Chem. Pharm. Bull. 34(11)4523—4526(1986)

An Efficient Total Synthesis of (\pm) -Epilupinine and (\pm) -Lupinine from a Common Quinolizidine Intermediate¹⁾

HIROKI TAKAHATA, KYOKO YAMABE, TOSHIAKI SUZUKI, and TAKAO YAMAZAKI*

Faculty of Pharmaceutical Sciences, Toyama Medical & Pharmaceutical University, Sugitani 2630, Toyama 930-01, Japan

(Received April 23, 1986)

A common intermediate, the quinolizidine (5) obtained by annulation of a cyclic thioimidate (3) with Nazarov's reagent (4) in the presence of mercuric chloride, was stereospecifically transformed into (\pm) -epilupinine (1) and (\pm) -lupinine (2).

Keywords— (\pm) -epilupinine; (\pm) -lupinine; quinolizidine; cyclic thioimidate; Nazarov's reagent; annulation; mercuric chloride; Lawesson's reagent

Various total syntheses of the quinolizidine alkaloids, (\pm) -epilupinine (1) and (\pm) -lupinine (2) have still been well reported in recent years.²⁾ We are interested in exploring the utilization of cyclic thioimidates in a heterocyclic synthesis,^{3,4)} and we describe here the use of a quinolizidine intermediate obtained by Robinson annulation of a cyclic thioimidate with methyl 3-oxo-4-pentenoate (Nazarov's reagent) in the efficient syntheses of 1 and 2.

Robinson annulation using Nazarov's reagent (4) is an attractive synthetic route to functionalized 6-membered ring compounds.⁵⁾ Although it is already known that the treatment of a cyclic imidate (3b) with 4 gives a quinolizidine product (5) (54%),^{5e)} we found that the reaction of 3a with 4 in the presence of mercuric chloride afforded 5 in 62% yield, together with methyl 5-methylthio-3-oxopentenoate (6) (25%). Mercuric chloride might have worked as a scavenger of methanethiol. We were able to develop efficient total syntheses of 1 and 2 from the readily accessible 5.

Transformation of 5 into 1 is shown in Chart 2. Stereospecific reduction of 5 with diisobutylaluminum hydride (DIBAL) in the presence of triethylamine (Et₃N) in tetrahydrofuran (THF) at -50 °C gave the saturated β -ketoester (7) in 55% yield. ⁶⁾ The proton nuclear magnetic resonance (¹H-NMR) spectrum showed a signal due to C₁-H at δ 3.24 (d, J=11 Hz) and its coupling constant strongly indicated *trans* relative configuration between C₁-H and C_{9a}-H. This selective conversion (5 \rightarrow 7) can be attributed to thermodynamic control.

Chart 1

Treatment of 7 with 1,2-ethanedithiol in the presence of boron trifluoride etherate in trifluoroacetic acid afforded the thioketal (8) in 81% yield. Desulfurization of 8 with Raney Ni (W-2) in ethanol produced the ester (9) in 51% yield. The methoxycarbonyl group of 9 underwent reduction with lithium aluminum hydride (LiAlH₄) in boiling THF to give (\pm)-epilupinine (1) in 75% yield.

Conversion of 5 to 2 is depicted in Chart 3. Chemoselective thionation of 5 with Lawesson's reagent⁷⁾ in benzene at room temperature gave the enaminothioketone (10) in 79% yield, and this underwent desulfurization with Raney Ni (W-2) in ethanol to produce the enaminoester (11) in 65% yield. According to the method described previously⁸⁾ stereospecific reduction of 11 was performed with sodium borohydride in methanol to afford methyl lupinate (12) in 69% yield. Compound 12 was subsequently converted to (\pm) -lupinine (2) by reduction with LiAlH₄. Both 1 and 2 thus prepared had spectra (NMR and infrared (IR)) identical with those of the corresponding authentic materials.^{2c)}

Experimental

All melting points were determined on a Yanaco micro melting point apparatus and are uncorrected. IR spectra were taken with a JASCO A-102 spectrophotometer, NMR spectra with Varian EM360, JEOL PMX60SI, Varian XL200, and JEOL JNM-FX270 spectrometers (SiMe₄ as an internal reference and CDCl₃ as the solvent), and mass spectra (MS) with a JEOL JMS-D200 spectrometer. Column chromatography was performed on Silica gel 60 (230—400 mesh), under pressure.

Methyl 3,4,6,7,8,9-Hexahydro-2-oxo-2*H*-quinolizine-1-carboxylate (5)—The thioimidate (3a) (1.28 g) was added dropwise to a stirred solution of mercuric chloride (3 g) and Nazarov's reagent (4)⁹⁾ (1.29 g) in methanol (30 ml) at 0 °C. The mixture was stirred overnight at room temperature and filtered through celite. The filtrate was evaporated. Saturated aqueous NaHSO₄ was added to the residue, and the mixture was extracted with CH₂Cl₂. The extract was washed with water, dried over MgSO₄, and evaporated. The residue was chromatographed on silica gel (ethyl acetate-methanol) to give methyl 5-methylthio-3-oxopentenoate (6) (436 mg, 25%) and the quinolizidine (5) (1.29 g, 62%). 6: bp 110—115 °C/5 mmHg (Kugelrohr). IR $\nu_{\text{max}}^{\text{film}}$ cm⁻¹: 1740, 1710. ¹H-NMR δ: 2.10 (3H, s, SMe), 3.70 (3H, s, COOMe). High-resolution MS Calcd for C₇H₁₂O₃S: 176.0507. Found: 176.0509. 5: mp 75—78 °C (from hexane-ethyl acetate); in lit., ^{5e)} 5 was isolated as an oil. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 1690, 1620. ¹H-NMR δ: 3.80 (3H, s, COOMe). *Anal.* Calcd for C₁₁H₁₅NO₃: C, 63.14; H, 7.23; N, 6.69. Found: C, 63.07; H, 7.15; N, 6.68.

Methyl (1 R^* ,9a R^*)-1,3,4,6,7,8,9,9a-Octahydro-2-oxo-2H-quinolizine-1-carboxylate (7)—Diisobutylaluminum hydride (1.0 M solution in n-hexane) (3.6 ml) was added to THF (3 ml) under argon at -50 °C, then triethylamine (0.52 ml) was added to that solution. The resulting mixture was stirred for 0.5 h, then added to a solution of the quinolizidine (5) (540 mg) in THF (13 ml) at -50 °C via a syringe. The whole was stirred at the same temperature for 4 h, then the reaction was quenched by the addition of 2 N NaOH. The mixture was extracted with ether and successively with CH₂Cl₂. Each extract was washed with brine, dried over K₂CO₃, and evaporated. The combined residues were chromatographed on silica gel (ethyl acetate-methanol) to give the β -ketoester quinolizidine (7) (300 mg, 55%) as a viscous oil. IR v_{max}^{flim} cm⁻¹: 2860, 2810, 2770, 1740, 1710. ¹H-NMR δ : 3.24 (1H, d, J=11 Hz, 1-H), 3.73 (3H, s, COOMe). High-resolution MS Calcd for C₁₁H₁₇NO₃: 211.1206. Found: 211.1164.

Methyl (1.5*,9a.R*)-1,3,4,6,7,8,9,9a-Octahydro-2,2-ethylenedithio-2*H*-quinolizine-1-carboxylate (8) — A mixture of the quinolizidine (7) (197 mg), ethanedithiol (1.1 ml), and BF₃ · Et₂O (1.1 ml) in trifluoroacetic acid (11 ml) was stirred at room temperature for 15 h. The reaction mixture was evaporated, and ice cold saturated aqueous NaHCO₃ was added to the residue. The mixture was extracted with CH₂Cl₂. The extract was washed with brine, dried over MgSO₄, and evaporated. The residue was chromatographed on silica gel (ethyl acetate-methanol) to give the thioketal (8) (216 mg, 81%) as a solid, mp 59—62 °C (from diisopropyl ether). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 2810, 1730. ¹H-NMR δ : 2.79 (1H, d, J=11 Hz, 1-H), 3.23 (4H, s, SCH₂CH₂S), 3.71 (3H, s, COOMe). *Anal.* Calcd for C₁₃H₂₁NO₂S: C, 54.32; H, 7.36; N, 4.87. Found: C, 54.34; H, 7.39; N, 4.89.

Methyl (15*,9aR*)-1,2,3,6,7,8,9,9a-Octahydro-4H-quinolizine-1-carboxylate (9)——A mixture of the thioketal (8) (216 mg) and W-2 Raney Ni (2.2 g) in ethanol (10 ml) was refluxed for 0.5 h, then filtered, and the filtrate was evaporated. The residue was chromatographed on silica gel (ethyl acetate-methanol) to give the ester (9) (75 mg, 51%) as a viscous oil. IR $v_{\text{max}}^{\text{film}}$ cm⁻¹: 2800, 1720. ¹H-NMR δ : 3.69 (3H, s, COOMe). High-resolution MS Calcd for $C_{11}H_{19}NO_2$: 197.1415. Found: 197.1395.

(\pm)-Epilupinine (1)—A solution of the ester (9) (74 mg) in THF (1 ml) was added to a suspension of lithium aluminum hydride (9.1 mg) in THF (1 ml). The reaction mixture was refluxed for 0.5 h, cooled, then treated sequentially with water (0.1 ml), 15% aqueous NaOH (0.1 ml), and water (0.1 ml), and dried over Na₂SO₄. The mixture was filtered and the solid was washed with THF. The combined solutions were evaporated. The residue was chromatographed on alumina (ethyl acetate-methanol) to give (\pm)-epilupinine (1) (51 mg, 76%) as a solid, mp 79—80 °C (from hexane-diisopropyl ether) (lit., 2c) 81—82.5 °C). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm $^{-1}$: 3640, 2870, 2820, 2780, 2690, 1640. 1 H-NMR δ : 1.20—2.10 (15H, m), 2.84 (2H, m), 3.66 (2H, m). The synthetic compound was completely identical with an authentic sample^{2c)} in terms of spectral properties (NMR and IR).

Methyl 3,4,6,7,8,9-Hexahydro-2-thioxo-2*H*-quinolizine-1-carboxylate (10) — A mixture of the quinolizidine (5) (1.4 g) and *p*-methoxyphenylthiophosphine sulfide dimer (707 mg) in dry benzene (30 ml) was stirred at room temperature for 1 h and filtered through celite. The filtrate was evaporated. The residue was chromatographed on silica gel (ethyl acetate) to give the thione (10) (887 mg, 79%) as a solid, mp 156—160 °C (from diisopropyl ether—CH₂Cl₂). IR $v_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 1720, 1590. ¹H-NMR δ: 3.80 (3H, s, COOMe). *Anal*. Calcd for C₁₁H₁₅NO₂S: C, 58.64; H, 6.71; N, 6.22. Found: C, 58.47; H, 6.74; N, 5.99.

Methyl 2,3,6,7,8,9-Hexahydro-4*H*-quinolizine-1-carboxylate (11)——A mixture of the thione (10) (450 mg) and W-2 Raney Ni (2.25 g) in ethanol (22.5 ml) was stirred at room temperature for 3 min, then filtered, and evaporated. The residue was chromatographed on silica gel (hexane–ethyl acetate) to give the enamino ester (11) (254 mg, 65%) as a syrup. IR $v_{\text{max}}^{\text{film}}$ cm⁻¹: 1670. ¹H-NMR δ : 3.62 (3H, s, COOMe). High-resolution MS Calcd for $C_{11}H_{17}NO_2$: 195.1259. Found: 195.1265.

Methyl (1 R^* ,9a R^*)-1,2,3,6,7,8,9,9a-Octahydro-4H-quinolizine-1-carboxylate (12)— The enaminoester (11) (224 mg) was dissolved in methanol (2 ml) and while the solution was stirred and kept between 5 and 10 °C, sodium borohydride (435 mg) was added in small portions during 1 h. The reaction mixture was stirred at room temperature for 15 min, diluted with water (5 ml), and extracted with ether. The extract was washed with brine, dried over K_2CO_3 , and evaporated. The residue was chromatographed on silica gel (ethyl acetate-methanol) to give the ester (12) (155 mg, 69%) as a syrup. IR $v_{\rm max}^{\rm film}$ cm⁻¹: 2850, 2800, 2750, 1740. ¹H-NMR δ : 3.67 (3H, s, COOMe). High-resolution MS Calcd for $C_{11}H_{19}NO_2$: 197.1416. Found: 197.1439.

(\pm)-Lupinine (2)—A solution of the ester (12 (99 mg) in THF (1.2 ml)) was added to a suspension of lithium aluminum hydride (11.4 mg) in THF (1.2 ml). The reaction mixture was refluxed for 0.5 h, cooled, then treated sequentially with water (0.1 ml) and 10% aqueous NaOH (0.1 ml), and then dried over Na₂SO₄. The mixture was filtered and the solid was washed with THF. The combined solutions were evaporated. The residue was chromatographed on alumina (ethyl acetate-methanol) to give (\pm)-lupinine (2) (63 mg, 74%) as a solid, mp 57—59 °C (from hexane) (lit.,^{2c)} 59—61 °C). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3170, 2850, 2800, 2770, 2680. ¹H-NMR δ : 1.20—2.30 (14H, m), 2.83 (2H, m), 3.70 (1H, m), 4.17 (1H, m). The compound prepared was identical with an authentic sample^{2c)} in terms of spectral data (NMR and IR).

Acknowledgements We are grateful to Professor Y. Ban for providing spectral data of (\pm) -epilupinine and (\pm) -lupinine, and to the Ministry of Education, Sciences, and Culture (Japan), and Toyama Prefecture Centennial Foundation for research grants.

References

- 1) This work was presented at the 106th Meeting of the Pharmaceutical Society of Japan, Chiba, April 1986. Part of this work has appeared as a preliminary communication: H. Takahata, K. Yamabe, T. Suzuki, and T. Yamazaki, *Heterocycles*, 24, 37 (1986).
- a) J. J. Tufariello and J. J. Tegeler, Tetrahedron Lett., 1976, 4037; b) T. Iwashita, T. Kusumi, and H. Kakisawa, J. Org. Chem., 47, 230 (1982); c) M. Okita, T. Wakamatsu, and Y. Ban, Heterocycles, 20, 401 (1983); d) M. L. Bremmer, N. A. Khatri, and S. M. Weinreb, J. Org. Chem., 48, 3661 (1983); e) A. R. Chamberlin, H. D.

- Nguyen, and J. Y. L. Chung, *ibid.*, **49**, 1682 (1984); f) M. Ihara, T. Kirihara, K. Fukumoto, and T. Kametani, *Heterocycles*, **23**, 1097 (1985); g) T. Nomoto and H. Takayama, *ibid.*, **23**, 2913 (1985); h) H. Hiemstra, M. H. A. M. Sno, R. J. Vijn, and W. N. Speckamp, *J. Org. Chem.*, **50**, 4014 (1985).
- 3) H. Takahata, T. Yamazaki, and K. Aoe, J. Org. Chem., 50, 4648 (1985).
- 4) H. Takahata, N. Hamada, and T. Yamazaki, Synthesis, 1986, 388.
- a) I. N. Nazarov and S. I. Zavyalov, Zh. Obsch. Khim., 23, 1703 (1953); b) E. Wenkert, A. Afonso, J. B. Bredenberg, C. Kaneko, and A. Tahara, J. Am. Chem. Soc., 86, 2038 (1964); c) G. Stork and R. N. Guthikonda, ibid., 94, 5109 (1972); d) J. E. Ellis, J. S. Dutcher, and C. H. Heathcook, Synth. Commun., 4, 71 (1974); e) B. M. Trost and R. A. Kunz, J. Am. Chem. Soc., 97, 7152 (1975); f) T. Watanabe, Y. Nakashita, S. Kataya, and M. Yamauchi, Heterocycles, 14, 1433 (1980); g) Y. Nakashita and M. Hesse, Helv. Chim. Acta, 66, 845 (1983).
- 6) R. B. Gammill, D. M. Sobieray, and P. M. Gold, J. Org. Chem., 46, 3555 (1981).
- 7) M. P. Cava and M. I. Levinson, Tetrahedron, 41, 5041 (1985).
- 8) a) S. I. Goldberg and I. Ragade, J. Org. Chem., 32, 1046 (1967); b) G. C. Gerrans, A. S. Howard, and B. S. Orlek, Tetrahedron Lett., 1975, 4171.
- 9) B. M. Trost and R. A. Kunz, J. Org. Chem., 39, 2648 (1974).