoxycarbonyloxy-4,6-dimethoxybenzofuran** (14)—A solution of LDA (8.6 mmol) in tetrahydrofuran (THF, 22 ml) was added dropwise to a solution of **13** (2 g, 7.5 mmol) in dry THF (100 ml) at -78 °C in an Ar atmosphere. The mixture was stirred for 30 min at -78 °C, then ethyl chloroformate (0.98 g, 9 mmol) was added at -78 °C. The mixture was stirred for an additional 1 h at room temperature, neutralized with 10% NH<sub>4</sub>Cl, and extracted with AcOEt. The AcOEt layer was dried over MgSO<sub>4</sub> and the solvent was removed. Chromatography of the residue on silica gel with AcOEt–petr. ether (1:10) gave **14** (2.13 g, 84%), mp 55—56 °C (from Et<sub>2</sub>O). *Anal.* Calcd for C<sub>16</sub>H<sub>18</sub>O<sub>8</sub>: C, 56.84; H, 5.36. Found: C, 56.72; H, 5.49. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1765 (CO), 1720 (CO). NMR (CDCl<sub>3</sub>)  $\delta$ : 1.42 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 3.89, 4.04 (each 3H, each s, each OMe), 4.44, 4.49 (each 2H, each q, J=7 Hz, each OCH<sub>2</sub>CH<sub>3</sub>), 6.86 (1H, s, 7-H), 7.90 (1H, s, 2-H).

7-Chloro-5-ethoxycarbonyl-3-hydroxy-4,6-dimethoxy-2-methoxycarbonylbenzofuran (4b) — A solution of LDA (3.7 mmol) in THF (9 ml) was added dropwise to a solution of 3b (1 g, 3.3 mmol) and hexamethylphosphoramide (HMPA, 5 ml) in THF (50 ml) at -78 °C under an Ar atmosphere. The mixture was stirred for 30 min at -78 °C, then methyl cyanoformate (0.32 g, 3.8 mmol) was added. The mixture was stirred for an additional 30 min at -78 °C, neutralized with 10% NH<sub>4</sub>Cl and extracted with AcOEt. The AcOEt layer was dried over MgSO<sub>4</sub> and concentrated. Chromatography of the residue on silica gel with AcOEt–petr. ether (1:10) gave 4b (1.1 g, 92%), mp 117—118 °C. Anal. Calcd for C<sub>15</sub>H<sub>15</sub>ClO<sub>8</sub>: C, 50.22; H, 4.21. Found: C, 50.41; H, 4.18. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 3350 (OH), 1725 (CO), NMR (CDCl<sub>3</sub>)  $\delta$ : 1.45 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 4.09, 4.11, 4.19 (each 3H, each s, each OMe), 4.44 (2H, q, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 8.60 (1H, s, OH).

7-Chloro-5-ethoxycarbonyl-2-methoxycarbonyl-4,6-dimethoxy-2-(4-oxopentan-2-yl)-3(2*H*)-benzofuranone (15)—A solution of 4b (2.12 g, 5.9 mmol), 3-penten-2-one<sup>16</sup>) (1.6 g, 18 mmol), KF (0.17 g, 3 mmol), and 18-crown-6-ether (0.16 g, 0.6 mmol) in CH<sub>3</sub>CN (50 ml) was refluxed for 4h under an Ar atmosphere, poured into ice-water, and extracted with AcOEt. The AcOEt layer was dried over MgSO<sub>4</sub> and concentrated. Chromatography of the residue on silica gel with AcOEt–hexane (1:9) gave 15 (2.56 g, 98%) as an oil. *Anal.* Calcd for C<sub>20</sub>H<sub>23</sub>ClO<sub>9</sub>: C, 54.23; H, 5.23. Found: C, 54.43; H, 5.11. IR  $\nu_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1755 (CO); 1735 (CO), 1725 (CO), 1720 (CO), NMR (CDCl<sub>3</sub>) δ: 0.84 (3H × 1/2, d, J = 7 Hz, C-Me), 1.03 (3H × 1/2, d, J = 7 Hz, C-Me), 1.33 (3H, t, J = 7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 2.20—3.40 (3H, m, CHCH<sub>2</sub>), 3.79 (3H, s, COOMe), 4.04, 4.10 (each 3H, each s, each OMe), 4.35 (2H, q, J = 7 Hz, OCH<sub>2</sub>CH<sub>3</sub>). MS m/z: 344 (M<sup>+</sup> + 2), 342 (M<sup>+</sup>).

dl-7-Chloro-5-ethoxycarbonyl-6'-methyl-2',4,6-trimethoxyspiro[benzofuran-2(3H),1'-(2-cyclohexene)]-3,4'-dione (dl-1b)—Potassium tert-butoxide (0.32 g, 2 mmol) was added to a solution of 3b (0.6 g, 2 mmol) and 1-methoxy-3-

oxo-4-hexen-1-yne<sup>1b</sup>) (17) (0.3 g, 2.4 mmol) in dry diglyme (45 ml) at 0 °C under a N<sub>2</sub> atmosphere, and the mixture was stirred for 24 h at room temperature, poured into ice-water and extracted with AcOEt. The AcOEt layer was dried over MgSO<sub>4</sub> and concentrated. Chromatography of the residue on silica gel with Et<sub>2</sub>O-petr. ether (1:4) gave *dl*-1b (0.21 g, 25%), mp 138—140 °C (from CH<sub>2</sub>Cl<sub>2</sub>-petr. ether). *Anal.* Calcd for C<sub>20</sub>H<sub>21</sub>ClO<sub>8</sub>: C, 56.50; H, 4.98. Found: C, 56.41; H, 5.03. IR  $v_{\rm max}^{\rm Nujol}$  cm <sup>-1</sup>: 1740 (CO), 1715 (CO), 1660 (CO). NMR (CDCl<sub>3</sub>)  $\delta$ : 0.97 (3H, d, J=6 Hz, 6'-Me), 1.41 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 2.20—3.10 (3H, m, 5'-H<sub>2</sub> and 6'-H), 3.72 (3H, s, 2'-OMe), 4.10, 4.18 (each 3H, each s, each OMe), 4.48 (2H, q, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 5.65 (1H, s, 3'-H).

7-Chloro-4,6-dimethoxy-5-ethoxycarbonyl-2-(1-methoxy-3-oxo-4-hexenylidene)-3(2*H*) benzofuranone (18) — Potassium *tert*-butoxide (0.19 g, 1.7 mmol) was added to a solution of **3b** (0.5 g, 1.7 mmol) and **17** (0.33 g, 2.7 mmol) in dimethoxyethane (40 ml) at room temperature, then the mixture was stirred for 1 h at room temperature and worked up as described for **1b**. The crude product (**18**) was recrystallized from benzene, yield 0.23 g (32%), mp 92—94 °C. *Anal.* Calcd for  $C_{20}H_{21}ClO_8$ . C, 56.54; H, 4.98. Found: C, 56.24; H, 4.90. IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1720 (CO), 1690 (CO), 1670 (CO). NMR (CDCl<sub>3</sub>) δ: 1.40 (3H, t, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 1.95 (3H, dd, J=7, 2 Hz, C-Me), 4.06 (3H, s, OMe), 4.22 (6H, s, OMe × 2), 4.39 (2H, s, CH<sub>2</sub>CO), 4.48 (2H, q, J=7 Hz, OCH<sub>2</sub>CH<sub>3</sub>), 6.34 (1H, d, J=18 Hz, CH=CH-Me), 7.15 (1H, dd, J=18, 2 Hz, CH=CH-Me). MS m/z: 426 (M<sup>+</sup>+2), 244 (M<sup>+</sup>).

Cyclization of 18 to dl-1b—A mixture of 18 (97 mg, 0.23 mmol), tert-BuOK (26 mg, 0.23 mmol), and 18-crown-6-ether (20 mg, 0.08 mmol) in dry benzene (10 ml) was heated at 60 °C for 4 h, poured into ice-water, made acidic with 10% HCl, and extracted with  $CH_2Cl_2$ . The organic layer was dried over MgSO<sub>4</sub> and concentrated. Chromatography of the residue on silica gel with AcOEt-petr. ether (1:4) gave dl-1b (87 mg, 90%).

Acknowledgement The authors are very grateful to Morishita Seiyaku Co., Ltd., for carrying out the biological activity screening.

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