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Studies on Diazepines. XXIV.¹⁾ Reactions of Monocyclic 1*H*-1,3-Diazepines. (2).¹⁾ Photo-Sensitized Oxygenation

Jyoji Kurita, Hirokazu Kojima, and Takashi Tsuchiya*

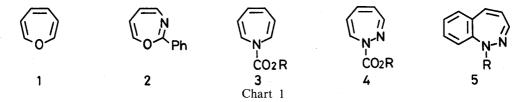
School of Pharmacy, Hokuriku University, Kanagawa-machi, Kanazawa 920-11, Japan

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The photo-sensitized oxygenation of monocyclic 1*H*-1,3-diazepines (6) gave several fragment products (7—12). 3-Pyrrolin-2-one derivatives (7 and 8) and ethyl aminoformates (9 and 10) are assumed to originate from the initially formed 4,7-endoperoxides (13), and vinylaminoformates (11 and 12) from the 4,5-dioxetanes (14).

Keywords—sensitized photooxygenation; 1*H*-1,3-diazepine; endoperoxide; dioxetane; 3-pyrrolin-2-one; aminoformate

The reaction of photochemically generated singlet oxygen with a variety of five- and six-membered heterocycles²⁾ and seven-membered conjugated trienes such as cycloheptatrienes³⁾ and tropolones⁴⁾ has been well studied. However, only a few reports have appeared on seven-membered heterocyclic compounds. The photooxygenation of oxepin (1) gives only phenol,⁵⁾ while that of 1-benzoxepin affords the relatively stable 2,5-endoperoxide.⁶⁾ The 1,3-oxazepine (2) has been shown to undergo initial 1,2- (or 1,6-) and 1,4-cycloaddition of singlet oxygen, followed by decomposition to give rise to several products.⁷⁾ Therefore, we were interested in examining such a reaction of fully unsaturated seven-membered N-heterocyclic systems. We have already reported that the photo-sensitized oxygenation of monocyclic azepines (3)⁸⁾ and 1H-1,2-diazepines (4)^{8,9)} gave the relatively stable 2,5- and 4,7-endoperoxides, respectively, as the sole oxidized products, while the 1H-1,2-benzodiazepines (5) afforded several fragment products derived from the initially formed 3- or 5-hydroperoxides.¹⁰⁾ These results prompted us to examine the photooxygenation behavior of the monocyclic 1H-1,3-diazepines (6), which are new N-heterocycles prepared recently by us,¹¹⁾ and we present the results here.



A solution of a 1*H*-1,3-diazepine (6) in carbon tetrachloride was irradiated with a halogen lamp in the presence of *meso*-tetraphenylporphine as a sensitizer for 0.5—2 h while oxygen was slowly passed through the solution. The photolysate, which gave a positive test¹²⁾ for peroxide with potassium iodide, was treated with water in tetrahydrofuran and then chromatographed on silica gel to give the products 7 to 12 shown in Chart 2. The yields of these products from the diazepines (6a—c) are summarized in Table I. These products were characterized by elemental analyses and infrared (IR), mass (MS), and proton nuclear magnetic resonance (¹H-NMR) spectral analyses. The structure of 7 and 8 were confirmed by spectral comparison with various 3-pyrrolin-2-ones already reported.¹³⁾ Although the geo-

metry of 11 and 12 was not examined in detail, their double bonds are tentatively assigned as Z-forms based on the mechanism shown in Chart 3.

When the photolysate was chromatographed without treatment with water, the yields of products were greatly decreased; this fact clearly indicates that the initially formed peroxides are susceptible to decomposition, and water is required for the formation of the products obtained. Although no adducts of the 1,3-diazepines (6) with singlet oxygen could be isolated, the formation of the products (7—12) in the present photooxygenation can be understood in terms of the initial formation of the 4,7-endoperoxides (13) and 4,5-dioxetanes (14), which are then transformed to the products presumably via the pathways shown in Chart 3.

The ring-opened intermediates (16) formed from the 4,7-endoperoxides (13) via 15 undergo cyclization with addition of water to give the 3-pyrrolin-2-one intermediates (17). The cleavage of the N-C bond (pathway a) of 17 gives the N-unsubstituted pyrrolinones (8) and ethyl formylaminoformate (9), and the N-C bond fission (pathway b) affords the N-formylpyrrolinones (7) and ethyl aminoformate (10). On the other hand, the dioxetanes (14) give the ring-opened products 11 and 12 via the intermediates (18) by bond fission similar to

6	Yield (%)					
	7	8	9	10	11	12
6a	19	53	32	18	_	
6b	8 .	40	28	11	2	2
6c		47	40		11	8

TABLE I. The Yields of Products (7—12) Obtained by Photooxygenation of 6

those in the case of 17. The result for the 5-methoxydiazepine (6c), in which the yields of 11 and 12 are higher than those for 6a, b, may show that the electron-donating methoxy group increases the formation of the dioxetane (14). The photo-sensitized 1,2-cycloaddition of singlet oxygen to electron-rich olefins is well known.¹⁴⁾

It should be noted that no product assumed to originate from the other possible dioxides such as 2,5- (19) and 2,7-endoperoxides (20) and 6,7-dioxetanes (21) has been detected. Therefore, in the present photooxygenation of 1H-1,3-diazepines, neither 1,4- nor 1,6-cycloaddition of singlet oxygen occurs in the aza-diene and aza-triene moieties. This behavior is analogous to that of monocyclic 1H,1,2-diazepines, 8,9 and 1H-1,2-benzodiazepines, 10 but is in contrast to that of 1,3-oxazepines, 7 imidazoles, 15 and pyrazines. 16

When 1H-1,3-benzodiazepine $(22)^{17}$ was photooxygenated under similar conditions, it decomposed, giving no characterizable products.

Experimental

The general experimental procedures were the same as in Part XXIII.¹⁾ Photolyses were carried out using an immersion apparatus equipped with a halogen lamp (Ushio JCV100-200 GC), which was cooled internally with running water.

Starting Materials—The 1H-1,3-diazepines (6a—c) were prepared by the reported method. 11)

Photooxygenation of 1-Ethoxycarbonyl-5-methyl-1H-1,3-diazepine (6a)—A solution of 6a (900 mg) in carbon tetrachloride (300 ml) containing *meso*-tetraphenylporphine (60 mg) was irradiated with a halogen lamp for 30 min while a steady stream of oxygen was bubbled through the solution, and then concentrated *in vacuo* below 30 °C. The resulting residue gave a positive test for peroxide with acidified starch-iodide paper. The residue was dissolved in tetrahydrofuran (50 ml) and water was added to the solution. The mixture was stirred for 6 h at room temperature, and then diluted with CH₂Cl₂ (100 ml), dried over MgSO₄ and evaporated to dryness *in vacuo*. The residue was chromatographed on silica gel using CH₂Cl₂-MeOH as an eluent to give 9, 7a, 10, and 8a, successively.

1-Formyl-5-hydroxy-3-methyl-3-pyrrolin-2-one (7a): 144 mg, 19% yield, mp 89—90 °C (dec.), colorless plates (from *n*-hexane–ether). MS m/z: 141 (M⁺). IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1735 and 1705 (C=O). ¹H-NMR (CDCl₃) δ : 1.96 (3H, br d, J = ca. 1 Hz, 3-Me), 4.6 (1H, br, OH), 6.07 (1H, br s, 5-H), 6.86 (1H, m, 4-H), 9.01 (1H, s, CHO). *Anal.* Calcd for C₆H₇NO₃: C, 51.07; H, 5.00; N, 9.93. Found: C, 51.01; H, 5.00; N, 10.02.

5-Hydroxy-3-methyl-3-pyrrolin-2-one (8a): 300 mg, 53% yield, mp 119—121 °C, colorless prisms (from AcOEt). MS m/z: 113 (M⁺). IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3300 (OH), 3200 (NH), 1680 (C=O). ¹H-NMR (acetone- d_6) δ : 1.78 (3H, br d, J = 1 Hz, 3-Me), 4.9 (1H, br, OH), 5.48 (1H, br s, 5-H), 6.60 (1H, m, 4-H), 7.5 (1H, br, NH). *Anal.* Calcd for C₅H₇NO₂: C, 53.09; H, 6.24; N, 12.38. Found: C, 52.90; H, 6.23; N, 12.34.

Ethyl *N*-Formylaminoformate (9): 185 mg, 32% yield, mp 47.5—48.5 °C, colorlss needles (from *n*-hexane—isopropyl ether). MS m/z: 89 (M—CO). IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3250 (NH), 1740 and 1690 (C=O). ¹H-NMR (CDCl₃) δ : 1.35 and 4.31 (3H, t, and 2H, q, CO₂Et), 8.5 (1H, br, NH), 8.97 (1H, d, J=9 Hz, CHO). *Anal.* Calcd for C₄H₇NO₃: C, 41.02; H, 6.03; N, 11.96. Found: C, 40.93; H, 5.98; N, 11.92.

Ethyl Aminoformate (10): 81 mg, 18% yield, mp 48—50 °C; this compound was identical with an authentic sample obtained from Tokyo Kasei Kogyo Co., Ltd., Tokyo.

Photooxygenation of 1-Ethoxycarbonyl-5,7-dimethyl-1*H*-1,3-diazepine (6b)—A solution of 6b (570 mg) in CCl₄ (400 ml) containing *meso*-tetraphenylporphine (50 mg) was photooxygenated for 2 h and worked up as described for 6a to give 12b, 11b, 9, 7b, 10, and 8b, successively.

1-Formyl-3,5-dimethyl-5-hydroxy-3-pyrrolin-2-one (7b): 37 mg, 8% yield, colorless viscous oil. MS m/z: 155 (M⁺). IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 3400 (OH), 1730 and 1700 (C=O). ¹H-NMR (CDCl₃) δ : 1.83 (3H, s, 5-Me), 1.94 (3H, br s, 3-Me), 4.3 (1H, br, OH), 6.83 (1H, m, 4-H), 9.01 (1H, s, CHO). *Anal*. Calcd for C₇H₉NO₃: C, 54.19; H, 5.85; N, 9.03.

Found: C, 53.92; H, 5.91; N, 8.87.

3,5-Dimethyl-5-hydroxy-3-pyrrolin-2-one (8b): 139 mg, 40% yield, mp 152—154 °C (dec), colorless needles (from CH₂Cl₂). MS m/z: 127 (M⁺). IR $v_{\rm max}^{\rm KBr}$ cm⁻¹: 3350 (OH), 3200 (NH), 1685 (C=O). ¹H-NMR (methanol- d_4) δ : 1.52 (3H, s, 5-Me), 1.83 (3H, d, J=1 Hz, 3-Me), 6.58 (1H, m, 4-H). *Anal*. Calcd for C₆H₉NO₂: C, 56.68; H, 7.14; N, 11.02. Found: C, 56.59; H, 7.30; N, 11.05.

Ethyl N-Formylaminoformate (9): 96 mg, 28% yield.

Ethyl Aminoformate (10): 29 mg, 11% yield.

Ethyl (1-Methoxy-2-acetylvinyl)aminoformate (11b): 12 mg, 2% yield, colorless solid (mp < 40 °C), MS m/z: 171 (M⁺). IR $v_{max}^{CHCl_3}$ cm⁻¹: 3350 (NH), 1740 and 1640 (C=O). 1 H-NMR (CDCl₃) δ: 1.30 and 4.17 (3H, t, and 2H, q, CO₂Et), 2.12 (3H, s, COMe), 2.33 (3H, br s, 1-Me), 5.32 (1H, br s, 2-H), 9.2 (1H, br, NH). *Anal*. Calcd for C₈H₁₃NO₃: C, 56.12; H, 7.65; N, 8.18. Found: C, 56.09; H, 7.78; N, 8.01.

Ethyl (1-Methyl-2-acetylvinyl)formylaminoformate (12b): 11 mg, 2% yield, colorless viscous oil. MS m/z: 199 (M⁺). IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 1750 and 1700 (C=O). ¹H-NMR (CDCl₃) δ : 1.35 and 4.40 (3H, t, and 2H, q, CO₂Et), 2.09 (3H, d, 1-Me), 2.19 (3H, s, COMe), 6.41 (1H, m, 2-H), 9.21 (1H, s, CHO). *Anal*. Calcd for C₉H₁₃NO₄: C, 54.26; H, 6.58; N, 7.03. Found: C, 53.89; H, 6.53; N, 6.83.

Photooxygenation of 1-Ethoxycarbonyl-5-methoxy-1*H*-1,3-diazepine (6c)—A solution of 6c (1.17 g) in CCl₄ (400 ml) containing *meso*-tetraphenylporphine (50 mg) was photooxygenated for 1 h and worked up as described for 6a to give 12c, 11c, 9, and 8c, successively.

5-Hydroxy-3-methoxy-3-pyrrolin-2-one (8c): 353 mg, 47% yield, mp 143.5—145.5 °C, colorless prisms (from methanol–isopropyl ether). MS m/z: 129 (M⁺). IR $v_{\rm max}^{\rm KBr}$ cm⁻¹: 3300 (OH), 3200 (NH), 1690 and 1710 (C=O). ¹H-NMR (methanol- d_6) δ : 3.76 (3H, s, OMe), 5.47 (1H, d, J=1.5 Hz, 5-H), 5.72 (1H, d, J=1.5 Hz, 4-H). *Anal*. Calcd for $C_5H_7NO_3$: C, 46.51; H, 5.47; N, 10.85. Found: C, 46.61; H, 5.40; N, 10.54.

Ethyl Formylaminoformate (9): 279 mg, 40% yield.

Ethyl (2-Methoxycarbonylvinyl)aminoformate (11c): 110 mg, 11% yield, colorlss solid (mp < 30 °C). MS m/z: 173 (M⁺). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3300 (NH), 1740 and 1685 (C=O). ¹H-NMR (CDCl₃) δ : 1.31 and 4.26 (3H, t, and 2H, q, CO₂Et), 3.71 (3H, s, OMe), 5.06 (1H, d, J=9 Hz, 2-H), 7.27 (1H, dd, J=11 and 9 Hz, 3-H), 9.7 (1H, br, NH). *Anal*. Calcd for C₇H₁₁NO₄: C, 48.55; H, 6.40; N, 8.09. Found: C, 48.51; H, 6.22; N, 7.86.

Ethyl (2-Methoxycarbonylvinyl)formylaminoformate (12c): 97 mg, 6% yield, colorless viscous oil. MS m/z: 201 (M⁺). IR $v_{max}^{CHCl_3}$ cm⁻¹: 1750 and 1720 (C=O). ¹H-NMR (CDCl₃) δ : 1.33 and 4.29 (3H, t, and 2H, q, CO₂Et), 3.66 (3H, s, OMe), 5.83 (1H, d, J=9 Hz, 2-H), 6.53 (1H, d, J=9 Hz, 1-H), 8.98 (1H, s, CHO). *Anal.* Calcd for C₈H₁₁NO₅: C, 47.76; H, 5.51; N, 6.96. Found: C, 47.68; H, 5.61; N, 6.72.

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