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STUDIES ON MACROCYCLIC LACTONE ANTIBIOTICS. ${\rm IX}^1$) NOVEL MACROLIDES FROM THE FUNGUS RHIZOPUS CHINENSIS: PRECURSORS OF RHIZOXIN

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Five novel 16-membered macrolides $(\underline{1b}-\underline{3b})$, homologues of rhizoxin, were isolated and their structures were determined from their physicochemical and spectral properties.

KEYWORDS — 16-membered macrolide; Rhizopus chinensis; phytotoxin; antifungal activity; anti-tumor activity

A novel 16-membered macrolide, rhizoxin ($\underline{1a}$), was previously isolated as a toxin produced by Rhizopus chinensis, the causal agent of rice seedling blight occured in nursery cases. The potent antifungal and anti-tumor activities of this compound were established, and its absolute structure has also been determined.

This paper deals with the isolation and the structures of five additional compounds related to rhizoxin produced by \underline{R} . chinensis.

Figure 1

R. chinensis Rh-2 strain was cultivated in a medium composed of 3% maltose, 1% polypepton, 0.25% $\rm KH_2PO_4$, 0.75% $\rm K_2HPO_4$, 0.25% $\rm MgSO_4$ $\rm 7H_2O$, 0.20% $\rm (NH_4)_2SO_4$ and 4% pharmamedia, at 28°C with shaking for 92 hours or in large fermentation tanks for 72 hours. The culture broth was first extracted with twice as much volume of acetone as broth. The filtrate after removal of cultured mycelia was then extracted with ethyl acetate. The extract was separated by successive silicagel and LH-20 column chromatographies to isolate rhizoxin (1a) and its related compounds 1b, 2a, 2b, 3a and 3b. The compounds were finally purified by HPLC using a Whatman M9 ODS-3 column eluted with mixtures of acetonitlile-water. These compounds could not be crystallized. Evaporation of the solvent from their acetone-hexane solution gave respective white powders. Their isolated yields per liter were calculated to be ca 20 mg, 0.01 mg, 2 mg, 0.04 mg, 0.1 mg and 0.1 mg for 1a, 1b, 2a, 2b, 3a and 3b, respectively.

The structures of compounds $\underline{1b-3c}$ were determined by comparing their physicochemical properties (Table 1) and their $^1\text{H-NMR}$ data, listed in Table 2 in which the multiplicities and chemical shift values of all the proton signals are given. Proton assignments for each compound were made by analysis of their spin-spin couplings and also by comparisons with the $^1\text{H-NMR}$ data of rhizoxin ($\underline{1a}$) previously reported in detail. 1,2)

The UV-absorption spectra of compounds $\underline{1b}-\underline{3b}$ showed the presence of the same chromophore as in the rhizoxin molecule. The NMR signals of H-18a to H-26a of these compounds (Table 2) also proved the identity of the partial structure of C-18 through C-26a.

The molecular formulas of these compounds determined by high resolution electron impact mass spectroscopy (HREIMS) indicate that the <u>b</u>-series compounds are composed of one CH₂ unit less than the corresponding <u>a</u>-series compounds. The $^1\text{H-NMR}$ spectra of compounds $\underline{1b}$, $\underline{2b}$ and $\underline{3b}$ exhibited no signal due to the OCH₃ group which appeared in the spectra of $\underline{1a}$, $\underline{2a}$ and $\underline{3a}$ at 6 3.15, 3.15 and 3.17 ppm respectively, and the H-17 signals appearing at 6 3.23-3.26 ppm for the <u>a</u>-series compounds shifted to 6 3.89-3.91 ppm for the <u>b</u>-series compounds. This shows that the 17-OCH₃ group present in $\underline{1a}$, $\underline{2a}$ and $\underline{3a}$ is replaced by an OH group in $\underline{1b}$, $\underline{2b}$ and $\underline{3b}$. Since the $^1\text{H-NMR}$ spectra of each pair of compounds showed all the other proton signals at the comparable positions, it was concluded that the only structural difference in the pair of compounds is the substituent at C-17.

On the other hand, comparison of the molecular formulas of $\underline{1a}$, $\underline{2a}$ and $\underline{3a}$ (and also of $\underline{1b}$, $\underline{2b}$ and $\underline{3b}$) show that their atomic composition differs only in the numbers of oxygen atoms and that $\underline{2a}$ and $\underline{3a}$ are composed of one atom and two atoms respectively less than $\underline{1a}$.

In the spectrum of <u>la</u>, H-2 and H-3 signals appeared at & 2.96 and 3.27 ppm as the protons on the carbon bearing an epoxy group, while those signals for <u>2a</u> and <u>3a</u> appeared at & 5.61 and 6.82 ppm and & 5.61 and 6.77 ppm in the respective spectra. These signals of <u>2a</u> and <u>3a</u> were assignable to the olefinic protons of disubstituted & unsaturated ester moieties, and an <u>E</u>-orientation of the olefinic linkages were indicated by large coupling constants between H-2 and H-3 (15.8 Hz for <u>2a</u>, and 15.2 Hz for <u>3a</u>). This partial structure change at C-2,3 also caused lower field shifts of the H-4 signals (see Table 2). On comparison of the ¹H-NMR spectra of <u>1a</u> and <u>3a</u> the H-11 and H-12a signals shown at & 3.20 and 1.45 ppm in the former shifted to & 5.81 and 1.79 ppm respectively in the latter. This suggested the presence of a

Table 1. Physico-Chemical Characteristics of $\underline{1a}$ - $\underline{3b}$

Compd.	Mol. Formula	HREIMS (calcd. val.)	$I\alpha I_D^{24}(MeOH)$	$UV(MeOH) \lambda_{max} nm(\epsilon)$	
<u>la</u>	C ₃₅ H ₄₇ O ₉ N	M ⁺ 625.3233 (625.3250)	+155.5 ⁰ (c=0.80)	295(42300) 308(54000) 325(39000)	
<u>lb</u>	^C 34 ^H 45 ^O 9 ^N	M ⁺ -1 610.3009 (610.3012)	+116.3 ⁰ (c=0.11)	297(32300) 309(41000) 323(30100)	
<u>2a</u>	C ₃₅ H ₄₇ O ₈ N	M ⁺ 609.3300 (609.3299)	+136.8 ^O (c=0.71)	297(46000) 308(64400) 323(46300)	
<u>2b</u>	C ₃₄ H ₄₅ O ₈ N	M ⁺ 595.3135 (595.3142)	+99.3 ⁰ (c=0.61)	297(39100) 308(50300) 323(36600)	
<u>3a</u>	^C 35 ^H 47 ^O 7 ^N	M ⁺ 593.3382 (593.3414)	+287.1 ^O (c=0.48)	216(25900) 235(24700) 238(24500) 297(38700) 309(49900) 323(36500)	
<u>3b</u>	с ₃₄ н ₄₅ 0 ₇ N	M ⁺ 597.3165 (597.3137)	+246.0 ^O (c=0.66)	215(28400) 232(26200) 238(25700) 297(42500) 308(55100) 323(40300)	

Table 2. Chemical Shifts of ${}^1\text{H-Signals}$ of $\underline{1a}$ - $\underline{3b}$ (in CDCl $_3$, in & Value)

Protons	<u>la</u>	<u>1b</u>	<u>2a</u>	<u>2b</u>	<u>3a</u>	<u>3b</u>
H-2 (d)	2.96	2.98	5.61	5.72	5.61	5.63
H - 3 (ddd)	3.27	3.29	6.82	6.85	6.77	6.79
H-4 (ddd)	2.33	2.35	2.56	2.58	2.53	2.55
(ddd)	0.79	0.82	1.80	1.79	1.74	1.75
H-5 (m)	2.05	2.08	1.80	1.83	1.75	1.80
H-5a (dd)	2.72	2.74	2.78	2.79	2.76	2.77
(dd)	2.10	2.07	2.10	2.10	2.09	2.10
H-6 (ddd)	1.93	1.99	1.98	1.98	1.98	1.97
(ddd)	0.93	0.95	0.72	0.76	0.68	0.70
H-7 (ddd)	3.87	3.89	3.79	3.78	3.67	3.70
H-8 (m)	2.30	2.32	2.32	2.13	2.27	2.29
H-8a (d)	1.20	1.22	1.19	1.20	1.19	1.20
H-9 (dd)	5,66	5.63	5.55	5.57	5.15	5.18
H-10 (dd)	5.38	5.37	5.34	5.36	6.23	6.24
H-11 (d)	3.20	3.23	3.24	3.27	5.81	5.79
H - 12a (s)	1.45	1.43	1.42	1.44	1.79	1.81
H-13 (dd)	3.02	3.07	3.07	3.15	3.90	3.99
H-14 (ddd)	2.05	2.08	2.10	2.13	2.14	2.16
(dd)	1.88	1.95	1.81	1.94	1.70	1.83
H-15 (dd)	4.63	4.66	4.61	4.69	4.58	4.69
H-16 (m)	2.37	2.34	2.40	2.34	2.27	2.12
H-16a (d)	1.00	1.02	1.00	1.00	1.00	0.97
H-17 (d)	3.23	3.89	3.25	3.91	3.26	3.89
17-OCH ₃ (s)	3.15	and the same	3.15		3.17	
H-18a ³ (s)	1.82	1.89	1.84	1.92	1.89	1.88
H-19 (d)	6.08	6.11	6.09	6.16	6.12	6.21
H-20 (dd)	6.60	6.53	6.59 °	6.55	6.63	6.58
H-21 (d)	6.38	6.37	6.37	6.39	6.41	6.40
H-22a (s)	2.12	2.12	2.34	2.13	2.15	2.13
H-23 (s)	6.27	6.25	6.25	6.25	6.26	6.25
H-25 (s)	7.58	7.54	7.54	7.53	7.55	7.54
H-26a (s)	2.45	2.46	2.46	2.46	2.46	2.46

double bond at C-11, 12 in 3a instead of an epoxy group in 1a. Evidence for a conjugated diene system in 3a (C-9 through C-12) was also provided by its UV spectrum (see Table 1). An E-orientation of the C-11, 12 double bond was determined by an NOE-enhancement of the H-10 signal observed on irradiation of H_3 -12a.

On the basis of the evidence described above, it was concluded that the only structural difference between $\underline{1a}$ and $\underline{2a}$ is at the C-2,3 position and between $\underline{1a}$ and $\underline{3a}$ at the C-2,3 and C-11, 12 positions. The structures of $\underline{2a}$ (accordingly of $\underline{2b}$) and of $\underline{3a}$ (accordingly of $\underline{3b}$) were thus established as shown in Figure 1.

Assignments of the relative stereochemistry of compounds $\underline{1b} - \underline{3b}$ were based on comparisons of the coupling modes of their proton signals with those of $\underline{1a}$, although the details of the NMR data of compounds $\underline{1b} - \underline{3b}$ could not be presented in this communication. The absolute structures of these compounds were also tentatively proposed as shown in Figure 1 in consideration of their biogenetic relationship to rhizoxin, since these compounds can be regarded as the biogenetic precursors of rhizoxin ($\underline{1a}$).

The potent activities of these compounds against rice seedling roots, fungi and tumour cells were also established by assaying their growth inhibitory activity. Such biological data will be reported elsewhere and will provide very useful information about the structure-activity relationship of the rhizoxinoid compounds.

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