

## Communications to the Editor

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FACILE SYNTHESIS OF THE CARBOCYCLIC ANALOGUE OF  $\beta$ -RIBOFURANOSYL-  
MALONATE, NEW SYNTHON FOR CARBOCYCLIC C-NUCLEOSIDE FORMATION  
BY RETROGRADE ALDOL C-C BOND FISSION<sup>1)</sup>

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A new carbocyclic C-nucleoside precursor, the carbocyclic analogue of  $\beta$ -ribofuranosylmalonate, has been synthesized through Diels-Alder addition of cyclopentadiene to dimethyl acetoxy-methylene-malonate, followed by retrograde C-C bond fission under reductive conditions.

KEYWORDS — carbocyclic C-nucleoside; carbocyclic  $\beta$ -ribofuranosylmalonate; Diels-Alder adduct; cyclopentadiene; dimethyl acetoxy-methylenemalonate; retrograde aldol reaction; sodium borohydride

In the synthesis of C-nucleosides starting from non-carbohydrate precursors, most methods have utilized [4+2] or [4+3] cycloadducts (A or B) between furan and an appropriate dienophile (either 3-substituted acrylate or oxyallyl species) as intermediates.<sup>2)</sup> From these bicycles to  $\beta$ -ribofuranosylacetates (G), the C-C bond originating from the dienophile must be cleaved by oxidation reactions either directly (D→F)<sup>3)</sup> or after regeneration of a double bond (E→G).<sup>4)</sup> In these methods, the carbon-carbon double bond in A and B (formed in the cycloaddition step) must be protected as a diol (C and D) before the above oxidation reactions are carried out.

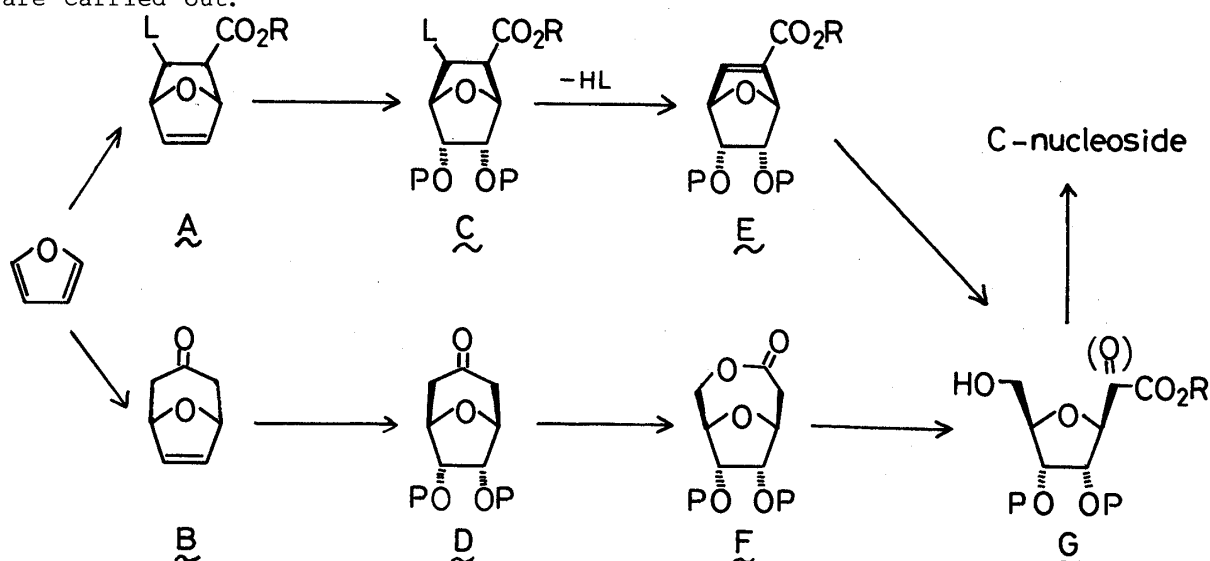
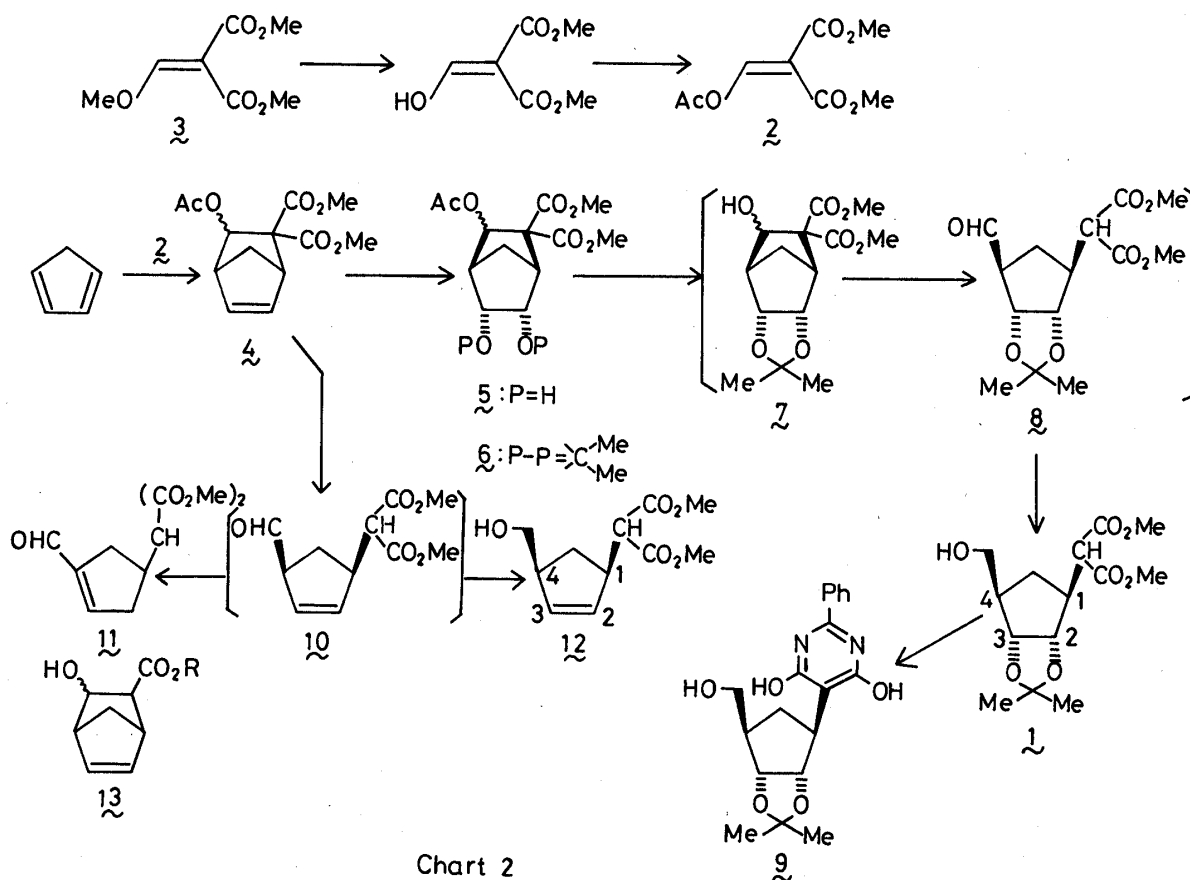


Chart 1. General Strategies for the Synthesis of C-Nucleosides from Furan (L and P Designate Appropriate Leaving and Protecting Groups, Respectively)

We now report a novel synthesis of carbocyclic analogue (1) of dimethyl  $\beta$ -ribofuranosylmalonate<sup>5)</sup> by retrograde aldol C-C bond fission under reductive conditions which defines a simple and new route to introduce all required functions into cyclopentadiene in a stereospecific manner without using any oxidation reaction.<sup>6)</sup>

In our synthesis, which obeys the general strategy for the construction of A-type compound by using Diels-Alder addition of cyclopentadiene to acrylate derivatives, we chose dimethyl acetoxy methylenemalonate (2) as a dienophile. This is because not only would 2 be a suitable dienophile due to its high electron deficiency, but also the adduct thus formed cleaves the C-C bond by retrograde aldol reaction. 2 (bp 92-95°C/2 mmHg) was synthesized from readily available dimethyl methoxymethylenemalonate (3)<sup>7)</sup> by basic hydrolysis followed by acetylation. The sequential reactions were the same as the synthesis of corresponding diethyl ester reported recently by Wolff et al.<sup>8)</sup> Cycloaddition of cyclopentadiene to 2 proceeded smoothly (80°C, PhH, 72 h) to give the [4+2] adduct (4)<sup>9)</sup> as a chromatographically separable mixture of endo and exo isomers (ca. 1:3 ratio) in 80% yield [ $4_{\text{endo}}$ : mp 92-94°C (ether-hexane); IR (CHCl<sub>3</sub>) 1735 cm<sup>-1</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 6.25 (1H, d, J=4 Hz, >CHOAc), and  $4_{\text{exo}}$ : mp 113-114°C (ether-hexane); IR (CHCl<sub>3</sub>) 1735 cm<sup>-1</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 5.62 (1H, d, J=2 Hz,



$>\text{CHOAc}$ ].<sup>10</sup>) The adduct (4) was hydroxylated ( $\text{OsO}_4$ , 60% aqueous 4-methylmorpholine N-oxide, acetone, 2 h, room temperature) to give quantitatively the exo diol (5) and the diol was protected as its acetonide (6) (acetone,  $\text{Me}_2\text{C}(\text{OMe})_2$ , TsOH, 2 h, room temperature, quantitative yield). When 6 was treated with methanol containing  $\text{K}_2\text{CO}_3$  and  $\text{NaBH}_4$  for 1 h at room temperature, the desired compound [1: IR ( $\text{CHCl}_3$ ) 3500, 1750 (sh), 1730  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ) $\delta$ : 1.30 (3H, s, isopropylidene-Me), 1.50 (3H, s, isopropylidene-Me), 3.50 (1H, d,  $J=8$  Hz,  $-\text{CH}(\text{CO}_2\text{Me})_2$ ), 4.3-4.6 (2H, m,  $-\text{CH-O} \times 2$ )] was obtained in quantitative yield. Obviously, the alcohol (7) primarily formed undergoes a facile C-C bond fission (retrograde aldol reaction) and the aldehyde (8) thus formed is reduced to the final product (1). As an example illustrating the construction of a heterocyclic subunit from this product, 1 was treated with benzamidine in the presence of sodium methoxide in methanol at 40°C for 48 h to give a pyrimidine carbocyclic C-nucleoside [9: mp 248-250°C ( $\text{CHCl}_3$ )].<sup>11,12)</sup>

Facile C-C bond scission by retrograde aldol reaction also occurred, when the adduct (4) was directly subjected to basic hydrolysis. Thus, treatment of 4 in methanol containing  $\text{K}_2\text{CO}_3$  at room temperature (1 h) afforded the aldehyde [11: IR ( $\text{CHCl}_3$ ) 1750, 1730, 1675  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ) $\delta$ : 2.0-3.7 (6H, m), 3.72 (6H, s,  $\text{CO}_2\text{Me} \times 2$ ), 6.70 (1H, m,  $=\text{C-H}$ ), 9.70 (1H, s,  $-\text{CHO}$ )]. The same reaction of 4 in the presence of sodium borohydride gave the alcohol [12: IR ( $\text{CHCl}_3$ ) 3500, 1750 (sh), 1730  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ) $\delta$ : 3.56 (2H, d,  $J=5$  Hz, 5-H), 3.75 (6H, s,  $\text{CO}_2\text{Me} \times 2$ ), 5.74 (2H, s, 2- and 3-H)] in quantitative yield. These experiments tell clearly that the configuration of C-4 in both 1 and 12 is retained under these reductive conditions ( $\text{NaBH}_4/\text{K}_2\text{CO}_3/\text{methanol}$ ).

It was revealed that such retrograde aldol reaction did not occur in the adduct (13).<sup>14)</sup> Therefore, the presence of two electron-withdrawing groups as neighbors of the hydroxyl group seems to be essential for this retrograde aldol reaction. The present method seems not only to provide a new route for the synthesis of C-nucleoside precursors but also to be applicable for the introduction of formyl and acetic acid appendages at the 1- and 4-positions of a 1,3-diene system. Further work is now in progress to extend our scheme to the synthesis of C-nucleosides by treating furan with 2 and its analogues.

## REFERENCES AND NOTES

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  - 8) I.A. Wolff, D.W. Olds, and G.E. Hilvert, *Synthesis*, 1984, 732.
  - 9) All new compounds herein reported were supported either by elemental analyses or by high-resolution mass spectra and showed acceptable spectral data.
  - 10) 4<sub>endo</sub>: <sup>1</sup>H-NMR (CDCl<sub>3</sub>) $\delta$ : 1.50 (2H, m), 1.93 (3H, s), 3.18 (1H, br s), 3.33 (1H, br s), 3.65 (3H, s), 3.77 (3H, s), 6.08 (1H, dd, J=3, 6 Hz), 6.65 (1H, dd, J=3, 6 Hz). 4<sub>exo</sub>: <sup>1</sup>H-NMR (CDCl<sub>3</sub>) $\delta$ : 1.7-2.4 (2H, m), 2.02 (3H, s), 2.92 (1H, br s), 3.40 (1H, br s), 3.70 (3H, s), 3.75 (3H, s), 6.0-6.4 (2H, m).
  - 11) In the <sup>13</sup>C-NMR spectrum (CDCl<sub>3</sub>-CD<sub>3</sub>OD) of 9, there are signals due to two methyl carbons of the isopropylidene group at  $\delta$  25.48 and 27.83, indicating the  $\beta$ -configuration at the 1-position.<sup>5b,13)</sup>
  - 12) Carbocyclic analogues of nucleosides have been of interest for many years and they have been synthesized to develop pharmacologically active substances. M. Ohno, "Anticancer Agents based on Natural Product Models," ed. by M.J. Cassady and J.D. Douros, Academic Press, New York, 1980, pp. 73-130.
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