## Communications to the Editor

Chem. Pharm. Bull. 34(10)4406-4409(1986)

## INTRAMOLECULAR PHOTOCYCLIZATION OF 3-SUBSTITUTED ALICYCLIC DITHIOIMIDES: FACILE SYNTHESIS OF 2-AZABICYCLOALKANES VIA THE NORRISH TYPE II PROCESS 1)

Kazuaki Oda, a Minoru Machida, \*, a and Yuichi Kanaoka b Faculty of Pharmaceutical Sciences, Higashi-Nippon-Gakuen University, a Ishikari-Tobetsu, Hokkaido 061-02, Japan and Faculty of Pharmaceutical Sciences, Hokkaido University, b Sapporo 060, Japan

Upon irradiation, dithiosuccinimides ( $\underline{1a-c}$ ) and dithioglutarimides ( $\underline{1d-f}$ ) with an aralkyl group at the  $\alpha$ -position of the thiocarbonyl group undergo the Norrish type II reaction to give the cyclized products ( $\underline{2a-e}$ ),  $\underline{2-azabicycloalkanes}$  with varying ring size.

KEYWORDS — dithiosuccinimide; dithioglutarimide; intramolecular
photocyclization; Norrish type II reaction; 2-azabicycloalkane

A new combination of functional groups often leads to a new reaction system, also in excited state pathways. We have recently shown that the change of a key element in the cyclic imide chromophore from oxygen to sulfur leads to unique thioimide photochemistry. The excited cyclic thioimides react with olefins only by photoaddition. Other photoreactions such as the Norrish type I ( $\alpha$ -cleavage) and type II reactions, which are common to the parent thione and imide families, are less important. For example, although 3-substituted alicyclic imides  $\underline{1}$  (Y=O) undergo Norrish type I reaction and subsequent rearrangements, neither  $\alpha$ -cleavage nor the rearrangements are competitive in the thioimide substrates ( $\underline{1}$ ; Y=S) in which  $\alpha$  is substituted for thiocarbonyl (Chart 1). As a result, such inertness of the thioimides to the Norrish type I reaction gives the alicyclic thioimide system interesting synthetic capacities; side reactions due to photolytic cleavages are negligible during the course of their

Chart 1

photoadditions. 2) Indeed, an intramolecular Paterno-Buchi reaction of certain dithiosuccinimides gave rise to a highly strained, fused-ring system. 4) present communication, we wish to report the first example of an intramolecular Norrish type II reaction of the 3-substituted alicyclic thioimides  $\underline{1}$  (Y=S). The substrates selected in this work are a series of 3-( $\omega$ -phenylalkyl)cyclic dithioimides <a>1a-e</a>, with which the special reactivity of the excited thiocarbonyl group of the thioimides to abstract benzylic hydrogens<sup>5)</sup> was expected to lead to a facile synthesis of 2-azabicycloalkanes since side reactions such as  $\alpha\text{-cleavage}$ would be ineffective.

Irradiation of these 3-( $\omega$ -phenylalkyl)cyclic dithioimides <u>1a-e</u> was carried out in benzene (10 mM) using a 1-kW high-pressure mercury lamp (Pyrex filter) in a stream of nitrogen at room temperature for 1-5 h. In all cases a pair of stereoisomers  $\underline{2-i}$  and  $\underline{2-ii}$  were obtained as a result of the C-C bond-formation between the thiocarbonyl and the benzylic carbons. The results are listed in Table I.

Table I. Photoproducts from 1

c) A mixture of two stereoisomers. Identified by 1H-NMR spectroscopy. The stereochemistry of the ring juncture is unknown.

The structural assignment for all of the products was based on elemental analyses and spectral data. For example, the  $^{13}\text{C-NMR}$  spectra of  $\underline{2-i}$  and  $\underline{2-ii}$ showed the presence of a newly formed quaternary carbon instead of the disappearance of one thiocarbonyl group. Here the presence of one doublet at 51.3-60.5 ppm indicated the presence of benzylic carbon. The stereochemistry of 2-i and  $\underline{2-ii}$  was determined on the basis of the  ${}^{1}\text{H-NMR}$  spectra by considering the anisotropic shielding effects of the phenyl ring on the chemical shifts of the

N-methyl group ( $\underline{2a-ii}$  -  $\underline{2a-i}$  = 0.9 ppm). Thus,  $\underline{2-i}$  and  $\underline{2-ii}$  are respectively of  $\underline{trans}$ - and  $\underline{cis}$ -configuration with respect to a thiol and a phenyl group.

As shown in Table I, in the photolysis of a series of thiosuccinimides (n=1;  $\underline{1a-c}$ ) having an aralkyl group at the 3-position of imide ring, hydrogen abstraction took place most effectively at the benzylic position which is  $\gamma$  and  $\delta$  to the thiocarbonyl group, respectively. But the ability of this abstraction at the benzylic  $\epsilon$ -position ( $\underline{1c}$ ) strikingly decreased and unchanged  $\underline{1c}$  was recovered in a 75% yield. Similarly, in the cases of thioglutarimides (n=2;  $\underline{1d-f}$ ), only the  $\gamma$ -hydrogen abstraction reaction at the benzylic carbon proceeded effectively giving the type II products, while  $\underline{1e}$  with a benzylic  $\delta$ -hydrogen gave an unseparable mixture of  $\underline{2e-i}$  and  $\underline{2e-ii}$  in poor yield, along with recovery of unchanged 1e (73%). Further, when 1f had a benzylic hydrogen available for  $\epsilon$ -hydrogen abstraction, no cyclized products due to the Norrish type II reaction were isolated. Instead the substrate  $\frac{1f}{2}$  was recovered in a 91% yield, even after irradiation for 5 h. In addition, the dithiosuccinimides  $(\underline{1a},\underline{b})$  gave the minor product N-methyl dithiosuccinimide (from  $\underline{1a}$ ) and the reduced thiol (from  $\underline{1b}$ ). probably results initially from  $\delta$ -hydrogen abstraction followed by elimination and  $\gamma$ -hydrogen transfer. Apparently the difference in photochemical behavior in these thiosuccinimides and thioglutarimides is due to an unfavorable geometrical distance between the thiocarbonyl group of the latter and the benzylic-hydrogen to be abstracted.

Much attention has been paid to the construction of azabicycloalkane skeletons in view of the biological interest in their 4.5-.7 4.6-.8 and 5.6-9 ring-fused systems. Very recently 2-azaallyl anion cycloaddition has been reported. This photocyclization also adds a new entry in the synthesis of some 2-azabicycloalkane systems.

## REFERENCES

- 1 a) Photochemistry of the Nitrogen-Thiocarbonyl Systems. Part 10. Part 9: M. Machida, K. Oda, E. Yoshida, and Y. Kanaoka, submitted; b) Photoinduced Reactions. Part 91. Part 90: see ref. 1a.
- 2 a) M. Machida, K. Oda, E. Yoshida, and Y. Kanaoka, J. Org. Chem., <u>50</u>, 1681 (1985); b) M. Machida, K. Oda, E. Yoshida, S. Wakao, K. Ohno, and Y. Kanaoka, Heterocycles, <u>23</u>, 1615 (1985); c) M. Machida, K. Oda, and Y. Kanaoka, Chem. Pharm. Bull., <u>33</u>, 3552 (1985); d) K. Oda, M. Machida, and Y. Kanaoka, Synthesis, in press.
- 3 a) Y. Kanaoka, H. Okajima, and Y. Hatanaka, J. Org. Chem., <u>44</u>, 1749 (1979);
   b) K. Maruyama, T. Ishitoku, and Y. Kubo, J. Am. Chem. Soc., <u>101</u>, 3670 (1979).
- 4 K. Oda, M. Machida, K. Aoe, Y. Nishibata, Y. Sato, and Y. Kanaoka, Chem. Pharm. Bull., <u>34</u>, 1414 (1986).
- 5 M. Machida, K. Oda, and Y. Kanaoka, Tetrahedron Lett., 26, 5173 (1985).
- 6 a) T. H. Jones and M. S. Glum, "Alkaloids: Chemical and Biological Perspectives," Vol. 1-3, Ed. by S. D. Pelletier, John Wiley and Sons, Inc.,

- New York, 1983; b) R. B. Herbert, "Rodd's Chemistry of Carbon Compounds," Vol.  ${\rm IV}^{\rm L}$ . 2nd. Ed. by S. Coffey, Elsevior Scientific Publishing Co., Amsterdam, 1980, p. 291.
- 7 a) T. S. Cantrell, J. Chem. Soc., Chem. Commun., <u>1972</u>, 155; b) G. Buhr, Chem. Ber., <u>106</u>, 3544 (1973).
- 8 G. Adembri, D. Donati, S. Fusi, and F. Ponticelli, Heterocycles, <u>23</u>, 2885 (1985).
- 9 a) L. E. Overman and L. T. Mendelson, J. Am. Chem. Soc., <u>103</u>, 5579 (1981);
  b) J. C. Gramain and R. Remuson, Tetrahedron Lett., <u>26</u>, 4083 (1985).
- 10 W. H. Pearson, M. A. Walters, and K. D. Oswell, J. Am. Chem. Soc., <u>108</u>, 2769 (1986).

(Received July 14, 1986)